Chemical data assimilation with aqueous chemistry in WRF-Chem coupled with WRFDA (V4.4.1)

Soyoung Ha¹, Rajesh Kumar¹, Gabriele G. Pfister¹, Yonghee Lee², Daegyun Lee², Hyun Mee Kim³, and Young-Hee Ryu³

¹National Center for Atmospheric Research (UCAR) ²National Institute of Environmental Research ³Yonsei University

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Abstract

This study introduces a new chemistry option in the Weather Research and Forecasting model data assimilation (WRFDA) system, coupled with the WRF-Chem model (Version 4.4.1), to incorporate aqueous chemistry (AQCHEM) in the assimilation of ground-level chemical measurements. The new data assimilation capability includes the integration of aqueous-phase aerosols from the Regional Atmospheric Chemistry Mechanism (RACM) gas chemistry, the Modal Aerosol Dynamics Model for Europe (MADE) aerosol chemistry, and the Volatility Basis Set (VBS) for secondary organic aerosol (SOA) production. The RACM-MADE-VBS-AQCHEM scheme facilitates aerosol-cloud-precipitation interactions by activating aerosol particles in cloud water during the model simulation. With the goal of enhancing air quality forecasting in cloudy conditions, this new implementation is demonstrated in the weakly coupled three-dimensional variational data assimilation (3D-Var) system through regional air quality cycling over East Asia. Surface particulate matter (PM) concentrations and four gas species (SO\$_2\$, NO\$_2\$, O\$_3\$, and CO) are assimilated every 6 h for the month of March 2019. The results show that including aqueous-phase aerosols in both the analysis and forecast can represent aerosol wet removal processes associated with cloud development and rainfall production. During a pollution event with high cloud cover, simulations without aerosols defined in cloud water exhibit significantly higher values for liquid water path (LWP), and surface PM\$_{10}\$ (PM\$_{2.5}\$) concentrations are overestimated by a factor of 10 (3) when wet scavenging processes dominate. On the contrary, aqueous chemistry proves to be helpful in simulating the wet deposition of aerosols, accurately predicting the evolution of surface PM concentrations without such overestimation.















Day (March 2019)

NODA AC



18 24 30 Forecast Hour

36

42 48

0 6 12

Г. 0

6 12

18 24 30 Forecast Hour 36 42

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3	Soyoung Ha
4	¹ National Center for Atmospheric Research, Boulder, CO, USA
5	Rajesh Kumar
6	¹ National Center for Atmospheric Research, Boulder, CO, USA
7	Gabriele Pfister
8	¹ National Center for Atmospheric Research, Boulder, CO, USA
9	Yonghee Lee
10	2 Air Quality Forecasting Center, National Institute of Environmental Research, Incheon,
11	Republic of Korea
12	Daegyun Lee
13	² Air Quality Forecasting Center, National Institute of Environmental Research, Incheon,
14	Republic of Korea
15	Hyun Mee Kim
16	³ Department of Atmospheric Sciences, Yonsei University, Seoul, Republic of Korea
17	Young-Hee Ryu
18	³ Department of Atmospheric Sciences, Yonsei University, Seoul, Republic of Korea

Key Points:

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20	•	The WRF-Chem/WRFDA 3D-Var system (V4.4.1) is extended for chemical data
21		assimilation with aqueous chemistry using the RACM-MADE-VBS scheme.
22	•	Surface $PM_{2.5}$, PM_{10} , SO_2 , NO_2 , O_3 , and CO concentrations are assimilated in
23		the coupled system.
24	•	The inclusion of aqueous-phase aerosols in air quality cycling enhances the sim-
25		ulation of aerosol wet deposition in cloudy conditions.

Corresponding author: Soyoung Ha, syha@ucar.edu

26 Abstract

This study introduces a new chemistry option in the Weather Research and Forecast-27 ing model data assimilation (WRFDA) system, coupled with the WRF-Chem model (Ver-28 sion 4.4.1), to incorporate aqueous chemistry (AQCHEM) in the assimilation of ground-29 level chemical measurements. The new data assimilation capability includes the integra-30 tion of aqueous-phase aerosols from the Regional Atmospheric Chemistry Mechanism (RACM) 31 gas chemistry, the Modal Aerosol Dynamics Model for Europe (MADE) aerosol chem-32 istry, and the Volatility Basis Set (VBS) for secondary organic aerosol (SOA) produc-33 tion. The RACM-MADE-VBS-AQCHEM scheme facilitates aerosol-cloud-precipitation 34 interactions by activating aerosol particles in cloud water during the model simulation. 35 With the goal of enhancing air quality forecasting in cloudy conditions, this new imple-36 mentation is demonstrated in the weakly coupled three-dimensional variational data as-37 similation (3D-Var) system through regional air quality cycling over East Asia. Surface 38 particulate matter (PM) concentrations and four gas species (SO₂, NO₂, O₃, and CO) 30 are assimilated every 6 h for the month of March 2019. The results show that includ-40 ing aqueous-phase aerosols in both the analysis and forecast can represent aerosol wet 41 removal processes associated with cloud development and rainfall production. During 42 a pollution event with high cloud cover, simulations without aerosols defined in cloud 43 water exhibit significantly higher values for liquid water path (LWP), and surface PM_{10} 44 $(PM_{2.5})$ concentrations are overestimated by a factor of 10 (3) when wet scavenging pro-45 cesses dominate. On the contrary, aqueous chemistry proves to be helpful in simulating 46 the wet deposition of aerosols, accurately predicting the evolution of surface PM con-47 centrations without such overestimation. 48

⁴⁹ Plain Language Summary

Major air pollution events over the Korean peninsula are often observed in association with significant cloud cover, especially over the Yellow Sea to the west of the peninsula. Cloudy conditions pose challenges for both remote sensing observations and model predictions, but the inclusion of aqueous-phase (or cloud-borne) aerosols in the WRF-Chem/WRFDA system improves the simulation of aerosol wet scavenging, leading to improved predictions of surface particulate matter concentrations that were otherwise substantially overestimated.

57 Keywords

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Aerosol data assimilation, aqueous chemistry, wet deposition

59 1 Introduction

Poor air quality, characterized by high concentrations of particulate matter (PM) 60 at ground level, is often accompanied by extensive cloud cover Eck et al. (2018, 2020). 61 posing challenges for both observation and prediction. Given the short lifetime of aerosol 62 species (up to one week) and the large uncertainties in modeling atmospheric compo-63 sition, improving initialization is crucial for enhancing short-range air quality forecast-64 ing. Data assimilation (DA) incorporates available observations into a numerical predic-65 tion model to produce initial conditions that can lead to accurate forecasts. The qual-66 ity of the initial condition, or the analysis, largely depends on the quantity of reliable 67 observations and the accuracy of the forecast model. 68

Various efforts have been recently made towards chemical data assimilation, but the utilization of advanced chemistry schemes remains limited, especially for the prediction of particulate matter concentrations (Chen et al., 2019; Sun et al., 2020; Ha, 2022). Hence, it is fair to say that chemical data assimilation, especially aerosol data assimi-

lation focusing on aerosol species, is still in its early stages, lagging behind meteorolog-73 ical or oceanographic applications (Baklanov et al., 2014; Bocquet et al., 2015). Major 74 challenges specific to aerosol data assimilation can be described as follows: 1) The lim-75 ited information content of atmospheric composition observations, which often lack ac-76 curacy and coverage, especially in cloudy conditions. 2) Large uncertainties or system-77 atic errors in chemical transport models, partly due to significant uncertainties in forc-78 ing parameters such as emissions and partly due to imperfect representation of complex 79 chemical processes and their interactions with the atmospheric environment. 3) Surface 80 PM concentrations, major indicators of the air quality index, are only computed diag-81 nostically at the end of the model integration to account for all contributions from mul-82 tiple aerosol species. Therefore, the data assimilation system should handle all the prog-83 nostic aerosol species that contribute to the estimation of PM concentrations and de-84 termine how to distribute analysis increments in ground PM concentrations back to the 85 three-dimensional aerosol variables. Since the number of aerosol species predicted in the 86 model is typically larger than the number of observed variables, it becomes an under-87 constrained problem where a unique solution is not guaranteed. Moreover, the use of so-88 phisticated chemistry schemes means a large number of prognostic variables, rendering 89 aerosol analysis a high-dimensional problem that requires substantial computational re-90 sources. This makes the three-dimensional variational data assimilation (3D-Var) algo-91 rithm still attractive due to its speed and operational simplicity, despite limitations such 92 as static background error covariance and the ignorance of model errors. 4) The model 93 configuration for aerosol chemistry coupled with meteorology, the first step toward aerosol cycling, is typically more complicated than that of a weather prediction model alone. In 95 an online coupled system, the dynamics and physics configurations for meteorology can 96 greatly impact the reliability and performance of chemical simulations. Therefore, care-97 ful consideration should be given to the weather component of the configuration as well. 98 5) Most interfaces to input data and chemical processes are highly customized for spe-99 cific gas and aerosol chemistry schemes. This is because each chemistry parameteriza-100 tion defines its own unique set of prognostic variables that are not interchangeable with 101 other schemes (Pfister et al., 2020). Consequently, each chemistry scheme requires its 102 own preparation of input forcing data, such as anthropogenic, biogenic and biomass burn-103 ing emissions. In the context of data assimilation, this also entails developing a new in-104 terface within the variational data assimilation system. This includes creating new ob-105 servation operators (that compute the model correspondents from the aerosol species de-106 fined in the scheme), as well as their tangent linear and adjoint models, and estimating 107 background error covariance. 108

Attempts to develop coupled data assimilation between chemistry and meteorol-109 ogy for regional forecast applications have been limited, despite the widespread recog-110 nition of high correlations between the two components in the modeling community (Baklanov 111 et al., 2017). In the context of 3D-Var, fully (or strongly) coupled data assimilation is 112 still challenging because many factors contributing to cross-covariance between meteo-113 rological and chemical variables are highly variable in time and space. When it comes 114 to real observations, developments and research on coupled data assimilation have pri-115 marily focused on trace gases such as ozone and carbon monoxide, often using simpli-116 fied background error covariance (Ménard et al., 2019). On the other hand, in weakly 117 coupled data assimilation, the coupling occurs during the model integration by using a 118 two-way coupled forecast model, but not through the analysis. Observations are assim-119 ilated in each component (e.g., meteorology or chemistry) to update the analysis vari-120 ables independently, which are then used together to initialize the coupled modeling sys-121 tem for prediction. The background error covariance is estimated from the forecasts pro-122 duced by the fully coupled model, capturing the coupling aspects from the model sim-123 ulations rather than the cross-covariance component. As a result, the direct influence 124 of observations is limited to each component of the model. 125

Aerosol particles play a crucial role in various key processes related to atmospheric 126 chemistry and physics (Rosenfeld et al. (2008); Stevens and Feingold (2009); Tao et al. 127 (2012); Baklanov et al. (2014)). These particles directly scatter and absorb incoming so-128 lar radiation, leading to changes in the atmospheric radiation reaching the Earth's sur-129 face (e.g., "aerosol direct effects"). They are either suspended in the air or attached to 130 hydrometeors such as cloud droplets or ice crystals, acting as cloud condensation nuclei 131 (CCN) or ice nuclei (IN). This interaction modifies the formation, lifetime, and optical 132 properties of clouds, including cloud albedo, as well as precipitation rates, indirectly af-133 fecting the radiative transfer (e.g., "aerosol indirect effects"). During cloud processes, 134 aerosol particles undergo physical and chemical changes in their composition and mass 135 concentrations. They are also redistributed by clouds and convection, which serve as trans-136 port media over time scales ranging from minutes to hours (Ervens, 2015). Aerosol-cloud 137 interactions, which encompass both the effects of aerosols on clouds and the cloud ef-138 fects on aerosol particles, can play a crucial role in daily air quality, especially in regions 139 with polluted aerosols and cloudy conditions. These interactions can influence both in-140 cloud and below-cloud wet scavenging of aerosol particles, leading to changes in PM con-141 centrations reaching the ground. 142

From a data assimilation perspective, cloudy conditions are particularly challeng-143 ing because ground-based or remote-sensing retrievals are often missing or significantly 144 degraded in quality due to cloud contamination. When only limited observations are avail-145 able for data assimilation, the quality of the analysis depends heavily on the numerical 146 prediction system, especially its systematic errors. Variational data assimilation algo-147 rithms typically assume a perfect model without any systematic errors. Although it is 148 not straightforward to detect and correct model error sources (Dee and Da Silva (1998); 149 Dee (2005)), the analysis can benefit from advanced features that can enhance physical 150 and chemical mechanisms, thereby improving the model performance. Eck et al. (2018, 151 2020) have reported that major air pollution events in East Asia are frequently associ-152 ated with significant cloud cover, highlighting the importance of aerosol-cloud interac-153 tions compared to pristine conditions. This study is motivated by the haze events in Ko-154 rea, which coincide with extensive cloud cover and rainfall. 155

The Weather Research and Forecasting model coupled with Chemistry (WRF-Chem; 156 Grell et al. (2005)) facilitates simulations of real-time interactions between aerosol, cloud, 157 radiation, and precipitation for regional applications. To account for aerosol effects on 158 clouds and simulate aerosol-cloud interactions within the system, it is necessary to de-159 fine cloud-borne aerosols (particles attached to cloud droplets) as well as interstitial aerosols 160 (particles suspended in the air). When aerosol particles are represented in the aqueous 161 phase (or in cloud water), cloud droplet number concentrations are prognostically treated 162 through processes such as droplet activation, scavenging, and resuspension, allowing for 163 real-time feedback between aerosols and clouds. Wet deposition, a major sink process 164 for aerosol particles, involves the transport and removal of soluble or scavenged constituents 165 by precipitation. It encompasses in-cloud scavenging and removal by rain and snow (rain-166 out), release by evaporation of rain and snow, and below-cloud scavenging by precipi-167 tation falling through without formation of precipitation (wash-out) (Seinfeld & Pan-168 dis, 2006). Subgrid-scale convective transport and in-cloud scavenging can be activated 169 for aerosols in both the interstitial and aqueous phases (as well as tracers). However, to 170 simulate wet scavenging by grid-resolvable precipitation and below-cloud scavenging through 171 cloud chemistry, one needs to choose a chemistry option that includes aqueous chemistry 172 with aerosols in the aqueous phase (as well as the interstitial phase) in the WRF-Chem 173 model. Under cloudy conditions, the absence of these features can impact cloud forma-174 tion, growth, and wet deposition of air pollutants, resulting in systematic forecast errors 175 in surface PM concentrations. 176

Tuccella et al. (2015) implemented aqueous chemistry (AQCHEM) in the Regional Atmospheric Chemistry Mechanism (RACM; Stockwell et al. (1997)) gas-phase chemistry, coupled with the Modal Aerosol Dynamics Model for Europe (MADE; Ackermann
et al. (1998)) inorganic aerosol mechanism and a secondary organic aerosol (SOA) scheme
based on a four-bin volatility basis set (VBS) (Ahmadov et al., 2012) in the WRF-Chem
model. They demonstrated that the RACM-MADE-VBS-AQCHEM scheme, coupled with
cloud microphysics and radiation parameterization schemes, could characterize aerosolcloud feedbacks, reducing large uncertainties in the prediction of microphysical and optical properties of clouds.

The WRFDA system has been recently updated to incorporate the RACM-MADE-VBS scheme (chem_opt=108) in the WRF-Chem model for chemical data assimilation (Ha, 2022). A case study conducted during the Korea–United States Air Quality (KORUS-AQ) period demonstrated that the 3D-Var aerosol analysis resulted in systematic improvements in the prediction of surface PM concentrations over Korea. However, forecast errors tended to increase in cloudy conditions.

This study further extends the WRFDA 3D-Var system to incorporate aerosol-cloud interactions with aqueous chemistry using the RACM-MADE-VBS-AQCHEM scheme (chem_opt=109) in aerosol analysis and forecast cycling. The objective is to enhance the short-range prediction of surface PM concentrations over South Korea, particularly during cloudy conditions characterized by wet scavenging of polluted aerosols.

Section 2 provides an overview of the WRF-Chem cycling system, describing the
new implementation in the WRFDA system, which includes the introduction of new forward operators and background error statistics designed for aerosols in the aqueous phase.
In Section 3, cycling experiments are conducted, and the forecast results are verified against
independent observations to evaluate the reliability and performance of the system. Finally, in Section 4, we draw conclusions, discuss the limitations of this study, and provide suggestions for future research.

²⁰⁴ 2 The WRF-Chem cycling system

The Weather Research and Forecasting (WRF) system consists of three main com-205 ponents: the WRF Preprocessing System (WPS), the WRF model coupled with Chem-206 istry (WRF-Chem), and the WRF Data Assimilation (WRFDA) system. Through cy-207 cling (e.g., conducting analysis and forecast consecutively), the observed information is 208 incorporated into the WRF-Chem model at certain time intervals (ex. every 6 h) to ini-209 tialize the simulations. By pulling out the model trajectory towards observations every 210 cycle, we re-initiate dynamical, physical and chemical mechanisms to be close to the ob-211 served state. This is a unified system in a sense that the forecast error is incorporated 212 into the analysis (through background error covariance) and the prediction is initiated 213 from its own analysis every time, thereby the error of the model itself is constantly re-214 flected throughout the cycling system. Figure 1 shows a flowchart of the WRF-Chem/WRFDA 215 cycling system with chemical data assimilation. Dotted lines imply optional input data 216 while solid lines the mandatory inputs for WRF-Chem/WRFDA cycling, accompanied 217 by typical input file names (with no specification of domain ID or time) used in the WRF 218 system. As a first step, WPS is run to configure the model domain using geographical 219 data for land use and soil categories (geogrid.exe), ungrib meteorological data (e.g., the 220 UK Met Office analysis; UM MET) in the grib format (ungrib.exe), and transform the 221 three-dimensional data into the WRF domain (metgrid.exe). Through the WRF initial-222 ization step (real.exe), the data is then converted to the initial condition (wrfinput) and 223 lateral boundary condition (wrfbdy) files for meteorological variables in each domain. 224 For chemical simulations, emissions data should be prepared based on the wrfinput file 225 to define land use categories consistent with those used for the meteorological initial con-226 dition. As soon as the WRF-Chem model starts, atmospheric physics and chemistry pa-227 rameterizations are initialized based on the land use categories (e.g., mminlu) in the lookup 228 tables such as LANDUSE.TBL and VEGPARM.TBL. It is thus critical to use the same 229

wrfinput file in producing all the emissions data. By default, WRF-Chem regional simulations use an idealized gas profile for some chemical species at the lateral boundaries,
as used in this study. It should be noted that the diagram shown here is not meant to
describe all the possible data input in the WRF-Chem model. Optional input data, such
as an upper boundary condition for some gas species, biomass burning (e.g., fire), or aircraft emissions data, are not included because they were not considered here.

Without data assimilation, wrfinput $(\mathbf{x}_{\mathbf{b}})$ and wrfbdy files are used directly to ini-236 tialize the model simulation, bypassing the WRFDA processes (e.g. da_wrfvar.exe and 237 da_update_bc.exe). But if one wants to update the initial condition in the variational data 238 assimilation, at least three input files are required for each model domain - a first guess 239 $(\mathbf{x}_{\mathbf{b}}; \text{wrfinput or fg})$, background error covariance (**B**; be.dat), and observations (**y**; ob.ascii 240 or ob. bufr) that usually come with the specification of observation errors (\mathbf{R}) . Before 241 incorporating observations into the DA system, data collection and processing should 242 be carefully carried out, including data quality check. Since WRF Version 4 (including 243 WRFDA), simultaneous data assimilation has been available for a few chemical options 244 in WRF-Chem to update meteorological and chemical fields at the same time. In the cur-245 rent implementation, chemical observations are designed to be available in ascii format 246 (ob_chemsfc.ascii), separate from meteorological (MET) data provided in BUFR format. 247 When data assimilation (da_wrfvar.exe) is run for each domain, the initial condition is 248 updated as the analysis $(\mathbf{x}_{\mathbf{a}})$ in each domain. The lateral boundary condition in the mother domain also needs to be updated (through da_update_bc.exe) to be consistent with the 250 analysis in the boundary zone. 251

To compute the background error covariance (**B**), WRF-Chem forecasts should be run in advance, typically cycling without data assimilation using the same model configuration for a long period of time (at least for one month). In the National Meteorological Center (NMC) method (Parrish and Derber (1992)), forecast differences between two different forecast leads at the same validation time are used to estimate the background error covariance for all the analysis variables in each domain.

Once the WRF-Chem model is integrated from the initial condition, the output forecast reached in the next cycle is reused to provide the next first guess with the simulated chemical species (e.g., wrf_chem_input). By repeating the WRF initialization, WRFDA, and WRF-Chem simulations with the recycled chemical species at the cycling frequency, WRF-Chem/WRFDA cycling can be carried out continuously.

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2.1 WRF-Chem for aerosol effects

The WRF-Chem model has long been used to study a wide range of atmospheric 264 phenomena associated with atmospheric chemistry and aerosols over regional domains 265 (e.g., Ntelekos et al. (2009), Grell and Baklanov (2011), Pfister et al. (2011), Ahmadov 266 et al. (2012), Saide et al. (2012), Yang et al. (2015)). The online-coupled chemical trans-267 port model numerically solves for the concentration of chemical species through various 268 processes. Included are emissions, advection, vertical mixing with dry deposition, con-269 vective transport, gas chemistry, aerosol chemistry, cloud chemistry (for activated aerosols 270 in cloud water), and wet scavenging. At the end of each time step, PM concentrations 271 272 are computed diagnostically as the sum of all aerosol species defined in the chemistry option. 273

In the WRF-Chem model, aerosol effects are simulated through various processes, including activation, resuspension, aqueous reactions, and wet removal of aerosol particles. These processes are mostly controlled through namelist options for each model domain. As such, we specify the namelist parameters in parentheses corresponding to the description of each process. To account for aerosol-cloud interactions, or aerosol indirect effects, the direct effects of aerosols on incoming solar radiation must be activated (e.g., aer_ra_feedback = 1), which involves relating aerosol sizes and compositions to aerosol optical properties (Fast et al., 2006). In this study, aerosol particles within a certain size range or mode are assumed to have the same composition. This allows for the averaging of refractive indices of spherical particles over all the species within each mode (e.g., volume averaging; aer_op_opt = 1). Subsequently, extinction coefficients due to aerosol scattering and absorption are calculated using Mie theory (Ghan et al., 2001) and incorporated into shortwave and photolysis schemes.

While aerosol direct effects can be accounted for by all the aerosol chemistry schemes 287 when using either the RRTMG or the Goddard shortwave radiation scheme, aerosol in-288 direct effects are only supported by a limited number of modal and sectional aerosol chem-289 istry options in WRF-Chem. For cloud microphysics, it is recommended to employ double-290 moment schemes such as Lin (Lin et al., 1983) or Morrison (Morrison et al., 2009) to take 291 indirect aerosol effects into account. In this study, the Morrison two-moment scheme is 202 utilized, which predicts the mass and number concentrations of five species (i.e., cloud 293 droplets, cloud ice, snow, rain, and graupel). It should be noted that the new implemen-294 tation in DA is applicable to any double-moment scheme without any modifications. The 295 double-moment microphysics accounts for the autoconversion of cloud droplets to rain-296 water based on the droplet number concentrations and interacts with prognostic aerosols, 297 altering their size and composition through aqueous processes and wet scavenging (Yang 298 et al., 2011). To represent aerosol effects on cloud chemistry and grid-scale precipitation 299 $(cldchem_onoff = 1 and wetscav_onoff = 1, respectively)$, it is necessary to simulate an 300 aerosol activation process that enables aerosol particles to grow by water condensation, 301 forming cloud droplets based on supersaturation and particle size (Abdul-Razzak & Ghan, 302 2002). For aerosol indirect effects, cloud droplet number concentrations should be prog-303 nostically treated (progn = 1). Meanwhile, the Grell-Devenyi cumulus scheme simulates convective precipitation, which is recognized by atmospheric radiation and photolytic 305 processes in the model. It also parameterizes convective transport, enabling the displace-306 ment of chemical species. Although aqueous chemistry in subgrid-scale convection (conv_tr_aqchem 307 = 1) and wet scavenging within subgrid-scale clouds (conv_tr_wetscav = 1) can be ac-308 counted for as part of the cumulus parameterization, it does not explicitly consider cloud-309 borne aerosols or their impact on cloud chemistry. For brevity, we refer to aqueous chem-310 istry (AQCHEM) only when aerosols are defined in the aqueous phase for the chosen chem-311 istry scheme (e.g., chem_opt = 109) and aerosol indirect effects are simulated. 312

Tuccella et al. (2015) implemented the RACM-MADE-VBS-AQCHEM scheme (chem_opt=109) 313 for the simulation of aerosol-cloud-radiation interactions, following Fast et al. (2006) and 314 Chapman et al. (2009), with simple aqueous reactions. The MADE-VBS aerosol scheme 315 defines the particle size distribution as a superposition of three log-normal modes: an 316 Aitken mode with a median diameter of 0.01 μ m, an accumulation mode ranging between 317 0.01 and 1 μ m, and a coarse mode for particles typically larger than 1 μ m (with a me-318 dian around 10 μ m). All aerosol particles are assumed to be spherical and internally mixed 319 (Aquila et al., 2011). The aerosol species treated are sulfate (SO_4^-) , nitrate (NO_3^+) , am-320 monium (NH_4^+) , elemental carbon (EC), primary organic matter (POA), anthropogenic 321 and biogenic secondary organic aerosol (SOA), chloride (Cl), sodium (Na), unspeciated 322 $PM_{2.5}$, unspeciated coarse fraction of PM_{10} (antha), soil dust (soila), and sea salt (seas). 323 The unspeciated $PM_{2.5}$ includes the fine fraction of sea salt and mineral dust aerosols. 324

325 For aqueous processes, each aerosol species is defined in the aqueous (or cloud-borne) phase as well as in the interstitial (or non-activated) state. The number and mass con-326 centrations of activated aerosols are calculated for each mode in the presence of water 327 supersaturation. In this study, we identified and addressed several bugs in the RACM-328 MADE-VBS-AQCHEM scheme in the previous versions of the WRF-Chem model (e.g., 329 prior to V4.4.1), primarily related to a simple sulfuric oxidation. These bug fixes were 330 incorporated into the released version of WRF V4.4.1. Also, some deposition variables 331 have been added for diagnostics purposes specific to this study. 332

2.2 WRFDA updates for aqueous chemistry

An interface between the WRF-Chem model and the WRFDA 3D-Var system in version 4.4.1 has been extended for the RACM-MADE-VBS-AQCHEM option to assimilate ground-level measurements of $PM_{2.5}$, PM_{10} , SO_2 , NO_2 , O_3 , and CO concentrations. The RACM-MADE-VBS scheme without aqueous chemistry (chem_opt=108) was previously implemented in the WRF-Chem/WRFDA system by Ha (2022). In this study, the interface is further expanded to include aqueous-phase aerosols (e.g., chem_opt=109) in the aerosol analysis.

$2.2.1 \quad Cost function$

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In the 3D-Var system, the cost function $J(\mathbf{x})$ is minimized to find an optimal so-342 lution for the model state (\mathbf{x}) that best fits to all the observations (\mathbf{y}) available at the 343 analysis time. The minimization is performed based on the background and observation 344 error covariance matrices (**B** and **R**, respectively), assuming Gaussian error distributions 345 (Lorenc, 1986). In the incremental formulation (Courtier et al., 1994) adopted in WRFDA, 346 analysis increments $\delta \mathbf{x} (= \mathbf{x} - \mathbf{x}_{\mathbf{b}})$ are computed at each iteration using the background 347 forecast $(\mathbf{x}_{\mathbf{b}})$ from the previous analysis $(\mathbf{x}_{\mathbf{a}})$ or the previous iteration step. The con-348 trol vector (v) is defined as $\delta \mathbf{x} = \mathbf{B}^{1/2} \mathbf{v}$ to construct the cost function as below. 349

$$J(\mathbf{v}) = \frac{1}{2}\mathbf{v}^{\mathbf{T}}\mathbf{v} + \frac{1}{2}(\mathbf{d} - \mathbf{H}\mathbf{B}^{1/2}\mathbf{v})^{\mathbf{T}}\mathbf{R}^{-1}(\mathbf{d} - \mathbf{H}\mathbf{B}^{1/2}\mathbf{v})$$
(1)

where the innovation vector is defined as $\mathbf{d} = \mathbf{y} - \mathbf{H}(\mathbf{x}_{\mathbf{b}})$ and the observation operator 350 **H** transforms the model states (\mathbf{x}) to the observed quantities (\mathbf{y}) at observation loca-351 tions. In the chemical data assimilation presented in this study, all the chemical species 352 defined in the model states (\mathbf{x}) are also used as control variables (\mathbf{v}) , which are the same 353 as analysis variables. These control variables have univariate error covariances, mean-354 ing that there are no static cross-correlations between chemical species or between chem-355 ical and meteorological variables. A list of 32 three-dimensional aerosol species defined 356 in the analysis includes the following: aerosol sulfate (so4ai and so4aj), nitrate (no3ai 357 and no3aj), ammonium (nh4ai and nh4aj), chloride (clai and claj), primary organic mat-358 ter (orgpai and orgpaj), elemental carbon (eci and eci), sodium (naai and naaj), unspe-359 ciated $PM_{2.5}$ (p25ai and p25aj), 4-bin anthropogenic and biogenic SOA (asoali, asoalj, 360 asoa2i, asoa2j, ..., bsoa4i, bsoa4j). Each variable name in the parenthesis ends with i or 361 *j* to indicate Aitken or accumulation mode. Also included are three coarse-mode variables - non-reactive anthropogenic primary aerosol (antha), marine aerosol concentra-363 tion (seas), soil-derived aerosol particles such as dust (soila). In summary, there are 35 364 aerosol species defined in three modes (= 16 Aitken + 16 accumulation + 3 coarse modes). 365 and their aqueous phase counterparts are also defined with 'cw' added to their names. Four gas species, namely SO_2 , NO_2 , O_3 , and CO, can be also assimilated at the ground 367 level, depending on the assimilation option. This results in a total of 74 three-dimensional 368 chemical species in the control vector. 369

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2.2.2 Forward operator and observation errors

In the assimilation of surface PM observations, **H** is calculated as the sum of each 371 aerosol species defined in the control vector (\mathbf{v}) , interpolated at each observation site. 372 This approach follows the way the MADE-VBS aerosol scheme in the model estimates 373 PM concentrations based on individual aerosol species, ensuring that they are treated 374 consistently between the analysis and forecast. For the activation of aqueous chemistry, 375 $\mathbf{H}(\mathbf{x})$ is extended to include cloud-borne (activated) as well as interstitial (non-activated) 376 aerosol species in all three modes. The PM_{2.5} concentrations in the model space $(y_{pm_{2.5}})$ 377 are computed as the sum of all the aerosol species listed above in accumulation (j) and 378

379 Aitken (i) modes.

$$\mathbf{y}_{pm_{2.5}} = \rho_d \sum_{p=1}^{N} (\sum_{m=i}^{j} \mathbf{y}_m^p + \sum_{m=i}^{j} \mathbf{y}_m^{*p}),$$
(2)

where N is 32, ρ_d dry air density ([kg m⁻³]) for unit conversion from aerosol mixing ratios ([μ g kg-1]) to mass concentrations ([μ g m⁻³]), \mathbf{y}_m^p and \mathbf{y}_m^{*p} representing each aerosol species in the interstitial and aqueous phases, respectively.

When assimilating PM_{10} alone, the model correspondent is computed by adding 383 three coarse-mode variables - antha, seas, and soila - to the simulated $PM_{2.5}$. For the 384 aqueous chemistry option, the three coarse-mode variables in the aqueous phase (anthcw, 385 seascw, and soilcw) are also included in the observation operator. If PM_{10} is assimilated 386 together with $PM_{2.5}$, the residuals from $(PM_{10} - PM_{2.5})$ are assimilated as the sum of 387 the three coarse-mode aerosols, following Ha (2022). It should be noted that aerosol num-388 ber concentrations, which are not directly associated with PM mass concentrations, are 389 not included as analysis variables or in the observation operators in this study. 390

The assimilation of trace gases is straightforward because each gas species is explicitly defined in the model prognostic variables. The control variables for the assimilation are the same four gas species (SO₂, NO₂, O₃, and CO). The observation operator for trace gases involves a simple horizontal interpolation of the corresponding variable at the lowest model level.

The observation error covariance matrix **R** remains unchanged, regardless of the inclusion of aqueous chemistry. It uses the same uncorrelated observation errors for each observation (y_o) . Following Ha (2022), the observation errors for surface PM consist of measurement errors (ϵ_o) and representative errors (ϵ_r) : $\epsilon_y = \sqrt{\epsilon_o^2 + \epsilon_r^2}$ where $\epsilon_o =$ $1.5+0.0075*y_o$ and $\epsilon_r = \gamma \epsilon_o \sqrt{\frac{\Delta x}{L}}$. Here, γ is set to be 0.5, Δx is grid spacing (27 km for domain 1 and 9 km for domain 2), and the scaling factor L (defined as 3 km).

For system reliability, data quality control (QC) is performed by applying maxi-402 mum thresholds to observation values and innovations ((o - f))'s) during the assimila-403 tion process. Surface $PM_{2.5}$ and PM_{10} observations are rejected if they exceed 300, 500 404 $\mu g m^{-3}$, respectively, or if they differ from their corresponding model equivalents (e.g., 405 $H(x_b)$ by more than 100 μg m⁻³. Regarding gas species, the maximum threshold val-406 ues are set at 2 ppmv for observed SO_2 , NO_2 , and O_3 , and 50 ppmv for CO. They are 407 also rejected if their innovations exceed the threshold values of 0.2 ppmv for SO₂, NO₂, 408 and O_3 , and 20 ppmv for CO. It is noted that these threshold values are set to be con-409 servative for the sake of the system reliability, especially in the operational environment. 410 Although none of the observations were rejected based on the thresholds during the month-411 long period in our case study, it could limit the applicability of WRFDA to heavy pol-412 lution events or wildfire episodes. In the future, it would be nice to move these param-413 eters out of the codes and place them in the namelist such that users can easily mod-414 ify them for their specific applications. In the current implementation, ground-level gas-415 phase pollutants are assimilated together as a group rather than individually, using the 416 corresponding model variables as their analysis (or control) variables. 417

418

2.2.3 Background error covariance

In the WRFDA system, the square root of the **B** matrix $(\mathbf{B} = \mathbf{B}^{1/2} (\mathbf{B}^{1/2})^{\mathrm{T}})$ is decomposed into a series of sub-matrices, eliminating the need of computing the inverse of the large **B** matrix, as below.

$$\mathbf{B}^{1/2} = \mathbf{U}_{\mathbf{p}} \mathbf{S} \mathbf{U}_{\mathbf{v}} \mathbf{U}_{\mathbf{h}} \tag{3}$$

 $_{422}$ Here, the matrix $\mathbf{U}_{\mathbf{p}}$ is called physical or balance transformation (via linear regression),

 \mathbf{S} a diagonal matrix of forecast error standard deviation, $\mathbf{U}_{\mathbf{v}}$ the vertical transform, and $\mathbf{U}_{\mathbf{h}}$ the horizontal transform matrix.

In this study, the WRF-Chem model is configured with two domains at grid res-425 olutions of 27 and 9 km, respectively, in a one-way nesting mode, as illustrated in Fig. 426 2 (a). Vertically, a total of 31 model levels are used up to 50 hPa, with upper level jets 427 located around level 23 (\sim 12 km) and low level jets (LLJ) situated around level 9 (\sim 1.2 428 km). At Seoul (37.5N, 127.0E), for instance, a total of 8 model levels are configured un-429 der 1 km, with the lowest level at 72 meters (height above mean sea level). Chemical sim-430 ulations are cycled without data assimilation every six hours using the RACM-MADE-431 VBS-AQCHEM scheme, starting from 21 Feb to 31 March 2019, producing 48-hour fore-432 casts from 00Z each day for the month of March. The differences between the 24-hour 433 and 48-hour forecasts at the same validation time are then computed as a proxy for fore-434 cast errors in each domain. In total, 31 sample forecasts in March, following an 8-day 435 spin-up in Feb 2019, were used to construct the **B** matrix through the NMC method. 436 To generate the **B** matrix in the WRFDA system, the GENBE 2.0 software (Descombes 437 et al., 2015) is extended to incorporate all the aerosols in the aqueous phase, as spec-438 ified in the chosen chemistry scheme. In line with previous 3D-Var studies (Kumar et 439 al. (2019) and Ha (2022)), we binned all the grid points together for each latitudinal band 440 (every 2°) and each model level, but with no longitudinal dependencies in the background 441 error covariance. To examine the impact of aqueous-phase aerosols in the B matrix, we 442 conducted one-month cycling forecasts with chem_opt = 108 and 109, referred to as NO_AQ 443 and AQ, respectively. 444

Figure 3 illustrates the comparison of the square root of the background error co-445 variance $(\mathbf{B}^{1/2})$ between the two experiments for each aerosol component in the inter-446 stitial state. As the control vector (v) is multiplied by $\mathbf{B}^{1/2}$ to convert it to analysis in-447 crements $(\delta \mathbf{x})$ after the minimization process, this figure shows the relative weights of 448 each species, indicating their contributions to atmospheric constituents and their ver-449 tical distribution when assimilating ground-level PM concentrations. It is evident that 450 there is considerable variability in the vertical distribution of aerosol species, and their 451 vertical structure undergoes distinct changes when simulating aerosols in cloud water (AQ). 452 These changes in the interstitial state can be interpreted in conjunction with the back-453 ground error covariance $(\mathbf{B}^{1/2})$ for aqueous-phase aerosols (AQ case), as displayed in Fig. 454 4. Sulfate aerosols are found to have the largest weights in the accumulation mode (cwj) 455 and Aitken mode (cwi), while sea salt aerosols dominate the coarse mode (cw). These 456 findings are consistent with the significant reductions observed for the same species in 457 the interstitial phase across the entire troposphere, as depicted in Fig. 3. These changes 458 can be attributed to the aqueous chemistry implemented in the WRF-Chem model, which primarily involves the oxidation of sulfur dioxide (SO_2) to dissolved sulfur in oxidation 460 state 4, S(IV). It is also noteworthy that aerosols in cloud water ("cw") are concentrated 461 below level 15 (e.g., below 5 km), with peak concentrations estimated around level 5, likely 462 in association with low-level clouds. In contrast, most interstitial aerosols (as shown in Fig. 3) tend to increase as the level goes down, but their distribution extends up to level 464 24. Only dust aerosols (soila in Fig.3 (l)) dominate at higher altitudes, with a peak around 465 level 13 (around 3 km), possibly due to long-range transport of dust during the spring 466 month. Another thing to note is that organic aerosols (both primary and secondary) are 467 not updated through aqueous chemistry in the current version, showing little changes 468 in the aqueous phase, as indicated by the pink solid line for "POA" and blue dotted line 469 for "SOA" in Fig. 4 (a) and (b). The specific mechanisms underlying all the changes in 470 each species are not fully understood, but it is apparent that the weights for soluble species 471 are partitioned between the interstitial (dry) and the aqueous phase (cloud water), chang-472 ing the structure of the background error covariance with aqueous chemistry. This high-473 lights the complex interactions and transformations that occur between different phases 474 of aerosols and the role of aqueous chemistry in modifying their behavior and represen-475 tation in the model. 476

⁴⁷⁷ **3** Chemical analysis and forecast cycling

3.1 Cycling experiments

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To examine the impact of chemical data assimilation with aqueous chemistry, WRF-479 Chem and WRFDA cycling experiments are conducted every 6 h from Feb 21 to Mar 480 31, 2019 over the East Asian region (with a 27-km grid resolution) nested down to the 481 Korean peninsula (at a 9-km resolution). This study uses the Morrison two-moment scheme 482 (Morrison et al., 2009) for cloud microphysics, Grell-3 for cumulus parameterization (Grell 483 & Dévényi, 2002), the YSU scheme (Hong et al., 2006) for the planetary boundary layer 484 (PBL), and the rapid radiative transfer model for GCMs (RRTMG) for both shortwave 485 and longwave radiation (Iacono et al., 2008). As described in the previous section, di-486 rect aerosol effects are accounted for through interactions with atmospheric radiation and 487 photolysis while indirect aerosol effects are represented through interactions with cloud 488 microphysics. Dust and sea salt emissions are simulated online, following the GOCART 489 mechanism (e.g., dust_opt = 13 and seas_opt = 2). Photolysis rates of chemical species 490 are computed in a simplified version of the National Center for Atmospheric Research 491 (NCAR) Tropospheric Ultraviolet-Visible (TUV) model (phot_opt=1) (Madronich, 1987). 492 A list of the physics and chemistry schemes used in this study is summarized in Table 493 S1. 494

The anthropogenic emissions data for chemical species defined in the RACM-MADE-VBS scheme are obtained from the National Institute of Environment Research (NIER), which operates daily air quality forecasting in South Korea. These data are provided at a single level and do not include information on plume rise or vertical distribution. Biogenic emissions, on the other hand, are generated online using the Model of Emission of Gases and Aerosol from Nature (MEGAN; Version 2) (Guenther et al., 2006), but biomass burning emissions are not used in this study. All WRF files, including anthropogenic and biogenic emissions, are processed based on MODIS land use dataset (Friedl et al., 2002).

The initial and lateral boundary conditions for meteorological variables are derived from global analyses and forecasts from the UK Met Office's Unified Model (UM) operated by the Korea Meteorological Administration (KMA) every 6 hours. However, the chemical lateral boundary conditions for the outer domain are not considered, while the chemical boundaries for the inner domain (D2) are provided during the one-way nested model simulations. Upper boundary conditions are also not provided in this particular study. The impacts of chemical boundary conditions on internal physical mechanisms, such as wet scavenging, are reserved for future studies.

Hourly surface observations of PM_{2.5}, PM₁₀, SO₂, NO₂, O₃, and CO are collected 511 from 379 South Korean sites operated by AIRKOREA (http://www.airkorea.or.kr, last 512 access: 27 April 2023) and 765 Chinese sites of the China National Environmental Mon-513 itoring Center (CNMEC; http://www.cnemc.cn, last access: 27 April 2023) within model 514 domain 1 (D1). Since measurements are mainly concentrated in large cities, Korean sites 515 are randomly divided into assimilation and verification datasets. Each dataset is then 516 averaged over the 9-km model grid. As a result, 279 Korean sites are processed into 219 517 stations for assimilation, while the remaining 100 sites are averaged to form 71 indepen-518 dent observations for evaluation over South Korea. The Chinese data are used for both 519 assimilation and verification without any additional data processing because the focus 520 of this study is to examine the aerosol impact over South Korea. Figure 2 depicts the 521 surface network used for assimilation (panel (a)) and for evaluation (light blue dots in 522 (b)). In panel (b), three surface stations of the EANET (Acid Deposition Monitoring 523 Network in East Asia; https://www.eanet.asia; last access: 13 June 2023) are also rep-524 resented by a pink "+" symbol along with their respective site names, which are utilized 525 for evaluating the monthly wet deposition simulations. For meteorological data assim-526 ilation, all the conventional observations available in the National Centers for Environ-527 mental Prediction (NCEP) prepbufr data (https://rda.ucar.edu/datasets/ds337.0/; last 528

access: 13 June 2023) are employed. To verify the model performance against weather
observations, a total of 699 surface Automatic Weather System (AWS; https://www.weather.go.kr/weather/observatic
last access: 13 June 2023) sites in South Korea (marked as blue dots in Fig. 2c) are employed.

As summarized in Table S2, two baseline experiments are performed with and without aqueous chemistry (NODA and NODA_AQ, respectively). Using the background error statistics computed from each of these experiments, two corresponding DA cycling runs are then conducted with the same model configuration, assimilating surface concentrations of all six air pollutants as well as conventional meteorological data (DA and DA_AQ, respectively).

Figure 5 illustrates time series of observation-minus-background (o-b; blue solid lines) 539 and observation-minus-analysis (o-a; red dashed lines) averaged over 219 Korean stations 540 (red dots in Fig. 2) for $PM_{2.5}$ (top) and PM_{10} (bottom) at the ground level in the DA_AQ 541 cycling experiment in domain 2. The light blue shading indicates the standard devia-542 tion in (o-b)'s across the stations. Although the deviations of 6-hour background fore-543 casts from observations exhibit some fluctuations with cycles, they remain stable, and 544 the analyses closely match the observations, confirming that the system runs reliably through-545 out the entire period. As for four gas species, they tend to be slightly overestimated dur-546 ing the cycling period, but runs stably as well, as shown in the supplement (Fig. S1). 547 As this study focuses on the prediction of surface particulate matter, our discussion will 548 remain focused on surface $PM_{2.5}$ and PM_{10} concentrations. 549

Figure 6 compares the analysis and background forecast profiles of each aerosol com-550 ponent, averaged over the 71 verification sites in Korea, for the month of March 2019, 551 in two DA experiments (DA and DA_AQ). Aerosols are combined for each species regardless of the aerosol phase or mode. Panels (a) - (i) show the sum of Aitken and ac-553 cumulation mode particles, and cloud-borne (cw) aerosols are also included in the cor-554 responding chemical species in the case of DA_AQ. In the 3D-Var analysis, analysis in-555 crements in PM concentrations are distributed across aerosol species based on the back-556 ground error covariance, resulting in the vertical structure of each species generally fol-557 lowing their background error structures illustrated in Figs. 3 and 4. Therefore, even though 558 only surface concentrations are assimilated, their impact goes up to the boundary layer. 559 Comparing the analysis with and without aqueous chemistry (red and black lines, re-560 spectively), sulfate (a), ammonium (b), sea salt (k), and soil dust (l) are reduced with 561 aqueous chemistry throughout the troposphere, while nitrate (b) and chloride (i) are mostly 562 increased in the low troposphere. Primary organic carbon (f), elemental carbon (g), and 563 the unspeciated coarse fraction of PM_{10} (j) tend to be redistributed toward the surface 564 with aqueous chemistry. Regarding analysis increments, DA with aqueous chemistry seems 565 to produce slightly larger increments than without AQ, especially in coarse-mode sea salt 566 (k) in the boundary layer, followed by nitrate (b) and sulfate aerosols (a) in the accu-567 mulation mode. Averaged over the month of March 2019, the analyses of most aerosol 568 species show clear differences depending on whether aqueous chemistry is activated (e.g., 569 red vs. black). However, the analysis increments are relatively small in both DA exper-570 iments, implying that the physics mechanism plays a more crucial role in simulating at-571 mospheric composition than the initialization through data assimilation. 572

573

3.2 Air pollution events in cloudy conditions

Air pollutants transported to the Korean peninsula are susceptible to the moist environments above the Yellow Sea, which lies between Korea and China. The extent to which aerosol particles interact with moisture or cloud droplets is subject to the moving speed and direction of the synoptic weather systems, such as fronts or troughs, that traverse the sea. To examine the effect of data assimilation with aqueous chemistry, an air pollution case with substantial cloud cover and wet scavenging was selected during

the cycling period. Figure 7 (a) illustrates the Level 3 (gridded) daily mean aerosol op-580 tical depth (AOD) retrieved from the Visible Infrared Imaging Radiometer Suite (VI-581 IRS) aboard the Suomi National Polar-Orbiting Partnership (Suomi NPP) spacecraft 582 (https://ladsweb.modaps.eosdis.nasa.gov/missions-and-measurements/viirs/, last access: 583 27 April 2023) on 19 March 2019. The total column AOD indicates high aerosol load-584 ing over Korea, especially along the west coast near the Seoul Metropolitan Area. On 585 that day, a surface high pressure center is situated southwest of Jeju island (around 33.4°N, 586 $126.5^{\circ}E$), and the associated anticyclonic circulation over the sea brings air pollutants 587 and moisture into South Korea along with southwesterly winds (not shown). The areas 588 without colors in the satellite image are commonly affected by cloud contamination, re-589 sulting in missing data. 590

On the following day, a low surface pressure system further approaches, leading to 591 significant cloud cover over a wide area surrounding the Korean peninsula. As shown in 592 Fig. 7 (b), the low cloud top pressure retrieved from MODIS sensors onboard the Aqua 593 satellite (DOI: 10.5067/MODIS/MYD06_L2.061) indicates the development of convec-594 tive clouds (shown in white) over the Yellow Sea near Seoul, South Korea. This cloudy 595 condition leads to the absence or poor quality of most remote sensing retrievals across 596 the entire region, making in-situ surface measurements the primary source of observed 597 information. In association with the extensive cloud cover and long-range transport em-598 bedded in the synoptic atmospheric flows, air quality experiences large variations before and after the two days. Figure 8 displays the boxplots representing surface PM_{10} (top) 600 and $PM_{2.5}$ (bottom) concentrations for 24-hour forecasts from the analysis at 00 UTC 601 every day for five consecutive days (17-22 March 2022). Each boxplot indicates either 602 daily observations or daily forecasts from each cycling experiment at 71 verification sites. 603 The time series of observations (shown in a blue boxplot) illustrates that both PM_{10} and 604 $PM_{2.5}$ concentrations increase until 19 March 2019, accompanying the low pressure sys-605 tem with extensive clouds, and then gradually decreases in the following days. 606

The largest differences between experiments and observations are found in surface 607 PM_{10} concentrations on 20 March 2019, depending on whether aqueous chemistry (AQ) 608 is employed. The experiments with AQ (e.g., NODA_AQ in red and DA_AQ in purple) 609 accurately simulate the decreasing trend in mass concentrations owing to the wet scav-610 enging of aerosol particles. However, the experiments without AQ (e.g., NODA in or-611 ange and DA in green) predict substantial increases in concentrations. This implies that 612 the absence of aqueous chemistry in the model can lead to a statistically significant over-613 estimation of surface concentrations. Without the capability to simulate below-cloud scav-614 enging of air pollutants by precipitation, the model misses the observed wash-out fea-615 ture, predicting an opposite trend to the observed one. Surface $PM_{2.5}$ concentrations ex-616 hibit similar trends, but the differences between experiments are much smaller. Also, the 617 differences in experiments without DA (e.g., NODA-NODA_AQ) are larger than those 618 with DA (e.g., DA-DA_AQ) on 20 March, indicating that model forecasts are more vul-619 nerable to model errors when initialized without data assimilation. 620

These daily trends can provide a statistical overview of each experiment's behav-621 ior, but it would be also interesting to investigate hourly variations associated with the 622 wet scavenging process. Figure 9 depicts 48-hour forecasts from the 00 UTC analysis for 623 624 each experiment, comparing them to the observed surface PM_{10} (top) and $PM_{2.5}$ (bottom) concentrations averaged over the 71 verification sites. Hourly rainfall accumula-625 tions (top) and mean sea level pressure (bottom) are also plotted as gray dotted lines 626 with the right y-axis, averaged over 699 automated weather stations in South Korea. The 627 time series of hourly rainfall (in the top panel) shows light precipitation reported overnight 628 (from 10 to 15 UTC; 19-24 KST), while a surface low pressure system (in the bottom 629 panel) passes over Korea. As aerosol particles are removed through wet scavenging, the 630 observed PM concentrations continue to decrease until the next morning. The experi-631 ments with aqueous chemistry (NODA_AQ represented by an orange plus sign and DA_AQ 632

marked as a red circle) accurately capture the decreasing concentrations for most of the 633 48-hour forecast period. In contrast, experiments without aqueous chemistry (NODA 634 in green "x" and DA in a blue square) predict substantial increases in hourly concen-635 trations even after the rainfall. This is likely because the model simulations without AQ 636 assume all the aerosol particles as suspended in the air without any loss through wash-637 out. In this particular case, data assimilation does not make any significant differences, 638 even at the analysis time, suggesting that the forecast error is primarily driven by model 639 errors related to the parameterized wet scavenging mechanism. And it should also be 640 noted that the difference between NODA and NODA_AQ in surface PM_{10} concentrations 641 (green "x" vs. orange "+"), as the mean over all the Korean sites, corresponds to the 642 prediction of air quality changing from very unhealthy (> 150 μ g kg-1) to clear condi-643 tion (< 30 μ g kg-1), which is almost comparable to a situation of rain or no rain. This 644 suggests that wet deposition with aqueous chemistry should be included to correctly sim-645 ulate the evolution of aerosols and prevent false alarms in air quality forecasting, espe-646 cially when associated with precipitating clouds. 647

These distinctive differences are also observed in the meteorological variables. Fig-648 ure 10 displays the same 48-hour forecasts in AOD, liquid water path (LWP), planetary 649 boundary layer height (PBLH), and hourly rain accumulations. Note that this figure is 650 not intended for evaluation, but rather for demonstrating the sensitivity of these mete-651 orological fields to the wet scavenging of aerosols in the online-coupled system. The column-652 integrated AOD is expected to follow the hourly trends in surface PM_{10} concentrations 653 in each experiment. As observed in the cloud image shown in Fig. 7 (b), the LWP $(=\int_0^z \rho q_c dz)$ 654 where ρ stands for dry density, q_c cloud water mixing ratio, dz height differences between 655 two adjacent levels in the model) increases overnight. However, depending on whether 656 aqueous chemistry is activated with aqueous-phase aerosols, the simulated LWP can vary 657 by as much as three times. The PBL height is also considerably different between the 658 experiments, in association with the development of clouds and rain that affect the wet 659 removal of aerosol species. During the nighttime (around 12 UTC; 21 KST), the height 660 of the stable boundary layer exceeds 1 km in experiments without aqueous chemistry. 661 Since we lack in-situ measurements for the PBLH, we are unable to directly verify these 662 modeling behaviors. However, it is apparent that the simulated PBL height becomes sys-663 tematically higher under stable boundary conditions (e.g., at night) when aqueous-phase 664 aerosols are not considered in cloud chemistry and wet scavenging. In terms of hourly 665 total precipitation, which includes both convective and nonconvective rainfall amounts, 666 wet deposition of aerosols does not significantly alter the predicted rainfall in the cycling experiments. But the total precipitation is mostly consistent with the observation (rep-668 resented by a "x" marker) for this particular case of high cloud cover producing light pre-669 cipitation. Whether or not aqueous-phase aerosols are represented, it not only affects 670 the vertical mixing and the structure of the boundary layer but also affects surface con-671 ditions. While surface winds are not very sensitive to the use of aqueous chemistry in 672 this particular study, which is partly attributed to the large uncertainties in simulating 673 the nocturnal stable boundary layer, there are noticeable changes in 2-m temperature 674 and relative humidity, up to 5%, especially during the nighttime (not shown). As Saide 675 et al. (2015) pointed out, aerosols can play an important role in modifying severe weather 676 conditions or outbreaks. But in the weakly coupled DA system used in this study, aerosol 677 and weather data assimilation only indirectly affect each other through aerosol feedbacks 678 in the forecast model, and the assimilation of surface weather observations is not effec-679 tively performed owing to the specification of large observation errors. A thorough in-680 vestigation of the influence of aerosol data assimilation on meteorological conditions and 681 the optimization of weather data assimilation is left behind for future studies. 682

To further explore the relationship between the vertical distribution of PM concentrations and the development of clouds and rain, we examine hourly PM_{10} concentrations with a few meteorological variables at Seoul (37.5°N, 127.0°E) in 24-hour forecasts from the 00 UTC analysis on 20 March 2019 for each experiment, as shown in Fig.

11. In the WRF-Chem model V4.4.1, since aerosols can act as cloud condensation nu-687 clei, but not as ice nuclei, the simulation of aerosol-cloud-precipitation interactions is lim-688 ited to warm rain processes (Tuccella et al., 2015). Also, the background error covari-689 ance defines most of the aerosol impacts below the model level 20 (as depicted in Figs. 690 3 and 4), so it is desirable to focus on the vertical structure up to level 20. Contours of 691 cloud water and rain water mixing ratios are represented in white and pink, respectively, 692 and the LWP is overlaid as a black dashed line with the y-axis on the right. In the com-693 parison, it is evident that the use of aqueous chemistry can play an important role in the 694 formation and development of clouds as well as the wet deposition of aerosols. Without 695 AQ and DA (Fig. 11 (a)), the simulation is initiated with high concentrations of PM_{10} 696 in the boundary layer at 00 UTC (09 KST), which remains for most of the day (until 697 08 UTC; 17 KST). Clouds start to form around level 15 (\sim 5 km) in the late afternoon 698 and undergo some autoconversion and accretion processes (pink contours), but they mostly 699 persist through the night, moving down to the ground. Autoconversion refers to the pro-700 cess where cloud droplets collide and coalesce to form raindrops, while accretion denotes 701 the collection of cloud droplets by falling raindrops. With the development of low clouds 702 in this simulation, the air quality is predicted to reach very unhealthy conditions (sur-703 face $PM_{10} > 150 \ \mu g \ m^{-3}$) for the following hours starting from 18 UTC (03 KST). In Fig. 11 (b), data assimilation effectively mitigates the overestimation of low-level PM 705 concentrations from the initial time to the late afternoon. But it cannot compensate for 706 the model error resulting from the absence of the wet removal mechanism in the later 707 forecast lead times, where a similar pattern of high PM concentrations is simulated. In Fig. 11 (c), however, the activation of aerosols in clouds drastically changes the model 709 behavior to enhance rain water mixing ratios in the low troposphere (between levels 8-710 13, corresponding to 1-3 km) starting from the late afternoon (around 08 UTC) and the 711 wet removal of aerosol particles in the troposphere (around level 10) for the next 6 hours 712 or so. In the case of DA_AQ (Fig. 11 (d)), data assimilation initially suppresses the PM 713 overestimation again. Other than that, the vertical structure and temporal evolution re-714 main similar to the case without DA, highlighting the strong impact of aqueous chem-715 istry interacting with cloud microphysics and wet deposition processes on air quality fore-716 casting. As cloud droplets play an important role in aqueous-phase reactions, an accu-717 rate simulation of LWP is crucial. The time series of LWP reveals that the LWP can be 718 simulated up to three times larger when aerosols in clouds are not considered. The en-719 hanced mid-level clouds with little autoconversion of cloud to rain water droplets are closely 720 tied to the overprediction of PM concentrations. 721

The same figure for $PM_{2.5}$ concentrations, along with the PBL height, is provided 722 in the supplementary material (Fig. S2). The temporal and vertical distributions of PM_{2.5} 723 concentrations are similar to those of PM_{10} , with data assimilation suppressing initial 724 concentrations and aqueous chemistry contributing to the wet removal of aerosols overnight. 725 In the AQ experiments, cloud-borne sulfate aerosols are well overlaid with cloud water 726 mixing ratios (QCLOUD), illustrating their formation within the clouds through the pro-727 cess of aqueous chemistry. Also, the temporal variation of the boundary layer height dur-728 ing nighttime is largely influenced by the activation of aqueous chemistry. While the max-729 imum height in Seoul remains relatively consistent across the experiments, ranging from 730 1.1 to 1.5 km, the timing of the peak PBL height varies significantly. Both AQ exper-731 iments (NODA_AQ and DA_AQ) simulate the deepest boundary layer around 08 UTC 732 (17 KST), followed by a sharp decrease, which is a more realistic representation com-733 pared to the experiments without AQ, where the boundary layer continues to grow un-734 til the evening (11 UTC or 20 KST). Overall, it is apparent that clouds and precipita-735 tion can greatly alter the evolution of aerosols in the atmosphere, and aerosols can also 736 exert large influences on the cloud formation and development. Notably, whether aerosol 737 wet scavenging or removal is simulated in the model determines the prediction of PM 738 concentrations between heavy or no pollution events at the ground level. The significance 739 of aerosol wet scavenging by clouds aligns with findings from previous studies (Ryu et 740 al., 2022a, 2022b). 741

It is challenging to directly evaluate the aerosol wet removal mechanism at the pro-742 cess level, and a comprehensive investigation of all the physics mechanisms in the cou-743 pled system is beyond the scope of this study. However, we can compare our simulations 744 with the monthly observations provided by the Acid Deposition Monitoring Network in 745 East Asia (EANET). The EANET data includes monthly (dry and wet) deposition at 746 three Korean sites, as marked in Fig.2 (b). To facilitate the comparison, we have mod-747 ified the model codes to produce wet deposition for sulfate, nitrate, and ammonium aerosols 748 in the RACM-MADE-VBS-AQCHEM scheme during the model integration. Figure 12 749 shows that the daily rainfall predictions in both NODA_AQ and DA_AQ are in good agree-750 ment with the observation in Gangwha, but are overestimated in Imsil and underesti-751 mated in Jeju. Wet deposition at each site largely follows the bias in rainfall, with the 752 largest underestimation simulated in Jeju, except for sulfate in DA_AQ. Overall, data 753 assimilation tends to increase wet deposition at all three sites, slightly improving the un-754 derestimation in NODA_AQ. 755

The equilibrium between the gas and aqueous phase varies across aerosol species 756 depending on their solubility. The activation of aerosols is determined based on the hy-757 groscopicity of each aerosol component, and wet deposition is applied to individual aerosol 758 species. Hence, the processes of aerosol activation, scavenging, and chemical reactions 759 through aqueous chemistry can lead to changes in the aerosol size distribution and the 760 atmospheric composition. To investigate the impact of aqueous-phase aerosols on the 761 aerosol size distribution, Fig. 13 presents the vertical distribution of aerosol mass con-762 centrations by mode. Here, the 6-hour background forecasts from each experiment are 763 averaged over the 71 verification sites in domain 2. The top panel shows the monthly 764 mean for March 2019, while the bottom panel compares the forecasts after the aerosol 765 wet removal associated with grid-scale precipitation. In the far right in (d) and (h), PM_{10} 766 and $PM_{2.5}$ are also displayed for each experiment. In the month-long statistics, data as-767 similation mostly increases aerosol species in the accumulation mode, especially in the 768 boundary layer (panel (b)), leading to an increase in surface $PM_{2.5}$ concentrations by 769 up to 29%. On the other hand, the use of aqueous chemistry seems to have a greater im-770 pact (compared to data assimilation) on the simulation of coarse-mode aerosols. It re-771 duces the mass concentrations in the mid-to-low troposphere, contributing to the decrease 772 of PM_{10} concentrations in NODA_AQ and DA_AQ, as indicated by the lines with dots 773 in panel (d). In the Aitken mode, both aqueous chemistry and data assimilation verti-774 cally redistribute aerosol species, resulting in higher concentrations near the surface. Af-775 ter the wet removal associated with the rainfall event, however, the inclusion of aque-776 ous chemistry considerably decreases aerosol species in all three modes, regardless of data 777 assimilation. This leads to reduced PM concentrations, as illustrated in the bottom panel 778 for 00 UTC 21 March 2019. It is worth noting that without AQ, the DA experiment (rep-779 resented by blue dashed lines) attempts to reduce PM concentrations in both accumu-780 lation and coarse modes but data assimilation alone is unable to effectively reduce the 781 overestimation when there is a large model error due to the missing mechanism for the 782 below-cloud wet scavenging. 783

In terms of forecast skills over the entire month, the root-mean-square-error (RMSE) 784 of 24-hour forecasts in the 9-km domain (D2) in NODA is around 24 and 37 $\mu g \text{ kg}^{-1}$ for 785 surface $PM_{2.5}$ and PM_{10} concentrations, respectively, verified against independent ob-786 servations (e.g., at the 71 verification sites). Data assimilation improves the errors up 787 to $\sim 3 \ \mu g \ kg^{-1}$, but the use of aqueous chemistry does not change the errors in a statis-788 tically significant way. In terms of systematic normalized mean bias of 24-h forecasts (e.g., 789 $NMB = \sum (f - o) / \sum o \times 100\%$, however, surface PM_{2.5} concentrations without data 790 assimilation are significantly underestimated (up to -24 %) in both NODA and NODA_AQ, 791 which is improved by assimilation up to -4 %. Surface PM₁₀ concentrations are some-792 what overestimated by data assimilation in DA (15%), which is reduced to approximately 793 5% when aqueous chemistry is activated (e.g., DA_AQ). 794

⁷⁹⁵ 4 Conclusions and discussion

Aerosol-cloud interactions can play a crucial role in daily air quality, especially in
 regions with polluted aerosols under cloud conditions. Major air pollution events in East
 Asia are frequently associated with extensive cloud cover, which leads to the absence or
 poor quality of most remote-sensing retrievals across the entire region, making in-situ
 surface measurements the primary source of observed information.

With the goal of enhancing air quality forecasting in cloudy conditions associated with aerosol wet removals, the WRF-Chem/WRFDA system is extended for the RACM-MADE-VBS-AQCHEM scheme (chem_opt = 109) to assimilate surface measurements of PM_{2.5}, PM₁₀, SO₂, NO₂, O₃, and CO concentrations. The reliability and the effects of data assimilation using aqueous chemistry are demonstrated through regional air quality cycling where chemical and weather observations are concurrently assimilated over East Asia every 6 h from February 21 to March, 2019.

By introducing aerosols in the aqueous (or cloud water) phase in WRFDA, the regional cycling system using the online-coupled forecast model could represent the belowcloud wet scavenging of aerosol particles, leading to changes in PM concentrations reaching the ground. The use of aqueous chemistry for aerosol-cloud interactions requires a double-moment microphysics, for which the Morrison two-moment scheme is employed along with Grell-3 cumulus and the RRTMG short- and long-wave schemes.

For the entire cycling period, a time series of innovations ((o-b)'s) in surface PM concentrations remains stable, confirming the successful implementation of the data assimilation system for the particular chemistry option (chem_opt = 109).

Although only surface mass concentrations are assimilated, the impact was recog-817 nized throughout the troposphere based on the background error statistics of each aerosol 818 species. Both aqueous chemistry (AQ) and data assimilation (DA) systematically changed 819 the atmospheric composition and its vertical structure, increasing nitrate in the accu-820 mulation mode and sea salt aerosols in the coarse mode near the surface. During the month 821 of March 2019, data assimilation tend to considerably increase aerosol species in the ac-822 cumulation mode within the boundary layer while aqueous chemistry significantly re-823 duces coarse-mode aerosol particles up to mid troposphere. When accounted for in clouds, 824 sulfate aerosols experience large increases in the aqueous phase, in association with the 825 oxidation in low-level clouds. 826

In a pollution event with high cloud cover, data assimilation with aqueous chem-827 istry was particularly helpful in simulating wet deposition of aerosols to accurately pre-828 dict the evolution of surface PM concentrations. As the activation, resuspension, and 829 wet scavenging processes in association with cloud chemistry can be all simulated only 830 when aerosols in cloud water are defined through aqueous chemistry, DA without aque-831 ous chemistry treated all the aerosols as interstitial (e.g., suspended in the air) even when 832 precipitation occurred, leading to a significant overestimation of surface PM concentra-833 tions. At that time, large LWP was also produced over a wide range of the domain, demon-834 strating that the formation and development of clouds were also largely affected by aerosols 835 in the aqueous phase. 836

The use of aqueous chemistry in the aerosol cycling system is beneficial from sev-837 eral perspectives. First, when air pollutants are transported with significant cloud cover, 838 aerosols can be considerably affected by clouds, and in turn, they can influence cloud de-839 velopment and properties. As shown in the presented case study, such interactions can 840 affect wet scavenging of aerosol particles, significantly reducing false alarms in surface 841 PM concentrations. Second, cloudy conditions are generally hard to observe, especially 842 with remote-sensing data, making data assimilation more challenging or susceptible to 843 cloud contamination. In the strong constraint variational data assimilation system, where 844 model errors are not taken into account, the solution (e.g., analysis) might be subopti-845

mal when model errors become large. In other words, reducing model errors through advanced physics mechanisms (such as aqueous chemistry along with cloud chemistry and
wet scavenging) can make data assimilation more effective, as the assumption of no model
errors is violated to a lesser extent. Lastly, the effects of the analysis can accumulate over
time through cycling to make systematic improvements in air quality forecasting.

Aqueous chemistry currently implemented in the WRF-Chem model is designed 851 for warm-rain processes by treating aerosols only in the cloud-water phase. And aque-852 ous chemistry implemented for the RACM-MADE-VBS scheme in WRF-Chem does not 853 account for all the complex aqueous-phase reactions, either. For example, the contribu-854 tion to SOA concentration by cloud chemistry is missing, and aerosol interactions with 855 ice nuclei are not taken into account in this version of the model (Tuccella et al., 2015). 856 As such, the aqueous chemistry used in the chemical option might be overly simple to 857 represent all the physical processes for indirect aerosol effects, especially for mixed-phase 858 convective clouds with nonprecipitating supercooled liquid water near cloud tops (Rosenfeld 859 et al., 2008). 860

In the spring case examined here, however, the new chemistry option was clearly 861 helpful to simulate the reduction of PM concentrations due to wet removal of aerosol par-862 ticles. In cloudy conditions that do not result in precipitation, however, enhanced aerosol 863 concentrations in the atmosphere can act to reduce the mean size of cloud droplets and 864 suppress coalescence and warm-rain processes, while enhancing the growth of large hail 865 and cold-rain processes. Those cases cannot be simulated in the model with such sim-866 ple aqueous chemistry, which can mislead the analysis. As the strong-constraint 3D-Var 867 system used in this study does not include any specific model error term, model errors 868 are not investigated nor discussed in detail, but there is room for further improvements 869 for the RACM-MADE-VBS-AQCHEM option in the model to account for more sophis-870 ticated aerosol effects in clouds, radiation, and precipitation. 871

On the other hand, in the WRFDA system developed for this study, aerosol num-872 ber concentrations are not included as part of analysis (or control) variables so that they 873 are not changed through the assimilation. To fully describe aerosol impacts on clouds 874 or to handle complex cases with mix-phased clouds or cold-rain processes, we might need 875 to consider developing the assimilation system to reflect the changes in aerosol number 876 concentrations per aerosol (size) mode. Recently, cloud properties and/or atmospheric 877 constituents have increasingly been measured or derived from multiple platforms on ground-878 and space-born satellites. Two notable examples are the National Aeronautics and Space 879 Administration (NASA)'s Tropospheric Emissions Monitoring of Pollution instrument 880 (TEMPO; https://tempo.si.edu/index.html, last access: 5 July 2023) and South Korea's 881 Geostationary Environment Monitoring Spectrometer (GEMS; Kim et al. (2020)) on-882 board the Geostationary Korean Multi-purpose Satellite 2B (GEO-KOMPSAT-2B; https://nesc.nier.go.kr/product/vi 883 last access: 5 July 2023). Needless to say, these data can be extremely valuable to not 884 only evaluate the modeling system but also to better initialize the model through data 885 assimilation. 886



Figure 1. Flowchart of the WRF-Chem/WRFDA cycling system for chemical data assimilation. Gridded input data is indicated by rectangular boxes on the left, and all the software packages are filled in colors. Dotted lines imply optional input data, while solid lines are mandatory inputs for WRF-Chem/WRFDA cycling, accompanied by typical input file names (without specification of domain ID and time) used in the WRF system.

⁸⁸⁷ 5 Open Research

All meteorological observations assimilated in this study are obtained from the Na-

tional Centers for Environmental Prediction (NCEP) prepbufr data (https://rda.ucar.edu/datasets/ds337.0/)

- while the monthly wet deposition observations at Korean EANET (Acid Deposition Mon-
- itoring Network in East Asia) sites are read from the website (https://monitoring.eanet.asia/document/public/index/
- ⁸⁹² for March 2019. Chemical measurements at the surface are collected from http://www.airkorea.or.kr
- for Korean sites and from http://www.cnemc.cn for Chinse sites. The observation data
- processed for the experiments presented in this study are available in Ha (2023b). The
- emissions data used in the case study are accessible in Ha (2023a). The base version (V4.4.1)
- of the WRF system is publicly released in https://github.com/wrf-model/WRF/releases/tag/v4.4.1,
- and the updated codes introduced for the new features in WRF-Chem/WRFDA are ac-
- cessible in Ha (2023c). Figure 1 is produced in Microsoft Powerpoint and all other fig-
- ⁸⁹⁹ ures are produced in Python V3.8.13. VIIRS Level 3 gridded AOD data in netCDF4 are
- downloaded and displayed using the scripts provided in https://www.star.nesdis.noaa.gov/smcd/emb/viirs_aerosol/sof which is publicly available.

-19-



Figure 2. Surface observing network used in this study: a) Assimilated observation sites are marked as dots in red (black) for Korean (Chinese) sites, with terrain height colored in domain 1 (at 27 km resolution) and a black box over the Korean peninsula indicating domain 2 (D2; 9 km resolution) b) 71 evaluation sites are marked in light blue dots while three EANET sites - Ganghwa, Imsil, and Jeju - are displayed in pink ("+" sign) along with their site names and c) the Korean automated weather stations (AWS) measuring meteorological variables at the surface are shown in blue dots.



Figure 3. Vertical profile of each aerosol species in background error standard deviation estimated with and without aqueous chemistry (AQ and NO_AQ, respectively) over domain 2 (D2). Accumulation mode aerosols in AQ (NO_AQ) are depicted in red lines with dots (black solid lines) while Aitken mode aerosols in AQ (NO_AQ) in dashed orange (dotted gray) lines.



Figure 4. Vertical profile of background error standard deviation with aqueous chemistry (AQ) for aerosol species in the aqueous or cloud water ("cw") phase in domain 2. Domain-averaged height ([km]), corresponding to the model levels, is displayed on the right y-axis.



Figure 5. Time series of (o-a; red dashed lines) and (o-b; blue solid lines) in $PM_{2.5}$ (top) and PM_{10} concentrations (bottom) on the ground in DA_AQ in domain 2. The lines indicate the mean over all the assimilated stations in South Korea while the light blue shading area shows standard deviation in (o-b)'s across the surface stations. The numbers in the legend indicate the mean over the entire cycling period from Feb 21 to March 31, 2019.

Scheme	Option	Namelist_option
Chemistry driver	RACM-MADE-VBS-AQCHEM	$chem_opt = 109$
Photolysis	Madronich	$phot_opt = 1$
Grid-scale Wet scavenging	On	wetscav_onoff = 1
Grid-scale Cloud chemistry	On	$cldchem_onoff = 1$
Convective transport and wet deposition	On	$conv_tr_wetscav = 1$
Convective-scale aqueous chemistry	On	$conv_tr_aqchem = 1$
Dust emissions	On	$dust_opt = 13$
Sea salt emissions	On	seas_opt = 2
Prognostic Number Concentration	On	progn = 1
Microphysics	Morrison 2-moment	$mp_physics = 10$
Longwave radiation	RRTMG	$ra_lw_physics = 4$
Shortwave radiation	RRTMG	$ra_sw_physics = 4$
Surface layer	Monin–Obukhov	$sf_sfclay_physics = 1$
Land surface	Noah	$sf_surface_physics = 1$
Boundary layer	YSU	$bl_pbl_physics = 1$
Cumulus parametrization	Grell-3D	$cu_physics = 5$

Table S1. Physics and chemical options



Figure 6. Vertical profile of each aerosol species ($[\mu g/kg-dryair]$) in the analysis and background forecast in domain 2 with and without aqueous chemistry (DA_AQ and DA, respectively), averaged over 71 verification sites in Korea (light blue dots in Fig.2) from 6-hr cycling for the period of 1-31 March 2019.



Figure 7. (a) Aerosol optical depth retrieved from VIIRS onboard the Suomi NPP as a daily mean gridded data (level 3) on Mar 19, 2019 (b) Level 2 cloud top pressure retrieved from the MODIS sensors onboard the Aqua satellite, merged between 04:35:12 and 04:40:12 UTC on 20 March 2019.



Figure 8. Time series of boxplots for daily PM simulations in each experiment compared to observations. Each day displays 24-hour forecasts initiated from the 00Z analysis in each experiment at 71 verification sites in Korea.

Table S2. Experiments. DA stands for data assimilation and AQ aqueous chemistry.

Experiment	$chem_opt$	Assimilation
NODA	108	None
DA	108	CHEM+MET
NODA_AQ	109	None
DA_AQ	109	CHEM+MET


Figure 9. Time series of (top) surface PM_{10} and (bottom) $PM_{2.5}$ concentrations for 48 h forecasts from 00 UTC 20 March 2019, averaged over 71 Korean verification sites (marked in Fig. 2 (b)). In-situ observations (OBS; black triangle) are compared with hourly forecasts from their 00Z analysis in each cycling experiment. Gray dotted lines with the right y-axis are (top) hourly rainfall (mm) and mean sea level pressure (hPa) observations (bottom) averaged over 699 AWS sites over South Korea (marked in Fig. 2 (c)).



Figure 10. Same as Fig.9, but for (a) AOD (b) LWP (c) PBLH and (d) hourly rain accumulation for each experiment. In (d), hourly observations averaged over 699 AWS sites are also marked as a "x" symbol.



Figure 11. Time series of 24 h forecasts of PM_{10} concentrations (filled) simulated at Seoul, South Korea (in model levels up to 20) in each experiment. Cloud and Rain water mixing ratios (QCLOUD and QRAIN ([g kg-1])) are contoured in white and pink, respectively. Liquid Water Path (LWP) is also plotted in black dashed lines with the right y-axis.



Figure 12. Monthly data for March 2019, comparing two AQ experiments with measures from three EANET sites (marked in Fig. 2 (b)) in (a) daily rain accumulation and wet deposition in (b) SO4 (c) NO3 and (d) NH4.



Figure 13. Vertical profile of aerosol species in each mode from (a)-(c) and (e)-(g). $PM_{2.5}$ and PM_{10} are displayed in (d) and (h). Upper panels show the monthly mean for March 2019 while lower panels are 6 hour forecasts from the analysis at 18 UTC 20 March 2019.



Figure S1. Same as Fig.5, but for four gas species.



Figure S2. Same as Fig.11, but for $PM_{2.5}$ (filled). The sum of cloud-borne sulfate aerosols in the accumulation and Aitken modes (e.g., so4cwj+so4cwi) is contoured in pink, superimposed with cloud water mixing ratios (QCLOUD ([g kg-1]); white solid lines). The PBL height is also plotted in black dashed lines with the y-axis on the right.

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Figure 1.



Figure 2.



Figure 3.







Figure 4.



Figure 5.



Figure 6.



Figure 7.

S-NPP/VIIRS 0.10° Aerosol Optical Depth 19 Mar 2019



MYD06_L2.A2019079.0435.061.2019079194242.hdf 2019-03-20_04:35:12 - 2019-03-20_04:40:12 UTC

Figure 8.

PM10 [μ g m^{-3}]



Figure 9.



Figure 10.



Figure 11.



Figure 12.


Figure 13.

F6H mean for March 2019

