# Enhanced Aerosol Mass in the Tropical Tropopause Layer Linked to Ozone Abundance

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#### Abstract

Aerosol particles play a critical role in the tropical tropopause layer (TTL) through cloud formation and heterogeneous chemistry, influencing the radiative and chemical balance of the stratosphere. However, aerosol measurements in the TTL are sparse, resulting in poor knowledge of aerosol abundance and distribution in this important region. Here, we present in situ aircraft measurements over the western tropical Pacific, revealing a persistent and altitude-dependent enhancement of aerosol mass in the TTL compared to the convectively influenced troposphere below. Notably, our data demonstrate a striking positive correlation between aerosol mass and ozone. Model simulations suggest that organic materials constitute a substantial fraction of the total aerosol mass within the TTL. We further derived an empirical parameterization of TTL aerosol mass as a function of ozone. Given the relative ease of ozone measurements and modeling, the parameterization provides a promising framework for estimating TTL aerosol abundance and its effects on climate.

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Abundance 2 3

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- 13 \*Correspondence author: shang.liu@northeastern.edu, rushan.gao@noaa.gov
- 14 15 **Kev Points:**
- 16 • Aircraft measurements reveal persistent enhancement of aerosol mass in the TTL.
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- The TTL aerosol enhancement tightly correlates with ozone. An empirical parameterization of TTL aerosol as a function of ozone is derived. 18
  - Modeling suggests that TTL aerosol particles are mainly composed of organics and sulfate.
- 20
- 21 Abstract
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23 Aerosol particles play a critical role in the tropical tropopause layer (TTL) through cloud formation and heterogeneous chemistry, influencing the radiative and chemical balance of the stratosphere. 24 However, aerosol measurements in the TTL are sparse, resulting in poor knowledge of aerosol 25 abundance and distribution in this important region. Here, we present in situ aircraft measurements 26 over the western tropical Pacific, revealing a persistent and altitude-dependent enhancement of 27 aerosol mass in the TTL compared to the convectively influenced troposphere below. Notably, our 28 29 data demonstrate a striking positive correlation between aerosol mass and ozone. Model 30 simulations suggest that organic materials constitute a substantial fraction of the total aerosol mass 31 within the TTL. We further derived an empirical parameterization of TTL aerosol mass as a 32 function of ozone. Given the relative ease of ozone measurements and modeling, the 33 parameterization provides a promising framework for estimating TTL aerosol abundance and its effects on climate. 34

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#### 36 **Plain Language Summary**

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38 We investigated tiny particles called aerosols in a specific atmospheric layer called the tropical tropopause layer (TTL). These particles are crucial because they affect cloud formation and 39 chemical processes in the atmosphere, influencing how energy is distributed. Unfortunately, there 40 hasn't been much research on aerosols in the TTL, leading to gaps in our understanding of their 41 abundance and distribution in this important region. To fill this knowledge gap, we conducted 42 measurements using aircraft over the western tropical Pacific. Our findings revealed that aerosol 43

44 mass in the TTL is consistently higher compared to the lower troposphere, which is influenced by upward air movement. What's interesting is that we observed a clear connection between the 45 46 amount of aerosol and ozone. Our model simulations indicated that a significant portion of the aerosol mass in the TTL is made up of organic materials. To make it easier to estimate aerosol 47 levels and their impact on climate, we developed a way to predict TTL aerosol mass based on 48 ozone measurements. Since ozone is relatively straightforward to measure and model, our method 49 could provide a useful framework for understanding aerosol abundance in the TTL and its effects 50 on the climate. 51

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#### 53 **1 Introduction**

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As the main pathway for the transport of tropospheric air into the stratosphere, the TTL largely determines the entry values for the materials entering the stratosphere (Fueglistaler et al., 2009). The properties of the TTL air and the processes occurring in the TTL thus affect the global stratosphere and climate (Bondel and Jansen 2012; SDADC 2006).

- stratosphere and climate (Randel and Jensen, 2013; SPARC 2006).
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Aerosol particles in the TTL affect the stratospheric water vapor budget through TTL dehydration 60 processes by serving as nuclei for the formation of cirrus clouds (Penner et al., 2009). TTL circus 61 clouds have substantial impacts on the earth's radiative balance (Hong et al., 2016). By providing 62 condensed surface areas, TTL aerosol can also facilitate condensation of low vapor pressure gases 63 such as sulfuric acid (Brock et al., 1995) and promote heterogeneous chemistry that depletes ozone 64 once they are transported to the stratosphere (Tolbert et al., 1988). Despite the importance, the 65 abundance and properties of the TTL aerosol remain poorly characterized. Field observations have 66 suggested that new particle formation events frequently occur in the TTL, in particular the lower 67 TTL just below the tropopause (Brock et al., 1995; Weigel et al., 2011, 2021). The composition of 68 the TTL aerosol is influenced by tropical dynamics (e.g., transport) and regional continental air 69 sources (Froyd et al., 2009). Recent advancements in SO<sub>2</sub> measurement in the TTL suggests little 70 contribution of SO<sub>2</sub> to stratospheric aerosols, revealing a significant gap in the stratospheric 71 72 aerosol budget (Rollins et al., 2017). The complex dynamic and chemical processes in the TTL 73 make it difficult to elucidate the formation mechanisms of the TTL aerosol. In-situ measurements 74 of TTL aerosol are limited, hindering our understanding on the climate impacts of the TTL and 75 stratospheric aerosols.

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77 For this study, we carried out aircraft measurements of aerosols over the western Pacific warm 78 pool during the Pacific Oxidants, Sulfur, Ice, Dehydration, and cONvection (POSIDON) campaign 79 in October 2016. The western Pacific warm pool plays a leading role for transport of air into the 80 TTL (Fueglistaler et al., 2005). Our measurements were carried out during nine flights aboard a 81 NASA WB-57F high-altitude aircraft stationed in Guam. Throughout these flights, we extensively characterized TTL aerosols and trace gases (flight tracks are shown in Fig. 1). To complement the 82 measurements, we employed modeling techniques to obtain insights into the chemical composition 83 84 of the aerosols. The combined approach of aerosol measurements, tracer analysis, and modeling offers insight into aerosol abundance and formation mechanisms within the TTL. 85

#### 87 **2 Methods**

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#### 89 2.1 Measurements

The POSIDON campaign, took place during October 2016, aimed at improving our understanding 90 of the physical and chemical processes occurring in the TTL. A total of nine flights were carried 91 out on board the NASA WB-57 high altitude aircraft from Guam (13.5° N, 144.8° E). These flights 92 covered the region from 0 to 15°N and from 130 to 160° E (Fig. 1a), with vertical coverage 93 spanning 0 to19 km (Fig. 1b). Each flight path consisted of several upward and downward 94 segments between 14 and 18 kilometers in altitude, providing extensive sampling of the TTL. The 95 measurement region lies over the tropical warm pool, where high sea surface temperatures lead to 96 97 widespread deep atmospheric convection (Yan et al., 1992).

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The aerosol particles were sampled through a near-isokinetic inlet that reduces the ambient sample 102 speed from the aircraft speed of  $\sim 90-200$  m s<sup>-1</sup> to  $\sim 3$  m s<sup>-1</sup> while maintaining ambient aerosol 103 mixing ratios (Jonsson et al., 1995). This inlet has been employed in various aircraft campaigns 104 105 for aerosol sampling (e.g. Schwarz et al., 2006). Aerosol number size distribution was measured in situ by a custom-built optical particle counter, the portable optical particle spectrometer (POPS), 106 which was mounted in the fuselage bay of the aircraft. The POPS uses a 405 nm laser to count and 107 size individual aerosol particles with diameters from 140 to 3000 nm (Gao et al., 2016). The 108 scientific application of the POPS has been demonstrated by recent field campaigns (Cui et al., 109 2018; Liu et al., 2021; Yu et al., 2017). In the POPS instrument, each particle passing across the 110 laser beam produces a pulse by scattering the laser light. The particle number is determined by the 111 number of the pulses. The particle size is calculated from the intensity of the pulse, which was 112 calibrated using a series of differential mobility analyzer (DMA) size-selected dioctyl sebacate 113 (DOS) particles prior to the campaign. The aerosol mass was calculated from the aerosol number 114 115 and size assuming that the particles are spherical and have a constant density of 1.6 g cm<sup>-3</sup>. The mass mixing ratio (MMR) of aerosols was determined as the ratio of aerosol mass to the density

- 117 of ambient air, which was calculated from the measured air pressure and temperature. A lognormal
- 118 fit to the measured mass size distribution derived from the POPS measurements in the TTL (Fig.
- 119 S1) suggests that the POPS measurements captured approximately 50% of the total aerosol mass.
- 120 Laboratory tests suggest that the POPS instrument used during POSIDON can provide reliable
- measurements under pressures as low as 70 hPa, which corresponds to an altitude of 18.9 km in
- this study. Therefore the POPS data acquired above 18.9 km are not used in the analysis.
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The size distribution of particles with diameter larger than 3  $\mu$ m was measured with a fast cloud droplet probe (FCDP). The FCDP detects particle forward scattering to determine the number and size of particles (Lance et al., 2010; McFarquhar et al., 2007). We assume that all particles greater than 3  $\mu$ m are ice crystals as the number of large aerosol particles in the upper troposphere is negligible (Jensen et al., 2013). We found that ice crystals could likely abrade the inlet materials, resulting in artifacts in POPS-measured aerosol size distribution. Such interference has been observed in previous aircraft measurements (Murphy et al., 2004). For this reason, we excluded

- the POPS data when ice crystals were present from the analysis.
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Additional real-time measurements included: (i) ozone mixing ratio obtained using a custom UV spectrophotometer designed for high altitude airborne deployment with high accuracy and precision (Gao et al., 2012), (ii) water vapor mixing ratio measured by a two-channel tunable diode laser-based hygrometer, which is capable of accurately measuring low-concentration (below 1 ppm) water vapor in the upper troposphere and lower stratosphere (Thornberry et al., 2015), and (iii) ambient air pressure and temperature measured by the Meteorological Measurement System (MMS; Chan et al., 1989; Scott et al., 1990), which also records aircraft position with 1-second

- time resolution.
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# 142 **2.2 Modeling**

We employed the Community Aerosol and Radiation Model for Atmospheres (CARMA), an advanced sectional aerosol model (Toon et al., 1988; Yu et al., 2015). CARMA is coupled with the Community Earth System Model (CESM), allowing for comprehensive analysis of aerosols. The model operates at a spatial resolution of  $1.9^{\circ} \times 2.5^{\circ}$  and employs a time step of 30 minutes.

- 147 The model incorporates 35 vertical pressure levels, spanning from the Earth's surface up to 200
- hPa, and an additional 21 vertical pressure levels from 200 hPa to 2 hPa. Simulations are nudged
- to meteorology from Modern-Era Retrospective analysis for Research and Applications, Version
   2 (MERRA-2; Gelaro et al., 2017).
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152 CARMA tracks two groups of aerosols, with each group containing 20 size bins. The first group 153 consists of pure sulfate particles with aerosol diameter ranging from 0.4 nm to 2.6 μm. These 154 particles form through nucleation and condensation of water and sulfuric acid vapor (Zhao and 155 Turco, 1995). The second group comprises internally mixed aerosols with the diameters varying 156 from 100 nm to 17 μm. These mixed aerosols consist of particles that contain organic compounds,

- 157 black carbon (BC), sea salt, dust, and condensed sulfate.
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- 159 In addition, we employed a straightforward, observationally constrained chemical model to
- 160 characterize the vertical distribution of  $O_3$  within the TTL. This model is a one-dimensional
- 161 column model that allows updrafts and vertical mixing but assumes no horizontal mixing. The  $O_3$
- 162 formation process is represented using the Chapman mechanism, and O<sub>3</sub> profiles were calculated
- 163 for the altitude range of 14.5–18.9 km. A detailed description of the model is provided in the
- 164 Supporting Information.
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#### 166 **3 Results and discussion**

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## 168 **3.1 Determination of TTL**

We adopt the upper boundary of the TTL to be at 19 km, following SPARC (2006). This altitude is 2.5 km higher than the cold point tropopause (CPT) as shown in Fig. S2. The CPT corresponds to an altitude of 17.5 km and air temperature of approximately 190 K. These values are in line with previous measurements of TTL (Fueglistaler et al., 2009; Gettelman et al., 2004).

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174 The lower boundary of the TTL is determined by examining the vertical profile of potential temperature ( $\theta$ ; Fig. S3). The curvature of the altitude- $\theta$  profile changes at 14.5 km, which is 175 mathematically characterized as the lapse rate minimum (LRM) of  $\theta$ . This change reflects the 176 transition of stability regimes, i.e., deep convection dominates air stability below the LRM and 177 178 radiation starts to influence air temperature above the LRM. Accordingly, we identify the LRM level at 14.5 km (355 K, 140 hPa) as the base of the TTL (Gettelman and Forster, 2002), which is 179 consistent with previous studies (Fueglistaler et al., 2009; Sunilkumar et al., 2017). As a result, the 180 TTL spans from 14.5 km to 20 km, and the majority of our measurements were conducted within 181 the TTL (Fig. 1). 182

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# 184 **3.2 Vertical profile of aerosols and ozone**

The vertical profile of aerosol mass from the surface to the top of the TTL can be segmented in 185 three layers with distinct characteristics (Fig. 2a). Layer I spans from the surface to 5 km altitude 186 and represents the lower troposphere, in which the MMR of aerosols decreased logarithmically 187 with altitude from 2  $\mu$ g kg<sup>-1</sup> to 0.08  $\mu$ g kg<sup>-1</sup> likely because of the influence of surface emissions 188 from Guam. Layer II ranges from 5 km to 14.5 km altitude. In this layer, the aerosol MMR 189 remained at approximately 0.08 µg kg<sup>-1</sup> with small variability. Such low-concentration aerosol 190 191 layers immediately below the convection outflow have been observed previously over the Northern Indian Ocean (de Reus et al., 2001) and the rain forests in South America (Andreae et 192 193 al., 2018; Krejci et al., 2003). The observation indicates that deep convection serves as an effective 194 sink for aerosol particles (Yu et al., 2019). Layer III lies in the TTL, extending from 14.5 km to 18.9 km, with the upper boundary 1.4 km higher than the CPT. In this layer the aerosol MMR 195 increased rapidly from 0.08  $\mu$ g kg<sup>-1</sup> at 14.5 km to 1.5  $\mu$ g kg<sup>-1</sup> at 18.9 km. Our observations show 196 a sustained increase of aerosol MMR from the upper troposphere to lower stratosphere across the 197 tropopause, rather than abrupt transitions. This suggests that the CPT has no unique role for the 198 transport of aerosols. Accounting for the mass of <140 nm particles that were not measured by the 199 POPS instrument, the aerosol MMR at 18.9 km would be approximately 3 µg kg<sup>-1</sup>. This 200 concentration is close to the measurement during the POLARIS mission in 1997 (Mclinden et al., 201

1999), in which an aerosol MMR of approximately  $3.5 \,\mu g \, kg^{-1}$  was observed for aerosols with size range of 0.07–1  $\mu$ m at 18.9 km in eastern Pacific (17.5° N, 159.3° W).

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The ozone concentration remained below 35 ppb in layers I and II and began to increase at 14.5 km (Fig. 2b). This observation is consistent with previous ozonesonde measurements in the tropical Pacific (Folkins et al., 1999; Folkins and Martin, 2005). The concurrence of ozone minimum and LRM is in line with the analysis by Gettelman and Forster (2002) and Folkins et al. (2002).

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**Figure 2**. Potential temperature ( $\theta$ ) versus (**a**) O<sub>3</sub> and (**b**) MMR for data acquired from 0–5 km, 5–14.5 km, and 14.5–18.9 km measurements for all flights. Individual data points (5-s averages) are shown by the dots and the averages (by  $\theta$  of 2K) are shown by the solid squares.

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#### **3.3 Mechanisms for aerosol enhancement in the TTL**

We find that the aerosol MMR was tightly correlated with  $O_3$  in the TTL during all flights with small variation between flights. This can be clearly seen in the example time series of O3 and aerosol MMR for the flight on October 18 shown in Fig. 3a. The Pearson's correlation coefficient (*r*) for the campaign-average  $O_3$  and aerosol MMR (averaged into 2-K  $\theta$  intervals) was 0.98 in layer III. In contrast, the aerosol MMR and  $O_3$  were anticorrelated with an *r* value of -0.35 in layer I (Fig. 3b), and no correlation was observed between aerosol MMR and  $O_3$  in layer II.

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224 The main processes contributing to the increase in TTL  $O_3$  from the bottom to the top of the TTL 225 include chemical production via photolytic dissociation of molecular oxygen (O<sub>2</sub>) (Prather 2009; 226 Crutzen et al., 1999) and isentropic in-mixing of stratospheric air from the extratropical lower 227 stratosphere (Konopka et al., 2010; Ploeger et al., 2012). The relative contribution of these 228 processes to  $O_3$  remains unclear. While some studies argue that in situ chemical production dominates (Avallone and Prather 1996), others suggest that isentropic stratospheric in-mixing can 229 contribute to O<sub>3</sub> by as much as 40-60% (Abalos et al., 2013a, 2013b; Konopka et al., 2009, 2010; 230 Ploeger et al., 2011, 2012; Sargent et al., 2014). This contribution is most significant during the 231 boreal summer months (June-August), primarily driven by the Asian summer monsoon, and 232

233 gradually decreases during transition into the fall and winter months. In particular, these studies

suggest that the contribution from in-mixing falls within the range of 0-20% in October, averaged over a  $\pm 10^{\circ}$ N latitude range, with the contribution increasing with altitude.

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Figure 3. (a) Example time series of aerosol MMR and O<sub>3</sub> for the October 18 measurements. The gaps indicate ice-influenced data that were excluded from the analysis. (b) Scatter plot of MMR vs O<sub>3</sub> for the 0– 5 km (layer I), 5–14.5 km (layer II), and 14.5–18.9 km (layer III) measurements. Individual data points are shown by the dots and the averages are shown by the solid squares Logarithmic scales are used to make the small values visible.

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244 We used a one-dimensional column model (over the flight region) and tracer analysis to examine these processes. The model calculates the vertical distribution of O<sub>3</sub> within the TTL, considering 245 a constant slow vertical ascent rate of 0.25 mm s<sup>-1</sup> (Park et al., 2010; Avallone and Prather 1996) 246 of air in the TTL and assuming it evolves in isolation (Supporting Information). Remarkably, the 247 calculated vertical profile of O<sub>3</sub> closely aligns with the observed O<sub>3</sub> profile (Fig. 4a). This result is 248 in line with earlier investigations that yielded similar results using a column model (Avallone and 249 250 Prather 1996) and with research that assumed the tropics are isolated from extra-tropics when 251 explaining the annual cycles of ozone above the tropical tropopause (Randel et al., 2007; Schoeberl et al., 2008). However, we acknowledge the simplicity of the model. For example, the model does 252 not consider the chemistry of nitrogen oxides (NO and NO<sub>2</sub>; not measured during POSIDON) and 253 O<sub>3</sub>, which could be important in the TTL O<sub>3</sub> abundance (Nussbaumer et al., 2023), and 254 uncertainties exist in the assumed ascent rate. On the other hand, the tracer analysis shows evidence 255 of stratospheric in-mixing. Fig. S4 shows that N<sub>2</sub>O was anticorrelated with O<sub>3</sub> in the TTL during 256 POSIDON. The observed negative correlation indicates that the sampled TTL air included a 257 258 contribution from mixing of stratospheric origin (Folkins et al., 1999). This is because the strong UV radiation at higher altitudes in the stratosphere photolyzes N<sub>2</sub>O while leading to O<sub>3</sub> production, 259 resulting in an anticorrelation of N<sub>2</sub>O and O<sub>3</sub> (Assonov et al., 2013). In contrast, N<sub>2</sub>O in the 260 troposphere is inert and uniformly distributed, thus correlation of N<sub>2</sub>O with O<sub>3</sub> is not expected and 261 not observed below the TTL during our measurements. The synthesis of modeling and tracer 262 analysis indicates contributions from both chemical production and stratospheric in-mixing to TTL 263

 $O_3$ . Given the strong correlation between  $O_3$  and aerosols in the TTL, these results suggest that TTL aerosols may also come from a combination of these chemical and physical processes.



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Figure 4. Comparison of modeling results to measurements for (a)  $O_3$  and (b) aerosol MMR. Individual measurement points are shown by the grey dots. The median values and the 25<sup>th</sup> and 75<sup>th</sup> percentiles are shown by the blue line-square symbols and the error bars. The y-axis represents altitude relative to the tropopause.

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273 To investigate the chemical composition of TTL aerosol, we employed the CARMA model embedded in CESM to simulate the aerosol formation and growth processes. The model 274 reproduced the O<sub>3</sub> profile reasonably well (Fig. 4a). Our model results suggest that sulfate aerosol 275 exhibits a consistent increase with altitude, whereas organic aerosol displays an ascending trend 276 below the tropopause but declines above it. Moreover, our modeled vertical profile of the total 277 aerosol MMR demonstrates good agreement with the in situ measurements (Fig. 4b) below the 278 tropopause, with organic aerosol constituting the significant fraction of the aerosol mass. These 279 modeling results imply that sulfur alone is insufficient to explain the observed aerosol mass, and 280 organic precursors may play a pivotal role in the formation and/or growth of TTL aerosols under 281 low-temperature conditions. This finding aligns with previous findings from TTL measurements 282 during the Pre-AVE and CR-AVE campaigns over Southwest Central America in boreal winter 283 (Froyd et al., 2009), which highlighted the prevalence of organic-sulfate particles as the most 284 abundant particle type in the TTL and lower stratosphere. We note that the model underestimates 285 the aerosol mass above the tropopause, with the difference increasing with altitude within the TTL, 286

reaching 60% at an altitude of 1 km above the CPT. Additional research is needed to understandthis discrepancy.

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290 We postulate that TTL aerosols are likely generated through the processes of new particle formation and subsequent growth following convective outflow in the upper troposphere 291 (Williamson et al., 2019). The observed aerosol number size distribution in the TTL suggests 292 particle growth with increasing altitude (Fig. S5), although obtaining measurements of smaller 293 particles would enhance our understanding. During convective transport, soluble species are 294 effectively removed, whereas the insoluble and weakly soluble species can endure washout and 295 gradually ascend upward, with minimal loss, to reach the stratosphere (Bechara et al., 2010). The 296 TTL, characterized by low temperature, low particle surface area density, and high relative 297 humidity, provides ideal conditions for new particle formation and growth. Multiple studies have 298 299 proposed the upper troposphere as the primary nucleation region (Brock et al., 1995; Weigel et al., 2011), and investigations have revealed the involvement of organics in new particle formation and 300 initial growth in the remote tropical upper troposphere (Kupc et al., 2020). Future measurements 301 analyzing the composition of oxidized organic species will enhance our understanding of particle 302 formation and growth pathways. Exceptional overshooting that crosses the tropical tropopause 303 could also affect the aerosol abundance in the TTL (Vernier et al., 2011). This possibility is 304 examined using  $H_2O$  as a tracer. The  $\theta$ - $H_2O$  profile displayed no spikes above the tropopause (Fig. 305 306 S6), suggesting the absence of overshooting during the measurements. 307

- The prevalence of secondary aerosol formation as the primary factor influencing TTL aerosol 308 abundance is in line with the strong correlation between TTL aerosol mass and O<sub>3</sub> in our 309 measurements. Building upon this observation and the underlying mechanisms, we further derived 310 an empirical parameterization of aerosol MMR as a function of O<sub>3</sub> in the TTL using linear 311 regression. The derived relationship is expressed as MMR ( $\mu g kg^{-1}$ ) = 0.0074(±2.2x10<sup>-4</sup>) × O<sub>3</sub> (ppb) 312 + 0.23 (±0.049), accounting for the 50% mass that was not measured by the POPS instrument. 313 This parameterization could be used to estimate TTL aerosol abundance and for validation of 314 modeled results. 315
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#### 317 **4** Conclusions

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Our aircraft measurements conducted over the western Pacific revealed a vertically stratified 319 aerosol distribution comprised of three distinct layers. The lower-troposphere layer, ranging from 320 321 0 to 5 km in altitude, exhibited a decrease in aerosol mass with height, primarily influenced by 322 surface-level emissions. The layer between 5 and 14.5 km displayed consistently low aerosol 323 concentrations with minimal variability, indicating effective aerosol removal through deep 324 convection processes. In the TTL, we observed an enhanced aerosol mass that exhibited a strong 325 correlation with O<sub>3</sub>. The modeling and tracer analysis suggest that TTL O<sub>3</sub> and aerosols likely originate from a combination of chemical production and stratospheric in-mixing processes. 326 327 Furthermore, based on the linear relationship observed between aerosol MMR and O<sub>3</sub> in the TTL, 328 we derived an empirical parameterization that allows for the estimation of aerosol MMR as a function of  $O_3$ . This parameterization holds potential for validating the simulated TTL aerosol abundance in global models, enabling the simulation of the climate impacts of TTL aerosols.

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332 Our study also highlights the challenge of accurately predicting aerosol mass, particularly above the tropopause, even for the most advanced models. This underscores the existence of significant 333 334 gaps in our understanding of the origin and formation mechanisms of aerosols in the TTL. To achieve a more comprehensive understanding of TTL aerosol, it is crucial to conduct future 335 measurements that encompass detailed aerosol chemical composition and gas-phase precursors 336 across different locations and seasons. These endeavors will help diagnose potential deficiencies 337 in the models and validate the modeling results. Consequently, they will lead to an improved 338 comprehension of TTL aerosols and facilitate a predictive understanding of their effects on 339 stratospheric chemistry, clouds, and climate. 340

341

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- 345 providing the MMS data.
- 346

## 347 Data Availability Statement

- The aircraft measurements during the POSIDON field campaign are available through the NASA
   ESPO Data Archive: https://espoarchive.nasa.gov/archive/browse/posidon/WB57
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#### Enhanced Aerosol Mass in the Tropical Tropopause Layer Linked to Ozone 1

Abundance 2 3

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- 13 \*Correspondence author: shang.liu@northeastern.edu, rushan.gao@noaa.gov
- 14 15 **Kev Points:**
- 16 • Aircraft measurements reveal persistent enhancement of aerosol mass in the TTL.
- 17
- The TTL aerosol enhancement tightly correlates with ozone. An empirical parameterization of TTL aerosol as a function of ozone is derived. 18
  - Modeling suggests that TTL aerosol particles are mainly composed of organics and sulfate.
- 20
- 21 Abstract
- 22

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23 Aerosol particles play a critical role in the tropical tropopause layer (TTL) through cloud formation and heterogeneous chemistry, influencing the radiative and chemical balance of the stratosphere. 24 However, aerosol measurements in the TTL are sparse, resulting in poor knowledge of aerosol 25 abundance and distribution in this important region. Here, we present in situ aircraft measurements 26 over the western tropical Pacific, revealing a persistent and altitude-dependent enhancement of 27 aerosol mass in the TTL compared to the convectively influenced troposphere below. Notably, our 28 29 data demonstrate a striking positive correlation between aerosol mass and ozone. Model 30 simulations suggest that organic materials constitute a substantial fraction of the total aerosol mass 31 within the TTL. We further derived an empirical parameterization of TTL aerosol mass as a 32 function of ozone. Given the relative ease of ozone measurements and modeling, the 33 parameterization provides a promising framework for estimating TTL aerosol abundance and its effects on climate. 34

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#### 36 **Plain Language Summary**

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38 We investigated tiny particles called aerosols in a specific atmospheric layer called the tropical tropopause layer (TTL). These particles are crucial because they affect cloud formation and 39 chemical processes in the atmosphere, influencing how energy is distributed. Unfortunately, there 40 hasn't been much research on aerosols in the TTL, leading to gaps in our understanding of their 41 abundance and distribution in this important region. To fill this knowledge gap, we conducted 42 measurements using aircraft over the western tropical Pacific. Our findings revealed that aerosol 43

44 mass in the TTL is consistently higher compared to the lower troposphere, which is influenced by upward air movement. What's interesting is that we observed a clear connection between the 45 46 amount of aerosol and ozone. Our model simulations indicated that a significant portion of the aerosol mass in the TTL is made up of organic materials. To make it easier to estimate aerosol 47 levels and their impact on climate, we developed a way to predict TTL aerosol mass based on 48 ozone measurements. Since ozone is relatively straightforward to measure and model, our method 49 could provide a useful framework for understanding aerosol abundance in the TTL and its effects 50 on the climate. 51

52

#### 53 **1 Introduction**

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As the main pathway for the transport of tropospheric air into the stratosphere, the TTL largely determines the entry values for the materials entering the stratosphere (Fueglistaler et al., 2009). The properties of the TTL air and the processes occurring in the TTL thus affect the global stratosphere and climate (Bondel and Jansen 2012; SDADC 2006).

- stratosphere and climate (Randel and Jensen, 2013; SPARC 2006).
- 59

Aerosol particles in the TTL affect the stratospheric water vapor budget through TTL dehydration 60 processes by serving as nuclei for the formation of cirrus clouds (Penner et al., 2009). TTL circus 61 clouds have substantial impacts on the earth's radiative balance (Hong et al., 2016). By providing 62 condensed surface areas, TTL aerosol can also facilitate condensation of low vapor pressure gases 63 such as sulfuric acid (Brock et al., 1995) and promote heterogeneous chemistry that depletes ozone 64 once they are transported to the stratosphere (Tolbert et al., 1988). Despite the importance, the 65 abundance and properties of the TTL aerosol remain poorly characterized. Field observations have 66 suggested that new particle formation events frequently occur in the TTL, in particular the lower 67 TTL just below the tropopause (Brock et al., 1995; Weigel et al., 2011, 2021). The composition of 68 the TTL aerosol is influenced by tropical dynamics (e.g., transport) and regional continental air 69 sources (Froyd et al., 2009). Recent advancements in SO<sub>2</sub> measurement in the TTL suggests little 70 contribution of SO<sub>2</sub> to stratospheric aerosols, revealing a significant gap in the stratospheric 71 72 aerosol budget (Rollins et al., 2017). The complex dynamic and chemical processes in the TTL 73 make it difficult to elucidate the formation mechanisms of the TTL aerosol. In-situ measurements 74 of TTL aerosol are limited, hindering our understanding on the climate impacts of the TTL and 75 stratospheric aerosols.

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77 For this study, we carried out aircraft measurements of aerosols over the western Pacific warm 78 pool during the Pacific Oxidants, Sulfur, Ice, Dehydration, and cONvection (POSIDON) campaign 79 in October 2016. The western Pacific warm pool plays a leading role for transport of air into the 80 TTL (Fueglistaler et al., 2005). Our measurements were carried out during nine flights aboard a 81 NASA WB-57F high-altitude aircraft stationed in Guam. Throughout these flights, we extensively characterized TTL aerosols and trace gases (flight tracks are shown in Fig. 1). To complement the 82 measurements, we employed modeling techniques to obtain insights into the chemical composition 83 84 of the aerosols. The combined approach of aerosol measurements, tracer analysis, and modeling offers insight into aerosol abundance and formation mechanisms within the TTL. 85

#### 87 **2 Methods**

88

#### 89 2.1 Measurements

The POSIDON campaign, took place during October 2016, aimed at improving our understanding 90 of the physical and chemical processes occurring in the TTL. A total of nine flights were carried 91 out on board the NASA WB-57 high altitude aircraft from Guam (13.5° N, 144.8° E). These flights 92 covered the region from 0 to 15°N and from 130 to 160° E (Fig. 1a), with vertical coverage 93 spanning 0 to19 km (Fig. 1b). Each flight path consisted of several upward and downward 94 segments between 14 and 18 kilometers in altitude, providing extensive sampling of the TTL. The 95 measurement region lies over the tropical warm pool, where high sea surface temperatures lead to 96 97 widespread deep atmospheric convection (Yan et al., 1992).

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The aerosol particles were sampled through a near-isokinetic inlet that reduces the ambient sample 102 speed from the aircraft speed of  $\sim 90-200$  m s<sup>-1</sup> to  $\sim 3$  m s<sup>-1</sup> while maintaining ambient aerosol 103 mixing ratios (Jonsson et al., 1995). This inlet has been employed in various aircraft campaigns 104 105 for aerosol sampling (e.g. Schwarz et al., 2006). Aerosol number size distribution was measured in situ by a custom-built optical particle counter, the portable optical particle spectrometer (POPS), 106 which was mounted in the fuselage bay of the aircraft. The POPS uses a 405 nm laser to count and 107 size individual aerosol particles with diameters from 140 to 3000 nm (Gao et al., 2016). The 108 scientific application of the POPS has been demonstrated by recent field campaigns (Cui et al., 109 2018; Liu et al., 2021; Yu et al., 2017). In the POPS instrument, each particle passing across the 110 laser beam produces a pulse by scattering the laser light. The particle number is determined by the 111 number of the pulses. The particle size is calculated from the intensity of the pulse, which was 112 calibrated using a series of differential mobility analyzer (DMA) size-selected dioctyl sebacate 113 (DOS) particles prior to the campaign. The aerosol mass was calculated from the aerosol number 114 115 and size assuming that the particles are spherical and have a constant density of 1.6 g cm<sup>-3</sup>. The mass mixing ratio (MMR) of aerosols was determined as the ratio of aerosol mass to the density

- 117 of ambient air, which was calculated from the measured air pressure and temperature. A lognormal
- 118 fit to the measured mass size distribution derived from the POPS measurements in the TTL (Fig.
- 119 S1) suggests that the POPS measurements captured approximately 50% of the total aerosol mass.
- 120 Laboratory tests suggest that the POPS instrument used during POSIDON can provide reliable
- measurements under pressures as low as 70 hPa, which corresponds to an altitude of 18.9 km in
- this study. Therefore the POPS data acquired above 18.9 km are not used in the analysis.
- 123

The size distribution of particles with diameter larger than 3  $\mu$ m was measured with a fast cloud droplet probe (FCDP). The FCDP detects particle forward scattering to determine the number and size of particles (Lance et al., 2010; McFarquhar et al., 2007). We assume that all particles greater than 3  $\mu$ m are ice crystals as the number of large aerosol particles in the upper troposphere is negligible (Jensen et al., 2013). We found that ice crystals could likely abrade the inlet materials, resulting in artifacts in POPS-measured aerosol size distribution. Such interference has been observed in previous aircraft measurements (Murphy et al., 2004). For this reason, we excluded

- the POPS data when ice crystals were present from the analysis.
- 132

Additional real-time measurements included: (i) ozone mixing ratio obtained using a custom UV spectrophotometer designed for high altitude airborne deployment with high accuracy and precision (Gao et al., 2012), (ii) water vapor mixing ratio measured by a two-channel tunable diode laser-based hygrometer, which is capable of accurately measuring low-concentration (below 1 ppm) water vapor in the upper troposphere and lower stratosphere (Thornberry et al., 2015), and (iii) ambient air pressure and temperature measured by the Meteorological Measurement System (MMS; Chan et al., 1989; Scott et al., 1990), which also records aircraft position with 1-second

- time resolution.
- 141

# 142 **2.2 Modeling**

We employed the Community Aerosol and Radiation Model for Atmospheres (CARMA), an advanced sectional aerosol model (Toon et al., 1988; Yu et al., 2015). CARMA is coupled with the Community Earth System Model (CESM), allowing for comprehensive analysis of aerosols. The model operates at a spatial resolution of  $1.9^{\circ} \times 2.5^{\circ}$  and employs a time step of 30 minutes.

- 147 The model incorporates 35 vertical pressure levels, spanning from the Earth's surface up to 200
- hPa, and an additional 21 vertical pressure levels from 200 hPa to 2 hPa. Simulations are nudged
- to meteorology from Modern-Era Retrospective analysis for Research and Applications, Version
   2 (MERRA-2; Gelaro et al., 2017).
- 151

152 CARMA tracks two groups of aerosols, with each group containing 20 size bins. The first group 153 consists of pure sulfate particles with aerosol diameter ranging from 0.4 nm to 2.6 μm. These 154 particles form through nucleation and condensation of water and sulfuric acid vapor (Zhao and 155 Turco, 1995). The second group comprises internally mixed aerosols with the diameters varying 156 from 100 nm to 17 μm. These mixed aerosols consist of particles that contain organic compounds,

- 157 black carbon (BC), sea salt, dust, and condensed sulfate.
- 158

- 159 In addition, we employed a straightforward, observationally constrained chemical model to
- 160 characterize the vertical distribution of  $O_3$  within the TTL. This model is a one-dimensional
- 161 column model that allows updrafts and vertical mixing but assumes no horizontal mixing. The  $O_3$
- 162 formation process is represented using the Chapman mechanism, and O<sub>3</sub> profiles were calculated
- 163 for the altitude range of 14.5–18.9 km. A detailed description of the model is provided in the
- 164 Supporting Information.
- 165

#### 166 **3 Results and discussion**

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## 168 **3.1 Determination of TTL**

We adopt the upper boundary of the TTL to be at 19 km, following SPARC (2006). This altitude is 2.5 km higher than the cold point tropopause (CPT) as shown in Fig. S2. The CPT corresponds to an altitude of 17.5 km and air temperature of approximately 190 K. These values are in line with previous measurements of TTL (Fueglistaler et al., 2009; Gettelman et al., 2004).

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174 The lower boundary of the TTL is determined by examining the vertical profile of potential temperature ( $\theta$ ; Fig. S3). The curvature of the altitude- $\theta$  profile changes at 14.5 km, which is 175 mathematically characterized as the lapse rate minimum (LRM) of  $\theta$ . This change reflects the 176 transition of stability regimes, i.e., deep convection dominates air stability below the LRM and 177 178 radiation starts to influence air temperature above the LRM. Accordingly, we identify the LRM level at 14.5 km (355 K, 140 hPa) as the base of the TTL (Gettelman and Forster, 2002), which is 179 consistent with previous studies (Fueglistaler et al., 2009; Sunilkumar et al., 2017). As a result, the 180 TTL spans from 14.5 km to 20 km, and the majority of our measurements were conducted within 181 the TTL (Fig. 1). 182

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# 184 **3.2 Vertical profile of aerosols and ozone**

The vertical profile of aerosol mass from the surface to the top of the TTL can be segmented in 185 three layers with distinct characteristics (Fig. 2a). Layer I spans from the surface to 5 km altitude 186 and represents the lower troposphere, in which the MMR of aerosols decreased logarithmically 187 with altitude from 2  $\mu$ g kg<sup>-1</sup> to 0.08  $\mu$ g kg<sup>-1</sup> likely because of the influence of surface emissions 188 from Guam. Layer II ranges from 5 km to 14.5 km altitude. In this layer, the aerosol MMR 189 remained at approximately 0.08 µg kg<sup>-1</sup> with small variability. Such low-concentration aerosol 190 191 layers immediately below the convection outflow have been observed previously over the Northern Indian Ocean (de Reus et al., 2001) and the rain forests in South America (Andreae et 192 193 al., 2018; Krejci et al., 2003). The observation indicates that deep convection serves as an effective 194 sink for aerosol particles (Yu et al., 2019). Layer III lies in the TTL, extending from 14.5 km to 18.9 km, with the upper boundary 1.4 km higher than the CPT. In this layer the aerosol MMR 195 increased rapidly from 0.08  $\mu$ g kg<sup>-1</sup> at 14.5 km to 1.5  $\mu$ g kg<sup>-1</sup> at 18.9 km. Our observations show 196 a sustained increase of aerosol MMR from the upper troposphere to lower stratosphere across the 197 tropopause, rather than abrupt transitions. This suggests that the CPT has no unique role for the 198 transport of aerosols. Accounting for the mass of <140 nm particles that were not measured by the 199 POPS instrument, the aerosol MMR at 18.9 km would be approximately 3 µg kg<sup>-1</sup>. This 200 concentration is close to the measurement during the POLARIS mission in 1997 (Mclinden et al., 201

1999), in which an aerosol MMR of approximately  $3.5 \,\mu g \, kg^{-1}$  was observed for aerosols with size range of 0.07–1  $\mu$ m at 18.9 km in eastern Pacific (17.5° N, 159.3° W).

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The ozone concentration remained below 35 ppb in layers I and II and began to increase at 14.5 km (Fig. 2b). This observation is consistent with previous ozonesonde measurements in the tropical Pacific (Folkins et al., 1999; Folkins and Martin, 2005). The concurrence of ozone minimum and LRM is in line with the analysis by Gettelman and Forster (2002) and Folkins et al. (2002).

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#### 211

**Figure 2**. Potential temperature ( $\theta$ ) versus (**a**) O<sub>3</sub> and (**b**) MMR for data acquired from 0–5 km, 5–14.5 km, and 14.5–18.9 km measurements for all flights. Individual data points (5-s averages) are shown by the dots and the averages (by  $\theta$  of 2K) are shown by the solid squares.

215

#### **3.3 Mechanisms for aerosol enhancement in the TTL**

We find that the aerosol MMR was tightly correlated with  $O_3$  in the TTL during all flights with small variation between flights. This can be clearly seen in the example time series of O3 and aerosol MMR for the flight on October 18 shown in Fig. 3a. The Pearson's correlation coefficient (*r*) for the campaign-average  $O_3$  and aerosol MMR (averaged into 2-K  $\theta$  intervals) was 0.98 in layer III. In contrast, the aerosol MMR and  $O_3$  were anticorrelated with an *r* value of -0.35 in layer I (Fig. 3b), and no correlation was observed between aerosol MMR and  $O_3$  in layer II.

223

224 The main processes contributing to the increase in TTL  $O_3$  from the bottom to the top of the TTL 225 include chemical production via photolytic dissociation of molecular oxygen (O<sub>2</sub>) (Prather 2009; 226 Crutzen et al., 1999) and isentropic in-mixing of stratospheric air from the extratropical lower 227 stratosphere (Konopka et al., 2010; Ploeger et al., 2012). The relative contribution of these 228 processes to  $O_3$  remains unclear. While some studies argue that in situ chemical production dominates (Avallone and Prather 1996), others suggest that isentropic stratospheric in-mixing can 229 contribute to O<sub>3</sub> by as much as 40-60% (Abalos et al., 2013a, 2013b; Konopka et al., 2009, 2010; 230 Ploeger et al., 2011, 2012; Sargent et al., 2014). This contribution is most significant during the 231 boreal summer months (June-August), primarily driven by the Asian summer monsoon, and 232

233 gradually decreases during transition into the fall and winter months. In particular, these studies

suggest that the contribution from in-mixing falls within the range of 0-20% in October, averaged over a  $\pm 10^{\circ}$ N latitude range, with the contribution increasing with altitude.

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Figure 3. (a) Example time series of aerosol MMR and O<sub>3</sub> for the October 18 measurements. The gaps indicate ice-influenced data that were excluded from the analysis. (b) Scatter plot of MMR vs O<sub>3</sub> for the 0– 5 km (layer I), 5–14.5 km (layer II), and 14.5–18.9 km (layer III) measurements. Individual data points are shown by the dots and the averages are shown by the solid squares Logarithmic scales are used to make the small values visible.

243

244 We used a one-dimensional column model (over the flight region) and tracer analysis to examine these processes. The model calculates the vertical distribution of O<sub>3</sub> within the TTL, considering 245 a constant slow vertical ascent rate of 0.25 mm s<sup>-1</sup> (Park et al., 2010; Avallone and Prather 1996) 246 of air in the TTL and assuming it evolves in isolation (Supporting Information). Remarkably, the 247 calculated vertical profile of O<sub>3</sub> closely aligns with the observed O<sub>3</sub> profile (Fig. 4a). This result is 248 in line with earlier investigations that yielded similar results using a column model (Avallone and 249 250 Prather 1996) and with research that assumed the tropics are isolated from extra-tropics when 251 explaining the annual cycles of ozone above the tropical tropopause (Randel et al., 2007; Schoeberl et al., 2008). However, we acknowledge the simplicity of the model. For example, the model does 252 not consider the chemistry of nitrogen oxides (NO and NO<sub>2</sub>; not measured during POSIDON) and 253 O<sub>3</sub>, which could be important in the TTL O<sub>3</sub> abundance (Nussbaumer et al., 2023), and 254 uncertainties exist in the assumed ascent rate. On the other hand, the tracer analysis shows evidence 255 of stratospheric in-mixing. Fig. S4 shows that N<sub>2</sub>O was anticorrelated with O<sub>3</sub> in the TTL during 256 POSIDON. The observed negative correlation indicates that the sampled TTL air included a 257 258 contribution from mixing of stratospheric origin (Folkins et al., 1999). This is because the strong UV radiation at higher altitudes in the stratosphere photolyzes N<sub>2</sub>O while leading to O<sub>3</sub> production, 259 resulting in an anticorrelation of N<sub>2</sub>O and O<sub>3</sub> (Assonov et al., 2013). In contrast, N<sub>2</sub>O in the 260 troposphere is inert and uniformly distributed, thus correlation of N<sub>2</sub>O with O<sub>3</sub> is not expected and 261 not observed below the TTL during our measurements. The synthesis of modeling and tracer 262 analysis indicates contributions from both chemical production and stratospheric in-mixing to TTL 263

 $O_3$ . Given the strong correlation between  $O_3$  and aerosols in the TTL, these results suggest that TTL aerosols may also come from a combination of these chemical and physical processes.



#### 267

Figure 4. Comparison of modeling results to measurements for (a)  $O_3$  and (b) aerosol MMR. Individual measurement points are shown by the grey dots. The median values and the 25<sup>th</sup> and 75<sup>th</sup> percentiles are shown by the blue line-square symbols and the error bars. The y-axis represents altitude relative to the tropopause.

272

273 To investigate the chemical composition of TTL aerosol, we employed the CARMA model embedded in CESM to simulate the aerosol formation and growth processes. The model 274 reproduced the O<sub>3</sub> profile reasonably well (Fig. 4a). Our model results suggest that sulfate aerosol 275 exhibits a consistent increase with altitude, whereas organic aerosol displays an ascending trend 276 below the tropopause but declines above it. Moreover, our modeled vertical profile of the total 277 aerosol MMR demonstrates good agreement with the in situ measurements (Fig. 4b) below the 278 tropopause, with organic aerosol constituting the significant fraction of the aerosol mass. These 279 modeling results imply that sulfur alone is insufficient to explain the observed aerosol mass, and 280 organic precursors may play a pivotal role in the formation and/or growth of TTL aerosols under 281 low-temperature conditions. This finding aligns with previous findings from TTL measurements 282 during the Pre-AVE and CR-AVE campaigns over Southwest Central America in boreal winter 283 (Froyd et al., 2009), which highlighted the prevalence of organic-sulfate particles as the most 284 abundant particle type in the TTL and lower stratosphere. We note that the model underestimates 285 the aerosol mass above the tropopause, with the difference increasing with altitude within the TTL, 286

reaching 60% at an altitude of 1 km above the CPT. Additional research is needed to understandthis discrepancy.

289

290 We postulate that TTL aerosols are likely generated through the processes of new particle formation and subsequent growth following convective outflow in the upper troposphere 291 (Williamson et al., 2019). The observed aerosol number size distribution in the TTL suggests 292 particle growth with increasing altitude (Fig. S5), although obtaining measurements of smaller 293 particles would enhance our understanding. During convective transport, soluble species are 294 effectively removed, whereas the insoluble and weakly soluble species can endure washout and 295 gradually ascend upward, with minimal loss, to reach the stratosphere (Bechara et al., 2010). The 296 TTL, characterized by low temperature, low particle surface area density, and high relative 297 humidity, provides ideal conditions for new particle formation and growth. Multiple studies have 298 299 proposed the upper troposphere as the primary nucleation region (Brock et al., 1995; Weigel et al., 2011), and investigations have revealed the involvement of organics in new particle formation and 300 initial growth in the remote tropical upper troposphere (Kupc et al., 2020). Future measurements 301 analyzing the composition of oxidized organic species will enhance our understanding of particle 302 formation and growth pathways. Exceptional overshooting that crosses the tropical tropopause 303 could also affect the aerosol abundance in the TTL (Vernier et al., 2011). This possibility is 304 examined using  $H_2O$  as a tracer. The  $\theta$ - $H_2O$  profile displayed no spikes above the tropopause (Fig. 305 306 S6), suggesting the absence of overshooting during the measurements. 307

- The prevalence of secondary aerosol formation as the primary factor influencing TTL aerosol 308 abundance is in line with the strong correlation between TTL aerosol mass and O<sub>3</sub> in our 309 measurements. Building upon this observation and the underlying mechanisms, we further derived 310 an empirical parameterization of aerosol MMR as a function of O<sub>3</sub> in the TTL using linear 311 regression. The derived relationship is expressed as MMR ( $\mu g kg^{-1}$ ) = 0.0074(±2.2x10<sup>-4</sup>) × O<sub>3</sub> (ppb) 312 + 0.23 (±0.049), accounting for the 50% mass that was not measured by the POPS instrument. 313 This parameterization could be used to estimate TTL aerosol abundance and for validation of 314 modeled results. 315
- 316

#### 317 **4** Conclusions

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Our aircraft measurements conducted over the western Pacific revealed a vertically stratified 319 aerosol distribution comprised of three distinct layers. The lower-troposphere layer, ranging from 320 321 0 to 5 km in altitude, exhibited a decrease in aerosol mass with height, primarily influenced by 322 surface-level emissions. The layer between 5 and 14.5 km displayed consistently low aerosol 323 concentrations with minimal variability, indicating effective aerosol removal through deep 324 convection processes. In the TTL, we observed an enhanced aerosol mass that exhibited a strong 325 correlation with O<sub>3</sub>. The modeling and tracer analysis suggest that TTL O<sub>3</sub> and aerosols likely originate from a combination of chemical production and stratospheric in-mixing processes. 326 327 Furthermore, based on the linear relationship observed between aerosol MMR and O<sub>3</sub> in the TTL, 328 we derived an empirical parameterization that allows for the estimation of aerosol MMR as a function of  $O_3$ . This parameterization holds potential for validating the simulated TTL aerosol abundance in global models, enabling the simulation of the climate impacts of TTL aerosols.

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332 Our study also highlights the challenge of accurately predicting aerosol mass, particularly above the tropopause, even for the most advanced models. This underscores the existence of significant 333 334 gaps in our understanding of the origin and formation mechanisms of aerosols in the TTL. To achieve a more comprehensive understanding of TTL aerosol, it is crucial to conduct future 335 measurements that encompass detailed aerosol chemical composition and gas-phase precursors 336 across different locations and seasons. These endeavors will help diagnose potential deficiencies 337 in the models and validate the modeling results. Consequently, they will lead to an improved 338 comprehension of TTL aerosols and facilitate a predictive understanding of their effects on 339 stratospheric chemistry, clouds, and climate. 340

341

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- 346

## 347 Data Availability Statement

- The aircraft measurements during the POSIDON field campaign are available through the NASA
   ESPO Data Archive: https://espoarchive.nasa.gov/archive/browse/posidon/WB57
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1	Supporting Information for
2	Enhanced Aerosal Mass in the Transcel Transpages I over Linked to Ozana
3	A hundance
4 5	Abundance
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16	
17	1. Model calculation of ozone profile in the TTL
18	
19	The natural ozone chemical cycle is described by the Chapman reactions as follows:
20	
	$O_2 \xrightarrow{h\nu} O + O \tag{R1}$
	$O + O_2 \xrightarrow{M} O_2 \tag{R2}$
	hv (DO)

$$O_3 \xrightarrow{h\nu} O + O_2 \tag{R3}$$

$$0 + O_3 \to O_2 + O_2 \tag{R4}$$

21 The rate equations for [O] and  $[O_3]$  are 22

$$\frac{d[0]}{dt} = 2j_{0_2} - k_2[0][0_2][M] + j_{0_3}[0_3] - k_4[0][0_3] \quad (1)$$
$$\frac{d[0_3]}{dt} = k_2[0][0_2][M] - j_{0_3}[0_3] - k_4[0][0_3] \quad (2)$$

23

24 Combining (1) and (2) we obtain25

$$\frac{d[0+O_3]}{dt} = 2j_{O_2}[O_2] - 2k_4[0][O_3]$$
(3)

26

Since  $[O] \ll [O_3]$ , equation (3) can be written as

 $\frac{d[O_3]}{dt} = 2j_{O_2}[O_2] - 2k_4[O][O_3] \tag{4}$ 

The fast interconversion between O and O<sub>3</sub> (R2 and R3) leads to the following relationship between O and O<sub>3</sub> concentrations (Seinfeld and Pandis 2016):

$$\frac{[O]}{[O_3]} = \frac{j_{O_3}}{k_2[O_2][M]}$$
(5)

32

33 Therefore the production rate of  $O_3$  is expressed as

$$\frac{d[O_3]}{dt} = 2j_{O_2}[O_2] - \frac{2k_4 j_{O_3}[O_3]^2}{k_2[O_2][M]}$$
(6)

34

35 Where

- $k_2 = 6.0 \times 10^{-34} (T/300)^{-2.4}$ , with an average of 1.7 x 10<sup>-33</sup> cm<sup>6</sup> molec<sup>-2</sup> s<sup>-1</sup> in the TTL.
- $k_4 = 8.0 \times 10^{-12} \exp(-2060/T)$ , with an average of 2.0 x  $10^{-16}$  cm<sup>3</sup> molec<sup>-1</sup> s<sup>-1</sup> in the TTL.
- $j_{O_3}$  ranges from 3 x 10<sup>-4</sup> s<sup>-1</sup> (SZA=85°) to 6.5 x 10<sup>-4</sup> s<sup>-1</sup> at SZA = 0° in the TTL (Seinfeld and Pandis, 2016, Fig. 4.13 on p. 112).
- 40  $j_{O_2}$  ranges with increasing altitude from 2.0 x 10<sup>-15</sup> s<sup>-1</sup> to 7 x 10<sup>-14</sup> s<sup>-1</sup> at SZA = 0° in the TTL 41 (2016, Fig. 4.12 on p.111).
- 42
- 43 From the observations we have the following average concentrations:
- 44 45  $[M] = 1.5 \times 10^{18}$  molecules cm<sup>-3</sup>
- 46  $[O_2] = 0.21 \text{ x} [M] = 3.2 \text{ x} 10^{17} \text{ molecules cm}^{-3}$
- 47  $[O_3] = 117 \text{ ppb} = 1.9 \text{ x } 10^{11} \text{ molecules cm}^{-3}$

#### 48

Using the these values, we find that the first term of equation (6) is  $2j_{O_2}[O_2] = 1.2 \times 10^3$ -4.4 x 10<sup>4</sup> molecules cm<sup>-3</sup>s<sup>-1</sup>, and the second term of equation (6) is  $\frac{2k_4 j_{O_3}[O_3]^2}{k_2[O_2][M]} = 5.3$ -11.5 molecules cm<sup>-3</sup>s<sup>-1</sup>, which can be neglected as it is about 1000 times smaller than the first term. Therefore the O<sub>3</sub> formation rate becomes

$$\frac{d[o_3]}{dt} = 2j_{o_2}[o_2] \tag{7}$$

54

We used the vertical profile from Seinfeld and Pandis (2016, Fig. 4.12 on p.111) and the vertical profile of  $[O_2]$  calculated from the measurements to derive the ozone production rate. The calculated vertical profile of the ozone production rate is shown in Fig. S7. The O<sub>3</sub> concentration in the TTL is calculated as

59

60

$$[O_3]_i = [O_3]_{i-1} + \left(\frac{d[O_3]}{dt}\right)_i \cdot \Delta t_i = [O_3]_{i-1} + \left(2j_{O_2}[O_2]\right)_i \cdot \Delta t_i$$

61

62 Where *i* indicates the parameters at altitude *i*, and  $\Delta t_i$  (5 s) represents the time it takes the air to 63 rise from altitude *i*-1 to altitude *i*, using a constant ascent rate of 0.25 mm s<sup>-1</sup> (Park et al 2010; 64 Availance and Prother 1006; Sainfold and Pandia 2016). The calculation is performed for the

- altitude range of 14.5-18.9 km, with a prescribed [O<sub>3</sub>] at 14.5 km of 28.8 ppb, which represents
- the average observed  $O_3$  concentration at that altitude.
- 67

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# 77 Supporting Figures

## 



**Figure S1**. Lognormal fit of the average mass size distribution for the TTL measurements.





**Figure S2**. Vertical profile of air temperature measured from all flights.



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**Figure S3**. Vertical profiles of potential temperature ( $\theta$ ) and the lapse rate of  $\theta$ . In the altitude- $\theta$ 

profile, individual data points (5-s averages) are shown by the dots and the averages (by  $\theta$  of 2K) are shown by the solid squares.



91
92 Figure S4. Scatter plot of O<sub>3</sub> versus N<sub>2</sub>O in the TTL. The dashed line indicates linear fit.





95 Figure S5. Image plot illustrating aerosol number size distribution with altitude in the TTL. The

96 line and markers represent averages at various altitudinal intervals.



**Figure S6**.  $\theta$  versus water vapor mixing ratio.





Figure S7. Calculated vertical profile of ozone production rate in the TTL.