

1 **Constraining global marine iron sources and ligand-mediated scavenging**
2 **fluxes with GEOTRACES dissolved iron measurements in an ocean**
3 **biogeochemical model**

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12 **Key Points:**

- 13 • Global marine iron model tests varying levels of atmospheric deposition, sedimentary release,
14 ligand distributions and scavenging rates
- 15 • Simulations that best reproduce observations include variable ligands and high rates of
16 atmospheric deposition and sedimentary release
- 17 • Simulations with high iron sources require high scavenging rates resulting in short residence
18 times

19 Abstract

20 Iron is a key micronutrient controlling phytoplankton growth in vast regions of the global ocean.
21 Despite its importance, uncertainties remain high regarding external iron source fluxes and
22 internal cycling on a global scale. In this study, we used a global dissolved iron dataset,
23 including GEOTRACES measurements, to constrain source and scavenging fluxes in the marine
24 iron component of a global ocean biogeochemical model. Our model simulations tested three key
25 uncertainties: source inputs of atmospheric soluble iron deposition (varying from 1.4–3.4
26 Gmol/yr), reductive sedimentary iron release (14–117 Gmol/yr), and compared a variable ligand
27 parameterization to a constant distribution. In each simulation, scavenging rates were tuned to
28 reproduce the observed global mean iron inventory for consistency. The variable ligand
29 parameterization improved the global model-data misfit the most, suggesting that heterotrophic
30 bacteria are an important source of ligands to the ocean. Model simulations containing high
31 source fluxes of atmospheric soluble iron deposition (3.4 Gmol/yr) and reductive sedimentary
32 iron release (114 Gmol/yr) further improved the model most notably in the surface ocean. High
33 scavenging rates were then required to maintain the iron inventory resulting in relatively short
34 surface and global ocean residence times of 0.83 and 7.5 years, respectively. The model
35 simulates a tight spatial coupling between source inputs and scavenging rates, which may be too
36 strong due to underrepresented ligands near source inputs, contributing to large uncertainties
37 when constraining individual fluxes with dissolved iron concentrations. Model biases remain
38 high and are discussed to help improve global marine iron cycle models.

39 1 Introduction

40 Iron is a critical micronutrient limiting primary productivity in vast ocean regions (Boyd
41 and Ellwood, 2010; Tagliabue et al., 2017). Iron limitation is responsible for the development of
42 so-called High Nitrate Low Chlorophyll (HNLC) regions of the Southern Ocean, Subarctic North
43 Pacific, Subarctic North Atlantic, and Eastern Equatorial Pacific (Moore et al., 2013). Since
44 dissolved iron (DFe) in the ocean exists in the picomolar (pM) to nanomolar (nM) concentration
45 range, historical measurements with higher detection limits and contamination issues have
46 hindered a robust global understanding of the marine iron cycle compared to macronutrients
47 (Bruland et al., 2014). However, over the past two decades, in large part due to the
48 GEOTRACES program, considerable progress has been made and reliable intercomparable iron

49 measurements have become available that permit a more synoptic view of the marine iron cycle
50 (Schlitzer et al., 2018).

51 The increasing number of robust iron measurements has sparked recent modeling efforts.
52 However, few observational constraints are provided on a global scale, and the degree of
53 complexity and assumptions on the mechanistic processes implemented in global marine iron
54 models have varied dramatically (e.g., Tagliabue et al. (2016)). For example, there is no
55 consensus on the rates of key source fluxes to the ocean, particularly from atmospheric
56 deposition (Anderson et al., 2016) and sedimentary release (e.g., Elrod et al. (2004); Dale et al.
57 (2015)) that vary between $1.4\text{--}30 \text{ Gmol yr}^{-1}$ and $0\text{--}194 \text{ Gmol yr}^{-1}$, respectively, in state-of-the-art
58 marine iron models (Tagliabue et al., 2016). Since uncertainties associated with scavenging and
59 removal of DFe are also high, global marine iron models can tune scavenging rates to reproduce
60 the global iron inventory with large ranges of sources fluxes (Frants et al., 2016; Pasquier and
61 Holzer, 2017).

62 Another key aspect of marine iron models is the representation of ligands which
63 organically bind DFe and thereby prevent it from being scavenged to sinking particulates. Some
64 models still prescribe a globally constant ligand concentration typically at 1 nM, while others
65 account for ligand distributions via a parameterization or directly simulating ligands as a
66 prognostic tracer. Ligands are thought to be produced by microbes as a by-product during the
67 production of organic matter (Gledhill and Buck, 2012), including by heterotrophic siderophores
68 that flourish when systems become iron stressed (Bundy et al., 2018). This has led modelers to
69 predict ligand concentrations by assuming they are produced during the production of organic
70 matter (e.g. Völker & Tagliabue (2015)) or by prescribing a relationship to other organic tracers
71 such as dissolved organic matter and apparent oxygen utilization (e.g., Tagliabue & Völker
72 (2011); Misumi et al. (2013); Pham and Ito (2018)).

73 The uncertainties associated with external source fluxes and scavenging represent key
74 gaps in understanding the global marine iron cycle. This hampers accurate estimates of the DFe
75 budget, residence time and, consequently, its sensitivity to environmental perturbations and
76 climate change. While the rapidly increasing amount of DFe measurements is improving our
77 knowledge of the distribution and inventory of dissolved iron in the ocean, constraining external
78 fluxes has proved to be more difficult. As a result, the range of residence times estimated by the

79 current global marine iron cycle models ranges from less than a decade to multiple centuries
80 (Tagliabue et al., 2016), which limits our ability to confidently predict the impact of changes to
81 the marine iron cycle on productivity in a future ocean. Observational estimates fall within a
82 similar range (Johnson et al., 1997), noting that more recent studies estimate much shorter
83 residence times in the upper ocean (~10 days–4 years) (Croot et al., 2004; Sarthou et al., 2003)
84 depending on the local dynamics, iron pools considered, and source inputs in different regions
85 (Black et al., 2020).

86 In this study, we use a global marine DFe dataset to constrain the iron cycle fluxes in a
87 global marine biogeochemical model. We analyze model sensitivity simulations that focus on
88 three key uncertainties: varying source fluxes of (1) atmospheric soluble iron deposition and (2)
89 reductive sedimentary iron release, as well as the role of a (3) variable ligand distribution on DFe
90 distribution and scavenging rates. The resulting DFe concentrations in each model simulation are
91 evaluated against observations to determine the most realistic marine iron cycle fluxes among the
92 model scenarios.

93 **2 Model Description**

94 We used the UVic Earth System Climate Model (Weaver et al., 2001) version 2.9 (Eby et
95 al., 2009). In the following section, we provide a general overview of the model components
96 then focus on improvements made to the marine iron cycle in this study, whereas other
97 modifications applied to all model simulations are described in the supplementary information.

98 **2.1 Physical Model**

99 The physical ocean-atmosphere-sea ice model includes a three-dimensional ($1.8^\circ \times 3.6^\circ$,
100 19 vertical levels) general circulation model of the ocean (Modular Ocean Model 2) with
101 parameterizations such as diffusive mixing along and across isopycnals and eddy-induced tracer
102 advection (Gent and Mcwilliams, 1990). The physical configuration is based on Somes et al.
103 (2017) and includes parameterizations such as computation of tidally-induced diapycnal mixing
104 over rough topography on the sub-grid scale (Schmittner and Egbert, 2014), anisotropic viscosity
105 (Large et al., 2001; Somes et al., 2010), and enhanced zonal isopycnal mixing schemes in the
106 tropics to better represent zonal equatorial undercurrents (Getzlaff and Dietze, 2013). A two-
107 dimensional, single level energy-moisture balance atmosphere and a dynamic-thermodynamic

108 sea ice model are used, forced with prescribed monthly climatological winds (Kalnay et al.,
109 1996) and constant ice sheets (Peltier, 2004).

110 **2.2 Marine Biogeochemical Model**

111 The updated marine ecosystem-biogeochemical model coupled within the ocean
112 circulation model is based on the Model of Ocean Biogeochemistry and Isotopes (MOBI),
113 version 2.0. Briefly, MOBI includes three prognostic inorganic nutrient tracers (nitrate (NO_3^-),
114 phosphate (PO_4^{3-}), iron (DFe)) and two organic phases (dissolved organic nitrogen (DON) and
115 dissolved organic phosphorus (DOP)), three phytoplankton (ordinary, N_2 -fixing diazotrophs,
116 calcifying coccolithophores), one zooplankton, sinking detritus (i.e. dead particulate organic
117 matter (POM)), as well as dissolved oxygen (O_2), dissolved inorganic carbon, alkalinity, and
118 $\Delta^{14}\text{C}$ (Figure S1). It combines latest features from previous studies focusing on the nitrogen
119 cycle (Somes and Oschlies, 2015), iron cycle (Muglia et al., 2017), and carbon chemistry (Kvale
120 et al., 2015), and is also constrained by isotope systems of ^{13}C and ^{15}N (Schmittner and Somes,
121 2016) (not shown here). Our model experiments were simulated for over 5,000 years under pre-
122 industrial boundary conditions as they approached their quasi steady-state.

123 **2.3 Marine Iron Cycle Model**

124 **2.3.1 Base Configuration**

125 The marine iron model configuration is based on the previous UVic Kiel Marine
126 Biogeochemistry Model (KMBM) (Nickelsen et al., 2015), including improvements
127 implemented in Muglia et al. (2017) (Figure 1). The marine iron model includes explicit tracers
128 for DFe and particulate iron (PFe). All phytoplankton grow with a constant elemental
129 stoichiometry ratio of iron relative to nitrogen. The sources of DFe to the ocean are atmospheric
130 soluble deposition (Luo et al., 2008), reductive dissolution and release from sediments (Elrod et
131 al., 2004; Moore and Braucher, 2008), and hydrothermal fluxes (Tagliabue et al., 2010) (Table 2,
132 Figure 2). The ligand concentration determines the fraction of DFe that is organically complexed
133 and thus unavailable for scavenging, whereas the remaining free DFe (DFe') pool can be
134 scavenged to PFe, which then sinks and remineralizes at the same rate as POM (Table S1). In the
135 base simulation #1, ligands are prescribed to be globally constant at 1 nM as in previous
136 iterations of the model. This simulation is given the name *SrcLow_LigCon* to reflect its

137 differences (i.e., low source inputs of atmospheric soluble deposition and reductive sedimentary
 138 iron release, and constant ligand distribution) from further changes made to the marine iron
 139 model in this study (see subsections below and Tables 1 and 2).

140 **2.3.2 Scavenging**

141 The formulation for scavenging and partitioning of free and organically-complexed DFe
 142 is based on from previous model parameterizations (Nickelsen et al., 2015; Galbraith et al.,
 143 2010). Scavenging of DFe' to PFe occurs via two mechanisms in the model: (1) absorption onto
 144 particulate organic matter following (Honeyman et al., 1988; Parekh et al., 2004)

145

146 | (1) $Fe_{OrgSc} = kFe_{org}DFe POC^{0.58},$

147

148 which is a function of particulate organic carbon (POC), free DFe (DFe'), and the particle
 149 scavenging rate constant (kFe_{org}); and (2) inorganic scavenging

150

151 | (2) $Fe_{InSc} = kFe_{prp}DFe^2,$

152

153 which depends only on DFe' and the inorganic scavenging rate constant (kFe_{prp}) following the
 154 scheme of Galbraith et al. (2010). This inorganic scavenging term primarily represents colloidal
 155 aggregation into larger, sinking particles as well as lithogenic scavenging not explicitly
 156 accounted for in our model. Here we use a non-linear formulation for inorganic scavenging
 157 following Galbraith et al. (2010) which was designed to account for high lithogenic scavenging
 158 rates to better reproduce DFe where atmospheric deposition is high (e.g., tropical and subtropical
 159 North Atlantic) (Pham and Ito, 2019; Ye and Völker, 2017). Note that we included a slightly
 160 higher non-linear exponent (2.) compared to Galbraith et al., 2010 (1.5) that better reproduced
 161 DFe in high atmospheric deposition areas in our model. This difference may be related to the fact
 162 that Galbraith et al., 2010 model included higher phytoplankton iron quotas when DFe is high
 163 which further reduces DFe in that model, whereas our model formulation assumes constant iron
 164 stoichiometry due to high uncertainties associated with this process. Thus, our model performed

165 better with higher scavenging rates to reduce the overestimation of DFe in these high deposition
 166 areas.

167 In each model simulation, the scavenging rate constants (kFe_{org} , kFe_{prp}) were manually
 168 tuned so that each simulation contains a nearly identical global iron inventory with an average
 169 global DFe concentration of 0.7 ± 0.03 nM (Table 2). The inorganic scavenging rate constant was
 170 adjusted until the model reproduced the mean observed DFe concentration in the ocean interior
 171 since it is the dominant form of scavenging there, whereas the POM scavenging rate constant
 172 was adjusted to reproduce declining DFe concentrations towards the surface ocean (Figure 4).
 173 The globally integrated rates of the different scavenging processes are shown in Table 2,
 174 vertically-integrated rates from high and low source input simulations in Figure 2, and total
 175 basin-scale averages in Figure 4.

176 2.3.3 Ligand Parameterization

177 In the base model configuration, a constant ligand concentration of 1 nM is applied
 178 globally, and thus has *LigCon* in its model name (see Table 1). However, the distribution of
 179 ligands in the real ocean is variable (e.g. Völker and Tagliabue (2015)). Since iron-binding
 180 ligands are thought to be produced during the production of organic matter (Gledhill and Buck,
 181 2012), which might explain why dissolved organic matter (DOM) and apparent oxygen
 182 utilization (AOU) may qualitatively reflect some observed ligand concentration patterns (Misumi
 183 et al., 2013; Pham and Ito, 2018; Tagliabue and Völker, 2011). However, a first global model-
 184 data comparison with ligands simulated as prognostic tracers found ligand distributions difficult
 185 to constrain with available observations and is further complicated by large variations in binding
 186 strength of different types of ligands (Völker and Tagliabue, 2015). Therefore, to maintain
 187 computational efficiency, we pragmatically chose to implement ligand concentrations as a
 188 function of existing tracers rather than include additional prognostic tracers.

189 We implemented a variable ligand parameterization to estimate ligand concentrations
 190 based on a function of dissolved organic nitrogen (DON) and apparent oxygen utilization
 191 (AOU):

192

193 | (3)

$$Lig = \alpha AOU^{0.8} + \beta DON^{0.8},$$

194

195 where α ($0.015 \text{ nmol ligand}/(\text{mmol O}_2 \text{ m}^{-3})^{0.8}$) and β ($0.21 \text{ nmol ligand}/(\text{mmol DON m}^{-3})^{0.8}$) are
 196 generic parameters that determine ligand concentration associated with the tracers AOU and
 197 DON, respectively. The parameters α and β were chosen so that the global ligand mean
 198 concentration remained at 1 nM, consistent with simulation #1 with constant ligands, but now
 199 reflects changes in their spatial distribution (Figure 3). Model simulations with this variable
 200 ligand parameterization (simulations #2-5, see Table 1) have *LigVar* in their respective model
 201 simulation name.

202 Although we follow previous studies for the variable ligand parameterization (Misumi et
 203 al., 2013; Pham and Ito, 2018; Tagliabue and Völker, 2011), a few notable changes have been
 204 made in our version. Since AOU can be negative in the surface ocean due to dissolved oxygen
 205 supersaturation, we applied a minimum ligand concentration of 0.5 nM. Previous ligand
 206 parameterizations have also applied minimum ligand concentrations to account for ligands
 207 associated with more refractory forms of DOM not explicitly included in our model (Aumont et
 208 al., 2015; Tagliabue and Völker, 2011). We also applied an exponential parameter (0.8) to the
 209 AOU and DON terms, which reduces ligands associated to these tracers particularly when their
 210 concentrations are high. This helped the model from overestimating DFe concentrations when
 211 AOU and DON concentrations are at their highest concentrations in the model.

212 2.3.4 Reductive Sedimentary Iron Release Parameterization

213 The base model version uses reductive sedimentary iron release based on the Moore and
 214 Braucher (2008) implementation of Elrod et al. (2004),
 215

216 | (4) $Fe_{sed} = \gamma_{FeSed} C_{ox},$
 217

218 where the Fe flux from the sediments (Fe_{sed}) is determined by the sedimentary iron release rate
 219 ($\square_{FeSed} = 0.27 \mu\text{mol Fe}/\text{mmol C}_{ox} \text{ m}^{-2} \text{ d}^{-1}$), and organic carbon oxidation (C_{ox}) in the sediments.
 220 The base model version uses the DFe flux rate from Nickelsen et al. (2015) that is lower than
 221 suggested by Elrod et al. (2004) ($0.72 \mu\text{mol Fe mmol C}_{ox}^{-1} \text{ m}^{-2} \text{ d}^{-1}$). Since this formulation

222 yields lower global rates of this source input in the model compared with other implemented
 223 sedimentary functions included in this study (described below), model simulations with this
 224 sedimentary iron release implementation (#1-2) contain the name *SrcLow*, noting they also
 225 includes a low source input of atmospheric soluble iron deposition (see section 2.3.5 below).

226 We also implemented the sedimentary iron release function proposed by Dale et al.
 227 (2015), who compiled a global dataset of sedimentary DFe fluxes to constrain their model
 228 estimate. While it has a strong dependence on the flux of particulate organic matter to the
 229 seafloor, similar to Elrod et al. (2004), the dataset in Dale et al. (2015) also revealed a strong
 230 dependence on bottom water oxygen concentration. Dale et al. (2015) thus parameterized
 231 sedimentary DFe release as

232

233 | (5)
$$F_{e_{sed}} = \gamma_{FeSedMax} \tanh(C_{ox}: bwO_2),$$

234

235 where $\gamma_{FeSedMax}$ is the maximum flux under steady-state conditions, and bwO_2 is dissolved oxygen
 236 concentration in bottom waters interacting with the sediments.

237 We test two scenarios with the Dale et al. (2015) parameterization by altering the
 238 maximum flux constant ($\gamma_{FeSedMax}$). The *SedHigh* simulations apply the value suggested by Dale
 239 et al. (2015) ($\gamma_{FeSedMax} = 170 \mu\text{mol m}^{-2} \text{ d}^{-1}$), whereas the *SedMid* simulation reduces the
 240 maximum flux value to $100 \mu\text{mol m}^{-2} \text{ d}^{-1}$ to test more a intermediate level of sedimentary DFe
 241 release (see Tables 1 and 2). This reduced value was chosen to test a global sedimentary DFe
 242 flux approximately halfway in between *SedHigh* and *SedLow* since their fluxes differ by a large
 243 amount. Note that the *SedMid* simulation does not produce a significantly different spatial
 244 distribution compared to *SedHigh*.

245 2.3.5 Atmospheric Soluble Iron Deposition

246 We applied the atmospheric soluble iron deposition mask from Luo et al. (2008) in model
 247 simulations #1–4. This atmospheric soluble iron deposition estimate delivers 1.4 Gmol yr^{-1} of
 248 soluble iron to the global ocean, which is on the low-end (AtmLow; see Figure 2) compared to
 249 other estimates applied in the marine iron model intercomparison study (Tagliabue et al., 2016).

250 This estimate from Luo et al. (2008) is one of the first deposition models that explicitly accounts
251 for the soluble iron deposition rather than assuming a constant solubility from total deposition.

252 Another estimate we test in this study applies the average flux from four recent
253 atmospheric soluble iron deposition models (Myriokefalitakis et al., 2018). The intermodel
254 average global soluble deposition rate is 3.4 Gmol yr^{-1} with similar patterns to Luo et al. (2008)
255 but higher rates most notably in the North Atlantic. This simulation with high atmospheric
256 soluble iron deposition (AtmHigh; Figure 2) is applied to the simulation with high sedimentary
257 release and variable ligands and is therefore named *Atm+SedHigh_LigVar*.

258 **3 Model Results and Data Comparison**

259 **3.1 Global Dissolved Iron Dataset**

260 The DFe database used in this study is a collection of observations from both
261 GEOTRACES Intermediate Data Product 2017 (7520 points; Schlitzer et al. (2018)) and prior
262 observations compiled by Tagliabue et al. (2012) (12371 points). Note that we excluded 37
263 measurements (19 from GEOTRACES, 18 from prior) with high DFe concentrations between 10
264 nM to 216 nM mainly from locations with high hydrothermal activities, but also some near-shore
265 settings (e.g. Laptev Sea, Bristol Bay, Peruvian coastal waters near urban area of Trujillo) and
266 around small islands not resolved in the model (e.g., Kerguelen, Indonesian and Coronation), and
267 thus the dataset used here contains concentrations up to 10 nM. We then interpolated the data
268 onto the UVic model grid using the PyFerret SCAT2GRIDGAUSS function developed by
269 NOAA's Pacific Marine Environmental Laboratory, which is a Gaussian interpolation function
270 based on Kessler and McCreary (1993). This gridded data was used for the model-data
271 comparison (Figures 3–7) and to calculate model-data statistical metrics (i.e. correlation
272 coefficient, (uncorrected) standard deviation, and root-mean-squared error) (Figure 8). It covers
273 5917 grid points since many observations overlap and thus are averaged on corresponding grid
274 points. Since we compare to annual model results, we interpolated all observations onto the grid
275 and thus temporal aspects and variability of the data is not taken into account or investigated in
276 this study.

277 Model-data misfit statistical metrics are sensitive to unresolved outlier concentrations and
278 spatial extent of the data interpolation onto the model grid. However, these aspects do not affect

which simulations best reproduce the global dataset according to statistical metrics. This is illustrated by comparing metrics calculated from all observations (triangles) to only GEOTRACES (circles) in Figure 8. The statistical metrics slightly improve when comparing against only GEOTRACES observations, with the only exception being root-mean-squared error for model simulation #1 in the surface ocean, but the relative improvements in the model simulations are nearly identical. The arbitrary exclusion concentration threshold of 10 nM was chosen as a balance between including as many observations as possible while still being able to calculate useful statistical metrics that are not dominated by these outlier concentrations.

3.2 Variable Ligand Distribution

The simulation with constant ligands does not reproduce the major basin-scale features of the observed DFe distribution, despite that its globally averaged depth profile is generally consistent with observations (Figure 4c). Most notably, simulations with constant ligands significantly overestimate the DFe in the interior Southern Ocean (Figure 4o), a critical ocean basin for Fe-limited phytoplankton growth. *LigCon* thus overestimates supply of DFe via upwelling, and underestimates Fe limitation of phytoplankton growth, which is a key deficiency in the base configuration and previous model versions (e.g. Muglia et al. (2017)). They also underestimate DFe in intermediate waters in the Indian and Pacific Ocean (Figures 4k, 5b), which we have averaged together since they have similar deep ocean biogeochemical tracer profiles relative to the global average (Figure S1).

The simulations with variable ligand concentrations (#2-5; *LigVar*) better reproduce the ocean interior distribution of DFe (Figure 5). This is primarily due to the AOU dependence of the variable ligand parameterization which mainly determines ligand concentrations in the deep ocean since semi-refractory DOM concentrations are low there in the model. This is most obvious when comparing intermediate depths of the Southern and Indian-Pacific Oceans, which contain relatively low and high values of AOU and thus ligand concentrations, respectively, according to our parameterization (see Figures 3, S1). Lower ligand concentrations in the Southern Ocean enhances scavenging causing lower DFe concentrations, with the opposite effect occurring in the Indian-Pacific Ocean, and better reproduces observations in both basins. Therefore, the interior DFe distribution with the variable ligand parameterization is better partitioned with respect to observations (Figures 4,5) and improves the global model-data misfit

309 by 9.2% when averaging across our three metrics (i.e. correlation coefficient (R), normalized
310 standard deviation ($nSTD$), and normalized root-mean-squared error ($nRMS$));
311 $(\Delta R + \Delta nSTD + \Delta nRMS) / 3 \times 100\%$) against all observations (Figure 8), which represents the
312 largest improvement from any individual simulation in this study.

313 The concentration of semi-refractory DON largely determines ligand concentrations in
314 the surface ocean (Figure 3a). DON concentrations are higher around the high productivity
315 regimes in the low latitudes with generally decreasing values towards higher latitudes (Somes
316 and Oschlies, 2015) (Figure S2). This pattern is reflected in the surface DFe distribution that
317 shows the same latitudinal trend in the variable ligand model (Figure 6c-d). While this
318 meridional DFe pattern better reproduces low DFe concentrations in the open Southern Ocean, it
319 creates larger model-data biases on high latitude continental shelves in the Bering Sea, Weddell
320 Sea, and European shelf seas (Figures 6a-d, 7c,e). This shows that while the overall variable
321 ligand effect significantly improves the global DFe distribution (Figure 7), model-data biases in
322 some regions (e.g. high latitude continental shelf seas) still increase, which contributes to a
323 smaller average metric improvement (3.9%) in the surface layer compared to the global ocean.

324 **3.3 Sedimentary Iron Release**

325 The simulations with low sedimentary source inputs (#1-2 *SrcLow*) provide a relatively
326 poor fit to observed DFe concentrations according to the statistical metrics (Figure 8). They fail
327 to reproduce the high DFe concentrations near continental margins (Figures 6, 7), suggesting
328 higher sedimentary release rates are necessary to explain these features. The simulated DFe
329 distribution also lacks the strong spatial gradient towards depleted concentrations in many open
330 ocean regions in the observations. These overly smooth gradients in *SrcLow* are the result of low
331 sedimentary release rates and subsequent low scavenging rates that are then required to
332 reproduce the global mean DFe inventory, resulting in a relatively long global mean residence
333 time of 35 years among our simulations (Table 2).

334 The simulations with higher sedimentary release rates (Figure 2e) produce higher DFe
335 concentrations in continental shelf seas (Figures 6,7), particularly where bottom water oxygen is
336 low in the low latitudes. The simulations applying high-end sedimentary Fe release rates
337 (*SedHigh*) modestly outperformed simulations assuming lower rates across all calculated

338 statistical metrics Figure 8)) on average by 3.3% in the global ocean and slightly higher by 3.8%
339 in the surface layer, with the intermediate release rate scenario *SedMid* performed between
340 *SedLow* and *SedHigh*. Therefore, our model-data analysis suggests that high-end estimates for
341 global reductive sedimentary iron release rates are the most realistic.

342 One region that was notably improved by high sedimentary release rates was the low
343 latitude margins near oxygen deficient zones (ODZs) (Figures 6, 7). Observations there in both
344 the eastern tropical South Pacific off Peru (Figure 7a), eastern tropical South Atlantic off
345 Namibia (Figure 7d), and northern Indian Ocean show high DFe concentrations that are best
346 reproduced in *SedHigh* scenarios. Since *SedHigh* simulations also contain high scavenging rates,
347 they better reproduce the lowest DFe concentrations in the offshore open ocean locations as well.

348 The high DFe concentrations on high latitude continental shelf systems (Figures 6, 7c,e)
349 are not improved in *SedHigh_LigVar* due to the interactions with ligands and scavenging.
350 Decreasing surface ligand concentrations towards high latitude systems (Figure 3) allow
351 scavenging to compensate the additional sediment-derived DFe more efficiently, in contrast to
352 low latitude systems near ODZs (e.g. Tropical Pacific) that contain higher ligands allowing DFe
353 to be retained in the water column. This causes the simulation with constant ligands to retain
354 slightly higher DFe compared to simulations with variable ligands in high latitude continental
355 shelf systems (e.g., Bering Sea (Figure 7c) and European Shelf Seas (Figure 7e)), despite that
356 these simulations with variable ligands include much higher sedimentary release rates there (e.g.
357 *SedHigh_LigVar*, Figure 2). This demonstrates that more efficient scavenging rates associated
358 with low ligands can overcompensate the high sedimentary release rates in determining DFe
359 concentrations in the model.

360 **3.4 Atmospheric Soluble Deposition**

361 The two soluble atmospheric deposition scenarios tested here predict similar spatial
362 depositional patterns (Figure 2), with the more recent GESAMP intermodel average
363 (Myriokefalitakis et al., 2018) providing a significantly higher global deposition rate (3.4 Gmol
364 yr^{-1}) relative to the low estimate from Luo et al. (2008) (1.4 Gmol yr^{-1}). These enhanced rates
365 cause higher DFe concentrations mainly from the Saharan dust plume in subtropical North
366 Atlantic, but also to a lesser degree in the Arabian Sea and North Pacific (Figure 6g,h, Figure
367 7c,f). The impact of including higher soluble deposition only slightly improves the global model-

368 data statistical metrics by 0.7% globally and 1.5% in the surface layer, making it difficult to
369 determine the most realistic rates based on our model-data DFe comparison alone.

370 **3.5 High Scavenging Effect**

371 In model simulations with high source fluxes (e.g. #5 *Atm+SedHigh_LigVar*), higher
372 scavenging rates are necessary to maintain a realistic global DFe inventory (Tables 1 and 2,
373 Figures 2h-i, 3). Scavenging is thus more efficient at reducing DFe concentrations in the high
374 source flux simulations. In regions far away from the source fluxes, particularly in the deep
375 ocean and open Southern Ocean (e.g. see Figure 6), the model simulations with higher source
376 fluxes actually contain lower DFe because the enhanced scavenging outweighs the source fluxes
377 in these areas (Figure 4). Lower DFe concentrations in these deep and open ocean regions better
378 reproduce observations further improving the model-data misfit metrics (Figure 8). The
379 combined effects of high atmospheric and sedimentary source inputs, which also includes
380 highest scavenging rates, contributed to the largest improvement in the surface ocean across our
381 metrics (5.5% improvement relative to *SrcLow_LigVar*).

382 **4 Discussion**

383 **4.1 Model-Data Constraints and Uncertainties**

384 The variable ligand parameterization improved the model's ability to reproduce the
385 global distribution of DFe observations the most. This is most evident in the interior ocean due to
386 AOU dependency of this parameterization. Since ligands are produced when dissolved oxygen is
387 consumed during the respiration of POM via heterotrophic microbes in the variable ligand
388 parameterization, their concentrations reach maximum values in old Pacific intermediate waters
389 (Figure 3). High ligands reduce scavenging that causes the model to better reproduce high
390 observed DFe concentrations there (Figures 4k, 5c), a feature that has also been demonstrated in
391 other models (e.g. see also (Misumi et al., 2013; Pham and Ito, 2018; Frants et al., 2016)). This
392 model improvement suggests that ligand production by heterotrophic bacteria is a key
393 mechanism maintaining the global marine iron cycle.

394 The model simulations that include higher source inputs and scavenging rates show a
395 subtle but continuous improvement in the model-data misfit metrics particularly in the surface
396 ocean (Figure 8). This is in contrast to the model comparison study of Tagliabue et al. (2016),

which showed no clear relationship between model performance and source inputs, as well as an inverse modeling study of Pasquier and Holzer (2017), which could not find an optimal solution among their large set of model simulations varying source inputs. However, Pasquier and Holzer (2017) only tested relatively low sedimentary release rates (up to 22 Gmol/yr compared to 117 Gmol/yr in this study) and also did not include an oxygen dependency that has a strong influence in our parameterization. Our analysis emphasizes that future modeling studies should test these important factors associated with reductive sedimentary DFe release that contributed to the model improvements in this study.

The ligand and high sedimentary DFe release effects have similar impacts on DFe spatial distributions making it difficult to constrain their individual impacts with DFe concentrations alone. This spatial overlap is most pronounced above ODZs in the eastern tropical Pacific, eastern tropical Atlantic, and Northern Indian Ocean (Figure 6). This spatial covariance occurs because when AOU is high, bottom water oxygen is typically low. Therefore, DFe concentrations are enhanced both by reduced scavenging due to high ligands where AOU is high, as well as by higher sedimentary DFe release rates where bottom water oxygen is low. Future studies should examine the integrative DFe cycling in these systems (e.g. sedimentary release and scavenging rates, ligand concentrations) to give additional insights on individual processes contributions to total DFe.

Despite high sedimentary release rates, the *SedHigh* model simulations still underestimate DFe on most continental shelf systems (Figure 7). The poorly resolved coastal dynamics in our coarse resolution circulation model is likely a key model deficiency preventing the model from representing many coastal dynamics where sedimentary DFe fluxes are high. Coarse resolution models underestimate coastal upwelling and the nutrient input on narrow shelf systems that drive productivity. This bias causes underestimated particulate organic matter production as well as overestimated dissolved bottom water oxygen concentrations, both of which would contribute to underestimating reductive sedimentary DFe release rates and from coastal shelf systems.

Further complicating matters are interactions between sedimentary DFe release rates, ligands, and scavenging. For example, our *SedHigh_LigVar* model simulation releases significantly higher DFe on high latitude shelves (Figure 2e-f). However, only a small part of

427 this DFe remains in the dissolved pool since scavenging efficiently converts it to particulate iron
428 that eventually sinks back to the sediments (Figure 2h-i). Therefore, our model underestimation
429 of DFe concentrations remains despite high DFe release rates. This strong spatial coupling
430 between source and scavenging fluxes has also been demonstrated in other modeling studies
431 (Frants et al., 2016; Pasquier and Holzer, 2017), which also found that this tight spatial coupling
432 significantly contributes to the difficulty in constraining source inputs. The exclusion of riverine
433 inputs that may also directly include ligands could also contribute to overly efficient scavenging
434 resulting in underestimated DFe. If our ligand parameterization predicted higher concentrations
435 on these high latitude shelf systems, which has been indicated by ligand observations (Völker
436 and Tagliabue, 2015), this would prevent rapid scavenging of DFe released from sediments and
437 better reproduce observations.

438 Sedimentary DFe release rates may still be underestimated even in our high release
439 scenario. Note that our highest tested global sedimentary release rate (117 Gmol yr^{-1}) was not the
440 highest from the marine iron model intercomparison (up to 194 Gmol yr^{-1}) (Tagliabue et al.,
441 2016), and every model scenario tested here with increased source fluxes improved the model-
442 data misfit metrics (Figure 8). Potentially important sedimentary processes not included in the
443 model are non-reductive dissolution and release from reactive sediments in tectonically active or
444 volcanic regions (Conway and John, 2014; Homoky et al., 2013) and sedimentary colloidal
445 production/release (Homoky et al., 2021), which could further contribute to higher total
446 sedimentary DFe release rates that may improve the model-data misfit.

447 An important limitation of applying these empirical functions of reductive sedimentary
448 DFe release (e.g. (Dale et al., 2015; Elrod et al., 2004)) in global models is that total iron balance
449 within the sediments is not explicitly accounted for. Thus, these parameterizations can
450 potentially represent an unlimited long-term supply of DFe to the ocean which is unrealistic.
451 This simplification can be justified because many important sources of particulate Fe to the
452 sediment are not yet included in the model, e.g. atmospheric and riverine input of lithogenic
453 material and in situ production at volcanic islands or active margins, which provide DFe for
454 release. Also note that the Dale et al. (2015) parameterization applied in the *SedHigh* simulations
455 sets a maximum rate determined under steady-state conditions which caps potentially unrealistic
456 high release rates. While this simplification is likely not a significant deficiency in steady-state

457 model simulations presented here, this should be considered in transient simulations with
458 substantial enhancement of sedimentary DFe fluxes.

459 Atmospheric deposition often occurs at high rates over continental shelves (e.g. North
460 Pacific, Patagonia) and ODZs (e.g. Arabian Sea), again making it difficult to constrain individual
461 processes driving DFe concentrations when multiple processes act together in close spatial
462 proximity. For example, our high atmospheric soluble deposition scenario helps reproduce high
463 DFe concentrations in the Arabian Sea (Figure 7f). However, our model underestimates the
464 extent of the Arabian Sea ODZ which could be the real cause driving high DFe concentrations
465 there via high sedimentary DFe release, reduced scavenging, and/or enhanced redox cycling
466 (Moffett et al., 2007). Instead the model ODZ is mostly misplaced to the Bay of Bengal, where
467 higher simulated DFe there in the model better matches observations within the real ODZ in the
468 Arabian Sea (see star symbols in Figure 7f).

469 The model simulations do not resolve the high variance of the observations which is
470 reflected in the underestimated standard deviation (Figures 4,8). This occurs everywhere in the
471 ocean and is most pronounced in the Southern Ocean due to it containing very low DFe in the
472 open ocean but also high concentrations near islands, continental margins, and hydrothermal
473 vents (Figures 4–6). Although not a focus of this study, the model was not able to reproduce the
474 full spatial extent of high DFe concentrations near hydrothermal vents at mid-ocean depths
475 (Figures 4,5), despite that this source is included (Table 2). Previous modeling studies were only
476 able to reproduce this high DFe extent when assuming that the hydrothermal vents were also a
477 significant source of ligands (Frants et al., 2016; Resing et al., 2015) or included stabilization via
478 reversible scavenging (Roshan et al., 2020), both of which we have not accounted for in our
479 model. This emphasizes that future model versions should include all important ligands and
480 scavenging dynamics to better represent their importance in marine iron models, but that a more
481 robust global database of ligand concentrations including their binding strength would be
482 required (Völker and Tagliabue, 2015).

483 High variance in the global dataset may not reflect mean climatological conditions
484 simulated by the preindustrial steady-state model results given the highly dynamic nature of DFe
485 cycling particularly in the surface ocean with short residence times (Black et al., 2020). The
486 spatial and temporal sparsity of the dataset likely contribute to high variance as well. But note

487 that the standard deviation was significantly improved in our best model simulation with variable
488 ligands and high source/scavenging fluxes (*Atm+SedHigh_LigVar*; see Figure 4, 8b,e)
489 suggesting that a model with low residence times can better reproduce the high variance and
490 strong gradients in the DFe observations. Since most DFe observations have been collected in
491 recent decades, there could already be a significant anthropogenic impact (e.g. enhanced
492 deoxygenation, atmospheric/riverine pollutants) on the global marine iron cycle not included in
493 these model simulations, especially if the marine DFe residence time operates on decadal
494 timescales or less. Future additions and expansion to the global DFe dataset as well as
495 comparison with transient model simulations at the same period of data collection will improve
496 uncertainties in future model-data analyses.

497 **4.2 A global marine iron cycle with a residence time under a decade?**

498 Our model simulations testing various external source fluxes in the global marine iron
499 cycle result in global average residence times ranging from 7.5 to 36 years. The simulation that
500 best reproduces the observations (*Atm+SedHigh_LigVar*) has the lowest residence time (global:
501 7.5 years; surface ocean: 0.83 years) among our model experiments. This low-end residence time
502 is caused in large part due to the high source fluxes, with the reductive sedimentary release being
503 the most important with the highest global rate in our simulations. These high source fluxes need
504 to be compensated by efficient scavenging and subsequent removal via burial in the sediments to
505 reproduce the distribution and global mean inventory in DFe observations, a model feature that
506 was also found in other modeling studies (e.g. see Frants et al. (2016); Pasquier and Holzer
507 (2017)).

508 This is in general agreement with observational studies focusing on the surface layer
509 (Black et al., 2020; Sarthou et al., 2003). For example, Black et al. (2020) estimated similar
510 residence times throughout the global surface ocean (0-250 meters) for DFe ranging from
511 approximately 1 month to 4 years depending on the region and specific iron pools considered,
512 although noting that the uncertainties remain large (i.e. equal or greater than the absolute value
513 of the estimate in each region). These generally low surface residence times are captured in our
514 model simulations that range from 0.83 to 3.12 years (Table 2). However, residence times of
515 individual molecules and regions can further vary depending on the local coupling of source
516 inputs, scavenging efficiency, and regeneration (e.g. Holzer et al. (2016; 2018); Tagliabue et al.

517 (2019)). For instance, DFe in the ocean interior is more stable and controlled by the amount of
518 ligands that reduces scavenging and removal to the sediments via sinking particulates,
519 contributing to the longer global residence times.

520 **4.3 Marine iron flux impacts on global ocean biogeochemistry**

521 An interesting feature of the model simulations is that there is surprisingly little change to
522 globally averaged marine productivity and export production (Table 3). This occurs in large part
523 in the model because scavenging was also increased in high sedimentary iron release scenarios,
524 and thus much of the additional DFe fluxes from the sediments is efficiently scavenged to
525 particulate iron that sinks back to the sediments before it can be transported to the surface ocean
526 where it may stimulate additional productivity. This general impact was also found in a model
527 study using a previous iteration of the model version used here but comparing different
528 complexities of the marine iron configurations (Yao et al., 2019) as well as other inverse
529 modeling studies (Pasquier and Holzer, 2017, 2018). However, it must be noted that all of these
530 model studies, including this study, only evaluated steady-state simulations in which uncertain
531 parameters were manually tuned or optimized to best reproduce observations. Therefore, they are
532 not necessarily indicative to how the iron dynamics in the model may respond to and impact
533 marine productivity in externally-forced transient scenarios.

534 There is a notable decrease in marine productivity and export production in the Southern
535 Ocean among our model simulations with better representations of the global iron distribution
536 (Table 3). The variable ligand parameterization predicts less ligands in the Southern Ocean
537 (Figure 3), which allows higher scavenging to reduce DFe that better reproduces observations.
538 Furthermore, since external iron sources in the Southern Ocean are small (Figure 2,4m), the
539 enhanced scavenging in the high source flux simulations removes more DFe than source fluxes
540 add to the Southern Ocean. Therefore, DFe levels further decrease in the Southern Ocean
541 (Figures 4o, 6) in the high source flux scenarios. The high scavenging in our best model
542 simulation with variable ligands and high source fluxes (*Atm+SedHigh_LigVar*) reduces DFe,
543 marine productivity and resulting oxygen consumption during remineralization of particulate
544 organic matter, thereby increasing dissolved oxygen concentrations at depth. This effect is
545 significant enough to increase average global dissolved oxygen concentrations by 8% in the
546 model because water masses formed in the Southern Ocean contribute to much of the global

547 deep ocean (Table 3). This emphasizes the importance of simulating a robust global marine iron
548 cycle most importantly in the Southern Ocean.

549 **5 Conclusions**

550 In this study we tested various rates of atmospheric soluble deposition, reductive
551 sedimentary release, and variable ligand distributions within a marine iron component in a global
552 ocean biogeochemical model. The simulations that best reproduce the global DFe observations
553 include highest tested source fluxes and a variable ligand parameterization. The most striking
554 feature in the global DFe observations that supports this hypothesis is the strong gradients that
555 often occur with high concentrations near source fluxes and low concentrations in adjacent open
556 ocean regions. This high source flux/scavenging iron cycling regime causes a relatively short
557 residence times of less than a decade in the global oceans and less than a year in the surface
558 ocean. The short residence time implies that the global marine iron cycle is highly sensitive to
559 environmental perturbations in the Anthropocene and geological past. Uncertainties remain high
560 due to model parameterizations of complex, poorly understood, and often intertwined processes
561 (e.g. ligand production and subsequent control on scavenging near source inputs) and the sparsity
562 of DFe and ligand measurements throughout the global ocean. Nevertheless, our model-data
563 analysis suggests the marine iron cycle operates with high global source inputs and scavenging
564 rates and low residence times compare to most previous estimates.

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583 GEOMAR open access repository (<https://thredds.geomar.de>).

584 **Table 1. Marine Iron Model Configurations**

#	Simulation Name	Atmospheric soluble deposition	Reductive sedimentary release	Ligand distribution	Inorganic Scavenging (kFe_{prp}^a)	Particle Scavenging (kFe_{org}^b)
1	SrcLow_LigCon	Low ^c	Low ^d	Constant ^e	0.0069	1.2
2	SrcLow_LigVar	Low	Low	Variable ^f	0.0052	1.5
3	SedMid_LigVar	Low	Mid ^g	Variable	0.0069	2.2
4	SedHigh_LigVar	Low	High ^h	Variable	0.0081	2.9
5	Atm+SedHigh_LigVar	High ⁱ	High	Variable	0.0098	2.9

585

586 a Inorganic scavenging parameter has units of $(\text{mmol Fe/m}^3)^{-2}\text{d}^{-1}$ 587 b Particle scavenging parameter has units of $(\text{gC/m}^3)^{-0.58}\text{d}^{-1}$

588 c (Luo et al., 2008)

589 d (Elrod et al., 2004) parameterization with low flux rate (see section 2.3.4)

590 e Constant concentration of 1 nM everywhere in the ocean

591 f Variable ligand parameterization (see section 2.3.3)

592 g Dale et al. (2015) parameterization with intermediate maximum flux rate $100 \mu\text{mol Fe m}^{-2} \text{d}^{-1}$ 593 h Dale et al. (2015) parameterization with suggested maximum flux rate $170 \mu\text{mol Fe m}^{-2} \text{d}^{-1}$

594 i (Myriokefalitakis et al., 2018)

595 **Table 2. Global Marine Iron Cycle Results**

#	Simulation Name	Atmospheric soluble deposition (Gmol yr ⁻¹)	Reductive Sedimentary release (Gmol yr ⁻¹)	Hydro-thermal (Gmol yr ⁻¹)	Inorganic Scavenging (Gmol yr ⁻¹)	Particle Scavenging (Gmol yr ⁻¹)	Dissolved Iron (nM)	Global Residence time ^a (yr)	Surface Residence time ^b (yr)
1	SrcLow_LigCon	1.4	15.1	11.4	34.3	22.5	0.68	33.3	3.12
2	SrcLow_LigVar	1.4	14.6	11.4	30.9	29.3	0.73	35.9	2.56
3	SedMid_LigVar	1.4	68.6	11.4	99.3	55.9	0.73	12.2	1.35
4	SedHigh_LigVar	1.4	117	11.4	159	83.9	0.73	7.66	0.87
5	Atm+SedHigh_LigVar	3.4	114	11.4	162	81.5	0.71	7.49	0.83

596

597 ^aSince our iron model simulates active (re)cycling between particulates and dissolved forms and thus scavenging does not permanently
 598 remove bioavailable iron from the system, we calculate residence time based on global external fluxes and bulk inventory, i.e. global
 599 Fe inventory/ \sum Source Inputs.

600 ^bFor surface residence time, we follow Black et al. (2020) by including the upper 250 meters and account for sinking particulate iron
 601 out of this layer as the sink flux. Since our particulate iron pool includes both biogenic (i.e. produced during primary production) and
 602 authigenic (i.e. produced by scavenging) iron in the model, this model residence time is comparable to their mean dissolved,
 603 biogenic+authigenic estimate, which ranges from 0.1 to 4 years depending on location.

604 **Table 3. Global Marine Biogeochemistry Results**

#	Simulation Name	Net Primary Production (Gt C yr ⁻¹)		Export Production (Gt C yr ⁻¹)		dissolved O ₂ (mmol m ⁻³)	
		Global	Southern	Global	Southern	Global	Southern
1	SrcLow_LigCon	47.0	8.11	8.1	2.12	167	206
2	SrcLow_LigVar	47.4	7.09	7.9	1.86	175	216
3	SedMid_LigVar	47.7	6.72	7.9	1.75	178	221
4	SedHigh_LigVar	48.0	6.67	7.9	1.74	179	222
5	Atm+SedHigh_LigVar	47.9	6.42	7.8	1.68	181	224

605

606 **Figure 1.** Schematic of the marine iron (Fe) model. See section 2.3 for a full description.

607

608 **Figure 2.** Vertically-integrated fluxes of atmospheric soluble iron deposition (top row)
609 prescribed on model simulations #1-4 from Luo et al. (2008) (*AtmLow*) (a), high scenario
610 (*AtmHigh*) from the GESAMP intermodel average (Myriokefalitakis et al., 2018) (b), and their
611 difference (c). Center row: Vertically-integrated sedimentary iron release using
612 parameterizations based on Elrod et al. (2004) (*SedLow* from simulation #2) (d) and Dale et al.
613 (2015) (*SedHigh* from simulation #4) (e), and their difference (f). Bottom row: Vertically-
614 integrated total scavenging rates from simulation #2 with low source input and scavenging rates
615 (*SrcLow*) (g) and simulation #5 with highest rates (*Atm+SedHigh*) (h), and their difference (i).

616

617 **Figure 3.** Distribution of variable ligand concentrations in the surface (0-250 meters) ocean (a),
618 and basin-scale averages in the Atlantic (b), Indian (c), Pacific (d), and Southern (e). Note that
619 the Southern Ocean region ($>40^{\circ}\text{S}$) from within the other basins (b-d) is excluded there since it is
620 shown in (e).

621

622 **Figure 4.** Annually averaged depth profiles of marine iron source inputs (left column),
623 scavenging rates (center-left column), dissolved iron concentrations (center-right column), and
624 dissolved iron (DFe) standard deviation (Std Dev) (right column) in the Global, Atlantic, Indian-
625 Pacific, and Southern Ocean for model simulations (colored symbols) and dissolved iron
626 observations (black circles). Source inputs (left column) are atmospheric soluble deposition as
627 large filled symbols in the low scenario (*AtmLow*, green down-pointing triangle) and high
628 (*AtmHigh*; red diamonds) scenarios, sedimentary iron release in the low (*SedLow*; blue
629 hexagons) and high scenarios (*SedHigh*; purple triangles), and hydrothermal flux (green square,
630 applied to all simulations). For dissolved iron concentrations (center-right column), lines show
631 model averages in the entire selected domain, while symbols include model results only where
632 dissolved iron observations exist. Note that the Southern Ocean region ($>40^{\circ}\text{S}$) from within the
633 Atlantic and Indian-Pacific basins is excluded there since it is shown in the Southern Ocean
634 panels.

635

636 **Figure 5.** Annual, zonally averaged dissolved iron concentrations in the Indian-Pacific and
637 Atlantic basins in observations (a), *SrcLow_LigCon* (b), *SrcLow_LigVar* (c), *SedHigh_LigVar*
638 (e), and *Atm+SedHigh_LigVar* (g). Right column highlights individual effects on dissolved iron
639 concentrations by showing model differences from variable ligands (i.e. *SrcLow_LigVar* –
640 *SrcLow_LigCon*) (d), high sedimentary iron release (i.e. *SedHigh_LigVar* – *SrcLow_LigVar*) (f),
641 and high atmospheric soluble deposition (i.e. *Atm+SedHigh_LigVar* – *SedHigh_LigVar*) (h). In
642 locations where no observations exist (black region in a), zonal model averages are shown
643 (b,c,e,g).

644

645 **Figure 6.** Annually averaged dissolved iron concentrations in the upper 250 meters in
646 observations (a), *SrcLow_LigCon* (b), *SrcLow_LigVar* (c), *SedHigh_LigVar* (e), and
647 *Atm+SedHigh_LigVar* (g). Right column highlights individual effects on dissolved iron
648 concentrations by showing model differences from variable ligands (i.e. *SrcLow_LigVar* –
649 *SrcLow_LigCon*) (d), high sedimentary iron release (i.e. *SedHigh_LigVar* – *SrcLow_LigVar*) (f),
650 and high atmospheric soluble deposition (i.e. *Atm+SedHigh_LigVar* – *SedHigh_LigVar*) (h).

651

652 **Figure 7.** Comparison of dissolved iron measurements from GEOTRACES (black circles) and
653 others (black down-pointing triangles) in the upper 250 meters with model simulations
654 *SrcLow_LigCon* (green squares), *SrcLow_LigVar* (blue hexagons), *SedHigh_LigVar* (purple
655 triangles), *Atm+SedHigh_LigVar* (red diamonds) across ocean the western equatorial Pacific ((10°S–10°N) (a); eastern tropical South Pacific (5°S–15°S) (b); and eastern North Atlantic
656 (30°W–0°) (c); eastern tropical South Atlantic (35°W–15°)(d); central North Pacific (175°–
657 150°W); Indian (zonal averaged from 20°–100°E) (e). The intersecting continental margin or
658 shelf sea at the end of the transect is given in parenthesis. Model results are included only at
659 locations where observations exist. Since the core of oxygen deficient zones in the model does
660 not directly overlap with the real ocean where high dissolved iron concentrations exist in the
661 eastern tropical South Pacific (b) and northern Indian Ocean (f), we added dissolved iron
662 concentrations directly above the core of the oxygen deficient zones ($O_2 < 5 \text{ mmol m}^{-3}$) in the
663 model as star symbols.

664

666 **Figure 8.** Model-data statistical misfit metrics calculated using all observations (triangles) and
667 using only GEOTRACES observations (circles). Correlation coefficient (left column), standard
668 deviation (center column), root-mean-squared error (right column) are calculated for the global
669 ocean (top rows) and upper 250 meters of the water column (bottom rows). Standard deviation
670 (b,e) and root-mean-squared error (c,f) are normalized by the standard deviation of observations.
671 The root-mean-squared error vertical axis has been inverted so the upwards direction represents a
672 better model misfit in all panels. Note a perfect representation of observations would yield the
673 value 1 for correlation coefficient, 1 from normalized standard deviation, and 0 for normalized
674 root-mean-squared error.

675 **References**

- 676 Anderson, R. F., Cheng, H., Edwards, R. L., Fleisher, M. Q., Hayes, C. T., Huang, K. F., Kadko,
677 D., Lam, P. J., Landing, W. M., Lao, Y., Lu, Y., Measures, C. I., Moran, S. B., Morton, P. L.,
678 Ohnemus, D. C., Robinson, L. F., and Shelley, R. U.: How well can we quantify dust deposition
679 to the ocean?, *Philos Trans A Math Phys Eng Sci*, 374, 10.1098/rsta.2015.0285, 2016.
- 680 Aumont, O., Ethé, C., Tagliabue, A., Bopp, L., and Gehlen, M.: PISCES-v2: an ocean
681 biogeochemical model for carbon and ecosystem studies, *Geoscientific Model Development*, 8,
682 2465-2513, 10.5194/gmd-8-2465-2015, 2015.
- 683 Black, E. E., Kienast, S. S., Lemaitre, N., Lam, P. J., Anderson, R. F., Planquette, H., Planchon,
684 F., and Buesseler, K. O.: Ironing Out Fe Residence Time in the Dynamic Upper Ocean, *Global
685 Biogeochemical Cycles*, 34, 10.1029/2020gb006592, 2020.
- 686 Boyd, P. W. and Ellwood, M. J.: The biogeochemical cycle of iron in the ocean, *Nature
687 Geoscience*, 3, 675-682, 10.1038/ngeo964, 2010.
- 688 Bruland, K. W., Middag, R., and Lohan, M. C.: 8.2 - Controls of Trace Metals in Seawater, in:
689 Treatise on Geochemistry (Second Edition), edited by: Holland, H. D., and Turekian, K. K.,
690 Elsevier, Oxford, 19-51, <https://doi.org/10.1016/B978-0-08-095975-7.00602-1>, 2014.
- 691 Bundy, R. M., Boiteau, R. M., McLean, C., Turk-Kubo, K. A., McIlvin, M. R., Saito, M. A., Van
692 Mooy, B. A. S., and Repeta, D. J.: Distinct Siderophores Contribute to Iron Cycling in the
693 Mesopelagic at Station ALOHA, *Frontiers in Marine Science*, 5, 10.3389/fmars.2018.00061,
694 2018.
- 695 Conway, T. M. and John, S. G.: Quantification of dissolved iron sources to the North Atlantic
696 Ocean, *Nature*, 511, 212-215, 10.1038/nature13482, 2014.
- 697 Croot, P. L., Streu, P., and Baker, A. R.: Short residence time for iron in surface seawater
698 impacted by atmospheric dry deposition from Saharan dust events, *Geophysical Research
699 Letters*, 31, 10.1029/2004gl020153, 2004.
- 700 Dale, A. W., Nickelsen, L., Scholz, F., Hensen, C., Oschlies, A., and Wallmann, K.: A revised
701 global estimate of dissolved iron fluxes from marine sediments, *Global Biogeochemical Cycles*,
702 29, 691-707, 10.1002/2014gb005017, 2015.
- 703 Eby, M., Zickfeld, K., Montenegro, A., Archer, D., Meissner, K. J., and Weaver, A. J.: Lifetime
704 of Anthropogenic Climate Change: Millennial Time Scales of Potential CO₂ and Surface
705 Temperature Perturbations, *Journal of Climate*, 22, 2501-2511, 10.1175/2008jcli2554.1, 2009.
- 706 Elrod, V. A., Berelson, W. M., Coale, K. H., and Johnson, K. S.: The flux of iron from
707 continental shelf sediments: A missing source for global budgets, *Geophysical Research Letters*,
708 31, L12307, 10.1029/2004gl020216, 2004.

- 709 Frants, M., Holzer, M., DeVries, T., and Matear, R.: Constraints on the global marine iron cycle
710 from a simple inverse model, *Journal of Geophysical Research: Biogeosciences*, 121, 28-51,
711 10.1002/2015jg003111, 2016.
- 712 Galbraith, E. D., Gnanadesikan, A., Dunne, J. P., and Hiscock, M. R.: Regional impacts of iron-
713 light colimitation in a global biogeochemical model, *Biogeosciences*, 7, 1043-1064, 10.5194/bg-
714 7-1043-2010, 2010.
- 715 Gent, P. R. and McWilliams, J. C.: Isopycnal Mixing in Ocean Circulation Models, *Journal of*
716 *Physical Oceanography*, 20, 150-155, doi:10.1175/1520-
717 0485(1990)020<0150:IMIOCM>2.0.CO;2, 1990.
- 718 Getzlaff, J. and Dietze, H.: Effects of increased isopycnal diffusivity mimicking the unresolved
719 equatorial intermediate current system in an earth system climate model, *Geophysical Research*
720 *Letters*, 10.1002/grl.50419, 2013.
- 721 Gledhill, M. and Buck, K. N.: The organic complexation of iron in the marine environment: a
722 review, *Frontiers in microbiology*, 3, 10.3389/fmicb.2012.00069, 2012.
- 723 Holzer, M., Frants, M., and Pasquier, B.: The age of iron and iron source attribution in the ocean,
724 *Global Biogeochemical Cycles*, 30, 1454-1474, 10.1002/2016gb005418, 2016.
- 725 Homoky, W. B., John, S. G., Conway, T. M., and Mills, R. A.: Distinct iron isotopic signatures
726 and supply from marine sediment dissolution, *Nat Commun*, 4, 2143, 10.1038/ncomms3143,
727 2013.
- 728 Honeyman, B. D., Balistrieri, L. S., and Murray, J. W.: Oceanic trace metal scavenging: the
729 importance of particle concentration, *Deep Sea Research Part A. Oceanographic Research*
730 *Papers*, 35, 227-246, [https://doi.org/10.1016/0198-0149\(88\)90038-6](https://doi.org/10.1016/0198-0149(88)90038-6), 1988.
- 731 Johnson, K. S., Gordon, R. M., and Coale, K. H.: What controls dissolved iron concentrations in
732 the world ocean?, *Marine Chemistry*, 57, 137-161, [https://doi.org/10.1016/S0304-4203\(97\)00043-1](https://doi.org/10.1016/S0304-4203(97)00043-1), 1997.
- 734 Kalnay, E., Kanamitsu, M., Kistler, R., Collins, W., Deaven, D., Gandin, L., Iredell, M., Saha,
735 S., White, G., Woollen, J., Zhu, Y., Leetmaa, A., Reynolds, R., Chelliah, M., Ebisuzaki, W.,
736 Higgins, W., Janowiak, J., Mo, K. C., Ropelewski, C., Wang, J., Jenne, R., and Joseph, D.: The
737 NCEP/NCAR 40-Year Reanalysis Project, *Bulletin of the American Meteorological Society*, 77,
738 437-471, 10.1175/1520-0477(1996)077<0437:tnyrp>2.0.co;2, 1996.
- 739 Kessler, W. S. and McCreary, J. P.: The Annual Wind-driven Rossby Wave in the
740 Subthermocline Equatorial Pacific, *Journal of Physical Oceanography*, 23, 1192-1207,
741 10.1175/1520-0485(1993)023<1192:tawdrw>2.0.co;2, 1993.
- 742 Kvale, K. F., Meissner, K. J., Keller, D. P., Eby, M., and Schmittner, A.: Explicit Planktic
743 Calcifiers in the University of Victoria Earth System Climate Model, Version 2.9, *Atmosphere-*
744 *Ocean*, 53, 332-350, 10.1080/07055900.2015.1049112, 2015.

- 745 Large, W. G., Danabasoglu, G., McWilliams, J. C., Gent, P. R., and Bryan, F. O.: Equatorial
746 Circulation of a Global Ocean Climate Model with Anisotropic Horizontal Viscosity, *Journal of*
747 *Physical Oceanography*, 31, 518-536, doi:10.1175/1520-
748 0485(2001)031<0518:ECOAGO>2.0.CO;2, 2001.
- 749 Luo, C., Mahowald, N., Bond, T., Chuang, P. Y., Artaxo, P., Siefert, R., Chen, Y., and Schauer,
750 J.: Combustion iron distribution and deposition, *Global Biogeochemical Cycles*, 22, GB1012,
751 10.1029/2007gb002964, 2008.
- 752 Misumi, K., Lindsay, K., Moore, J. K., Doney, S. C., Tsumune, D., and Yoshida, Y.: Humic
753 substances may control dissolved iron distributions in the global ocean: Implications from
754 numerical simulations, *Global Biogeochemical Cycles*, 27, 450-462, 10.1002/gbc.20039, 2013.
- 755 Moffett, J. W., Goepfert, T. J., and Naqvi, S. W. A.: Reduced iron associated with secondary
756 nitrite maxima in the Arabian Sea, *Deep Sea Research Part I: Oceanographic Research Papers*,
757 54, 1341-1349, 10.1016/j.dsr.2007.04.004, 2007.
- 758 Moore, C. M., Mills, M. M., Arrigo, K. R., Berman-Frank, I., Bopp, L., Boyd, P. W., Galbraith,
759 E. D., Geider, R. J., Guieu, C., Jaccard, S. L., Jickells, T. D., La Roche, J., Lenton, T. M.,
760 Mahowald, N. M., Marañón, E., Marinov, I., Moore, J. K., Nakatsuka, T., Oschlies, A., Saito, M.
761 A., Thingstad, T. F., Tsuda, A., and Ulloa, O.: Processes and patterns of oceanic nutrient
762 limitation, *Nature Geoscience*, 6, 701-710, 10.1038/ngeo1765, 2013.
- 763 Moore, J. K. and Braucher, O.: Sedimentary and mineral dust sources of dissolved iron to the
764 world ocean, *Biogeosciences*, 5, 631-656, 10.5194/bg-5-631-2008, 2008.
- 765 Muglia, J., Somes, C. J., Nickelsen, L., and Schmittner, A.: Combined Effects of Atmospheric
766 and Seafloor Iron Fluxes to the Glacial Ocean, *Paleoceanography*, 32, 1204-1218,
767 10.1002/2016pa003077, 2017.
- 768 Myriokefalitakis, S., Ito, A., Kanakidou, M., Nenes, A., Krol, M. C., Mahowald, N. M., Scanza,
769 R. A., Hamilton, D. S., Johnson, M. S., Meskhidze, N., Kok, J. F., Guieu, C., Baker, A. R.,
770 Jickells, T. D., Sarin, M. M., Bikkina, S., Shelley, R., Bowie, A., Perron, M. M. G., and Duce, R.
771 A.: Reviews and syntheses: the GESAMP atmospheric iron deposition model intercomparison
772 study, *Biogeosciences*, 15, 6659-6684, 10.5194/bg-15-6659-2018, 2018.
- 773 Nickelsen, L., Keller, D. P., and Oschlies, A.: A dynamic marine iron cycle module coupled to
774 the University of Victoria Earth System Model: the Kiel Marine Biogeochemical Model 2 for
775 UVic 2.9, *Geoscientific Model Development*, 8, 1357-1381, 10.5194/gmd-8-1357-2015, 2015.
- 776 Parekh, P., Follows, M. J., and Boyle, E.: Modeling the global ocean iron cycle, *Global*
777 *Biogeochemical Cycles*, 18, n/a-n/a, 10.1029/2003gb002061, 2004.
- 778 Pasquier, B. and Holzer, M.: Inverse-model estimates of the ocean's coupled phosphorus, silicon,
779 and iron cycles, *Biogeosciences*, 14, 4125-4159, 10.5194/bg-14-4125-2017, 2017.

- 780 Pasquier, B. and Holzer, M.: The number of past and future regenerations of iron in the ocean and
781 its intrinsic fertilization efficiency, *Biogeosciences*, 15, 7177-7203, 10.5194/bg-15-7177-2018,
782 2018.
- 783 Peltier, W. R.: Global glacial isostacy and the surface of the ice-age Earth: The ICE-5G (VM2)
784 model and GRACE, *Annual Review of Earth and Planetary Sciences*, 32, 111-149,
785 10.1146/annurev.earth.32.082503.144359, 2004.
- 786 Pham, A. L. D. and Ito, T.: Formation and Maintenance of the GEOTRACES Subsurface-
787 Dissolved Iron Maxima in an Ocean Biogeochemistry Model, *Global Biogeochemical Cycles*,
788 32, 932-953, 10.1029/2017gb005852, 2018.
- 789 Pham, A. L. D. and Ito, T.: Ligand Binding Strength Explains the Distribution of Iron in the
790 North Atlantic Ocean, *Geophysical Research Letters*, 10.1029/2019gl083319, 2019.
- 791 Resing, J. A., Sedwick, P. N., German, C. R., Jenkins, W. J., Moffett, J. W., Sohst, B. M., and
792 Tagliabue, A.: Basin-scale transport of hydrothermal dissolved metals across the South Pacific
793 Ocean, *Nature*, 523, 200-203, 10.1038/nature14577, 2015.
- 794 Roshan, S., DeVries, T., Wu, J., John, S., and Weber, T.: Reversible scavenging traps
795 hydrothermal iron in the deep ocean, *Earth and Planetary Science Letters*, 542, 116297,
796 <https://doi.org/10.1016/j.epsl.2020.116297>, 2020.
- 797 Sarthou, G., Baker, A. R., Blain, S., Achterberg, E. P., Boye, M., Bowie, A. R., Croot, P., Laan,
798 P., de Baar, H. J. W., Jickells, T. D., and Worsfold, P. J.: Atmospheric iron deposition and sea-
799 surface dissolved iron concentrations in the eastern Atlantic Ocean, *Deep Sea Research Part I:*
800 *Oceanographic Research Papers*, 50, 1339-1352, [https://doi.org/10.1016/S0967-0637\(03\)00126-2](https://doi.org/10.1016/S0967-0637(03)00126-2), 2003.
- 801 Schlitzer, R., Anderson, R. F., Dodas, E. M., Lohan, M., Geibert, W., Tagliabue, A., Bowie, A.,
802 Jeandel, C., Maldonado, M. T., Landing, W. M., Cockwell, D., Abadie, C., Abouchami, W.,
803 Achterberg, E. P., Agather, A., Aguliar-Islas, A., van Aken, H. M., Andersen, M., Archer, C.,
804 Auro, M., de Baar, H. J., Baars, O., Baker, A. R., Bakker, K., Basak, C., Baskaran, M., Bates, N.
805 R., Bauch, D., van Beek, P., Behrens, M. K., Black, E., Bluhm, K., Bopp, L., Bouman, H.,
806 Bowman, K., Bown, J., Boyd, P., Boye, M., Boyle, E. A., Branellec, P., Bridgestock, L.,
807 Brissebrat, G., Browning, T., Bruland, K. W., Brumsack, H.-J., Brzezinski, M., Buck, C. S.,
808 Buck, K. N., Buesseler, K., Bull, A., Butler, E., Cai, P., Mor, P. C., Cardinal, D., Carlson, C.,
809 Carrasco, G., Casacuberta, N., Casciotti, K. L., Castrillejo, M., Chamizo, E., Chance, R.,
810 Charette, M. A., Chaves, J. E., Cheng, H., Chever, F., Christl, M., Church, T. M., Closset, I.,
811 Colman, A., Conway, T. M., Cossa, D., Croot, P., Cullen, J. T., Cutter, G. A., Daniels, C.,
812 Dehairs, F., Deng, F., Dieu, H. T., Duggan, B., Dulaquais, G., Dumousseaud, C., Echegoyen-
813 Sanz, Y., Edwards, R. L., Ellwood, M., Fahrbach, E., Fitzsimmons, J. N., Russell Flegal, A.,
814 Fleisher, M. Q., van de Flierdt, T., Frank, M., Friedrich, J., Fripiat, F., Fröllje, H., Galer, S. J. G.,
815 Gamo, T., Ganeshram, R. S., Garcia-Orellana, J., Garcia-Solsona, E., Gault-Ringold, M.,
816 George, E., Gerringa, L. J. A., Gilbert, M., Godoy, J. M., Goldstein, S. L., Gonzalez, S. R.,
817 Grissom, K., Hammerschmidt, C., Hartman, A., Hassler, C. S., Hathorne, E. C., Hatta, M.,
818 Hawco, N., Hayes, C. T., Heimbürger, L.-E., Helgoe, J., Heller, M., Henderson, G. M.,

- 820 Henderson, P. B., van Heuven, S., Ho, P., Horner, T. J., Hsieh, Y.-T., Huang, K.-F., Humphreys,
821 M. P., Isshiki, K., Jacquot, J. E., Janssen, D. J., Jenkins, W. J., John, S., Jones, E. M., Jones, J.
822 L., Kadko, D. C., Kayser, R., Kenna, T. C., Khondoker, R., Kim, T., Kipp, L., Klar, J. K.,
823 Klunder, M., Kretschmer, S., Kumamoto, Y., Laan, P., Labatut, M., Lacan, F., Lam, P. J.,
824 Lambelet, M., Lamborg, C. H., Le Moigne, F. A. C., Le Roy, E., Lechtenfeld, O. J., Lee, J.-M.,
825 Lherminier, P., Little, S., López-Lora, M., Lu, Y., Masque, P., Mawji, E., McClain, C. R.,
826 Measures, C., Mehic, S., Barraqueta, J.-L. M., van der Merwe, P., Middag, R., Mieruch, S.,
827 Milne, A., Minami, T., Moffett, J. W., Moncoiffe, G., Moore, W. S., Morris, P. J., Morton, P. L.,
828 Nakaguchi, Y., Nakayama, N., Niedermiller, J., Nishioka, J., Nishiuchi, A., Noble, A., Obata, H.,
829 Ober, S., Ohnemus, D. C., van Ooijen, J., O'Sullivan, J., Owens, S., Pahnke, K., Paul, M., Pavia,
830 F., Pena, L. D., Peters, B., Planchon, F., Planquette, H., Pradoux, C., Puigcorbé, V., Quay, P.,
831 Queroue, F., Radic, A., Rauschenberg, S., Rehkämper, M., Rember, R., Remenyi, T., Resing, J.
832 A., Rickli, J., Rigaud, S., Rijkenberg, M. J. A., Rintoul, S., Robinson, L. F., Roca-Martí, M.,
833 Rodellas, V., Roeske, T., Rolison, J. M., Rosenberg, M., Roshan, S., Rutgers van der Loeff, M.
834 M., Ryabenko, E., Saito, M. A., Salt, L. A., Sanial, V., Sarthou, G., Schallenberg, C., Schauer,
835 U., Scher, H., Schlosser, C., Schnetger, B., Scott, P., Sedwick, P. N., Semiletov, I., Shelley, R.,
836 Sherrell, R. M., Shiller, A. M., Sigman, D. M., Singh, S. K., Slagter, H. A., Slater, E., Smethie,
837 W. M., Snaith, H., Sohrin, Y., Sohst, B., Sonke, J. E., Speich, S., Steinfeldt, R., Stewart, G.,
838 Stichel, T., Stirling, C. H., Stutsman, J., Swarr, G. J., Swift, J. H., Thomas, A., Thorne, K., Till,
839 C. P., Till, R., Townsend, A. T., Townsend, E., Tuerena, R., Twinning, B. S., Vance, D.,
840 Velazquez, S., Venchiarutti, C., Villa-Alfageme, M., Vivancos, S. M., Voelker, A. H. L., Wake,
841 B., Warner, M. J., Watson, R., van Weerlee, E., Alexandra Weigand, M., Weinstein, Y., Weiss,
842 D., Wisotzki, A., Woodward, E. M. S., Wu, J., Wu, Y., Wuttig, K., Wyatt, N., Xiang, Y., Xie, R.
843 C., Xue, Z., Yoshikawa, H., Zhang, J., Zhang, P., Zhao, Y., Zheng, L., Zheng, X.-Y., Zieringer,
844 M., Zimmer, L. A., Ziveri, P., Zunino, P., and Zurbrick, C.: The GEOTRACES Intermediate
845 Data Product 2017, *Chemical Geology*, 493, 210-223,
846 <https://doi.org/10.1016/j.chemgeo.2018.05.040>, 2018.
- 847 Schmittner, A. and Egbert, G. D.: An improved parameterization of tidal mixing for ocean
848 models, *Geoscientific Model Development*, 7, 211-224, 10.5194/gmd-7-211-2014, 2014.
- 849 Schmittner, A. and Somes, C. J.: Complementary constraints from carbon (13C) and nitrogen
850 (15N) isotopes on the glacial ocean's soft-tissue biological pump, *Paleoceanography*, 31, 669-
851 693, 10.1002/2015PA002905, 2016.
- 852 Somes, C., Schmittner, A., Muglia, J., and Oschlies, A.: A three-dimensional model of the
853 marine nitrogen cycle during the Last Glacial Maximum constrained by sedimentary isotopes,
854 *Frontiers in Marine Science*, 4, 10.3389/fmars.2017.00108, 2017.
- 855 Somes, C. J. and Oschlies, A.: On the influence of "non-Redfield" dissolved organic nutrient
856 dynamics on the spatial distribution of N2 fixation and the size of the marine fixed nitrogen
857 inventory, *Global Biogeochemical Cycles*, 29, 973-993, 10.1002/2014GB005050, 2015.
- 858 Somes, C. J., Schmittner, A., Galbraith, E. D., Lehmann, M. F., Altabet, M. A., Montoya, J. P.,
859 Letelier, R. M., Mix, A. C., Bourbonnais, A., and Eby, M.: Simulating the global distribution of
860 nitrogen isotopes in the ocean, *Global Biogeochem. Cycles*, 24, GB4019,
861 10.1029/2009gb003767, 2010.

- 862 Tagliabue, A. and Völker, C.: Towards accounting for dissolved iron speciation in global ocean
863 models, *Biogeosciences*, 8, 3025-3039, 10.5194/bg-8-3025-2011, 2011.
- 864 Tagliabue, A., Bowie, A. R., Boyd, P. W., Buck, K. N., Johnson, K. S., and Saito, M. A.: The
865 integral role of iron in ocean biogeochemistry, *Nature*, 543, 51-59, 10.1038/nature21058, 2017.
- 866 Tagliabue, A., Mtshali, T., Aumont, O., Bowie, A. R., Klunder, M. B., Roychoudhury, A. N.,
867 and Swart, S.: A global compilation of dissolved iron measurements: focus on distributions and
868 processes in the Southern Ocean, *Biogeosciences*, 9, 2333-2349, 10.5194/bg-9-2333-2012, 2012.
- 869 Tagliabue, A., Bowie, A. R., DeVries, T., Ellwood, M. J., Landing, W. M., Milne, A., Ohnemus,
870 D. C., Twining, B. S., and Boyd, P. W.: The interplay between regeneration and scavenging
871 fluxes drives ocean iron cycling, *Nature Communications*, 10, 10.1038/s41467-019-12775-5,
872 2019.
- 873 Tagliabue, A., Bopp, L., Dutay, J.-C., Bowie, A. R., Chever, F., Jean-Baptiste, P., Bucciarelli, E.,
874 Lannuzel, D., Remenyi, T., Sarthou, G., Aumont, O., Gehlen, M., and Jeandel, C.: Hydrothermal
875 contribution to the oceanic dissolved iron inventory, *Nature Geoscience*, 3, 252-256,
876 10.1038/ngeo818, 2010.
- 877 Tagliabue, A., Aumont, O., DeAth, R., Dunne, J. P., Dutkiewicz, S., Galbraith, E., Misumi, K.,
878 Moore, J. K., Ridgwell, A., Sherman, E., Stock, C., Vichi, M., Völker, C., and Yool, A.: How
879 well do global ocean biogeochemistry models simulate dissolved iron distributions?, *Global
880 Biogeochemical Cycles*, 30, 149-174, 10.1002/2015gb005289, 2016.
- 881 Völker, C. and Tagliabue, A.: Modeling organic iron-binding ligands in a three-dimensional
882 biogeochemical ocean model, *Marine Chemistry*, 173, 67-77, 10.1016/j.marchem.2014.11.008,
883 2015.
- 884 Weaver, A. J., Eby, M., Wiebe, E. C., Bitz, C. M., Duffy, P. B., Ewen, T. L., Fanning, A. F.,
885 Holland, M. M., MacFadyen, A., Matthews, H. D., Meissner, K. J., Saenko, O., Schmittner, A.,
886 Wang, H., and Yoshimori, M.: The UVic earth system climate model: Model description,
887 climatology, and applications to past, present and future climates, *Atmosphere-Ocean*, 39, 361 -
888 428, 2001.
- 889 Yao, W., Kvale, K. F., Achterberg, E., Koeve, W., and Oschlies, A.: Hierarchy of calibrated
890 global models reveals improved distributions and fluxes of biogeochemical tracers in models
891 with explicit representation of iron, *Environmental Research Letters*, 14, 114009, 10.1088/1748-
892 9326/ab4c52, 2019.
- 893 Ye, Y. and Völker, C.: On the Role of Dust-Deposited Lithogenic Particles for Iron Cycling in
894 the Tropical and Subtropical Atlantic, *Global Biogeochemical Cycles*, 31, 1543-1558,
895 10.1002/2017gb005663, 2017.
896

Figure 1.

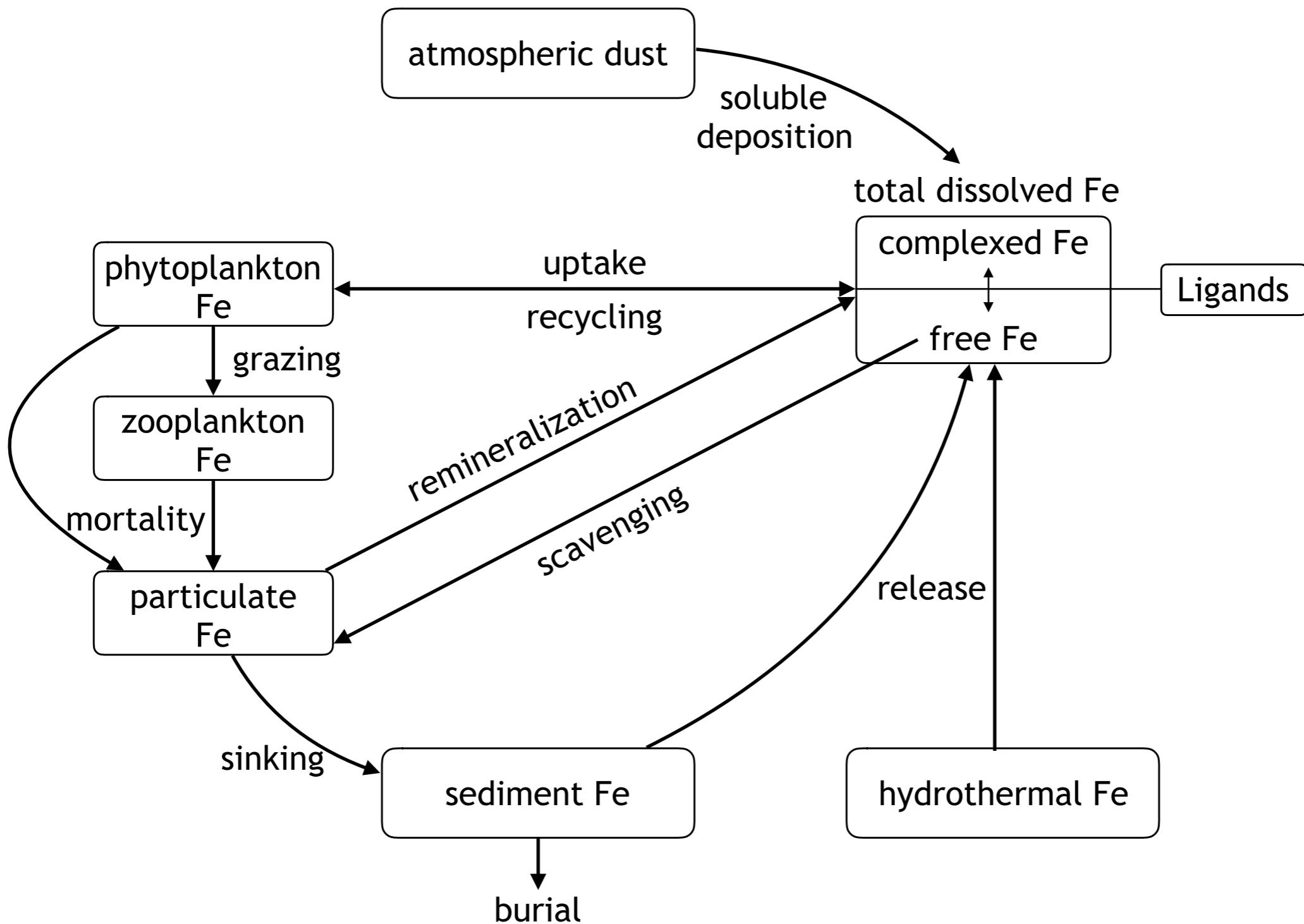
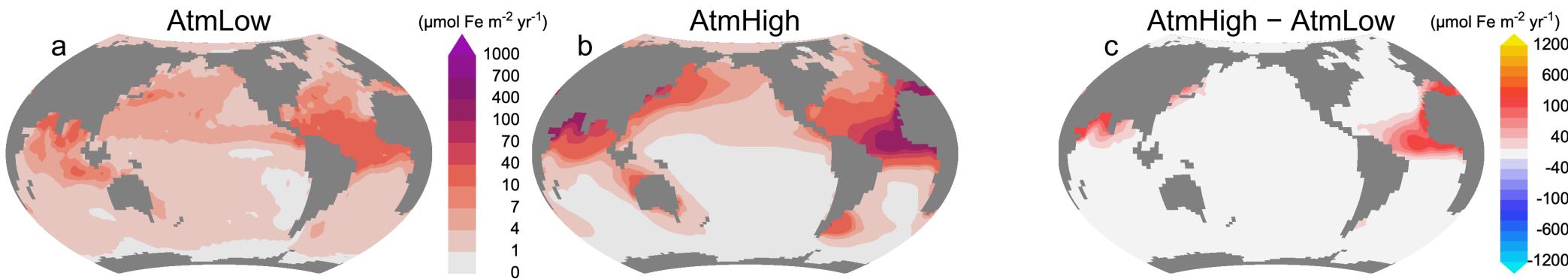
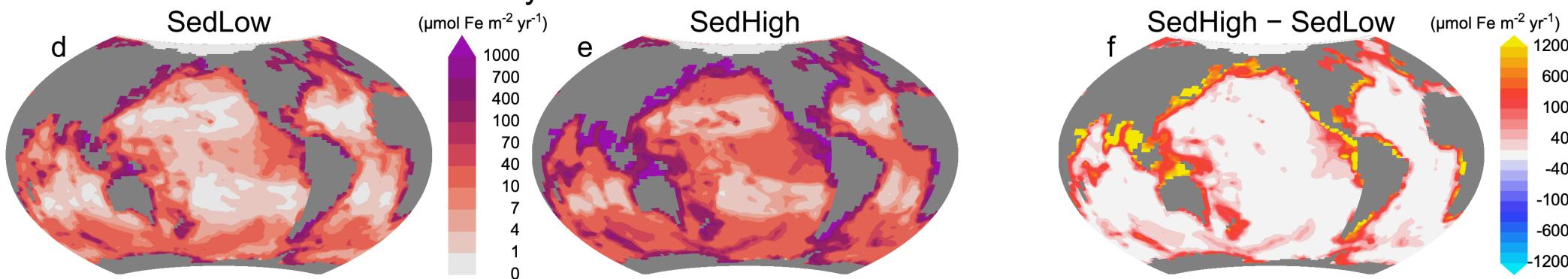


Figure 2.

Atmospheric Soluble Fe Deposition



Reductive Sedimentary Fe Release



Fe Scavenging

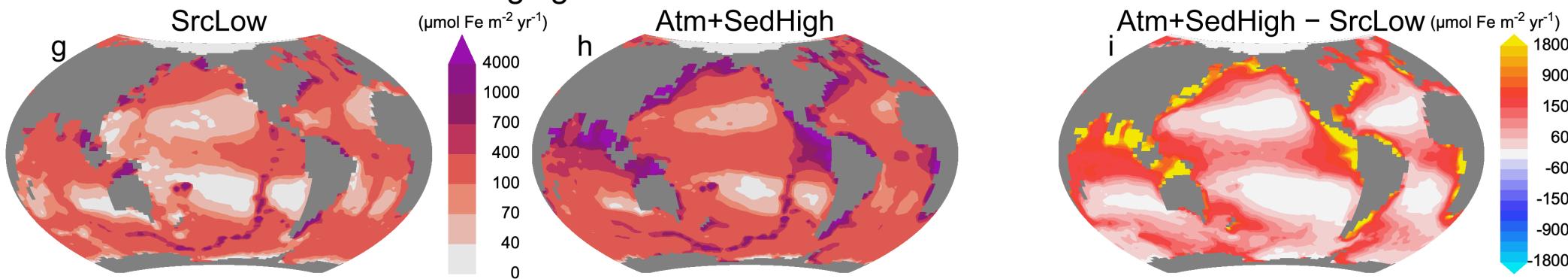
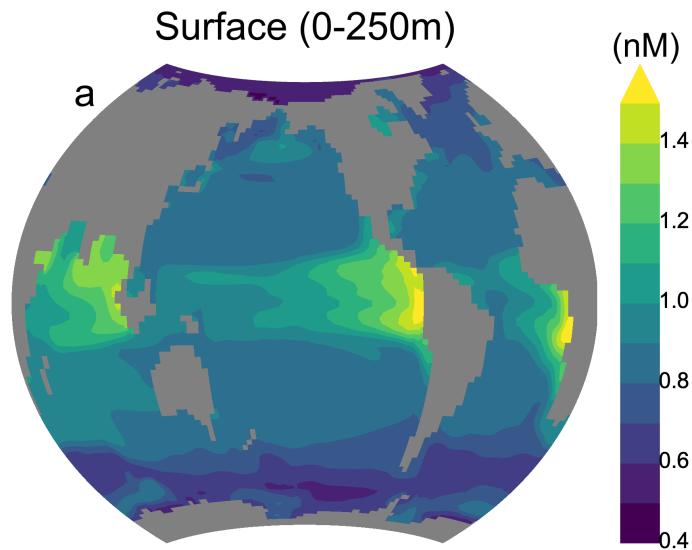


Figure 3.

Variable Ligand Distribution



Basin Depth Profiles

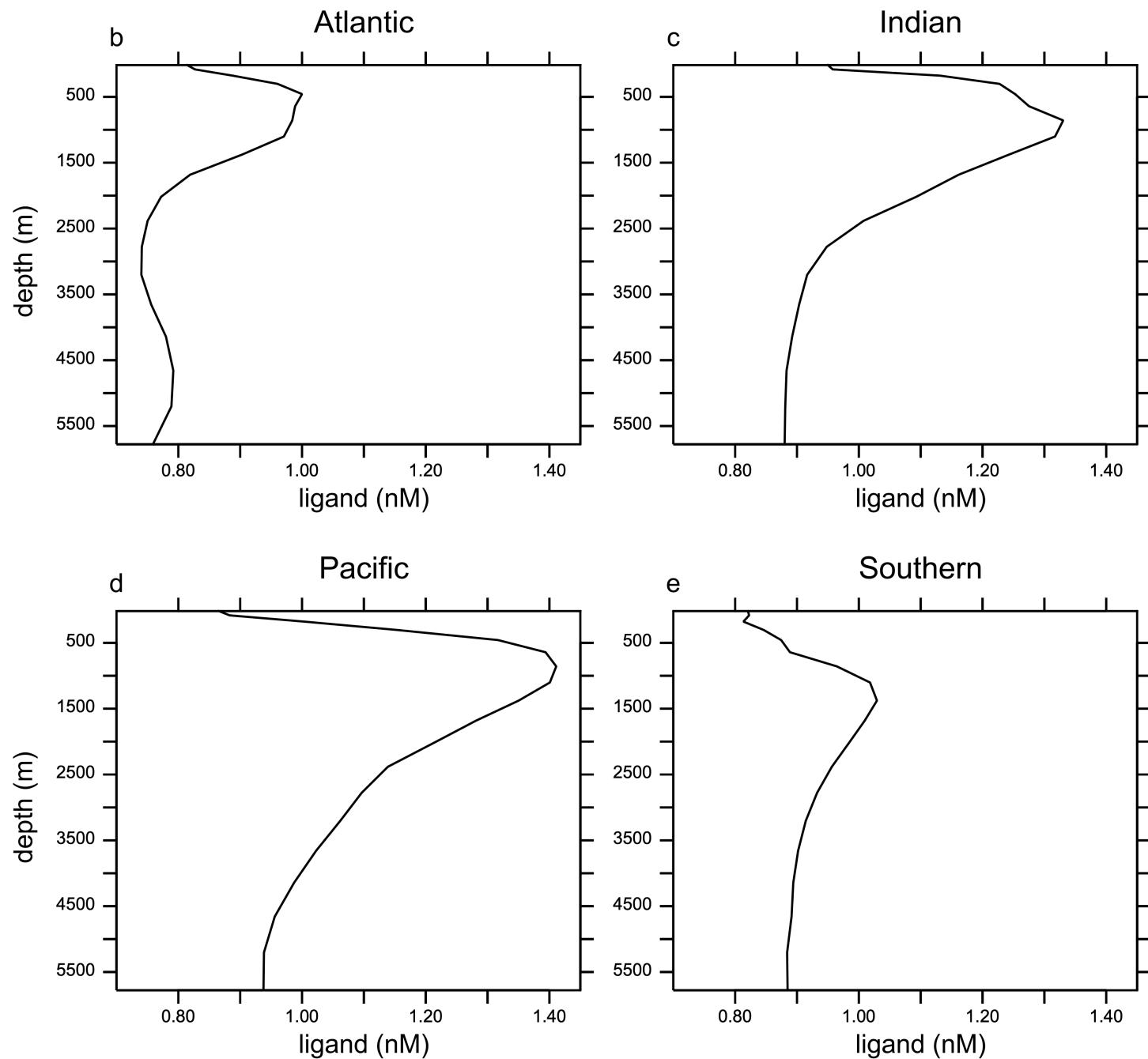


Figure 4.

SrcLow_LigCon(■) SrcLow_LigVar(●) SedHigh_LigVar(▲) Atm+SedHigh_LigVar(◆) Observations(●)

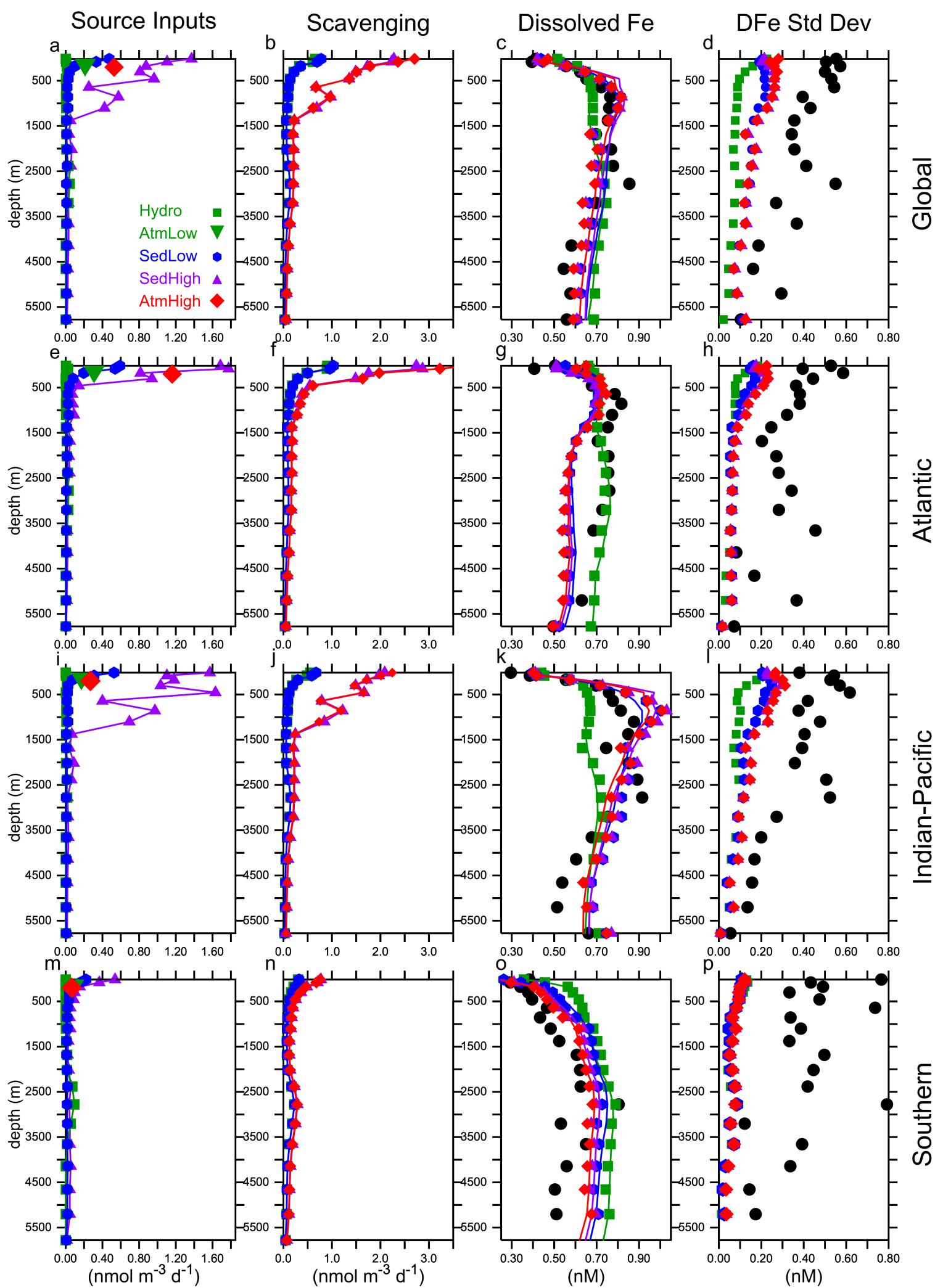


Figure 5.

Dissolved Fe

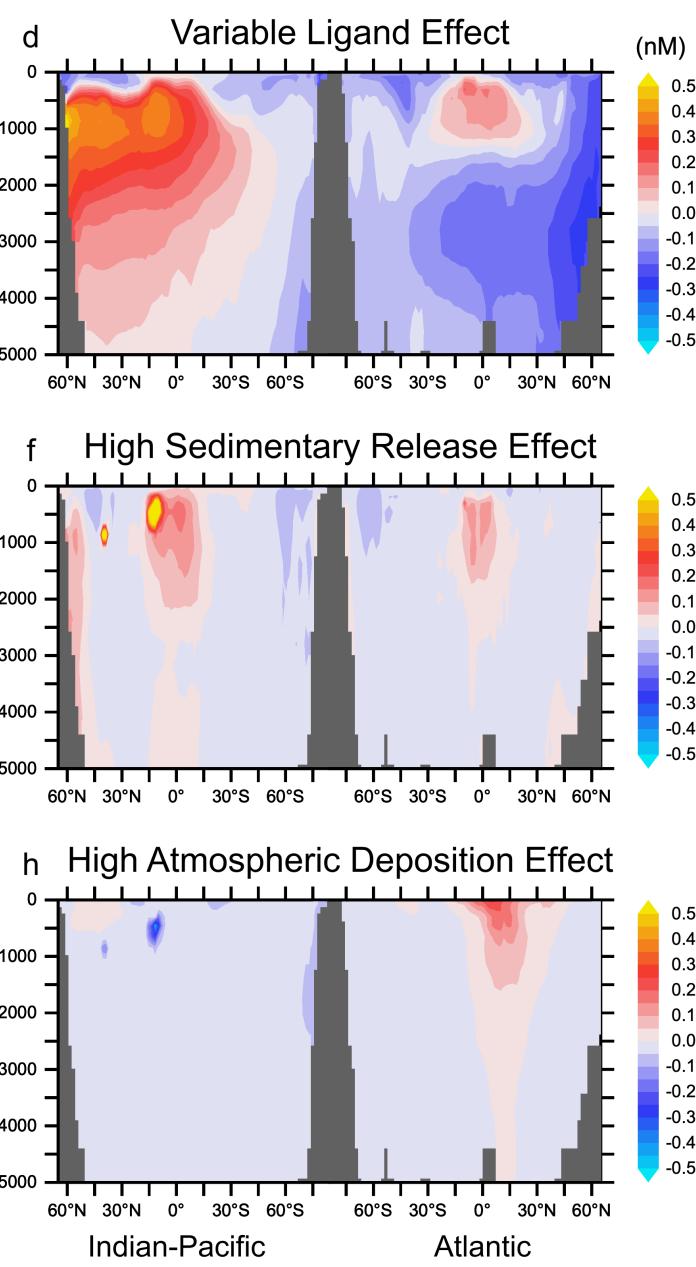
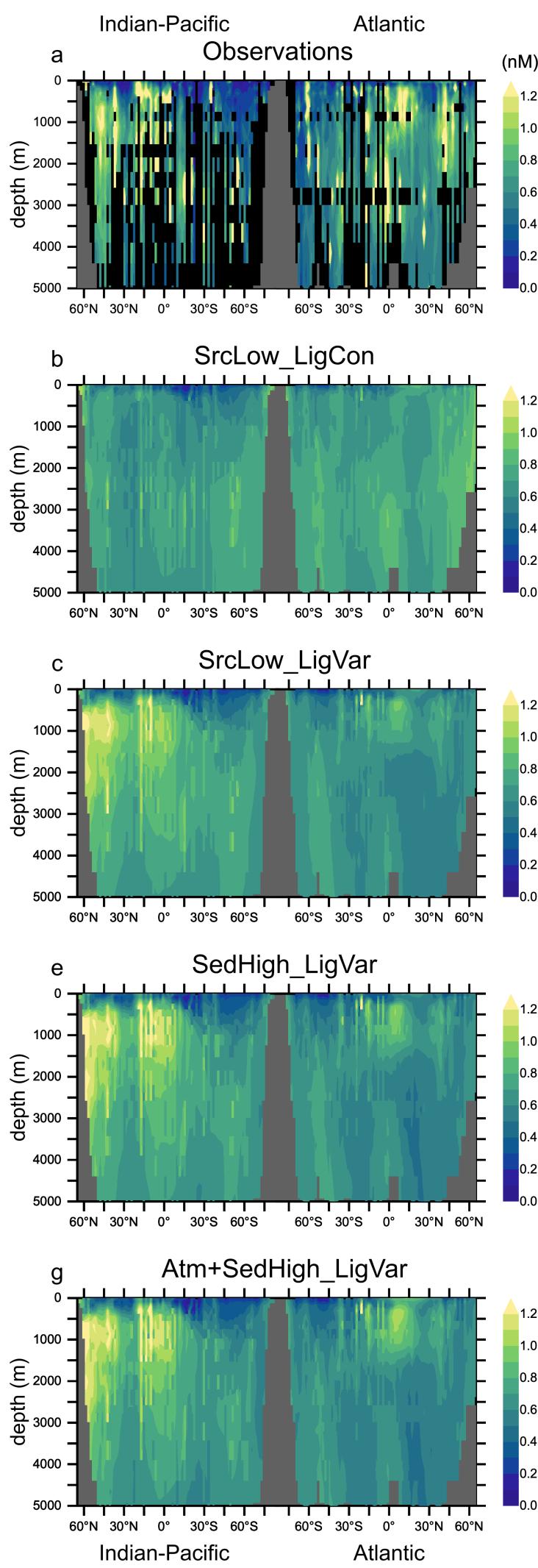


Figure 6.

Dissolved Fe (0-250m)

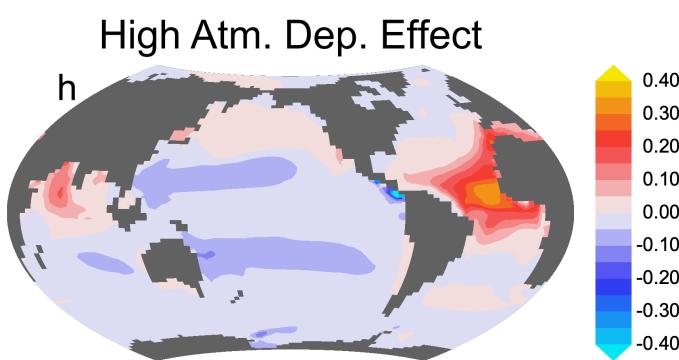
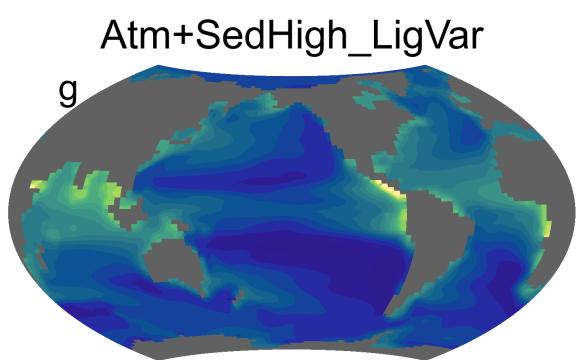
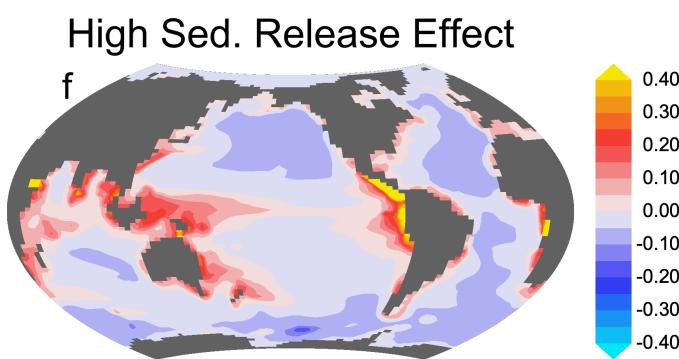
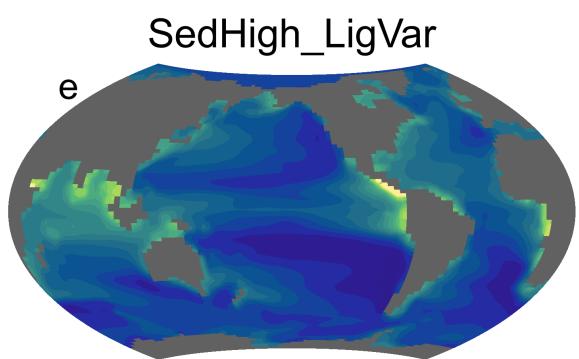
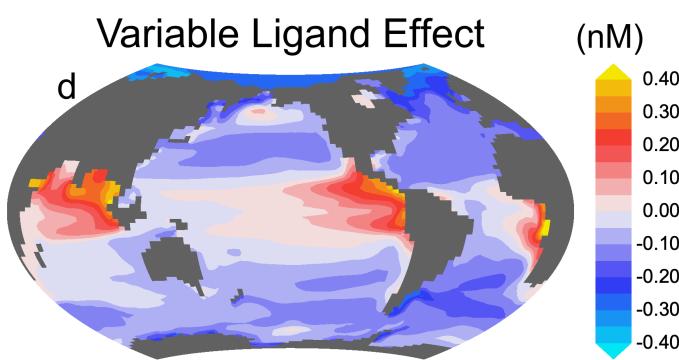
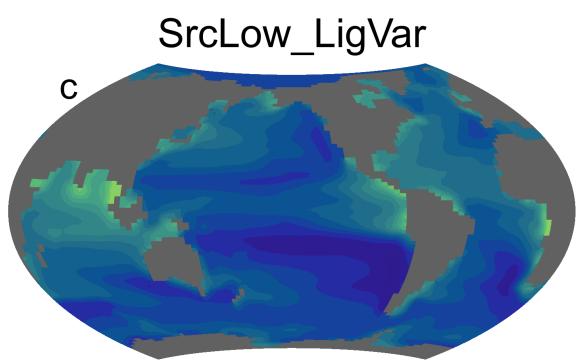
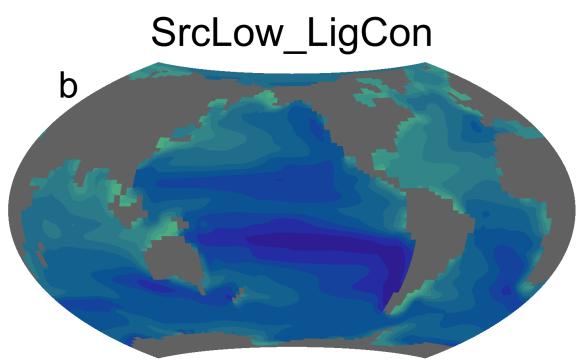
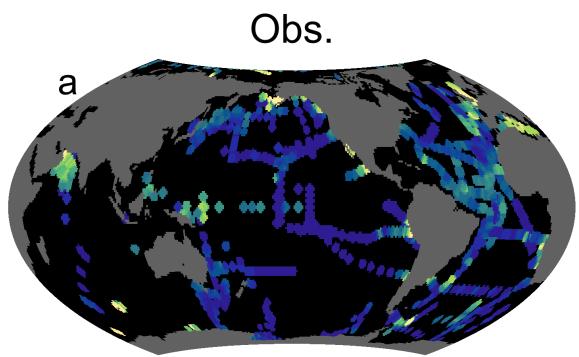


Figure 7.

Surface Sections (0-250m)

GEOTRACES Observations (●)

Other Observations (▼)

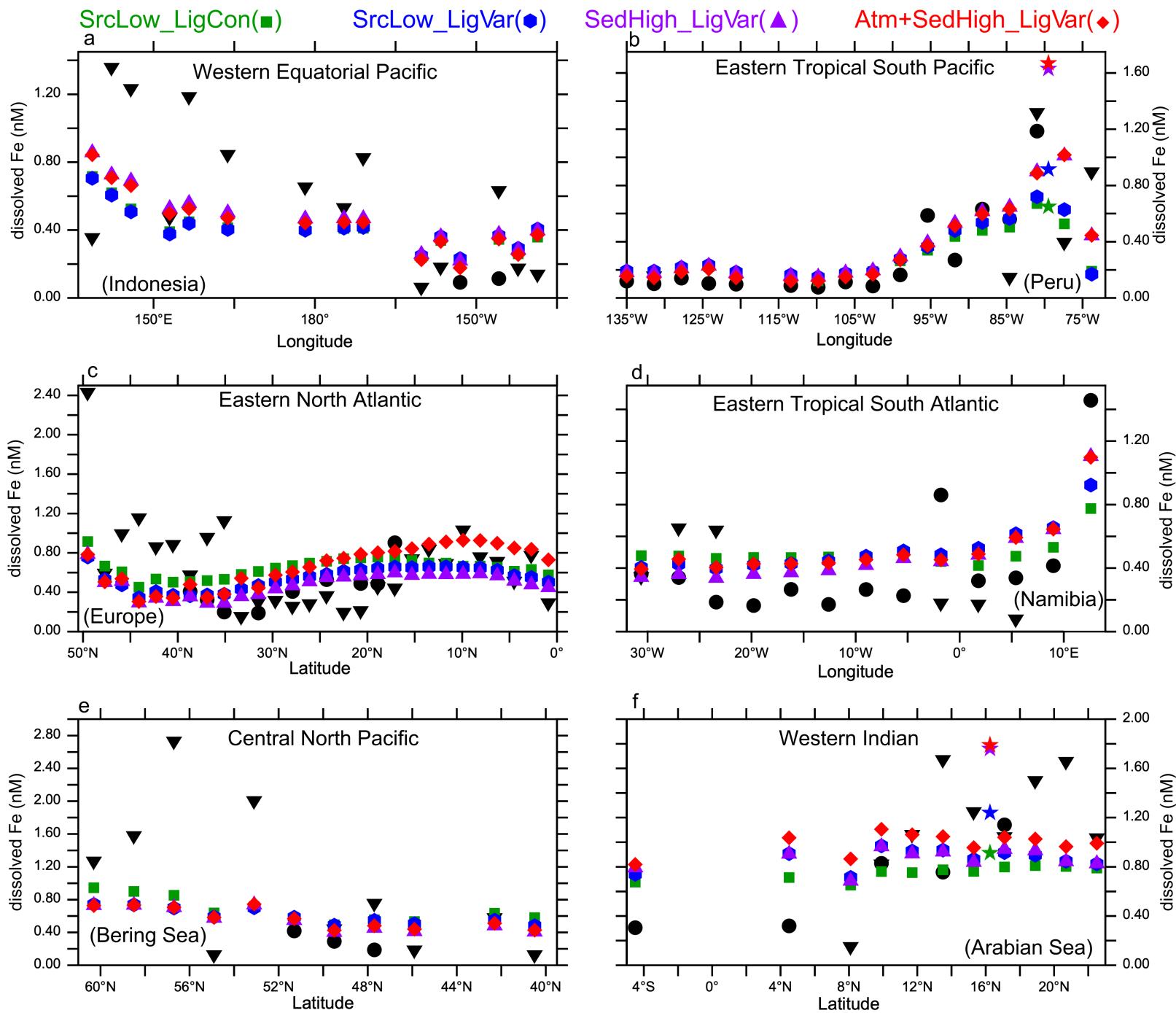


Figure 8.

Model-Data Statistical Metrics

All Observations(▼)

Only GEOTRACES Observations(●)

