# Geochemical Profiles Across the Listvenite-Metamorphic Transition in the Basal Megathrust of the Semail Ophiolite: Results from Drilling at Oman DP Hole BT1B

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**Key Points:**

- Listvenitization induce a mass increase of up to 51% (50 % volume) relative to the serpentinized mantle protolith

- FME distribution suggests reactions with several batches of $CO_2$-rich fluids over various flow paths parallel to the basal thrust

- Listvenitization could represent a major trap-and-release mechanism for water and FME as well as $CO_2$ along convergent margins

1  **Abstract**

2  The transition from the Semail ophiolite mantle to the underlying metamorphic sole was drilled
3  at ICDP OmanDP Hole BT1B. We analyzed the bulk major, volatile and trace element
4  compositions of the mantle-derived listvenite series and metamorphic rocks, with the aim to
5  constrain chemical transfers associated to peridotite carbonation along the ophiolite basal thrust.
6  The listvenite series comprise variously carbonated serpentinites and (fuchsite-bearing)
7  listvenites. They have high $CO_2$ (up to 43.2 wt.%) and variable $H_2O$ (0-12.1 wt.%). Yet, they
8  have compositions close to that of the basal banded peridotites for most major and lithophile
9  trace elements, with fuchsite-bearing listvenites overlapping in composition with amphibole-
10  bearing basal lherzolites (e.g., $Al_2O_3$= 0.1-2.2 wt.%; Yb= 0.05-1 x CI-chondrite), The protolith
11  of the listvenite series was likely similar in structure and composition to serpentinized banded
12  peridotites which immediately overlie the metamorphic sole elsewhere in Oman. The listvenite
13  series are enriched in fluid mobile elements (FME) compared to Semail peridotites (up to ~$10^3$-
14  $10^4$ x Primitive Mantle), with concentrations similar to the underthrusted metabasalts and/or
15  metasediments for Cs, Sr and Ca and sometimes even higher for Pb, Li, As, and Sb (e.g., Li up to
16  130 ppm; As up to 170 ppm). We also observe a decoupling between Sr-Ca enrichments and
17  other FME, indicating interactions with several batches of deep $CO_2$-rich fluids transported along
18  the basal thrust. These results suggest that peridotite carbonation could represent one of the
19  major trap-and-release mechanisms for carbon, water and FME along convergent margins.

20  **Plain Language Summary**

21  Ophiolites are sections of oceanic lithosphere emplaced on land as plates converge. The faults
22  developed at their base are analogues to plate interfaces in subduction zones, where mass
23  transfers occur and play a key role in the global cycling of elements. A core was drilled at the
24  base of the Semail Ophiolite, where variously hydrated and carbonated mantle rocks known as
25  serpentinites and listvenites witnessed major fluid fluxes. Reactions with $CO_2$-bearing fluids
26  (carbonation reaction) enhanced the mobility of elements during mass transfers along the basal
27  thrust. We measured the chemistry of 84 samples spaced along this core. Results indicate that
28  $CO_2$-bearing fluids derive from at least two sources or pathways. As peridotites reacted, their
29  volume increased, causing cracking, enhancing the ingress of reactive fluids and participating to
30  (almost) complete carbonation of the basal ophiolite mantle. Carbon as well as many elements
31  such as cesium, arsenic, antimony, lead became enriched in these rocks. If forming in subduction
32  zones, listvenites may act as temporary storage for these elements and impact global chemical
33  cycles.
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## 1 Introduction

Listvenites (or listwaenites) are produced by $CO_2$-metasomatism of mantle-derived ultramafic rocks [*Falk and Kelemen*, 2015; *Halls and Zhao*, 1995]. They are composed mainly of quartz and carbonate (magnesite and/or dolomite, ± Cr- or Mg-rich micas ± chlorite) and are often associated with serpentinites, ophicarbonates and/or talc. Since their first description in the literature [*Rose*, 1837], they have been investigated for one of their main characteristics: the occurrence of mineralizations concentrating economically-valuable metals, such as Au, Pb-Hg-Ag, Cu, Ni, Co or Sb (e.g., [*Belogub et al.*, 2017; *Buisson and Leblanc*, 1985; *Escayola et al.*, 2009; *Halls and Zhao*, 1995; *Laznicka*, 2010]). More recently, they have been studied also as natural analogues for industrial geological carbon storage in ultramafic basement [*Falk and Kelemen*, 2015; *Hansen et al.*, 2005; *Ulrich et al.*, 2014]. These studies have highlighted the structural, mineralogical and petrological complexity of listvenites, thus resulting in a wealth of different and sometimes contradictory genetic models [*Belogub et al.*, 2017; *Falk and Kelemen*, 2015; *Halls and Zhao*, 1995; *Nasir et al.*, 2007]. Some authors invoke serpentinization of mantle peridotites followed by the formation, concurrent or consecutive, of carbonate (carbonation) and quartz (silicification) [*Boschi et al.*, 2009; *Nasir et al.*, 2007; *Stanger*, 1985; *Ulrich et al.*, 2014], whilst some emphasize direct reactions with primary mantle minerals [*Hansen et al.*, 2005; *Kelemen et al.*, 2011; *Power et al.*, 2013]. Other authors stress the role of the composition of the infiltrating $CO_2$-bearing fluid rather than that of the protolith in the genesis of listvenites and associated ore-grade mineralizations [*Belogub et al.*, 2017; *Escayola et al.*, 2009; *Halls and Zhao*, 1995; *Menzel et al.*, 2020a; *Menzel et al.*, 2018; *Stanger*, 1985], often emphasizing a possible imprint of silica and other cations scavenged from neighboring lithologies [*Nasir et al.*, 2007; *Ulrich et al.*, 2014].

The estimated temperatures of formation and the proposed source(s) of fluids also differ: fluid-inclusions and thermodynamic analyses of carbonate-quartz assemblages suggest reaction within a temperature range of 120-280°C (down to 80°C and up to 400°C) and likely triggered by the infiltration of slab-derived fluids [*Belogub et al.*, 2017; *Boskabadi et al.*, 2020; *Escayola et al.*, 2009; *Hansen et al.*, 2005; *Menzel et al.*, 2018]. However, petrographic and geochemical analyses show also that some listvenite series were formed during weathering (<50°C) by silicification reactions triggered by interactions with fluids in equilibrium with atmospheric $CO_2$ (e.g., [*Stanger*, 1985; *Ulrich et al.*, 2014]) whilst other others show evidence for distinct fluid infiltration and weathering episodes (e.g., [*Nasir et al.*, 2007]). Finally, the mechanisms allowing the infiltration of the $CO_2$-rich metasomatic fluids remain speculative with suggested mechanisms including the role of large faults and deformation, reactive cracking and dissolution [*Escayola et al.*, 2009; *Falk and Kelemen*, 2015; *Nasir et al.*, 2007]. The first difficulty to unravel the genesis of listvenites stems from their environments: they outcrop along ophiolitic and orogenic belts as part of highly altered rock assemblages, which often have undergone several stages of metamorphism (e.g., [*Azer et al.*, 2019; *Halls and Zhao*, 1995; *Menzel et al.*, 2018; *Nasir et al.*, 2007]). For these reasons, both the protolith and potential $CO_2$ source(s) for listvenite-forming metasomatism are highly metamorphosed and difficult to identify, and display a large structural and compositional variability. Only a few sites allow sampling the transition from mantle peridotites to the possible source(s) of $CO_2$ metasomatism: the Semail ophiolite is one of them.

The Semail ophiolite is a fragment of Tethyan oceanic lithosphere tectonically emplaced on the Arabian plate at the end of the Cretaceous (Fig. 1a). Listvenite bodies occasionally crop

83    out along its basal thrust ([*Falk and Kelemen*, 2015; *Glennie et al.*, 1974; *Nasir et al.*, 2007]; Fig.
84    1a). In 2017, a 300m long section was drilled through one of the largest bodies, north of Wadi
85    Mansah (Fanjah, Sultanate of Oman; Fig 1a), as part of the ICDP Oman Drilling Project
86    (OmanDP, [*Kelemen et al.*, 2020b]). OmanDP Hole BT1B sampled the transition from listvenites
87    at the base of the ophiolite to the metamorphic sole: ~200 m of listvenites and variously
88    carbonated serpentinites overlapping ~100m of greenshists and greenstones were recovered (Fig.
89    1b). This sampling allows us to quantify the broad and complex range of mineralogy and rock
90    types that are typical of listvenite suites and, for the first time, to correlate these variations to the
91    distance to their possible source(s). In particular, the basal thrust, which intersects the core at 200
92    m depth, is considered as one of the major pathways for fluids triggering the formation of
93    listvenites, and neighboring lithologies [*Falk and Kelemen*, 2015; *Nasir et al.*, 2007].

94         We carried out a bulk rock geochemical study (major, volatile and trace elements) of 84
95    samples representative of the different lithologies recovered at OmanDP Hole BT1B. Our results
96    allow enable us to document the bulk composition of these different lithologies and their
97    downhole mineralogical and geochemical variations from the sample scale to that of the
98    borehole. This provides important new insights into the composition of the protolith of the
99    Semail listvenite series and of the underlying metamorphic lithologies, and highlights the
100   magnitude of fluid-mediated elemental transfers associated to the formation of listvenites along
101   the basal thrust and into the overlying ophiolitic mantle. It is suggested that such mass fluxes
102   may represent an important trap-release mechanism for fluid mobile elements in global
103   geochemical cycles.

## 104   2 Geological Setting

105        2.1 The Semail Ophiolite

106        The Semail ophiolite, composed of a dozen structural massifs outcropping over 500 km
107   along the north-eastern margin of the Arabian Plate (~20000 km$^2$), is regarded as one of the
108   world's largest and best documented ophiolites (Fig 1a, [*F Boudier and Nicolas*, 1988; *Coleman
109   and Hopson*, 1981; *Goodenough et al.*, 2010; *Lippard et al.*, 1986; *Peters et al.*, 1991; *Searle*,
110   2019]). The 5-7 km thick structural massifs expose continuous fragments of the Tethyan oceanic
111   lithosphere that were initially more than 12-15 km thick according to structural reconstitutions
112   [*Lippard et al.*, 1986; *Nicolas et al.*, 2000] suggesting tectonic erosion during emplacement
113   and/or post obduction. The ophiolitic sequence comprises a thick mantle section overlain by a
114   well preserved layered oceanic crust, from lower gabbros to upper volcanics [*Lippard et al.*,
115   1986; *Nicolas et al.*, 2000], formed at the end of the Cretaceous (96.12–95.50 Ma [*Rioux et al.*,
116   2012; *Rioux et al.*, 2016]). The geochemistry of the upper volcanics indicates a transition from
117   oceanic accretion to subduction related volcanism (e.g., [*Belgrano et al.*, 2019; *Ernewein et al.*,
118   1988; *Godard et al.*, 2003]), that has been interpreted as evidence for a spontaneously initiating
119   subduction (e.g., [*MacLeod et al.*, 2013; *Pearce et al.*, 1981]), or a change in geodynamic
120   settings, from mid-ocean ridge to intra-oceanic underthrusting and, possibly, incipient
121   subduction (e.g., [*Boudier et al.*, 1988; *Godard et al.*, 2006; *Goodenough et al.*, 2010]).

122        The mantle section is predominantly made up of moderately serpentinized (40-80 %)
123   depleted harzburgites with minor dunites [*Boudier et al.*, 2010; *Godard et al.*, 2000; *Hanghoj et
124   al.*, 2010; *Hopson et al.*, 1981], that preserve the microstructural fabric of asthenospheric
125   deformation below the oceanic spreading center [*Boudier and Coleman*, 1981]. Low temperature

126 ductile deformation overprints this early deformation towards the base of the mantle section and
127 has been ascribed to the onset of the ophiolite detachment (e.g., [*Boudier et al.*, 1988; *Linckens*
128 *et al.*, 2011; *Nicolas et al.*, 2000]). Local occurrences of lherzolites (>5 % clinopyroxene, Cpx)
129 and Cpx-harzburgites (>2 % Cpx) are also described in this lowermost mantle section, generally
130 in massifs where the thickest mantle sections were preserved (e.g. Fizh [*Takazawa et al.*, 2003];
131 Wadi Tayin [*Godard et al.*, 2000; *Hanghoj et al.*, 2010]) and/or close to areas where the
132 ophiolitic metamorphic sole outcrops [*Khedr et al.*, 2014]. These Cpx-rich basal peridotites have
133 been interpreted as the result of variable melt extraction along the palaeoridge [*Khedr et al.*,
134 2014; *Le Mée et al.*, 2004; *Monnier et al.*, 2006] or, alternatively, as due to refertilization
135 reactions at the base of the oceanic mantle lithosphere during off-axis cooling or early
136 intraoceanic thrusting [*Godard et al.*, 2000; *Lippard et al.*, 1986; *Takazawa et al.*, 2003].

137 The ophiolite lies upon a series of underthrusted sheets of pelagic and turbidititic
138 sediments, shelf carbonates, as well as volcanics, mainly of alkaline composition (sometimes
139 referred to as Haybi volcanics) with minor transitional to tholeiitic components [*Bechennec et*
140 *al.*, 1990; *Chauvet et al.*, 2011; *Lippard et al.*, 1986; *Maury et al.*, 2003]. These lithologies
141 commonly grouped as the Hawasina assemblages are the relicts of a wide oceanic basin (at least
142 540 km large), that formed during the breakup and thinning of the Arabian continental margin
143 during Permian to Trias [*Bechennec et al.*, 1990].

144 The ophiolite *sensu stricto* and the underlying allochtonous units were thrusted atop the
145 Arabian platform during late Cretaceous (e.g., [*Glennie et al.*, 1974; *Searle and Malpas*, 1980]).
146 Slivers of the ophiolite metamorphic sole are preserved at the transition from the allochtonous
147 units to the base of the ophiolite mantle section. They are locally overlain by <200m thick
148 (proto-)mylonitic "banded lherzolites", deformed parallel to the basal contact, comprising
149 alternating highly serpentinized lherzolites (or Cpx-harzburgites) and olivine rich harzburgites
150 and dunites [*Lippard et al.*, 1986; *Prigent et al*, 2018a]. These commonly preserve evidence of
151 secondary amphibole formed at the expense of Cpx and their constitutent minerals systematically
152 display preferential enrichments in highly incompatible and fluid mobile elements [*Khedr et al.*,
153 2013; *Khedr et al.*, 2014; *Prigent et al.*, 2018a]. The sole is composed of slivers of
154 metamorphosed mafic crust with subordinate metasediments and serpentinites stripped from/by
155 the understhrusted lithosphere. It is characterized by an inverted metamorphic gradient, from low
156 temperature (LT) greenschist facies at the base up to high temperature (HT)
157 amphibolite/granulite facies at the upper contact with the ophiolite mantle [*Ghent and Stout*,
158 1981; *Searle and Cox*, 1999; *Soret et al.*, 2017]. Geochronology indicates that peak HT
159 metamorphism is only 1-2 My younger than the ophiolite crustal sequence (96.16-94.82Ma
160 [*Rioux et al.*, 2016; *Warren et al.*, 2005]) suggesting that the overlying mantle lithosphere was
161 young and hot when the metamorphic sole formed.

162 Listvenites outcrop irregularly along the basal thrust of the the Semail Ophiolite, often as
163 2-50 m bodies within the highly altered ophiolitic and sedimentary mélange in contact with its
164 metamorphic sole (Fig 1a; [*Glennie et al.*, 1974; *Nasir et al.*, 2007; *Stanger*, 1985; *Wilde et al.*,
165 2002]). One of the largest listvenite bodies outcrops over ~1 x 2 km$^2$ in the Wadi Mansah area
166 (South of Muscat). It comprises large lenses of serpentinites and marks the transition from
167 mantle peridotites to greenschist facies meta-basalts and silicic sediments [*Falk and Kelemen*,
168 2015; *Villey et al.*, 1986]. This site was chosen to drill OmanDP Hole BT1B with the objective to
169 sample the transition from the ophiolite mantle section to its metamorphic sole.
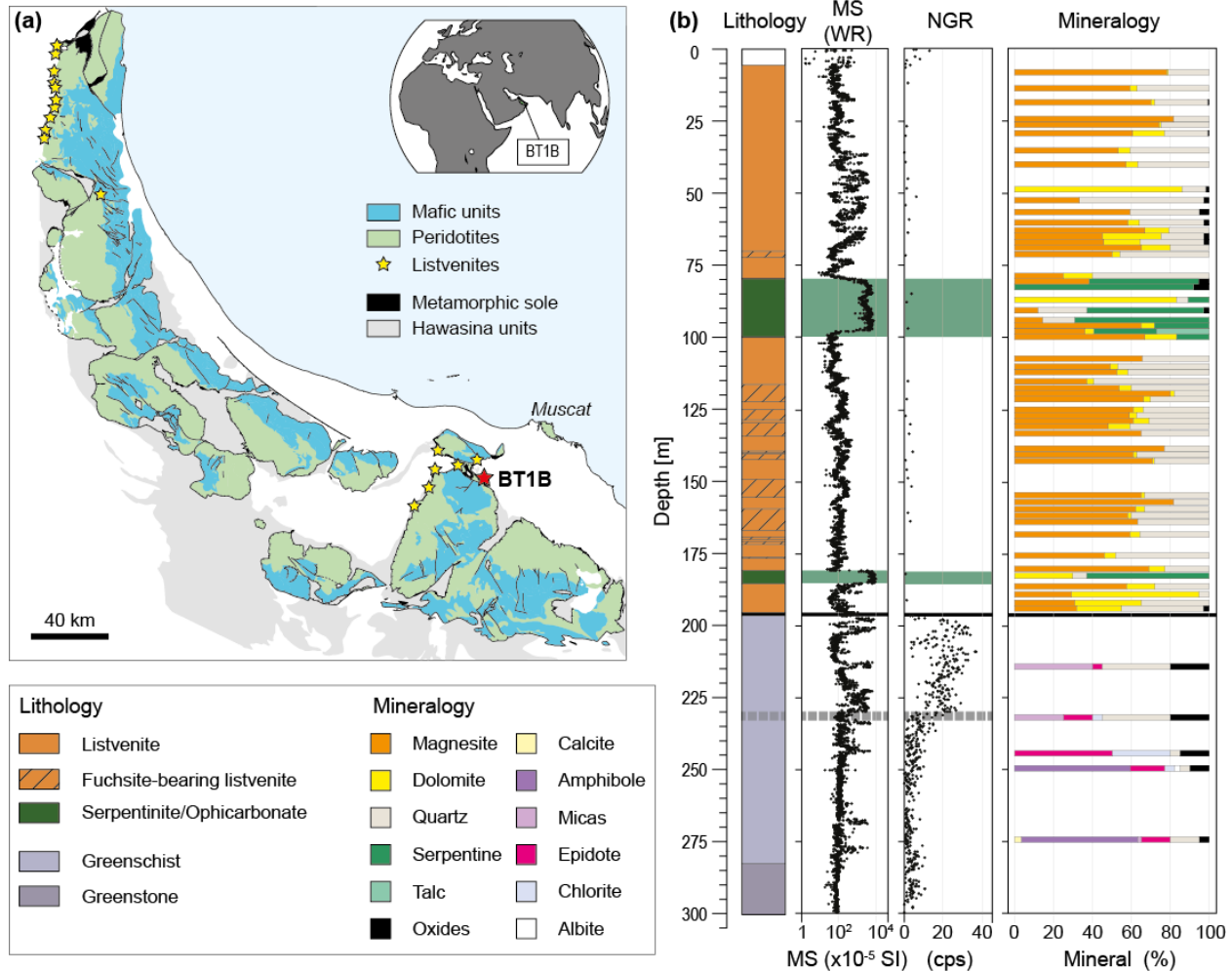
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**Figure 1**. Location and downhole plots of lithological and physical properties of OmanDP Hole BT1B. (a) Simplified geological map of the Samail ophiolite in the Sultanate of Oman. Location of OmanDP Hole BT1B and regional occurrences of listvenites (after Boudier and Nicolas [2018]) represented by red and yellow stars respectively. (b) Downhole plots of drilled lithologies, magnetic susceptibility (MS), natural gamma rays (NGR), and mineral proportions (data from Kelemen et al [2020b]). Plotted mineral proportions correspond to XRD analysis wherever it is available and otherwise, thin section observations.

2.2 OmanDP Hole BT1B and sampling

Drilling at OmanDP Hole BT1B (23°3607 N, 58°.1887 E) recovered 300.05 meters of continuous cores. The mineralogy, alteration and structure of the cores and their main physical and chemical properties were measured on-board D/V Chikyu on whole cores, core sections (Visual Core Description — VCD) and thin sections [*Kelemen et al.*, 2020b]. The main characteristics of the cores are summarized here.

Below a few meters of alluvial material, Hole BT1B drilled through first a series of listvenites interlayered with serpentinites (hereafter grouped as the *listvenite series*), then through the basal thrust at ~197 meters below ground (mbg) and, finally, into the metamorphic sole (Fig. 1b).

189        Listvenites comprise dominantly magnesite ± dolomite and quartz, and are characterized
190 by their pale yellow to dark reddish brown color in hand specimen. They contain relicts of
191 chromian spinel and magnetite alignments indicating that their protolith was a serpentinized
192 peridotite [*Beinlich et al.*, 2020; *Kelemen et al.*, 2020b]. Between ~ 110 and 182 mbg, fuchsite, a
193 chromian mica, is commonly observed; it occurs as light green quartz-fuchsite intergrowths
194 forming mm- to cm-size green spots easily recognizable on hand samples. Listvenites are highly
195 brecciated down to ~80 mbg and the presence of highly weathered fragmented veins and breccia,
196 associated with occurrences of hematite and goethite, results in variable and sometimes high
197 magnetic susceptibility (Fig. 1b).

198        Serpentinites have dark- to olive-green colors and high magnetic susceptibility values
199 (Fig. 1b). They were recovered at 80.28-100.23 mbg and 181.26-185.47 mbg. Serpentinites are
200 foliated to massive, they have mesh textures outlined by magnetite, and bastite is commonly
201 observed. They are crosscut by abundant light-green to white veins of dolomite (± magnesite)
202 and serpentine with, locally, carbonates replacing the serpentine matrix. Variously carbonated
203 serpentinites are distinguished from listvenites by the absence of quartz, but minor talc is found
204 locally at ~98–100 mbg, close to the contact with listvenites.

205        The metamorphic sole is composed of greenish, microcrystalline and finely laminated
206 schists (197.6-282.88 mbg) and greenstone (below 282.88 mbg) interpreted during shipboard
207 logging as metasediments and metabasalts respectively. It comprises epidote, chlorite, albite,
208 quartz, titanite, with carbonate and/or muscovite in schists and blue-green amphibole and minor
209 pumpellyite in greenstones. The contact with the upper listvenite series occurs at 196.56-197.6
210 mbg. This tectonic contact consists of a ~0.6 m thick layer of fault gouge mixed with clasts of
211 highly fragmented, fine-grained, chlorite- and epidote-bearing rocks. The thrust is characterized
212 by a strong and sharp increase in the Natural Gamma Radiation values (NGR), jumping from on
213 average <1 cps (counts/s) in the upper parts of Hole BT1B to >30 cps at the thrust, then
214 gradually decreasing downhole over ~30-40 m to relatively homogeneous and low values (<5
215 cps) (Fig. 1b).

216        84 samples were collected from Hole BT1B for bulk rock geochemical measurements.
217 Due to the high petrographic variability of the core, the VCD rock-names were used to designate
218 geochemistry samples. 51 listvenites, 14 serpentinites, and 19 greenshists and greenstones were
219 analyzed. 15 samples were collected on-site every 20m during the drilling operations. During the
220 description of the cores on board D/V Chikyu, 59 samples were selected by the shipboard
221 science party as representative of the different lithologies recovered from Hole BT1B. 10
222 additional listvenites and serpentinites were selected from Sections C5704B-73Z-1 to -75Z-2
223 (180.01-186.945mbg) for a coordinated on-shore study of the lower serpentinite intervals and
224 neighboring listvenites (thereafter referred to as consortium samples).

## 3 Methods

226        The major element composition of BT1B samples was determined by X-ray fluorescence
227 (XRF) on-board D/V Chikyu for the drillsite and shipboard samples and at GeoLabs (Ontario,
228 Canada) for consortium samples. Total $H_2O$ and $CO_2$ concentrations and abundances in inorganic
229 carbon of the drillsite and shipboard samples were determined by combustion CHNS elemental
230 analysis (EA) and coulometry, respectively, on-board D/V Chikyu. The FeO concentrations of a
231 subset of samples (8 serpentinites and 21 Listvenites) was quantified at the University of
232 Lausanne (ISTE) following the Fe-titration protocol of Wilson [1960]. The trace element

233    composition (Li, Sc, Ti, V, Mn, Co, Ni, Cu, Ga, As, Sn, Sb, Rb, Sr, Y, Zr, Nb, Cs, Ba, Rare
234    Earth Elements (REE), Hf, Ta, Pb, Th, U, W) of the drillsite, shipboard and consortium samples
235    was analyzed by Inductively-Coupled-Plasma-Mass Spectrometry (ICP-MS) at Géosciences
236    Montpellier (AETE-ISO Facility, University of Montpellier, France) using the protocol described
237    in Godard et al. [2000]. The preparation of the samples and the analytical procedures are detailed
238    in Supporting Information S1. The main lithological characteristics of the studied samples and
239    their major, trace and volatile element concentrations are reported in Supplementary Dataset S1.

## 4 Results

241    The listvenite series and the underlying metamorphic sole are distinguished by significant
242    downhole differences in the values of several geochemical indicators, such as loss on ignition
243    (LOI), Mg# (100 x cationic (Mg/(Mg+Fe), with all Fe as $Fe^{2+}$), $Al_2O_3$ and trace element contents
244    as illustrated on Figs. 2 and 3. These variations allow to discriminate several lithological and
245    geochemical domains along Hole BT1B.
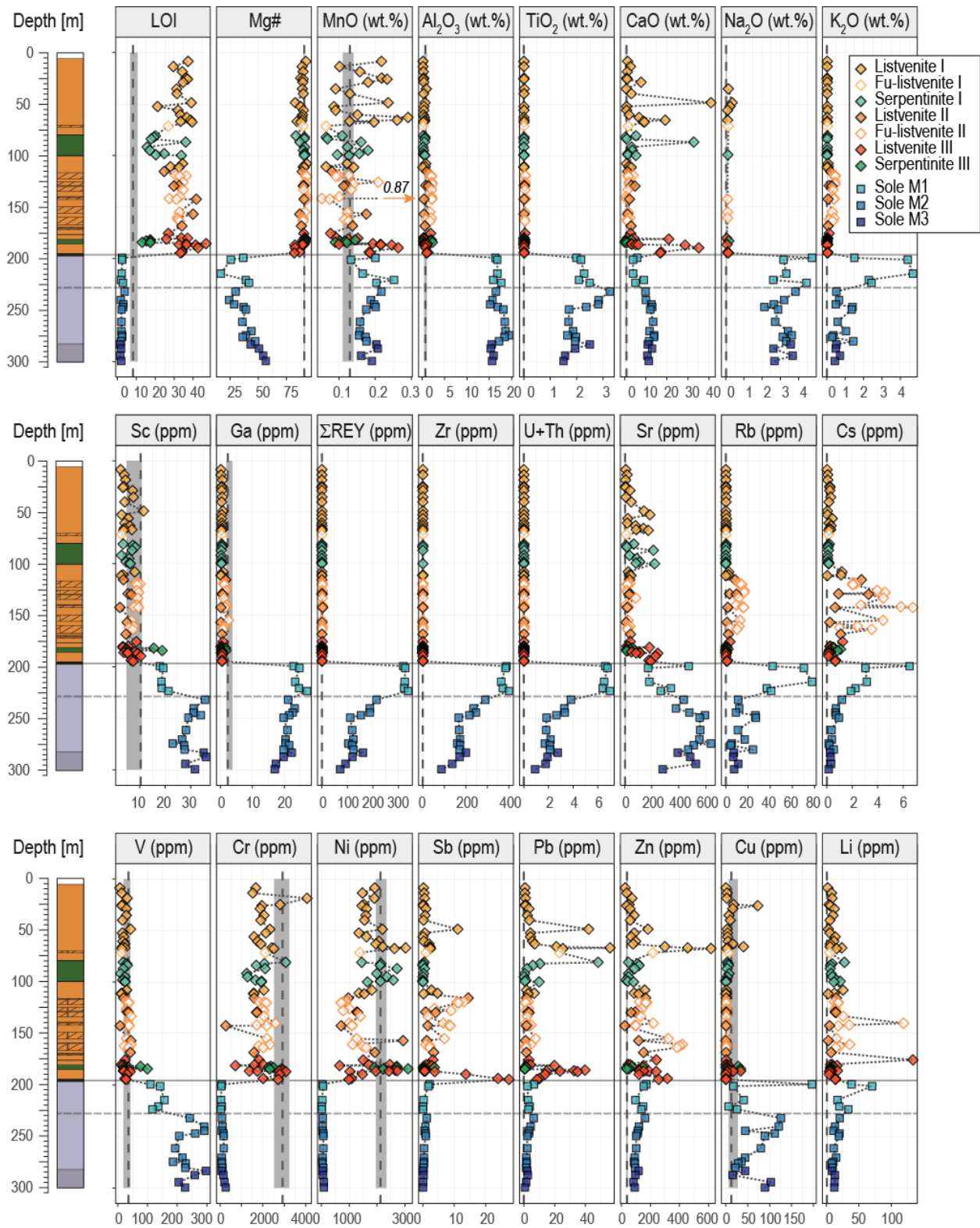
246    4.1 Geochemistry of the listvenite series (depth: 6.02-196.56 mbg)

247    The listvenite series are characterized by high LOI (13.1-46.8 wt.%), high Mg# (~90),
248    high concentrations in transition metals such as Ni (up to 3110 ppm), Co (up to 128 ppm) and Cr
249    (up to 4050 ppm), low concentrations in $Al_2O_3$ (<2 wt.%), $Na_2O$ (~0.1 wt.%) and $TiO_2$ (<0.1
250    wt.%) and in incompatible lithophile trace elements, such as Th (<<0.005 ppm), REE ($\Sigma$REE <1
251    ppm) and high field strength elements (HFSE, e.g., Zr~0.1 ppm). These compositions are highly
252    variable downhole at the sample- to the meter scale yet they overlap, on average, that of the
253    Semail ophiolite mantle (Figs. 2-5).

254    Volatile elements dominate the composition of the listvenite series (LOI = 13.1-46.8
255    wt.%). Listvenites have high $CO_2$ indicative of the predominance of carbonates (LOI>21.2 wt.%;
256    $CO_2$=21.3-43.2 wt.%) but no $H_2O$ (~0 wt.%). Serpentinites display the highest $H_2O$ contents (up
257    to 12.1 wt.%), primarily hosted by serpentines that structurally comprise ~13 wt.% $H_2O$ [*Deer et
258    al.*, 1996], and they have the lowest LOI (down to 13.1 wt.%) and $CO_2$ (down to 5.6 wt.%).
259    Carbonate-rich serpentinites have intermediate compositions with LOI up to 36.1 wt.%, $CO_2$ up
260    to 33.1 wt.% and $H_2O$ as low as 0.3 wt.%. Correlated TIC and total carbon values allow to
261    distinguish samples in which dolomite represent the dominant carbonate species as the shipboard
262    coulometry protocol did not allow the complete dissolution of magnesite (see Supporting
263    Information S1). Most of these samples are serpentinites ($CO_{2(TIC)}$ up to 33.3 wt.%; $CO_{2(TIC)}$: TIC
264    recalculated as $CO_2$).

265    **Figure 2**. (next page) Concentrations of selected geochemical species plotted by depth downhole
266    for all samples from Hole BT1B. Dashed line indicates the median concentration from literature
267    data for peridotites from the main mantle section (MMS) of the Semail Ophiolite (n=92;
268    [*Gerbert-Gaillard*, 2002; *Godard et al.*, 2000; *Hanghoj et al.*, 2010]), and the grey field indicates
269    $\pm$ 1 standard deviation (calculated as the 16th and 84th percentile) about the median
270    (Supplementary Table S2). Major oxides are plotted on a volatile free basis. Symbols are in inset.
271    Listvenites, fuchsite-bearing listvenites (Fu-listvenites), serpentinites and ophicarbonates
272    (grouped as Serpentinites) from Listvenite domains I, II and III (see text) are noted I, II and III
273    respectively. Samples from the metamorphic sole (noted sole) are grouped, from top to bottom,
274    as M1, M2 and M3 (see text)

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278       In addition to volatile elements, listvenite series are composed mainly of Si, Mg, Fe and
279  Ca: these elements calculated as oxides represent ~99 % of the volatile-free mass of samples.
280  Their distribution relate primarily to the mineralogy of the core (Fig. 4a): variations in $SiO_2$ (4.4–
281  70 wt.%) reflect changes in carbonate-quartz ratio in listvenites, and in carbonate-serpentine ratio
282  in serpentinites while downhole spikes in CaO (up to 32.8 wt.% in the serpentinites and up to
283  40.9 wt.% in listvenites) correspond to increasing TIC values ($CO_{2(TIC)}$ up to 40.1 wt.% in
284  listvenites), indicative of the presence of dolomite (Fig. 1; [*Kelemen et al.*, 2020a]).
285  Relationships to mineralogy are more difficult to identify for Mg, Fe and Mg#, except for local
286  decreases in MgO and Mg# associated to high CaO and TIC contents indicative of the presence
287  of dolomite (e.g. listvenite C5704B-23Z-1,37.0--42.0cm at 48.72 mbg; Figs. 2-3). In contrast,
288  $Al_2O_3$, transition metals (e.g., Ti, Ni and Co) and lithophile trace elements (e.g., REE, Y) do not
289  correlate with changes in mineralogy and volatile chemistry: in particular, we do not observe
290  major changes in their distribution between listvenites and serpentinites. These elements,
291  generally considered as fluid-immobile, display coherent tens of meter scale downhole trends
292  that allow to define three geochemical domains (Figs. 3-5), thereafter described as, from top to
293  bottom, listvenite domains I, II and III.

294       *Listvenite domain I* (top to~112 mbg) comprises listvenites (*listvenites I*), including one
295  fuchsite-bearing listvenite, and serpentinites (*serpentinites I*). It is characterized by low
296  concentrations of $Al_2O_3$ (0.14-0.92 wt.%) and $TiO_2$ (<0.01 wt.%) and of trace elements (e.g.
297  $Yb_N$=0.03-0.22; N=normalized to CI-Chondrite [*McDonough and Sun*, 1995]) that overlaps that
298  of the harzburgites and dunites from the main mantle section (MMS) of the Semail ophiolite
299  [*Godard et al.*, 2000; *Hanghoj et al.*, 2010; *Lippard et al.*, 1986]. It displays relatively linear
300  REE patterns similar to that of the MMS, with normalized REE abundances decreasing from
301  heavy (HREE) to light REE (LREE) ($[Ce/Yb]_N$=0.08 – 0.54). Several samples, in particular the
302  serpentinites and ophicarbonates, present minor LREE enrichments relative to middle REE
303  (MREE) (e.g., carbonate rich serpentinite C5704B-44Z-4, 50.0-55.0 cm with $(La/Sm)_N$=3.7),
304  similar to what observed in some Ca-bearing carbonates from oceanic and ophiolitic low
305  temperature peridotite hosted hydrothermal systems [*Noel et al.*, 2018; *Schroeder et al.*, 2015].

306       *Listvenite domain II* (~112 mbg to ~170 mbg) is composed of listvenites alternating with
307  fuchsite-bearing listvenites (all grouped as *listvenites II*). It is characterized by an elevated
308  concentrations of $Al_2O_3$ (0.4-2.23 wt.%) and $TiO_2$ (0.02-0.06 wt.%) compared to Listvenite
309  domain I. This domain has higher lithophile trace element contents (e.g. $Yb_N$=0.18-1.03) and
310  highly fractionated "spoon-shaped" REE patterns ($(Ce/Yb)_N$=0.007-0.061) with relatively flat
311  convex-upward MREE-HREE segments ($(Dy/Yb)_N$=0.49-0.96) and slight but systematic
312  enrichments of La relative to Ce ($(La/Ce)_N$=1.03-4.9), comparable to that of the basal lherzolites
313  and amphibole-bearing basal lherzolites from the northern Semail ophiolite (Figs. 4-5; [*Khedr et*
314  *al.*, 2014; *Takazawa et al.*, 2003]). These REE patterns are very similar in shape to those
315  obtained by Prigent et al [2018a] on clinopyroxene and amphibole from the banded lherzolite
316  units close to the metamorphic sole. Listvenite domain II is also distinguished by, on average,
317  low Ni (~1200 ppm) and Co (~68 ppm) concentrations and low $Fe^{3+}/Fe_{TOT}$ (0.1-0.6) compared to
318  Lisvenite domain I (Ni~1900 ppm; Co~90 ppm; $Fe^{3+}/Fe_{TOT}$~0.7).
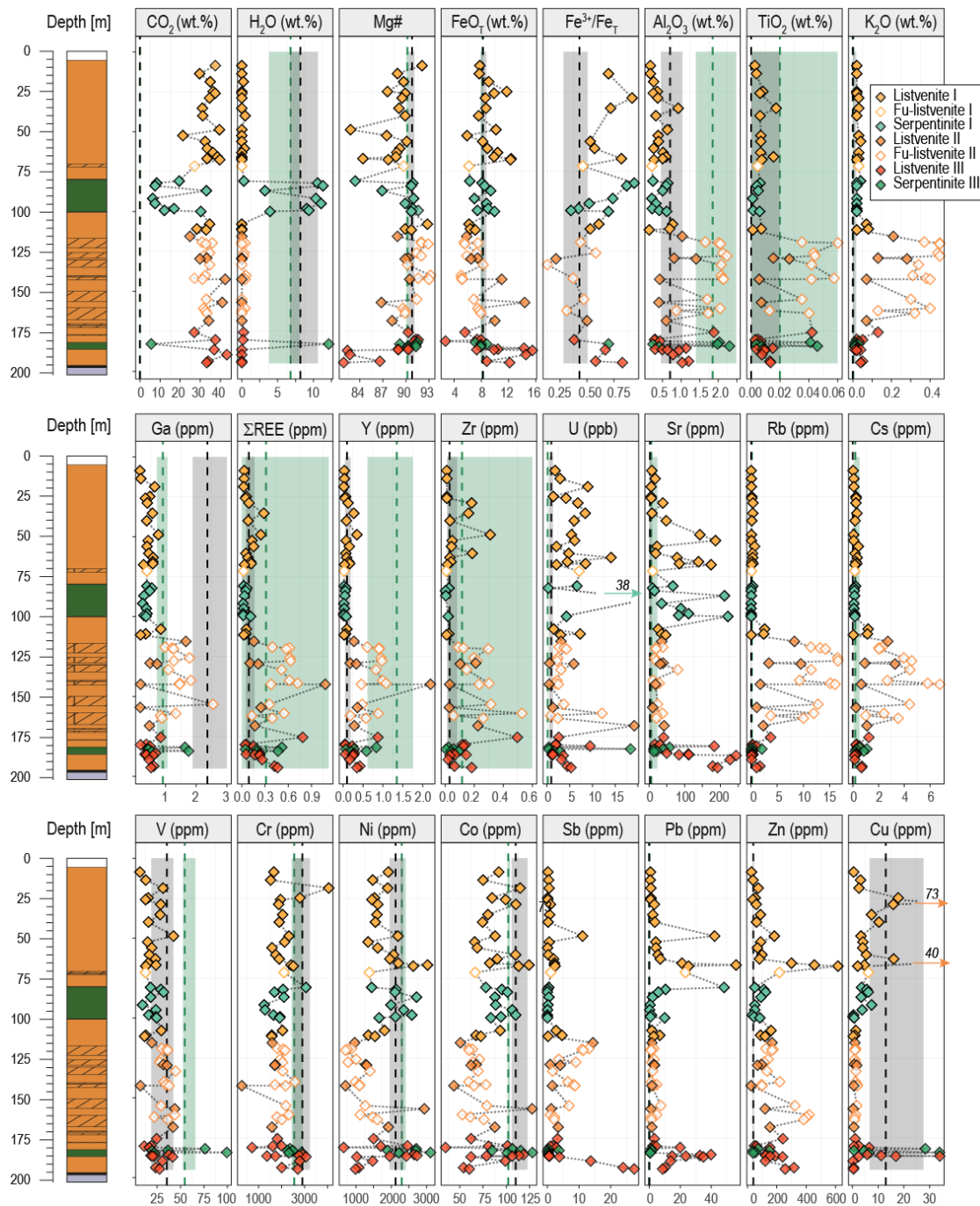
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320

**Figure 3**. Concentrations of selected geochemical species plotted by depth downhole for samples from the Hole BT1B listvenite series only. Dashed grey and green lines indicates the median concentration from literature data for peridotites from the main mantle section (MMS, n=92) and basal lherzolites ([*Lippard et al.*, 1986; *Takazawa et al.*, 2003]; n=13) of the Semail ophiolite, respectively. Fields of the corresponding color indicate ± 1 standard deviation (calculated as the 16th and 84th percentile) about the median (Supplementary Table S2). Symbols for Hole BT1B samples and literature data are in inset, abbreviations are as in Figure 2.

330    *Listvenite domain III* (~170 mbg to the basal thrust) is the closest to the metamorphic
331    sole. It comprises listvenites (*listvenites III*) and serpentinites (*serpentinites III*). It has
332    concentrations in $Al_2O_3$ (0.24-2.30 wt.%), $TiO_2$ (<0.02 wt.%) and of trace elements (e.g.
333    $Yb_N$=0.05-0.77) intermediate between that of the Lisvenite domains I and II, the most enriched
334    samples being serpentinites and listvenites close to Listvenite domain II. It displays linear REE
335    patterns mostly similar to that of Lisvenite domain I (($Ce/Yb)_N$=0.08-0.67) with the exception of
336    some samples close to Listvenite domain II that have similarly fractionated REE patterns
337    (($Ce/Yb)_N$=0.011-0.163). Listvenite domain III is characterized by systematic enrichments in
338    LREE relative to MREE (($La/Sm)_N$=0.34-3.29), this trend overall increasing towards the basal
339    thrust. This downhole trend goes with a decrease in Mg# (down to 82), an increase in $Fe^{3+}/Fe_{TOT}$
340    up to Listvenite domain I values (0.38-0.83) and an increasingly scattered distribution for several
341    elements such as Ni (650-3100 ppm) and Co (35-128 ppm), or Cu (0.2-34 ppm) and Pb (0.5-39.7
342    ppm) as well as a progressive increase in the concentrations of CaO (up to 35.1 wt.%), Sr (up to
343    244.3 ppm) and Sb (up to 27.4 ppm). These variations are associated to an increase in the
344    dolomite fraction towards the basal thrust (Fig. 1b).

345    The listvenite series display spiked U-shaped trace element patterns that reflect the
346    relative depletion of Zr-Hf relative to neighboring elements and significant enrichments in
347    incompatible fluid mobile elements (FME), such as alkali elements (Cs, Rb, Ba), Li, Sb, U, Pb
348    and Sr relative to Th, Nb, Ta and LREE (Fig. 5). These strong FME enrichments distinguish the
349    listvenite series from refractory peridotites from the main mantle section (Figs. 2, 3 and 5). High
350    FME concentrations appear as spikes on downhole plots (Fig. 3). These spikes are however
351    located preferentially in well-defined listvenite domains for most elements with a decoupling in
352    the downhole distribution of alkali elements and other FME such as Sb, Pb, Sr and U. High
353    concentrations of alkali elements including $K_2O$ (0.07-0.45 wt.%), Ba (up to 420 ppm), Rb (1.1-
354    17 ppm) and Cs (0.24-6.8 ppm) distinguish Listvenite domain II from Listvenite domains I and
355    III ($K_2O$=0.02-0.08 wt.%; Rb=0.006-2.6 ppm; Cs=0.02-1.19 ppm), with the highest
356    concentrations in fuchsite-bearing listvenites. In contrast, downhole spikes in concentrations for
357    Pb, Sr and U are mainly in Listvenite domains I and III (Pb up to 56 ppm, Sr up to 244 ppm, U
358    up to 0.04 ppm). Cu (up to 73 ppm) and to a lesser extent, Sb (up to 27.4 ppm) and Zn (up to 610
359    ppm) display similar trends with peaks in concentrations mainly in Listvenite domains I and III.

360    For some elements, their concentrations and distribution are highly scattered and become
361    virtually indistinguishable from the metamorphic sole, as for Li and MnO (e.g., Li=2.5-134 ppm
362    in listvenite series and 8-70 ppm in metamorphic rocks; Fig. 2). Some of these extreme
363    enrichments can be correlated on a case-by-case basis to lithological or structural features as for
364    MnO-rich sample C5704B-60Z-4-1,24.0--29.0 cm(V) (MnO=0.87 wt.%) identified as a
365    listvenite vein crosscutting a fuchsite-bearing listvenite (C5704B-60Z-4-1,24.0--29.0cm(H)).
366    This sample is also the most enriched in REE and Y (e.g., Y=2.2 ppm) and the most depleted in
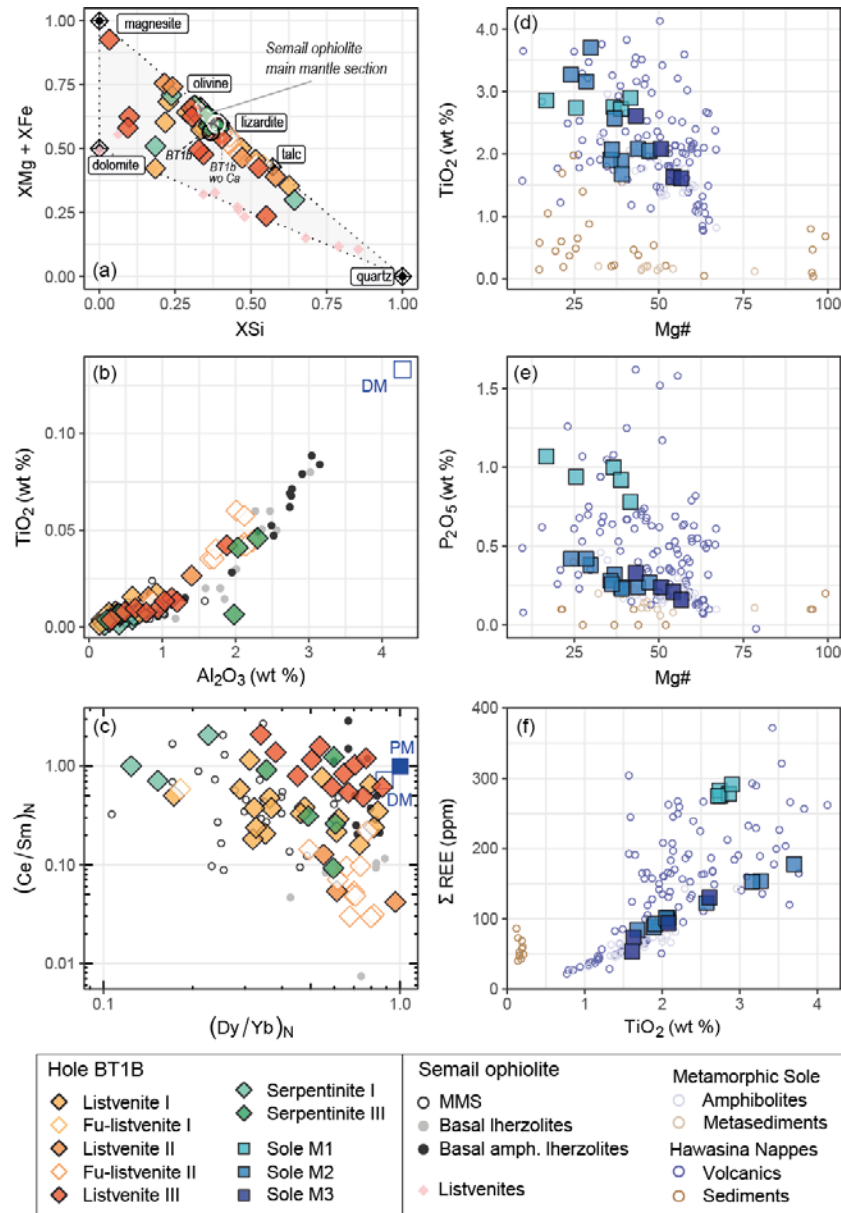367    Cr (271 ppm) indicating extensive elemental redistribution occurring at the sample scale.

368

**Figure 4**. Scatterplots of the composition of samples recovered from Hole BT1B. The composition of the listvenite series is plotted as (a) (XMg + XFe) versus XSi (X = molar proportion), (b) $TiO_2$ (wt.%) versus $Al_2O_3$ (wt.%), (c) Chondrite normalized (La/Sm) versus (Dy/Yb) diagrams. The composition of the metamorphic sole is plotted as (d) $TiO_2$ (wt.%) and (e) $P_2O_5$ (wt.%) versus Mg# and on (f) Total REE (ppm) versus $TiO_2$ (wt.%) diagrams. (a) The composition of the listvenite series is compared to the range of compositions of refractory peridotites from the Semail Ophiolite (white field) and to the composition of magnesite, dolomite, quartz, talc, lizardite and olivine [*Deer et al.*, 1996], the mean composition of listvenite series (black circle) also, recalculated as Ca free (white circle). (b)$TiO_2$ and $Al_2O_3$ compositions are plotted on a volatile-free basis and, when available, recalculated ICPMS data was used for plotting $TiO_2$. The composition of depleted mantle (DM [*Salters and Stracke*, 2004]) and primitive mantle (PM [*McDonough and Sun*, 1995]) and/or literature data from the Semail mantle (MMS [*Gerbert-Gaillard*, 2002; *Godard et al.*, 2000; *Hanghoj et al.*, 2010], basal

382     lherzolites [*Lippard et al.*, 1986; *Takazawa et al.*, 2003] and amphibole bearing basal lherzolites
383     – Basal amph. lherzolites [*Khedr et al.*, 2014]) and listvenites [*Falk and Kelemen*, 2015] are
384     shown for comparison on (a), (b) and (c). The composition of Permian and Triassic volcanics
385     (including Haybi) ([*Chauvet et al.*, 2011; *Lapierre et al.*, 2004; *Lippard et al.*, 1986; *Maury et*
386     *al.*, 2003]) and sediments [*Oberhänsli et al.*, 1999] from the underthrusted Hawasina nappes, and
387     of amphibolites and metasediments from the metamorphic sole [*Ishikawa et al.*, 2005] are shown
388     for comparison on  (d) , (e) and (f). Symbols are in inset and abbreviations are as in Figure 2.

389

390         4.2 Geochemistry of the metamorphic sole (depth: 196.56- 300.13 mbg)

391         BT1B metamorphic rocks have LOI of 1.3-3.8 wt.% related to the presence of $H_2O$ (>2
392     wt.%) in hydrous minerals (chlorite, amphibole …) and minor $CO_2$ (0.04-0.97 wt.%) in
393     carbonates, mainly calcite ($CO_{2(TIC)}$ 0.01-0.95 wt.%). Compared to listvenite series, they have
394     low Mg# (16.7-56.6), low concentrations of Cr (42-265 ppm) and Ni (15-86 ppm) and high
395     concentrations of $Al_2O_3$ (15.2-19.4 wt.%), $Na_2O$ (2.1-4.7 wt.%), $P_2O_5$ (0.16-1.07 wt.%) and $TiO_2$
396     (1.6-3.7 wt.%) and they display a relatively narrow range of $SiO_2$  concentrations (44.5-52.6
397     wt.%). They are characterized by high concentrations of V (110-297 ppm) and of incompatible
398     trace elements, such as Th (0.75-5.9 ppm), REE ($\Sigma$REE=54-291 ppm) and HFSE (e.g., Nb=9.5-
399     63 ppm), and by LREE-enriched chondrite–normalized REE patterns ($(Ce/Yb)_N$=3.4-8.2). They
400     overlap in composition with the amphibolites from the Semail ophiolite metamorphic sole
401     [*Ishikawa et al.*, 2005; *Lippard et al.*, 1986] and the volcanic rocks from the underthrusted
402     Hawasina assemblages [*Chauvet et al.*, 2011; *Lapierre et al.*, 2004; *Maury et al.*, 2003] (Figures
403     4-5). More particularly, they display trace element compositions similar to the transitional to
404     alkali basalt series forming the Hawasina-Haybi complex, suggesting that the metamorphic rocks
405     recovered at Hole BT1B comprise only metabasalts. We have subdivided the Hole BT1B
406     metamorphic rocks into three groups based on their lithology, physical properties, geochemistry
407     and depth (Fig. 3).

408         The first group (*M1*) corresponds to the high NGR cores (197.6-~230 mbg; Fig. 1b). It
409     represents the most enriched endmember of BT1B metabasalts for $P_2O_5$ (0.78-1.07wt.%), for
410     alkali elements ($K_2O$=1.5-4.7wt.%; Ba=192-598ppm; Rb=37-78ppm; Cs=1.9-6.5ppm) and for
411     moderately to highly incompatible lithophile elements, such as REE ($Yb_N$~23), Th (5.1-5.9ppm)
412     and U (1.0-1.3ppm). They have however middle range values for transition elements $TiO_2$ (1.9-
413     2.4wt.%), V (110-157ppm) and Sc (17.9-21.3ppm). They are overall depleted in CaO (3.95-
414     9.1wt.%) and in Sr (173-342ppm) although these elements increase with depth. *M1* metabasalts
415     display the most fractionated REE patterns (($(Ce/Yb)_N$=7.7-8.2) as well as slight enrichments in
416     Nb-Ta (e.g. Nb/Th~1.27xPM), and negative anomalies in Pb and Sr relative to neighboring
417     elements (Pb/Ce~0.25xPM; Sr/Ce 0.2xPM) on extended trace element diagrams (Fig. 5). Their
418     high concentrations in K, Th and U likely explain their high NGR values (Fig.1).
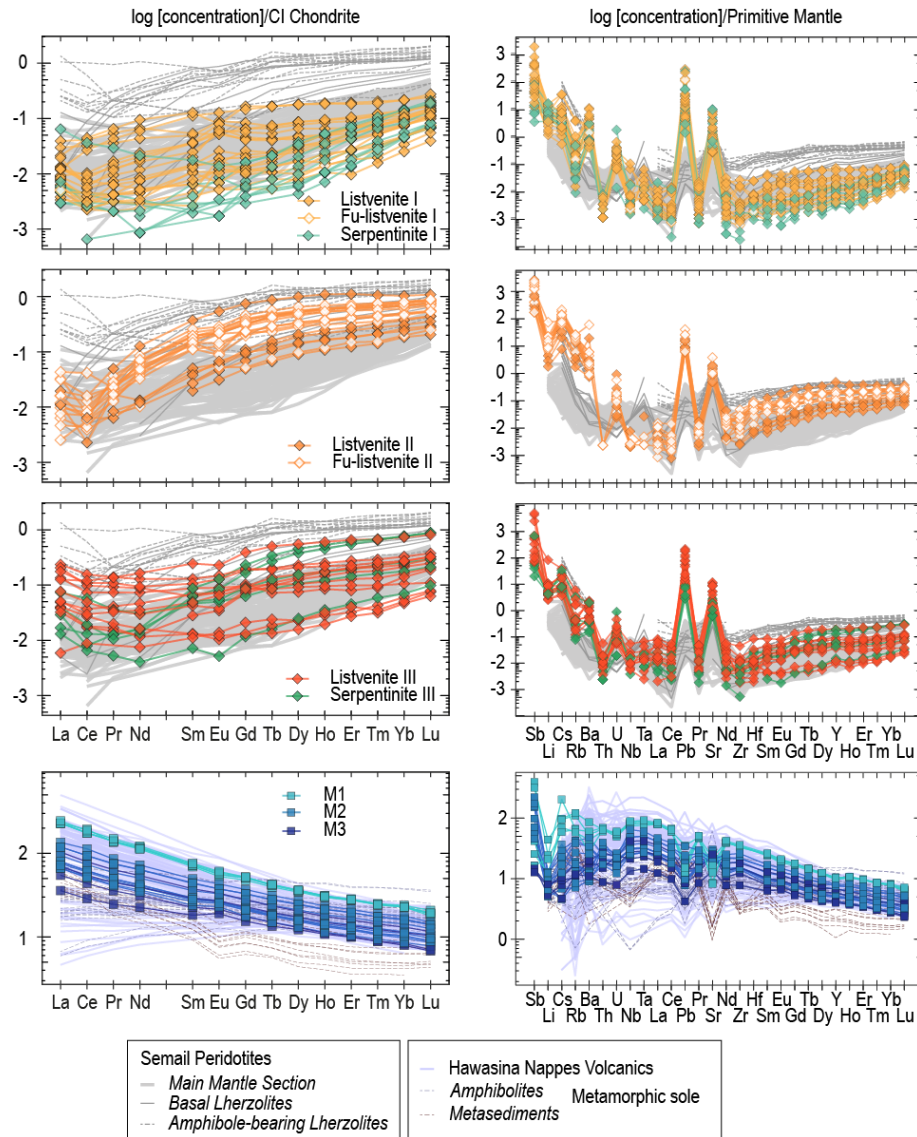
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**Figure 5**. Rare-earth element (REE) and trace element spider diagrams for samples from Hole BT1B. Separate panels are plotted for Listvenite domains I, II and III (see text), and for the metamorphic sole. Left: CI chondrite-normalised REE patterns. Right: Primitive mantle-normalized trace element plots; normalizing values from McDonough and Sun [1995] . Literature data from Semail Ophiolite peridotites are plotted for reference in the upper three panels, and from metamorphic sole and Hawasina nappes in the lowermost panel. Symbols are in inset. Abbreviations are in Figure 4.

The second and third groups, *M2* (~230–282.88 mbg) and *M3* (below 282.88 mbg) are composed of schists and greenstones respectively. The downhole transition from *M1* to *M2-M3* metabasalts is characterized by a sharp decrease in $K_2O$, and Rb, REE, HFSE, Th and U and a sharp increase in $TiO_2$ (up to 3.7 wt.%; Fig.2). *M2-M3* metabasalts record a continuous downhole increase in Mg# (up to 56.6), Sc (up to 36 ppm), V (up to 297 ppm), Co (up to 45 ppm) and Ni (up to 87 ppm) and a decrease in alkali elements (e.g., $K_2O$ down to 0.3 wt.%) and

435  in moderately to highly incompatible lithophile elements (e.g., $TiO_2$ down to 1.6 wt.%; Th down
436  to 0.7 ppm). They also show a continuous decrease in trace element concentrations and in
437  LREE/HREE ratios with depth (*M2:* $Yb_N$ = 17.7-8.5, $(Ce/Yb)_N$=6.4-4.5; *M3:* $Yb_N$ = 13.1-8.3,
438  $(Ce/Yb)_N$=6.0-3.4; Figs. 2, 4). Negative correlations between $TiO_2$, REE and HFSE and Co, Ni
439  and Mg# such as those observed downhole are typical of basaltic fractional crystallization trends
440  with the lowermost *M3* metabasalts having the least evolved compositions. Finally *M2-M3*
441  metabasalts display positive anomalies in Nb-Ta (e.g. Nb/Th~1.5xPM), minor negative
442  anomalies in Pb (Pb/Ce=0.43-0.88xPM) and variable Sr anomalies (Sr/Ce=0.5-1.7xPM) relative
443  to neighboring elements on extended trace element diagrams (Fig. 5). It should be noted also that
444  CaO and Sr downhole trends appear decoupled from other elements: they increase then stabilize
445  with depth in *M2* (CaO up to 14.0 wt.%; Sr up to 638 ppm) and decrease towards the bottom of
446  the borehole (CaO down to 10.75 wt.%; Sr down to 281 ppm).

**5 Discussion**

448  Drilling at Hole BT1B provides the first high resolution sampling of the transition from
449  the base of the Semail ophiolite to its metamorphic sole, thus allowing a detailed study of the
450  mineralogical and geochemical processes occurring across this major tectonic structure. The
451  basal thrust constitutes the ophiolite sole in this area and separates highly metasomatized,
452  deformed and fractured rocks of the mantle-derived listvenite series from the underlying
453  greenschist lithologies. The listvenite series recovered from the upper ~200 m of Hole BT1B
454  comprise highly fractured and veined listvenites and fuchsite-bearing listvenites and two minor
455  intervals of variously carbonated serpentinites (Fig. 1). Their main petrophysical and
456  geochemical characteristics reflect the dominant mineralogy of the cores: quartz + magnesite ±
457  dolomite ± fuchsite for listvenites; and serpentine ± dolomite ± magnesite for carbonated
458  serpentinites (Figs. 1-4). Specifically, they have high $CO_2$ contents (up to ~43 wt.%, Fig. 3)
459  showing the ubiquity of fluid-rock interactions with $CO_2$-bearing fluids (Kelemen et al [2021]
460  and references therein). In spite of the complete transformation of their mineralogy due to $CO_2$-
461  metasomatism, the listvenite series have average compositions comparable to that of the highly
462  serpentinized, often amphibole bearing, banded lherzolites forming a narrow zone at the basal
463  transition from the Semail mantle section to its metamorphic sole  (Fig. 2-5, [*Khedr et al.*, 2013;
464  *Khedr et al.*, 2014; *Prigent et al.*, 2018a; *Yoshikawa et al.*, 2015]). The BT1B metamorphic rocks
465  recovered in the lower ~100m of the borehole comprise finely laminated, fine grained greenish
466  mineral assemblages. They have basaltic compositions similar to that of the alkali basalts from
467  the underthrusted Hawasina assemblages (Figures 4-5) from which they likely derive (e.g., [*M P
468  Searle and Malpas*, 1980]). There is no compositional evidence for the presence of interlayered
469  meta-sediments, which are often observed associated with the mafic amphibolites composing the
470  metamorphic sole elsewhere in Oman (e.g., [*Agard et al.*, 2016; *Cowan et al.*, 2014; *Lippard et
471  al.*, 1986; *Soret et al.*, 2017]) including in Wadi Mansah [*Falk and Kelemen*, 2015], and as
472  previously hypothesized for BT1B cores [*P.B. Kelemen et al.*, 2020a]. They display decreasing
473  concentrations in alkali elements from *M1* to *M2-M3* metabasalts, similar to others have been
474  observed in the metamorphic sole, associated with the transition from HT to LT away from the
475  ophiolite contact ([*Ishikawa et al.*, 2005]; Fig. 2), although the mineralogy of those lithologies
476  indicate equilibration mainly under greenshist facies, indicative of retrograde metamorphism
477  (e.g., [*Ambrose et al.*, 2021; *Ghent and Stout*, 1981]).

478  We will use the high resolution geochemical database to evaluate the mass and volume
479  changes associated to the formation of the listvenite series during $CO_2$ metasomatism and discuss

480 elemental mobility during these processes. We will focus in particular on the role of the
481 composition of the (metasomatized) protolith(s), the main fluid pathways as recorded across the
482 basal thrust and possible changes in fluid source(s) and interactions with adjacent lithologies.

483     5.1 Mass changes and elemental mobility during listvenization

484     To evaluate the respective contribution of the host rock and incoming metasomatic
485 fluids(s) to the composition of the listvenite series and to constrain elemental mobility at the
486 scale of the borehole, we used the mass balance model of Baumgartner and Olsen [1995]. This
487 approach allows to evaluate the addition and removal of a broad range of chemical elements
488 during the alteration of a protolith of known composition and the resulting mass changes,
489 without a priori assumptions on elemental mobility. The main challenge for these calculations
490 was determining consistent and representative trace element compositions for the listvenite series
491 and for the model protoliths.

492     The listvenite series are characterized by strong downhole variations in mineralogy and
493 geochemistry from the sample to the meter scale. For simplicity, mass balance calculations
494 (detailed in Supplementary Information Text S2) were carried out for the two main rock types,
495 (listvenites including fuchsite-bearing listvenites, and the variously carbonated serpentinites), for
496 each of the downhole geochemical listvenite domains I, II and III. The results are illustrated as
497 isocon diagrams (Fig. 6) where average elemental concentrations of the altered rock are plotted
498 against those of the model protolith. Immobile elements were identified on the basis of the
499 largest number of elements which are consistent with a single isocon (line of immobility) to
500 within their uncertainty in the protolith and altered rock (i.e. collinear on the plot). The 1:1
501 reference line on isocon diagrams corresponds to zero mass change during alteration. Isocons
502 which lie above or below the 1:1 line indicate, respectively, overall mass loss or mass gain
503 during alteratio (reflecting overall concentration or dilution of immobile elements during each of
504 these scenarios).  Elements plotting above and below the isocon are enriched and depleted,
505 respectively, in the altered rocks compared to the model protolith.

506     The trace element compositions for the model protoliths were determined on the basis of
507 the published geochemical studies of Semail peridotites for which structural (localization,
508 distance to the Moho and/or to the sole) and petrological (lithology, mineralogy) information
509 were available. We defined two endmember compositions (Table S3): (1) a refractory protolith
510 composition; due to the scarcity of data on the composition of basal peridotites, we used the data
511 obtained on harzburgites ± dunites dominant in the main mantle section of the Semail ophiolite
512 [*Gerbert-Gaillard*, 2002; *Godard et al.*, 2000; *Hanghoj et al.*, 2010; *Lippard et al.*, 1986]; (2) a
513 fertile protolith, representative of the composition of basal lherzolites, including amphibole-
514 bearing samples [*Khedr et al.*, 2014; *Lippard et al.*, 1986; *Takazawa et al.*, 2003]. The model
515 refractory and fertile protoliths are noted MMS and basal lherzolite respectively in Fig. 6 and
516 Table 1. It should be noted that, for some trace elements, the published chemical database is
517 limited (e.g., Ga) and/or highly variable (e.g., LREE, Cs), in particular for basal lherzolites,
518 resulting in a large uncertainty in their distribution in the model protolith(s); however, these
519 elements represent a minor subset of the chemical database and therefore had little impact on the
520 evaluation of the overall mass changes resulting from $CO_2$ metasomatism.
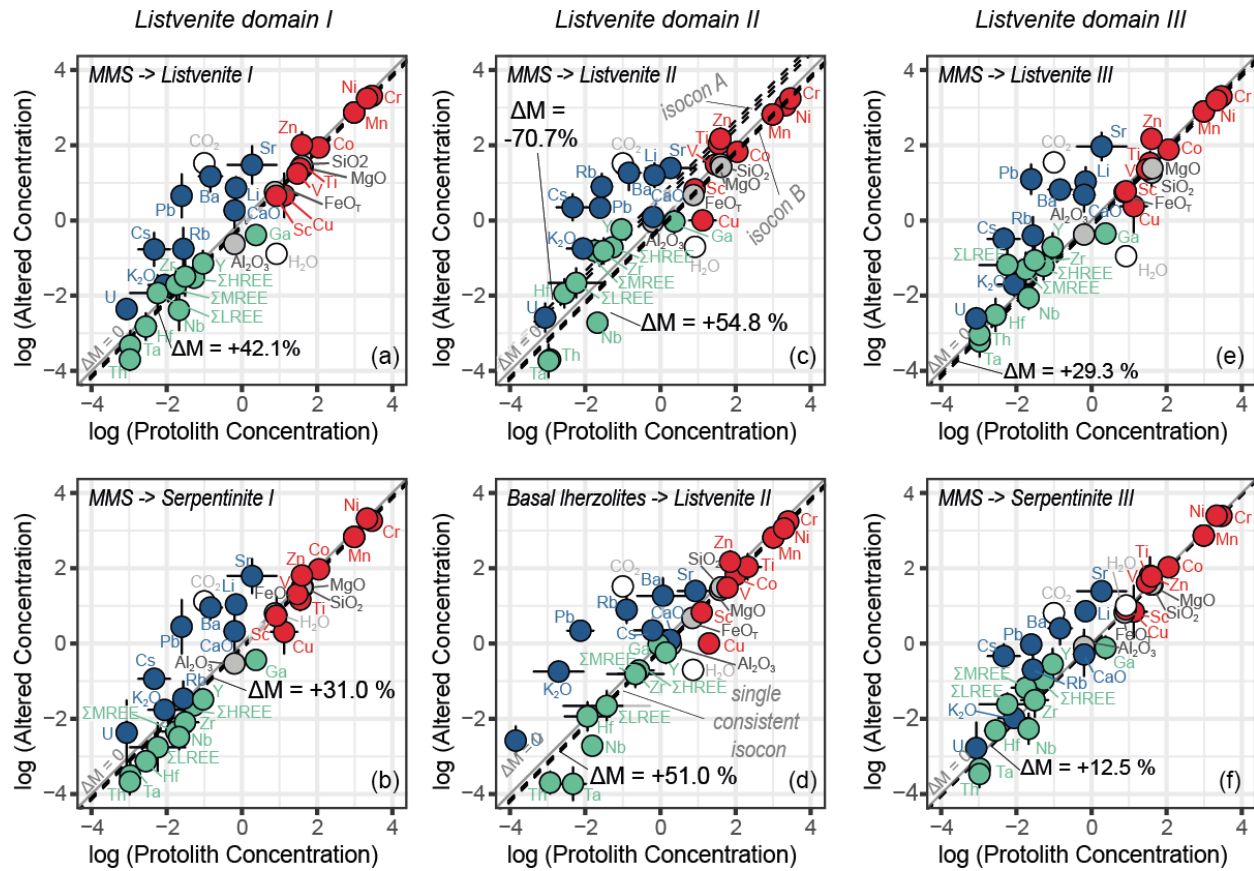
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**Figure 6**. Mass balance diagrams for listvenite series from Hole BT1B. Isocon plots comparing the average composition of listvenites and serpentinites – calculated as the log normal mean – from each listvenite domains with potential protolith compositions on log-log scales. Plotted on each panel, each corresponding to a protolith–altered rock pair, is an isocon (line of immobility). Isocons were calculated using the approach of Baumgartner and Olsen [1995]. The 1:1 reference line in grey on each panel corresponds to zero mass change during alteration. Isocons above and below this indicate mass loss and gain, respectively. Using the average composition of the MMS refractory peridotites (calculated as the log normal mean) as the protolith composition for each domain, this approach gives consistent patterns of mobile and immobile elements, with the exception of Listvenites II (panel c) where two potential isocons are apparent in the data (labeled A and B). Mass balance was repeated for Listvenite II with the average composition of basal lherzolites (including amphibole bearing lherzolites) which gives a single consistent isocon and similar patterns of mobility/immobility to the other panels. Modeled protolith compositions and calculated statistics are in Supplementary Table S3.

537

The mass balance calculations comparing the composition of listvenites and serpentinites to the refractory protolith showed co-linearity (within uncertainty) for most major and trace elements on isocon diagrams (Fig. 6, Supplementary Information Text S2). Only listvenites II displayed inconsistent results with two parallel 1:1 slopes (Fig. 6c). The same calculations using the fertile model protolith showed collinear trends and elemental variations similar to those obtained for the adjoining domains (Fig. 6d). This result supports the hypothesis that the

544 listvenite series were formed after a mantle section having a structure and composition analogous
545 to basal banded lherzolites, with Listvenite domains I and III being formed after a refractory
546 protolith end-member, and Listvenite domain II after a fertile protolith.

547     The listvenite series do not record major changes in $SiO_2$, MgO and $FeO_T$, the main
548 constituents of their mantle protolith(s), as already pointed out by Falk and Kelemen [2015] for
549 the Wadi Mansah listvenites. Our new data show that $Al_2O_3$ and $Na_2O$, and transition elements
550 generally concentrated in mantle peridotites (V, Cr, Co, Ni, Mn Ti, and Sc) are also immobile at
551 the scale of the borehole despite their meter-scale scattered downhole distribution (Fig. 2 and 3).
552 Similarly, moderately incompatible lithophile trace element such as HREE, MREE, Y, Zr and Hf
553 are aligned on the same 1:1 trend indicating that they were immobile at the scale of the borehole
554 during listvenitization. However, the most incompatible elements, even those generally
555 considered as fluid immobile, display minor differences from one rock type and domain to the
556 other (e.g., depleted LREE in Serpentinite I). In particular, Nb, Ta and Th are systematically
557 depleted relative to the model mantle protoliths. This could reveal trace element depleted
558 protoliths compared to the models but, we posit that it most likely relates to the uncertainty due
559 to the limited dataset on the composition of these elements in the Semail peridotites. The same
560 consideration could explain the apparent systematic loss of Ga observed for all calculations.

| Listvenite Domain | Lithology | Protolith | Mass change | Volume change |
|---|---|---|---|---|
| I | Listvenite | MMS | **42.1 ± 12.5** | 41.3 ± 14.4 |
| | Serpentinite | MMS | **31.0 ± 10.9** | 34.6 ± 13.6 |
| II | Listvenite | Basal lherzolites | **51.0 ± 13.3** | 50.1 ± 15.9 |
| III | Listvenite | MMS | **29.3 ± 20.2** | 28.8 ± 20.5 |
| | Serpentinite | MMS | **12.5 ± 4.5** | 13.9 ± 5.6 |

561

562 **Table 1.** Summary of mass and volume changes associated with $CO_2$ metasomatism. Changes in
563 mass were determined from a mass balance calculations using the approach of Baumgartner and
564 Olsen [1995]. These were converted to volume changes using a density of $2.81 \pm 0.50$ g cm$^{-3}$ for
565 serpentinised peridotite protoliths [*Dewandel*, 2002] and the mean density of listvenite and
566 serpentinite measured in BT1B cores, $2.86 \pm 0.07$ g cm$^{-3}$ and $2.52 \pm 0.05$ g cm$^{-3}$, respectively
567 [*Kelemen et al.*, 2020b] and propagating the associated uncertainties. For the listvenite or
568 serpentinite of each domains, results are shown for the most appropriate protolith only (see
569 Supporting Text S2 for details).

570

571     Only a limited number of elements show evidence for significant remobilization during
572 $CO_2$-metasomatism, relative to the model mantle protoliths. Enrichments are seen in CaO (where
573 dolomite is present), and most elements typically considered as fluid mobile: Li, K, Rb, Cs, Ba,
574 Sr, Pb and U. These enrichments vary from one domain and rock type to the other thus
575 suggesting downhole variability in fluid rock interactions (discussed in Section 5.2 below). The
576 most noticeable enrichment is however that of $CO_2$. It is associated to the development of two
577 reaction pathways for carbonation with, on one hand, the formation of variously carbonated
578 serpentinites (Serpentinites I and III, Figs. 6b, f) and, on the other hand, that of listvenites
579 (Listvenites I, II and III, Figs. 6a, d, e). Listvenites are distinguished from carbonated
580 serpentinites by their depletion in $H_2O$, suggesting that $CO_2$ addition induced the release of $H_2O$.

581 This process has been documented in listvenites from different orogens (e.g., Atlin listvenites
582 [*Hansen et al.*, 2005]): it could be a non-negligible dehydration mechanism for serpentinized
583 peridotites.

584 Mass balance calculations indicate that $CO_2$ metasomatism resulted in an increase of the
585 mass of the listvenite series compared to their refractory and/or fertile protoliths for serpentinites
586 as well as for listvenites. The estimated mass increase was slightly less for carbonated
587 serpentinites (12 to 31 %) than for listvenites (29.3 to 51.0 %). The density of carbonated
588 serpentinites, listvenites and model protoliths being relatively close, these mass changes induced
589 volume expansion of the same order of magnitude (Table 1). This massive increase in mass and
590 volume was likely accommodated by the abundant veining characterizing the listvenite series; it
591 suggests a strong coupling between carbonation reactions and fracturing during their formation.
592 The high density veining network could be associated to (1) crystallization forces produced by
593 the carbonation reactions (e.g., [*Jamtveit et al.*, 2008; *Lambart et al.*, 2018; *Ulven et al.*, 2014])
594 and/or (2) regional tectonics as, for instance, the development of local and transient extension
595 areas during the earliest stages of convergence in forearc and/or protoarc environments (e.g.
596 [*Reagan et al.*, 2017]), for which the Semail ophiolite could be an analogue (e.g., [*Belgrano et
597 al.*, 2019; *Kelemen et al.*, 2021]).

598 5.2 The BT1B listvenite series: sampling the final stage of a protracted suite of fluid rock
599 interactions

600 The BT1B listvenites series are characterized by strong mineralogical and compositional
601 variations from the sample to the meter-scale (Figs 2-4; Supplementary Dataset S1). Significant
602 variations in mass and volume are also observed at the tens of meter scale, from one Listvenite
603 Domain to the other (Table 1). In the same way, the transition from listvenite series to the
604 underlying metamorphic sole is sharp for structure, mineralogy and for geochemistry (with the
605 exception of Ca and Sr). These jagged distributions indicate that elemental transport was limited,
606 and fluid-rock interactions localized, along the section during $CO_2$ metasomatism and that $CO_2$-
607 bearing fluid(s) did not originate directly from the underlying metamorphic sole. The main flow
608 paths for $CO_2$ rich fluids were probably parallel to the basal thrust, in relation to the localization
609 of brittle and/or cataclastic deformation (e.g., [*Menzel et al.*, 2020b]). They thus followed and,
610 possibly reused, the same pathways as the fluids driving earlier metamorphic events affecting the
611 basal ophiolitic mantle (e.g., [*Lippard et al.*, 1986; *Prigent et al.*, 2018b; *Yoshikawa et al.*, 2015])
612 and its metamorphic sole (e.g., [*Ambrose et al.*, 2021; *Ghent and Stout*, 1981; *Ishikawa et al.*,
613 2005]). This implies that the basal thrust acted as a main fluid pathway for fluids of variable
614 compositions, over a broad range of pressure and temperature conditions over time. These
615 protracted fluid-rock processes produced a complex suite of metasomatic sequences. Hereafter,
616 we will focus on listvenitization and associated reactions and the respective contributions of the
617 (metasomatized) protolith(s) and fluid compositions in the geochemical signature of the
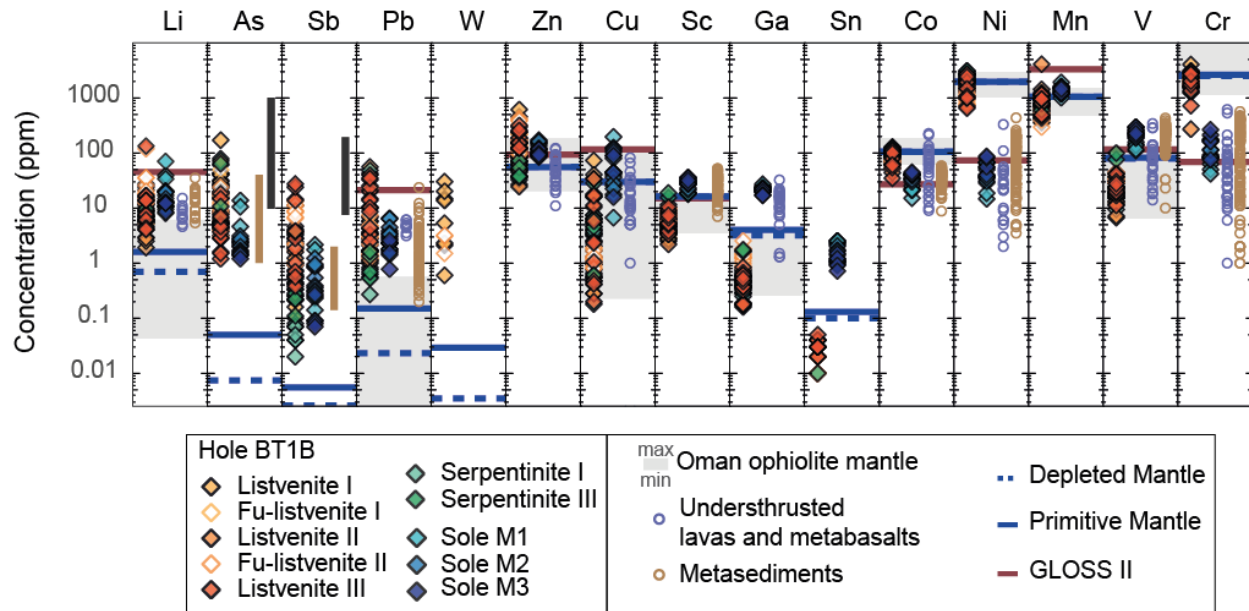618 listvenite series.

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620



**Figure 7**. Caltech diagram of the composition of elemental abundance of BT1B listvenites series and metamorphic sole samples. Compositions are plotted on a log scale and compared with the compositional range of Oman ophiolite (grey field), primitive mantle [*McDonough and Sun*, 1995], depleted mantle [*Salters and Stracke*, 2004], the GLOSS II compilation of subducting sediments [*Plank*, 2014],and lavas and sediments from the metamorphic sole and the understhrusted Hawasina assemblages. As and Sb compositions are compared to the composition of abyssal plain sediments (brown line, [*Plank and Ludden*, 1992]) and hydrothermal sulfides (black line, [*Fouquet et al.*, 2010]). Literature data sources for the Oman Ophiolite, metamorphic sole and Hawasina assemblages are as in previous figures. Symbols are in inset.

630

The BT1B listvenites series are systematically enriched in FME, particularly Cs, Rb, Ba and K, relative to the Semail ophiolite mantle section (Fig. 6). These alkali-rich compositions typically characterize the basal Semail ophiolite mantle and its metamorphic sole; they are interpreted as evidence for interactions with slab derived aqueous fluids along the slab-mantle interface in both the metamorphic sole (e.g., [*Ambrose et al.*, 2021; *Ghent and Stout*, 1981; *Ishikawa et al.*, 2005]) and the adjacent mantle section, where it induces amphibole precipitation at the expense of clinopyroxene in basal lherzolites during incipient subduction (T > 600°; e.g., [*Khedr et al.*, 2013; *Khedr et al.*, 2014; *Prigent et al.*, 2018a; *Yoshikawa et al.*, 2015]) and serpentinization during ophiolite cooling (e.g., [*Lippard et al.*, 1986; *Prigent et al.*, 2018b]). The formation of these alkali-rich aqueous fluids is attributed to the de-volatilization in granulite-amphibolite facies conditions of the altered oceanic crust and associated sediments, with a potential contribution from the metamorphic sole (e.g., [*Prigent et al.*, 2018b]). The $CO_2$-rich fluids triggering the formation of listvenites probably derive from the same deep processes [*Kelemen et al.*, 2021]. Downhole variations in mineralogy and geochemistry, however, indicate a complex relationship between fluid ingress and reactions. This variation may relate to: (1) variable permeability along the core with respect to $CO_2$-rich fluids; (2) variable contributions of earlier fluid-rock interactions and late $CO_2$-metasomatism; (3) variation in the fluid composition, temporal and/or spatial; or any combination of the above. The latter of these could induce the

649  two observed reaction pathways: carbonation only in serpentinites and coupled
650  carbonation/silicification reactions in listvenites.

651  The main mineralogical and petrophysical characteristics of the BT1B listvenite series
652  are manifested in alternating bodies of variously carbonated serpentinite intervals with listvenites
653  and fuchsite-bearing listvenites, a structural pattern also observed at the scale of the Wadi
654  Mansah listvenite massif [*Falk and Kelemen*, 2015]. BT1B listvenites preserve textures
655  indicating that they formed after serpentinites and, in spite of their strong differences in
656  mineralogy and volatile element compositions, serpentinites and listvenites overlap in
657  compositions for most major and trace elements, including alkali elements, suggesting that they
658  both interacted with the same fluids (Figs. 4, 7). A recent clumped isotope thermometry study of
659  BT1B listvenites indicates temperatures of listvenitization of 245-45°C [*Beinlich et al.*, 2020]).
660  These temperatures are low compared to those anticipated for the serpentinization of the basal
661  lherzolites by slab derived fluid (below 350°C) [*Prigent et al.*, 2018b] but there is a possible
662  overlap. We propose that the ingress of $CO_2$-rich fluids occurred after the onset of
663  serpentinization of basal lherzolites, and that the preservation of serpentinites indicates that flow
664  paths were highly localized (likely fractures and veins). We cannot preclude however that
665  serpentinization and the onset of listvenite formation were contemporaneous, with a possible
666  contribution of the water released by listvenitization to the serpentinization of neighboring
667  peridotites.

668  The main differences in the distribution of alkali elements, K, Ba, Cs and Rb, along the
669  listvenite series are their selective enrichments in Listvenite domain II compared to Listvenite
670  domains I and III. These enrichments are particularly prominent in the fuchsite-bearing
671  listvenites. They are associated also to relatively low $Fe^{3+}$/Fe, overlapping Semail ophiolite
672  mantle values and evidences of iron loss in individual samples (though not at the scale of the
673  entire borehole, see Section 5.1) suggesting that fluid-rock interactions occurred in relatively
674  reduced conditions (Fig. 2). The protolith of Listvenite Domain II was lherzolitic and contained
675  higher $Al_2O_3$ and likely higher concentrations in alkali elements due to the occurrence of
676  amphibole, compared to neighboring refractory peridotites. We propose that as $CO_2$-rich fluids
677  interacted with the base of the Semail ophiolite, the chemical components necessary for the
678  precipitation of fuchsite $((K,Na)(Al,Cr,Fe)_2(Si,Al)_3O_{10}(OH,F)_3)$ were abundant only in the fertile
679  peridotite end-members. This triggered a suite of metasomatic reactions different from those
680  taking place in the refractory protoliths, in turn possibly changing the local redox conditions
681  (buffering of local $fO_2$ by Fe trapping in the fuchsite structure).

682  In contrast, the development of dolomite-dominated Ca and Sr rich intervals in Listvenite
683  Domains I and III cannot be interpreted as solely resulting from changes in the protolith
684  composition. The Ca-Sr rich intervals are located within and at the transition of (talc-bearing)
685  serpentinites and listvenites, in the fractured upper part of Listvenite Domains I (48 to 53 mbg
686  and 63 to 67 mbg) and throughout Listvenite Domain III, with Ca-Sr enrichments increasing
687  toward the basal thrust. It is noteworthy that this downhole trend continues into the M1
688  metabasalts, drawing a shape similar to a diffusive front across the basal thrust (Fig. 2). Ca-Sr
689  rich intervals also show increasing $Fe^{3+}$/Fe$_T$ values, in particular towards the basal thrust,
690  systematically associated to the enrichments in (redox sensitive) U, suggesting fluid-rock
691  interactions in an oxidized environment (e.g., [*Paulick et al.*, 2006; *Peters et al.*, 2017]). We
692  interpret these compositions as resulting from interactions with Ca-, Sr and $CO_2$-rich fluids
693  which sample a different source than the dominantly alkali-rich fluids driving listvenitization and

694 potentially earlier metasomatic processes documented along BT1B cores. Fluid pathways for
695 these Ca-Sr-rich fluids were located mainly along the main lithological interfaces sampled at
696 Hole BT1B; interactions with these fluids also overprinted the composition of the sole (M1
697 metabasalts). The change of source could be associated to a final stage of listvenitization
698 associated with circulation of meteoric-derived fluids (e.g., New Caledonia [*Ulrich et al.*, 2014])
699 or evidence of an occasional change in the composition of the dehydrating slab at depth
700 (carbonate reef capped seamounts as preserved by the Hawasina assemblages [*Lippard et al.*,
701 1986] or variable abundances of carbonate veins in altered oceanic crust [*Alt and Teagle*, 1999]).
702 We favor the later interpretation as structural and lithological descriptions show that dolomitic
703 matrix and veins are commonly crosscut by late magnesite and/or magnesite-quartz veins.

704 Selective and variable enrichments in other FME, such as Li, Pb, Sb and As or U (Figs. 2,
705 3 and 7) are observed along the core, sometimes in association to slight changes in Zn and Cu
706 compositions or in $Fe^{3+}/Fe_T$. The most prominent occur at 18-35 mbg in the upper part of
707 Listvenite domain I, and show enrichments in Li, Mn, U, Zn and Cu, and high $Fe^{3+}/Fe_T$, also
708 independently of the presence of dolomite. This zone records extensive fracturing and re-
709 cementation events, with locally the precipitation of oxides [*Menzel et al.*, 2020b]. These
710 tectonic processes likely favored the development of localized fluid pathways and, in turn, more
711 effective fluid-rock interactions and elemental redistributions along these domains.

712 These heterogeneous chemical distributions along the core indicate  that solute transport
713 was mainly advective into highly localized fluid pathways, likely in relation with the ubiquitous
714 brittle and cataclastic fracturation that characterize the listvenite series. We posit that these
715 reactive, mechanical and hydrodynamic processes were coupled at the local scale thus explaining
716 the differences in calculated mass and volume changes along the core, resulting in the multiscale
717 and high density network of crosscutting veins charactering the BT1B listvenite series [*Kelemen
718 et al.*, 2021; *Kelemen et al.*, 2020b; *Menzel et al.*, 2020b]. The complex structure and mineralogy
719 characterizing the listvenite series reflects a suite of metasomatic reactions forming overlapping
720 reaction zones and triggered by interactions with fluids derived from (at least) two sources. The
721 development of this metasomatic sequence depends on several thermodynamic (temperature,
722 pressure), hydrodynamic (permeability, solute transport…) and chemical parameters
723 (composition of the fluid and rock, redox…) that will vary with local tectonics (cooling,
724 fracturing, …) and with the progress of the reactions (e.g., negative feedbacks between
725 permeability and carbonation reactions [*Peuble et al.*, 2019; *Peuble et al.*, 2015]). In this context,
726 determining the parameters controlling the sequence of fluid-rock reactions and the associated
727 elemental redistribution is challenging. It requires an integrated petro-structural, mineralogical
728 and in situ geochemical and isotopic investigation of the cores that is beyond the scope of this
729 study.

730 5.3 Contribution of listvenites to global chemical budgets

731 Numerous studies have demonstrated the important contribution of (de-)serpentinization
732 reactions to the deep Earth geochemical cycles of water and many fluid mobile elements (e.g.,
733 [*Deschamps et al.*, 2013; *Deschamps et al.*, 2011; *Kodolanyi et al.*, 2012; *Spandler and Pirard*,
734 2013]). Recent work has also shown the potential importance for the global carbon cycle of
735 carbonation reactions affecting peridotites, from oceanic to subduction environments (e.g.,
736 [*Cannaò et al.*, 2020; *Kelemen and Manning*, 2015]). However, the possible consequences of
737 peridotite carbonation, and in particular listvenitization, for the cycles of other chemical elements

738 is still little explored despite the fairly ubiquitous occurrence of listvenites in peridotite bodies
739 and subduction-related ophiolites globally. Our study provides one of the first insights on these
740 processes.

741 The BT1B listvenites are characterized by a dual geochemical signature: they preserve
742 the composition(s) of their serpentinized protolith(s) for major elements (except for Ca) and most
743 compatible (e.g., V, Sc, Ni, Cr, Co) and incompatible lithophile elements (e.g., REE, HFSE)
744 (Fig. 6) yet, for fluid mobile elements, and in particular alkali elements (Li, K, Cs, Rb, K, Ba,
745 Sr), Pb, As, Sb, and W, they have marked enrichments, with abundance similar or even enriched
746 relative to metabasalts and metasediments such as those composing the metamorphic sole (Fig.
747 4, 6, 7). These FME enrichments are comparable to, though towards the upper end of the range
748 of, those observed in most subduction related serpentinites (e.g., [*Deschamps et al.*, 2013; *Peters*
749 *et al.*, 2017]) and ophicarbonates [*Cannao et al.*, 2020] where, as proposed for the Semail
750 ophiolite-sole transition, they are attributed to interactions with fluids derived from de-
751 volatilization reactions at depths of subducted sedimentary sequences or previously
752 metasomatized altered oceanic lithosphere (Fig. 8). These compositions however differ from that
753 of the serpentinites and ophicarbonates formed in oceanic environments (e.g., [*Cannao et al.*,
754 2020; *Noel et al.*, 2018; *Peters et al.*, 2017]), that all show significant U enrichments and, when
755 carbonation occurs, selective enrichments in Ca and Sr, and in LREE, a trend observed in the
756 BT1B variously carbonated serpentinites. The remobilization (Zn, Cu) and enrichments (As, Sb,
757 Pb) of chalcophile elements and of selected siderophile elements (variable Ni, Co and Cr, and
758 enrichments in W) is a common characteristic of listvenites that typically show ore-grade
759 compositions for these elements (e.g., [*Belogub et al.*, 2017; *Buisson and Leblanc*, 1985;
760 *Escayola et al.*, 2009; *Halls and Zhao*, 1995; *Laznicka*, 2010]). It should be noted that it is not
761 solely related to subduction environments: such variations in compositions are observed also in
762 highly serpentinized ultramafic basement of high temperature oceanic hydrothermal vents (e.g.,
763 [*Andreani et al.*, 2014]). However, rather than differences in the tectonic setting, these variable
764 behavior in FME most likely reveal local changes in the conditions at which serpentinization and
765 carbonation reactions occur. The precipitation of Ca-bearing LREE-rich hydrothermal carbonates
766 and the mobility of U are favored in alkaline (serpentinization-derived) fluids at low temperature
767 oxidizing hydrothermal conditions (e.g., [*Noel et al.*, 2018; *Paulick et al.*, 2006; *Seyfried Jr. et*
768 *al.*, 1998]). In contrast, the remobilization of chalcophile and siderophile elements is commonly
769 associated to interactions with high temperature saline hydrothermal fluids occurring over a
770 broad range of redox conditions that will favor selective enrichments in serpentine (e.g.,
771 adsorption of As and Sb at oxidizing conditions) and the development of different suites of
772 sulfides, oxides and/or metals depending on local redox (e.g., [*Andreani et al.*, 2014; *Belogub et*
773 *al.*, 2017; *Cannao et al.*, 2020; *Fouquet et al.*, 2013; *Paulick et al.*, 2006]). As previously
774 discussed, these variations in serpentinization and carbonation conditions depend first the
775 distribution of fluid pathways, the timing of their formation, mostly in relation to the localization
776 of deformation during cooling (e.g., [*Menzel et al.*, 2020b]). In turn, serpentinization and
777 carbonation reactions modify the local chemical conditions by changing the mineralogy of the
778 reacted samples and fluid composition (pH, alkalinity, ionic strength; e.g., [*Andreani et al.*, 2009;
779 *Godard et al.*, 2013; *Grozeva et al.*, 2017; *Janecky and Seyfried*, 1986; *Peuble et al.*, 2019;
780 *Peuble et al.*, 2015; *Seyfried et al.*, 2007]). As noted by Frost [1985]), they could lead in
781 particular to extreme redox gradients as oxygen fugacity is buffered by the Fe-magnesite-Fe-
782 oxide assemblage, and thus can attain values above those of the hematite-magnetite buffer in

783 magnesite-bearing listvenites compared to neighboring serpentinites, which in turn will likely
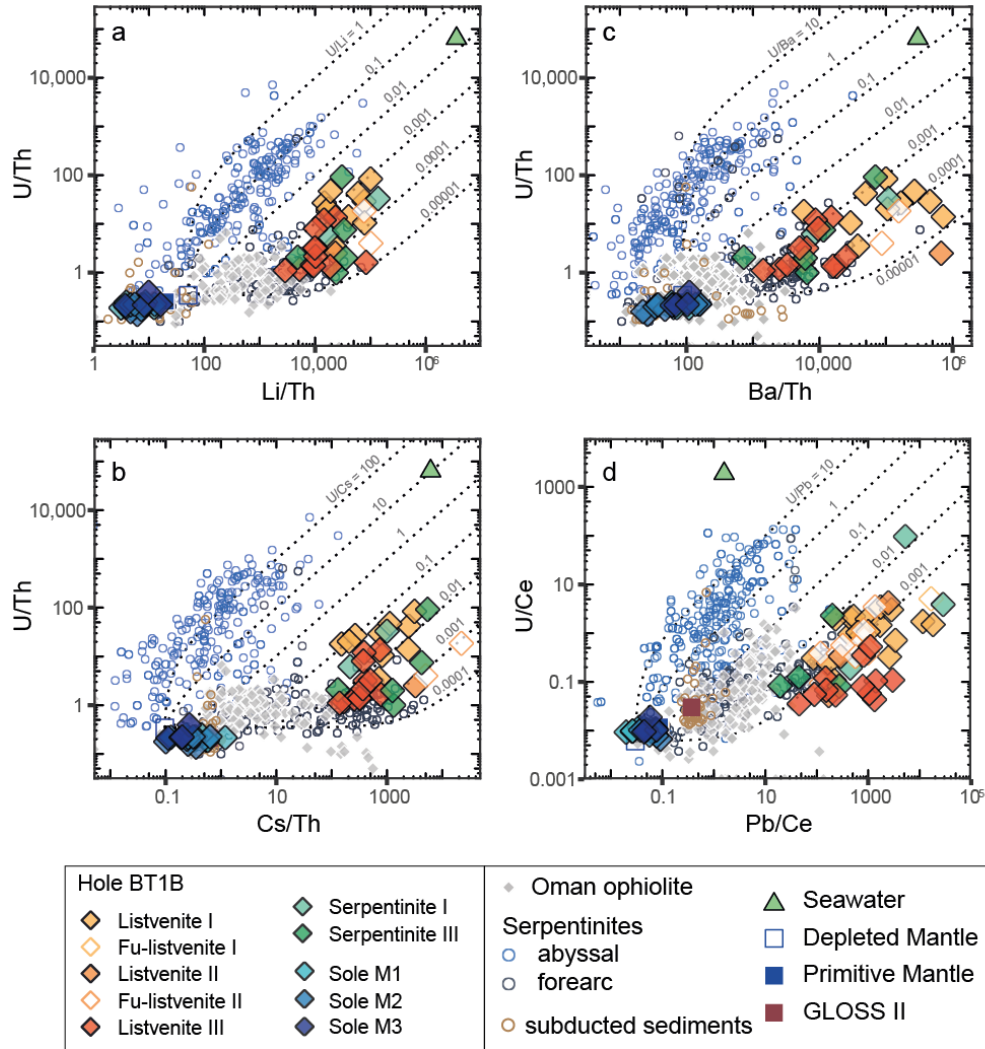784 impact the local sulfur speciation.



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**Figure 8**. Scatterplots showing fluid mobile element enrichment relative to immobile trace
elements in BT1B listvenites and serpentinites (a) U/Th versus Li/Th. (b) U/Th versus Cs/Th. (c)
U/Th versus Ba/Th. (d) U/Ce versus Pb/Ce. Plotted for comparison are: compiled serpentinite
compositions from abyssal (blue circles) and forearc (dark blue circles) settings (data from
compilations in *Peters et al.* [2017] and *Deschamps et al.* [2013]); compiled Oman ophiolite
data, sediments [*Plank et al.*, 2007], as well as the compositions of primitive mantle
[*McDonough and Sun*, 1995], depleted mantle [*Salters and Stracke*, 2004], GLOSS-II [*Plank*,
2014] and seawater [*Li*, 1991]. Also plotted are lines of addition of fluid mobile elements in
various ratios at fixed Th or Ce. Literature data sources for Oman ophiolite are as in previous
plots. Symbols are in inset.

796

797 This complex suite of thermodynamic and chemical processes coupled to fluid fluxes,
798 likely occurring far from equilibrium, explains the strong heterogeneity of geochemistry
799 observed along the BT1B core and likely the disequilibrium textures characterizing most

carbonate-quartz assemblages in BT1B listvenites (e.g., [*Beinlich et al.*, 2020]). In particular, fast growth of magnesite could efficiently trap FME in fluid inclusions (e.g., [*Cannaò and Malaspina*, 2018]) and in microphases (e.g., sulfides), thus inducing their extreme elemental enrichments when compared to previously analyzed serpentinites and ophicarbonates (e.g., [*Cannao et al.*, 2020; *Deschamps et al.*, 2013; *Peters et al.*, 2017; *Spandler and Pirard*, 2013]). The same reactive transport mechanisms likely explains the preservation of the mantle signature by listvenites. The quartz forming silicification reactions observed throughout the core are not related to interactions with incoming fluids having high silica activity, as suggested for the formation of some listvenite bodies (e.g., [*Nasir et al.*, 2007; *Ulrich et al.*, 2014]). Local silica enrichments are most likely due to its incomplete removal during coupled silicate dissolution – carbonate precipitation reactions, a process commonly observed during reactive percolation hydrothermal experiments when carbonation kinetics are fast compared to solute transport (low fluid renewal at the surface of reacting minerals and/or slow diffusivity of solutes [*Peuble et al.*, 2019; *Peuble et al.*, 2015]).

The dichotomy in the signatures of the listvenite series lead to extreme fractionation between fluid mobile elements and immobile incompatible elements (e.g., Th or Ce), in particular for alkali elements (Li, Ba, Cs, Rb) and chalcophile elements (Pb) as illustrated on Fig. 8 with Li/Th of $2000-10^5$, Ba/Th of $500-10^6$, Cs/Th of 100-5000 and Pb/Ce up to $5 \times 10^5$. Relative enrichments in U are comparatively less prominent with U/Th of 1-100 and U/Ce of 0.05-100. These values are however significantly higher than that of the possible sources of fluids such as the metamorphic sole and associated metasediments (e.g. Li/Th <30, Ba/Th<200, Cs/Th<20, U/Th<0.5, U/Ce<0.05 and Pb/Ce<0.5 for the metamorphic sole (this study) and GLOSS II [*Plank*, 2014]). These elemental fractionations follow trends similar to what observed for fore-arc serpentinites when compared to oceanic serpentinites [*Peters et al.*, 2017] but the degree of fractionation measured in the BT1B listvenite series is significantly higher. If recycled, these extreme compositions could affect that of the subduction-related volcanics (e.g., by inducing a prominent alkali-rich sedimentary signature) or, over longer time scales, the mantle isotopic signature, for instance for lead isotopes, due to their high Pb concentrations compared to U and Th (e.g., U/Pb <0.005). They could contribute to the development of a high Pb/Ce, low Th/Pb and U/Pb reservoir and be considered as a potential solution to the "first lead paradox" [*Hofmann*, 2008].

Further investigations are however required to assess the possible role of listvenitization reactions as an efficient elemental trap, and its contribution in global geochemical cycles. It should be noted that the formation of large listvenite massifs as drilled in Wadi Mansah is quite rare and that most listvenite bodies are embedded into serpentinite mélange (e.g., [*Nasir et al.*, 2007]). The change in rheology (from ductile serpentinite-talc to brittle quartz-dominated assemblages) associated to large and probably fast increases in mass and volume associated to listvenitization probably favor fracturation and mechanical erosion of the listvenite assemblages to accommodate volume changes. Such mechanisms would facilitate the entrainment downward along the subduction zone of listvenites within serpentinite mélanges.

## 6 Conclusions

During ICDP Oman Drilling Project, the transition from the base of the Semail ophiolite to the underlying metamorphic sole was drilled at Hole BT1B (Wadi Mansah). We analyzed the bulk major, volatile and trace element compositions of 65 variously carbonated peridotites

serpentinites and (fuchsite-bearing) listvenites, and 19 metamorphic rocks collected from recovered cores, with the aim to better constrain chemical transfers associated to peridotite carbonation along the ophiolite basal thrust.

The listvenite series record the formation of listvenites after a serpentinized peridotite protolith: this process is marked by $CO_2$ addition and $H_2O$ removal. Their bulk geochemistry is highly variable at the meter scale yet, on average, it is close to that of the refractory peridotites of the Semail mantle section for most major and lithophile trace elements, except of the fuchsite-bearing listvenite domain that has compositions overlapping that of the more fertile, often amphibole-bearing, basal lherzolites. This suggests preservation of the peridotite protolith geochemistry in spite of extensive peridotite carbonation. All samples are however enriched in fluid mobile elements compared to the composition of the Semail peridotites (up to ~$10^3$-$10^4$ x PM). They have concentrations similar to the metamorphic sole and/or associated metasediments for elements such as Cs, Sr and Ca and sometimes even higher for elements such as Pb, Li, As, and Sb. We also observe a decoupling between Sr-Ca enrichments and for other FME, indicating interactions with several batches of $CO_2$-rich fluids originating in neighboring lithologies or deeper along the basal thrust. These results suggest that peridotite carbonation could represent one of the major trap-and-release mechanisms for water and FME along convergent margins.

## Acknowledgments, Samples, and Data

Data will be available online on the Pangeae data archiving platform (https://www.pangaea.de/): the archiving of our data is in progress but the process is not complete. It is presently available as Supplementary Dataset S1.

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