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Supporting Information for

5   **Constraining global marine iron sources and ligand-mediated scavenging fluxes**  
6   **with GEOTRACES dissolved iron measurements in an ocean biogeochemical model**7   **Christopher J. Somes<sup>1</sup>, Andrew W. Dale<sup>1</sup>, Klaus Wallmann<sup>1</sup>, Florian Scholz<sup>1</sup>,**  
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20   **Introduction**21       This section documents minor changes made from previously published versions  
22   (Somes et al., 2017;Muglia et al., 2017) that were applied to all model simulations in this  
23   study. The core model code is based on the Model of Ocean Biogeochemistry and

24 Isotopes (MOBI), version 2.0 (<https://github.com/OSU-CEOAS-Schmittner/UVic2.9>),  
25 which is based on the University of Victoria (UVic) Earth System Model of intermediate  
26 complexity (Eby et al., 2013; Weaver et al., 2001).

27 **Text S1. Physical Model**

28 We applied the background vertical mixing setup from Somes et al. (2017) to the  
29 default MOBI 2.0 version. This setup applies background vertical mixing of  $0.15 \text{ cm}^2 \text{ s}^{-1}$   
30 in the ocean interior consistent with open ocean microstructure observations (Fischer et  
31 al., 2013), which caused a reduction in the large-scale overturning and an  
32 underestimation of  $\Delta^{14}\text{C}$  values. In order to reinvigorate the large-scale circulation,  
33 we increased the tidal mixing efficiency parameter to 0.28 (from 0.2), applied a  
34 background horizontal diffusivity of  $20 \text{ m}^2 \text{ s}^{-1}$ , and increased the atmospheric moisture  
35 diffusivity in the Southern Ocean by 20% (e.g. Muglia & Schmittner (2015)), all of  
36 which contributed to an improved representation of  $\Delta^{14}\text{C}$  (Figure S1).

37 **Text S2. Marine Biogeochemical Model**

38 Since MOBI version 2.0 integrated the latest improvements to the nitrogen  
39 (Somes and Oschlies, 2015), carbon chemistry (Kvale et al., 2015), and iron (Muglia et  
40 al., 2017), minor parameter changes were made to achieve a best fit to nutrient  
41 distribution (Figure S1, Table S1). Other structural changes are documented below.

42 The production of semi-refractory dissolved organic matter (DOM) has been  
43 modified to now include an additional source term from the remineralization of  
44 particulate organic matter (POM), along with phytoplankton mortality that previous  
45 versions Somes & Oschlies (2015) used. This new term represents DOM production by  
46 heterotrophic bacteria as they respire POM. The two DOM production factors have  
47 similar spatial patterns, but with the bacterial term based on POM remineralization  
48 extending to greater depths. The production fraction parameters (see Table S1) were  
49 chosen so they represent roughly equivalent total DOM production rate when integrated  
50 over the global ocean, and that they produce surface DON concentrations that are  
51 consistent with observations (Figure S2).

52 We have modified the low oxygen threshold including the reduction of dissolved  
53 iron (DFe) scavenging in the model. This parameterization was implemented to account  
54 for elevated DFe concentrations that exist in low oxygen waters associated with redox  
55 cycling including high nitrite concentrations, although it remains unclear exactly what  
56 processes contribute to these elevated low oxygen DFe concentrations (Moffett et al.,  
57 2015). Previous model versions applied a sharp threshold gradient at the dissolved O<sub>2</sub>  
58 concentration 5 mmol m<sup>-3</sup> (Figure S3). However, elevated DFe typically exists in lower  
59 dissolved O<sub>2</sub> concentrations <~2 mmol m<sup>-3</sup>, so in this study we apply a function that has a  
60 sharper gradient at lower dissolved O<sub>2</sub> concentrations (red line in Figure S3) using the  
61 equation tanh( $\kappa \cdot O_2$ ) where  $\kappa=0.25$ .

62 Sedimentary carbon oxidation (C<sub>ox</sub>) has been modified in all simulations  
63 following the Niemeyer et al. (2017) implementation of Flögel et al. (2011). This scheme  
64 estimates carbon oxidation from the difference between sinking particulate flux entering  
65 the sediment and burial. It has been constructed using a global compilation of  
66 sedimentary data that shows higher carbon burial efficiency, and thus lower carbon  
67 oxidation in continental margins (Burial=0.14·RR<sub>POC</sub><sup>1.11</sup>) compared to the deep-sea  
68 (Burial=0.014·RR<sub>POC</sub><sup>1.05</sup>) sediments. Instead of applying an abrupt transition at 1000  
69 meters depth as in Niemeyer et al. (2017) between these surface and deep sea systems,  
70 we applied a linear transition to the numerator and exponent coefficients from 500 meters  
71 to 1500 meters. Note that previous model marine iron versions (e.g. Nickelsen et al.  
72 (2015); Muglia et al. (2017)) applied the temperature-dependent water column  
73 remineralization rate to organic matter sinking into sediments to estimate carbon  
74 oxidation in the sediments which does not capture the sedimentary carbon dynamics  
75 shown in Flögel et al. (2011).

**Table S1. Marine Ecosystem-Biogeochemistry Parameters**

<i>Parameter</i>	<i>Symbol</i>	<i>Value</i>	<i>Units</i>
<i>Phytoplankton</i>			
Initial slope of P-I curve	$\alpha$	0.1	$(W\ m^{-2})^{-1}\ d^{-1}$
Photosynthetically active radiation	<i>PAR</i>	0.43	-
Light attenuation in water	$k_w$	0.04	$m^{-1}$
Light attenuation through phytoplankton	$k_c$	0.03	$m^{-1}(mmol\ m^{-3})^{-1}$
Light attenuation through sea ice	$k_i$	5	$m^{-1}$
$NO_3$ uptake half-saturation	$k_{NO3}$	0.7	$mmol\ m^{-3}$
$PO_4$ uptake half-saturation	$k_{PO4}$	0.044	$mmol\ m^{-3}$
DOP assimilation handicap	$h_{DOP}$	0.5	
minimum Fe uptake half-saturation	$k_{Femin}$	0.05	$nmol\ m^{-3}$
maximum Fe uptake half-saturation	$k_{Femax}$	0.5	$nmol\ m^{-3}$
Maximum growth rate (at 0°C)	$a_0$	0.6	$d^{-1}$
Phytoplankton fast-recycling rate (at 0°C)	$\mu_{PO0}$	0.001	$d^{-1}$
Phytoplankton specific mortality rate	$v_{P_0}$	0.03	$d^{-1}$
<i>Calcifying Phytoplankton (P<sub>C</sub>)</i>			
Maximum growth rate (at 0°C)	$a_0$	0.3	$d^{-1}$
CaCO <sub>3</sub> :POC production ratio	$R_{CaCO3:POC}$	0.065	0.065
NO <sub>3</sub> uptake half-saturation	$k_{NO3}$	0.35	$mmol\ m^{-3}$
PO <sub>4</sub> uptake half-saturation	$k_{PO4}$	0.022	$mmol\ m^{-3}$
minimum Fe uptake half-saturation	$k_{Femin}$	0.025	$nmol\ m^{-3}$
maximum Fe uptake half-saturation	$k_{Femax}$	0.25	$nmol\ m^{-3}$
<i>Diazotrophic Phytoplankton (P<sub>D</sub>)</i>			

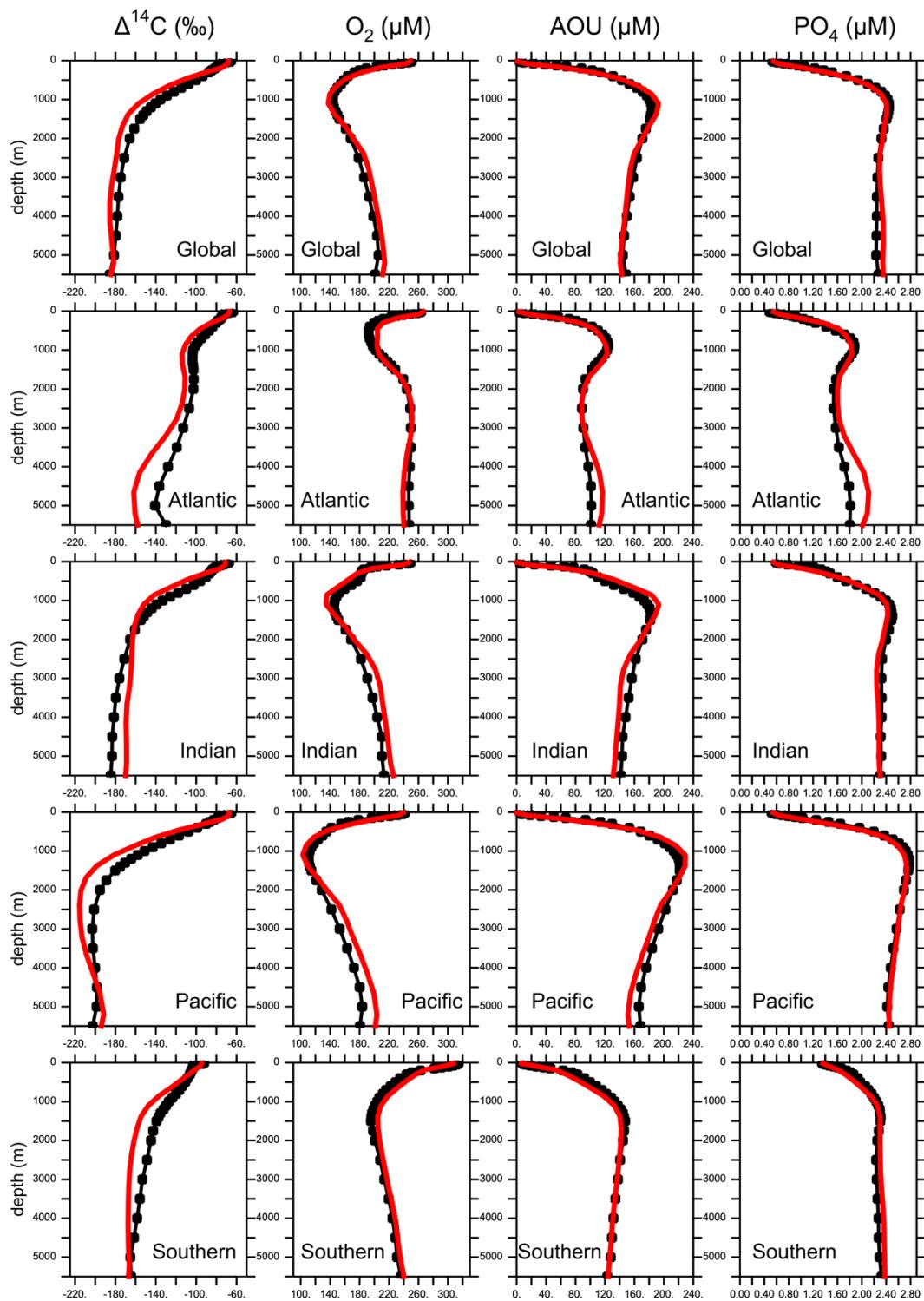
<i>Parameter</i>	<i>Symbol</i>	<i>Value</i>	<i>Units</i>
Diazotroph growth handicap	$h_{P_D}$	0.07	-
Fe uptake half-saturation	$k_{Fe}$	0.16	$\text{nmol m}^{-3}$
Diazotroph fast-recycling rate (at 0°C)	$\mu_{P_D0}$	0.004	$\text{d}^{-1}$
Diazotroph specialist grazing rate	$v_{P_D}$	0.7	$\text{d}^{-1}$
Diazotroph $\text{NO}_3$ uptake threshold	$U_{\text{NO}_3}$	5	$\text{mmol m}^{-3}$
<i>Zooplankton (Z)</i>			
Assimilation efficiency	$\gamma$	0.7	
Maximum grazing rate (at 0°C)	$g_z$	0.5	$\text{d}^{-1}$
Growth efficiency	$\varpi$	0.6	
Mortality	$m_z$	0.02	$\text{d}^{-1}$
Grazing preference $P_O$	$\Psi_{P_O}$	0.26	
Grazing preference $P_D$	$\Psi_{P_D}$	0.04	
Grazing preference $P_C$	$\Psi_{P_C}$	0.26	
Grazing preference $Z$	$\Psi_z$	0.18	
Grazing preference $D$	$\Psi_D$	0.26	
Grazing half-saturation	$k_{graz}$	0.15	$\text{mmol N m}^{-3}$
<i>Detritus (D)</i>			
Remineralization rate	$\mu_{D0}$	0.07	$\text{d}^{-1}$
Sinking speed at surface	$w_{D0}$	20	$\text{m d}^{-1}$
Increase of sinking speed with depth	$m_w$	0.05	$\text{d}^{-1}$

<i>Parameter</i>	<i>Symbol</i>	<i>Value</i>	<i>Units</i>
E-folding temperature of biological rates	$T_b$	15.65	°C
<i>Dissolved Organic Matter</i>			
phytoplankton DOM production factor	$\sigma_{PDOM}$	0.08	
bacterial DOM production factor	$\sigma_{DDOM}$	0.02	
DON remineralization rate (at 0°C)	$\lambda_{DON0}$	9.4E-6	d <sup>-1</sup>
DOP remineralization rate (at 0°C)	$\lambda_{DOP0}$	1.9E-5	d <sup>-1</sup>
<i>Elemental Ratios</i>			
Molar Oxygen:Nitrogen	$R_{O:N}$	11	
Molar Carbon:Nitrogen	$R_{C:N}$	7	
Molar Iron:Nitrogen	$R_{Fe:N}$	38.5	µmol Fe / mol N
Phytoplankton Nitrogen:Phosphorus	$R_{N:PP_O}$	16	
Diazotroph Nitrogen:Phosphorus	$R_{N:PP_D}$	28	
Detritus Nitrogen:Phosphorus	$R_{N:PD}$	16	
Zooplankton Nitrogen:Phosphorus	$R_{N:P_Z}$	16	

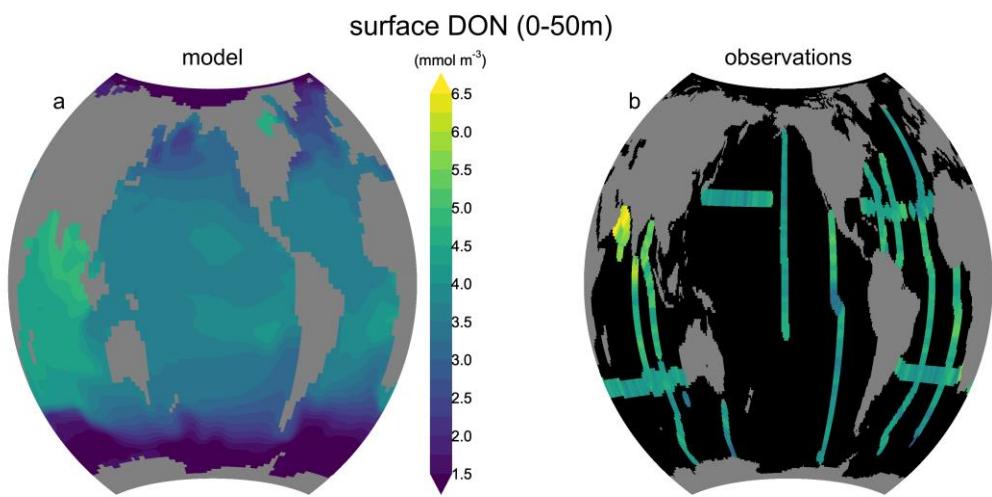
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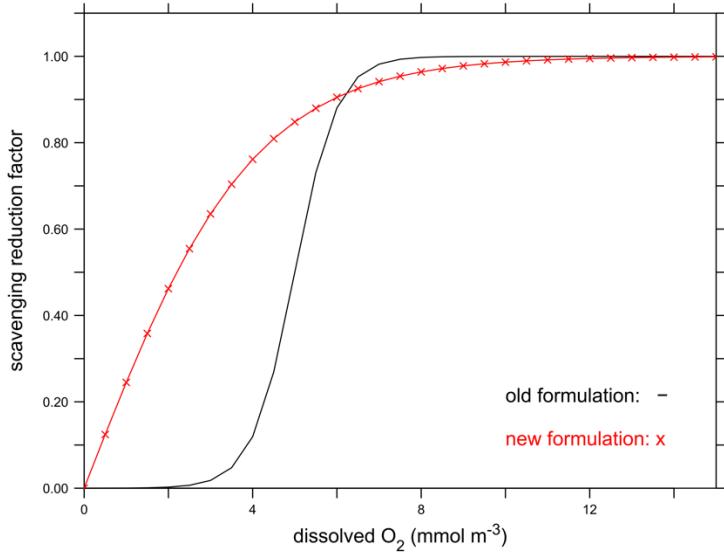
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81 **Figure S1.** Model-data comparison of basin scale average of radiocarbon ( $\Delta^{14}\text{C}$ ) with  
82 GLODAP observations (Key et al., 2004) (left column), and dissolved oxygen ( $\text{O}_2$ ),  
83 apparent oxygen utilization (AOU, center column), and phosphate ( $\text{PO}_4$ , right column)  
84 with World Ocean Atlas observations (Garcia et al., 2010a; Garcia et al., 2010b) (black  
85 circles) and the model simulation #5 *Atm+SedHigh\_LigVar* (red lines).



86  
87 **Figure S2.** Surface (0-50 meters) dissolved organic nitrogen (DON) concentrations in the  
88 model simulation #5 *Atm+SedHigh\_LigVar* and observations (Somes and Oschlies,  
89 2015; Letscher et al., 2013). Note that the model only includes semi-refractory DON,  
90 whereas the observations include total DON.



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**Figure S3.** Modified function that reduces scavenging in oxygen deficient zones.

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