The temperature and composition of the mantle sources of Martian basalts

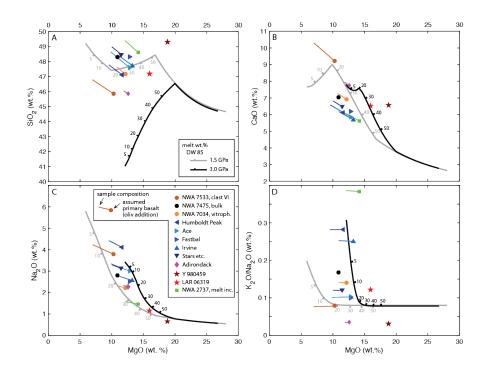
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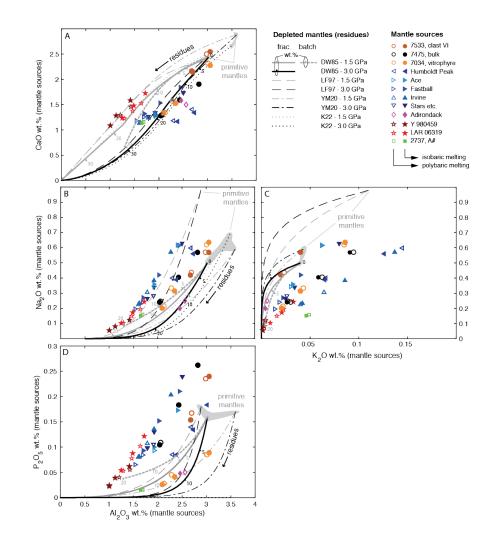
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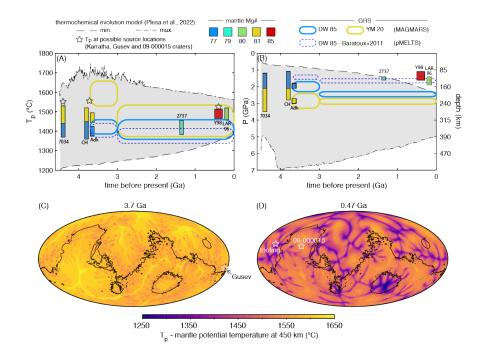
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Abstract

The composition of basaltic melts in equilibrium with the mantle can be determined for several Martian meteorites and in-situ rover analyses. We use the melting model MAGMARS to reproduce these primary melts and estimate the bulk composition and temperature of the mantle regions from which they originated. We find that most mantle sources are depleted in CaO and Al2O3 relative to models of the bulk silicate Mars and likely represent melting residues or magma ocean cumulates. The concentrations of Na2O, K2O, P2O5, and TiO2 are variable and often less depleted, pointing to the re-fertilization of the sources by fluids and low-degree melts, or the incorporation of residual trapped melts during the crystallization of the magma ocean. The mantle potential temperatures of the sources are 1400-1500 ^oC, regardless of the time at which they melted and within the range of the most recent predictions from thermochemical evolution models.







The temperature and composition of the mantle sources of Martian basalts

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Key Points:

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10	•	Basalts that sampled discrete mantle regions throughout Mars's history provide
11		information about the mantle composition and temperature
12	•	The mantle potential temperature of primitive basalts appears constant (1400–
13		1500 °C), yet is likely not representative of the average mantle
14	•	Incompatible element concentrations in the mantle vary due to magma ocean crys-
15		tallization, partial melting and metasomatism

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16 Abstract

¹⁷ The composition of basaltic melts in equilibrium with the mantle can be determined for

¹⁸ several Martian meteorites and in-situ rover analyses. We use the melting model MAG-

¹⁹ MARS to reproduce these primary melts and estimate the bulk composition and tem-²⁰ perature of the mantle regions from which they originated. We find that most mantle

sources are depleted in CaO and Al_2O_3 relative to models of the bulk silicate Mars and

likely represent melting residues or magma ocean cumulates. The concentrations of Na₂O,

 K_2O, P_2O_5 and TiO₂ are variable and often less depleted, pointing to the re-fertilization

of the sources by fluids and low-degree melts, or the incorporation of residual trapped

²⁵ melts during the crystallization of the magma ocean. The mantle potential temperatures

of the sources are 1400–1500 °C, regardless of the time at which they melted and within

the range of the most recent predictions from thermochemical evolution models.

²⁸ Plain Language Summary

Martian meteorites and rocks analyzed by rovers are witnesses of magmatic pro-29 cesses on Mars. In this study, we use the mantle melting model MAGMARS to deter-30 mine the composition and temperature of the mantle regions from which primitive basalts 31 have originated. Primitive basalts are closely related to mantle melts and hence record 32 the properties of their mantle source. We find that the mantle compositions needed to 33 explain these melts were poor in CaO and Al_2O_3 . They likely represent a mantle that 34 melted on several occasions or that crystallized from an early magma ocean. The com-35 position of these primitive basalts indicates that some elements (Na₂O, K₂O, P₂O₅ and 36 TiO_2) were subsequently added to the mantle source by fluids and low-degree melts. Al-37 ternatively, these elements can be explained by the trapping of melts during the evolu-38 tion and progressive crystallization of the magma ocean. The temperature of the man-39 the sources projected to the surface conditions for easier comparison, (i.e., potential tem-40 perature) was 1400–1500 °C, regardless of the time at which these sources melted and 41 is within the range of the most recent predictions from planetary-scale models of inte-42 rior dynamics. 43

44 **1** Introduction

Our knowledge of the thermal state, composition and structure of the Martian man-45 tle is derived from a diverse and continuously expanding array of geophysical and geo-46 chemical constraints. Early measurements of the moment of inertia factor, soil compo-47 sitions at the Viking landing sites, and the definitive recognition that the "SNC mete-48 orites" are from Mars (Baird et al., 1976; Johnston & Toksöz, 1977; Bogard & Johnson, 49 1983), unequivocally pointed to a FeO-rich mantle $(Mg/(Fe+Mg)\times 100 \text{ in moles or } Mg\#$ 50 = 75-81) compared to Earth (90). Model compositions of the "primitive mantle" were 51 rapidly put forth (e.g., Dreibus & Wänke, 1985) and allowed to create simple models of 52 the Martian interior structure (Longhi et al., 1992; Bertka & Fei, 1997; Elkins-Tanton 53 et al., 2003). Additional analyses of crustal rocks by subsequent orbiting probes and rovers, 54 the discovery of new Martian meteorites (Agee et al., 2013; Humayun et al., 2013), geode-55 tic and seismic data from the recent InSight mission (e.g., Khan et al., 2021; Huang et 56 al., 2022), and geodynamic modeling (e.g., Plesa et al., 2022), are now allowing to draw 57 ever improving representations of the interior structure of Mars and its evolution through 58 time. 59

Currently available compositions of the Martian mantle (e.g., Dreibus & Wänke, 1985; Lodders & Fegley, 1997; Yoshizaki & McDonough, 2020; Khan et al., 2022, abbreviated as DW85, LF97, YM20 and K22 hereinafter) represent average and idealized primitive compositions that are useful to derive average characteristics (density, solidus temperature, seismic wave velocity, etc.) but that probably do not represent actual regions of the mantle. The study of Martian meteorites has long shown that the mantle is highly heterogeneous—both in terms of isotopic composition and Mg#—and suggests that a
significant portion of the crust was formed very early (20–100 Myr; e.g., Borg et al., 1997;
Debaille et al., 2008; Humayun et al., 2013; Nyquist et al., 2016; Kruijer et al., 2017; Bouvier et al., 2018) during (or briefly after) the crystallization of a Martian Magma Ocean
(MMO). However, the major-element composition of the mantle reservoirs formed during the early differentiation of Mars is poorly constrained and model-dependent (e.g.,
Borg & Draper, 2003; Elkins-Tanton et al., 2005).

To derive more detailed models of the interior structure of Mars, independent con-73 straints on the composition and temperature of discrete regions of the Martian mantle 74 are desirable. A subset of Martian basalts, characterized by varied crystallization ages 75 and high Mg# have been suggested to represent primitive basalts in near-equilibrium 76 with their mantle sources and have been used to determine the P-T conditions of their 77 mantle source through experiments (Musselwhite et al., 2006; Monders et al., 2007; Fil-78 iberto et al., 2008; Filiberto, Dasgupta, et al., 2010; Filiberto, Musselwhite, et al., 2010) 79 or modeling (Lee et al., 2009; Filiberto & Dasgupta, 2011, 2015; Filiberto, 2017; Bara-80 toux et al., 2011; Balta & McSween, 2013a). Most of these basalts cannot be produced 81 by melting the primitive mantle and are instead expected to derive from mantle sources 82 of diverse compositions (e.g., Schmidt & McCoy, 2010; Collinet et al., 2015, Fig. 1). 83

Here, we use MAGMARS, a new model developed to simulate melting in the Mar-84 tian mantle (Collinet et al., 2021), to re-evaluate the melting conditions and the ther-85 mal state of the mantle sources of primitive Martian basalts, which crystallized at dif-86 ferent times and therefore represent snapshots of Mars' thermochemical evolution. In 87 addition, MAGMARS allows us to estimate for the first time the major-element com-88 position of these local mantle sources. We find that the P-T melting conditions appear 89 to have remained relatively stable through time and that mantle sources display vari-90 able CaO/Al_2O_3 , low overall abundances of incompatible elements but enrichment of al-91 kalis, P and Ti relative to Ca and Al. We discuss the implications of these findings for 92 the early differentiation of Mars and its long-lived magmatism. 93

⁹⁴ 2 Selected compositions of primitive Martian basalts

While the majority of mantle melts were modified by igneous differentiation as they 95 ascended through the crust (Udry et al., 2018; Payré et al., 2020; Ostwald et al., 2022; 96 Farley et al., 2022; Wiens et al., 2022), a limited number of Martian basalts bear wit-97 ness to the composition and temperature of the mantle at the time of their formation 98 (i.e., primitive basalts). To identify primitive basalts, we first make the assumption that qq the average Martian mantle contains olivine $Mg\# \ge 77$ (Table 1, Table S1 and Fig. 1), 100 and would produce primary melts with a Mg# ≥ 54 ($K_{D, \, \text{Fe}-Mg}^{\text{oliv}-\text{liq}}$ of 0.35; Filiberto & Das-101 gupta, 2011). A mantle of Mg# 77 is intermediate between the most commonly accepted 102 primitive mantle compositions (Dreibus & Wänke, 1985; Yoshizaki & McDonough, 2020). 103 Here, we only consider martian basaltic compositions with a Mg $\# \geq 48$, which could de-104 rive from primary mantle melts of Mg# ≥ 54 following a maximum of 10 wt.% of olivine 105 fractionation. 106

The Spirit rover analyzed numerous basalts with Mg# 48–55 at Gusev crater (McSween, 107 Wyatt, et al., 2006; Squyres et al., 2007; Ming et al., 2008) that could represent prim-108 itive basalts (Monders et al., 2007; Filiberto, Dasgupta, et al., 2010; Schmidt & McCoy, 109 2010). Among these, the Adirondack-class basalts are poor in K_2O and could derive from 110 a residual mantle depleted in incompatible elements by prior melting events (Schmidt 111 & McCoy, 2010; Collinet et al., 2021) while most of the basalts analyzed in the vicin-112 ity of the Columbia Hills are more enriched in alkali elements and poorer in CaO (Fig. 113 1). The ancient regolith breccia NWA 7034/7475/7533 (Humayun et al., 2013; Nyquist 114 et al., 2016; Cassata et al., 2018; Bouvier et al., 2018) is also characterized by a high Mg#115 (54; Wittmann et al., 2015) and, despite its complex history, could approach the com-116

position of a mantle melt based on trace (Humayun et al., 2013) and major element compositions (Collinet et al., 2015). We also test whether two individual clasts could be representative of primitive basalts later remelted by impacts: a vitrophyre (Udry et al., 2014)
and an alkali-rich microbasalt known as "Clast VI" (Humayun et al., 2013).

Recent geophysical constraints suggest that large portions of the mantle could be 121 more Mg-rich (Mg# = 81; Khan et al., 2022) than previously assumed (e.g., Dreibus & 122 Wänke, 1985; Yoshizaki & McDonough, 2020), as also evidenced by the study of Mar-123 tian meteorites. The most primitive depleted shergottite (Yamato 980459, nearly iden-124 tical to NWA 5789; Greshake et al., 2004; Gross et al., 2011) and the most primitive en-125 riched shergottite (LAR 06319, nearly identical to NWA 1068; Barrat et al., 2002; Pes-126 lier et al., 2010) have Mg# of 66 and 58, respectively. Y 980459 contains olivine Mg# 127 85–86 and is thought to represent a primary melt composition (e.g., Musselwhite et al., 128 2006; Matzen et al., 2022). The olivine megacrysts in LAR 06319 and NWA 1068 have 129 $Mg\# \leq 77$ (Basu Sarbadhikari et al., 2009) but were initially more magnesian (Mg# 80) 130 and were modified by Fe–Mg diffusion (Balta et al., 2013; Collinet et al., 2017). NWA 131 2737 is a dunitic cumulate (Mg# 79) with olivine-hosted melt inclusions. Its primary 132 melt is taken as the reconstructed composition of the parental trapped liquid (PTL; He 133 et al., 2013). Given the multitude of evidence of Mg-rich mantle reservoirs, we also cal-134 culated alternative primary melt compositions for the Gusev basalts and NWA 7034/7475/7533135 bulk rock and basaltic clasts that would be in equilibrium with a Mg# of 81. In this case, 136 larger amounts of olivine have to be added to the parental melt compositions (Table 1). 137

		age (Ga)	oliv (wt.%)	Mg#
NWA 7034	Vitrophyre [1]	4.49[2]	+10 / +26	77 / 81
NWA 7533	Clast VI [3]	4.49[2]	+9 / +24	77 / 81
NWA 7475	bulk [4]	4.49[2]	0	77
Adirondack-class basalts [5]		3.7[6]	+3 / +17	77 / 81
Columbia Hills	Humboldt Peak [7]	3.7[6]	+7 / + 20	77 / 81
	Fastball [8]	3.7[6]	0 / +13	77 / 81
	Stars, etc. [8]	3.7[6]	+5 / +17	77 / 81
	Ace [8]	3.7 [6]	+9 / +29	77 / 81
	Irvine [7]	3.7[6]	+8 / +25	77 / 81
chassignite	NWA 2737 [9]	1.3 [10]	+9	79
depleted shergottite	Y 980459 [11]	0.47 [12]	0 / +7	85 / 86
enriched shergottite	LAR 06319 [13]	$0.19 \ [14]$	0 / +5	80 / 81

Table 1. List of Martian primitive basalts, fraction of olivine addition required to reach mantle-melt equilibrium, and associated inferred mantle Mg#

Udry et al. (2014), [2] Costa et al. (2020), [3] Humayun et al. (2013), [4] Wittmann et al. (2015)
 McSween, Wyatt, et al. (2006), [6] Greeley et al. (2005), [7] Ming et al. (2008)

[8] Squyres et al. (2007), [9] He et al. (2013), [10] Udry and Day (2018)

[11] average of Misawa (2004), Shirai and Ebihara (2004) and Greshake et al. (2004)

[12] Shih et al. (2005), [13] Basu Sarbadhikari et al. (2009), [14] Shafer et al. (2010)

138 3 Methods

To constrain the mantle sources of the target basaltic compositions described above (Table 1 and S1), we first simulate the melting of various primitive mantle compositions (DW85, YM20 and K22) using MAGMARS (Collinet et al., 2021). We then adjust the mantle compositions incrementally (Mg#, TiO₂, Al₂O₃, CaO, Na₂O, K₂O, and P₂O₅ concentrations) until the liquids produced are identical to the target compositions (i.e.,

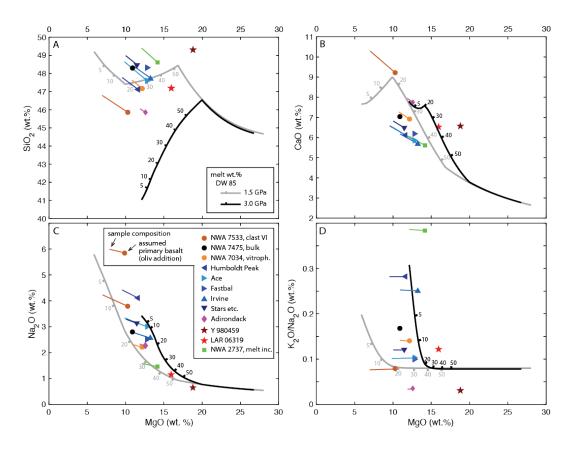


Figure 1. Comparison between the composition of Martian primitive basalts (left extremity of colored lines), their recalculated primary melts (symbols) and the melts produced by melting of the primitive mantle of Dreibus and Wänke (1985) at 1.5 (grey line) and 3.0 GPa (black line), as calculated by MAGMARS. The high SiO₂ content of primary melts (A) is consistent with shallow melting conditions (\ll 3.0 GPa). But compared to shallow DW85 melts (1.5 GPa), many primary basalts have either lower or higher CaO contents (B) and higher Na₂O and K₂O contents (C-D), and must therefore derive from mantle sources of contrasting compositions.

when the concentrations of all major and incompatible elements are within 1 wt.% rel-144 ative). Next, we mathematically remove a fraction (33 to 50 wt.%) of the melt (of com-145 position identical to the target compositions) and repeat the same procedure to iden-146 tify more refractory mantle compositions that can still produce identical melts. This ap-147 proach, in the absence of independent constraints on the melt fraction, leads to the iden-148 tification of several possible mantle sources for each target composition. To discuss the 149 non-uniqueness of the sources and quantify model uncertainties for the Fastball primary 150 melt (representative example), we performed \sim 500000 MAGMARS calculations by ran-151 domly varying the parameters around their average values. This automated search iden-152 tified slightly larger compositional trends compared to the manual search. However, the 153 mantle sources identified manually were found sufficient to discuss the mantle source ori-154 gin and melting temperature. It is this dataset (Table S2) that is described in the fol-155 lowing sections. 156

157 4 Results

The compositions of the mantle sources that can produce melts identical to the tar-158 get primary basalts (Table 1 and Fig. 1) are shown in Figure 2 and reported in Table 159 S2. Each primary basalt composition can be matched by melting a series of mantle sources 160 characterized by various concentrations of incompatible elements (Al₂O₃, CaO, Na₂O, 161 K_2O), both isobarically and polybarically. Despite the non-uniqueness of solutions, first-162 order chemical differences between the sources of the different basaltic compositions can 163 be identified. For example, the possible sources of shergottites are all notably poorer in 164 Al_2O_3 and Na_2O than the sources of the Gusev basalts (Fig. 2a). Among the latter, the 165 sources of the Columbia Hills basalts are characterized by high Na_2O , K_2O , and P_2O_5 166 concentrations (Fig. 2b-d) compared to the source of the Adirondack basalts. The source 167 of the NWA 2737 chassignite shows the highest K₂O/Na₂O ratio. Finally, one of the sources 168 that can match the composition of Clast VI (NWA 7533) is nearly identical to the DW85 169 primitive mantle. 170

The melt fractions required to produce the primary basalt compositions are com-171 prised between 5 and 30 wt.%. The associated mantle potential temperatures $(T_{\rm p})$ are 172 between 1320 and 1520 °C (Fig. 3a and Table S2). The average pressure of melting is 173 relatively low for all samples (1.1-2.0 GPa), and is largely constrained by the SiO₂ and 174 MgO concentrations of the target primary melts (Fig. 1a). If a Mg# of 81 (K22) is as-175 sumed instead of 77 for NWA 7034/7475/7533 and Gusev basalts, then the primary basalts 176 would contain a larger olivine component and the mantle $T_{\rm p}$ (1390–1570 °C) and aver-177 age pressure of melting (1.9–3.0 GPa) would both be higher (Fig. 3). 178

¹⁷⁹ 5 Discussion

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5.1 Thermal state of the Martian mantle

¹⁸¹ Compared to the $T_{\rm p}$ estimates of Filiberto (2017), and using the same starting as-¹⁸² sumptions (mantle of Mg# 77 and batch melting), we find that Gusev crater basalts are ¹⁸³ derived from slightly cooler mantle sources on average, with $T_{\rm p}$ of ~1400 °C (vs. ~1450 ¹⁸⁴ °C), but that the ranges of possible $T_{\rm p}$ largely overlap (1360–1460 vs. 1390–1550 °C, re-¹⁸⁵ spectively). Allowing for a higher Mg# of the mantle sources (77–81), we find that Gu-¹⁸⁶ sev basalts and all (pre-)Noachian to Hesperian samples point to a $T_{\rm p}$ of 1340–1520 °C ¹⁸⁷ (Fig. 3a).

¹⁸⁸ We calculate a $T_{\rm p}$ of 1420–1430 °C for the primary melt composition reconstructed ¹⁸⁹ from NWA 2737 melt inclusions (He et al., 2013), assumed to be parental to the middle-¹⁹⁰ Amazonian nakhlites and chassignites (1.34 Ga; Udry & Day, 2018). However, the man-¹⁹¹ tle source could have been metasomatized (Day et al., 2018, also see section 5.2) and could

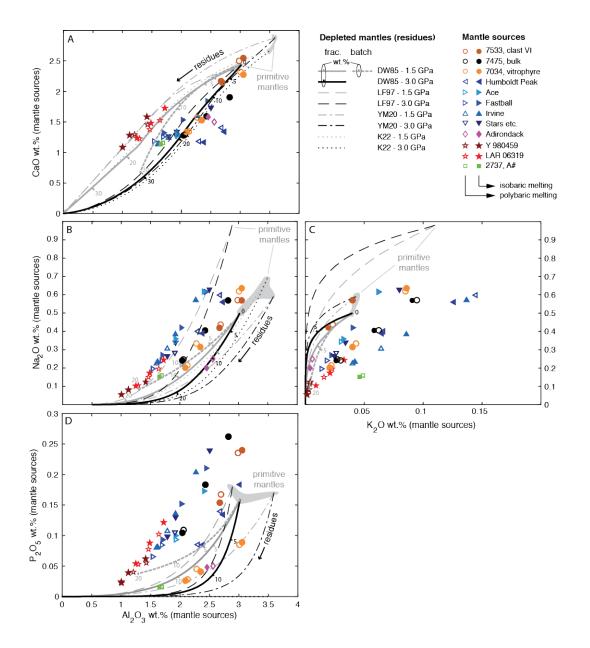


Figure 2. Incompatible element concentrations of the mantle sources of primary basalts (symbols) compared to residual model Martian mantles (lines). Each line represents the trajectory of residues produced by progressive melting of a primitive mantle composition (apex) at 1.5 (grey) and 3.0 GPa (black). For the DW85 model (solid lines), tick marks indicate the composition of residues after specific degrees of melting (in wt.%). All mantle source compositions are relatively poor in CaO and Al₂O₃ compared to the primitive mantle and are characterized by variable CaO/Al₂O₃ (A). The source of NWA 7533 clast VI is similar to a primitive mantle (DW85) and the source of the Adirondack basalts resemble a residual mantle following prior partial melting in all compositional spaces (A–D). All other sources are too rich in alkali elements—Na₂O (B) and especially K₂O (C)—and other incompatible elements, such as P_2O_5 (D), to derive from model Martian mantles by partial melting alone and other processes must be considered (see section 5.2).

have contained up to 250 ppm of water (McCubbin et al., 2016), which would translate into a lower T_p of 1380 °C (Katz et al., 2003).

The more recent olivine-phyric shergottites (160-500 Ma; Moser et al., 2013; Wu 194 et al., 2021; McFarlane & Spray, 2022) are picritic basalts that have been linked to plumes 195 with a $T_{\rm p}$ of at least 1480–1550 °C (e.g., Musselwhite et al., 2006; Filiberto & Dasgupta, 196 2015). The $T_{\rm p}$ of MAGMARS simulations (1470–1520 °C) are within error of these pre-197 vious constraints if batch melting is assumed, and slightly lower in the polybaric case 198 (1440–1450 °C). The presence of water in the source of shergottites could in principle lower 199 the minimum $T_{\rm p}$ and has been suggested to account for their relatively high SiO₂ con-200 centrations (Balta & McSween, 2013b). However, the small initial water concentration 201 of the source (14-73 ppm; McCubbin et al., 2016) and the fact that the SiO₂ concen-202 tration of shergottite melts can be reproduced with MAGMARS under nominally an-203 hydrous conditions preclude a significant effect of water. 204

Finally, we re-calculate using MAGMARS the $T_{\rm p}$ and pressures of melting of the 205 bulk volcanic provinces of Baratoux et al. (2011), as constrained by the Gamma Ray Spec-206 trometer (GRS) on board NASA's Mars Odyssey spacecraft. Baratoux et al. (2011) used 207 pMELTS in their analysis, which has since been shown to overestimate FeO and under-208 estimate SiO_2 concentrations by up to 8 wt.% (Collinet et al., 2021), significantly more 209 than anticipated by El Maarry et al. (2009). For Hesperian provinces, while the ranges 210 of $T_{\rm p}$ are similar (1390–1460 vs. 1370–1420 °C previously), MAGMARS predicts a slightly 211 higher pressure of melting (1.6–2.3 vs. 1.3–1.6 GPa). However, we find that only Ascraeus 212 and Elysium Mons (out of the 6 Amazonian volcanic provinces) can be matched with 213 a DW85 mantle composition using MAGMARS (Table S3). The composition of the other 214 215 4 provinces can either not be reproduced at all (Arsia and Pavonis Mons) or only with an extremely small melt fraction of <2 wt.% (Olympus Mons and Alba Patera). With 216 a YM20 composition (Mg# of 79, 81 after 15 wt.% of melting), a higher $T_{\rm p}$ of 1520–1660 217 $^{\circ}$ C and higher pressures of melting (2.3–3.5 GPa) are necessary to match the Hesperian 218 volcanic provinces. A higher Mg# mantle also allows to reproduce the composition of 219 a greater number of Amazonian volcanic provinces (5, all but Arsia Mons) with $T_{\rm p}$ of 220 1380–1460°C and pressures of 2.8–3.1 GPa. 221

The lack of temperature and pressure trends over time displayed by this set of con-222 straints renders it impossible to calculate rates of secular cooling or lithosphere thick-223 ening (Fig. 3a,b). This could be due to the limited number of primitive basalts avail-224 able that might not be representative of the average mantle. To test this possibility, we 225 compare the mantle temperature estimates derived from MAGMARS to the results of 226 a global convection model incorporating the most recent interior structure constraints 227 from InSight (Plesa et al., 2022). The maximum temperature (and minimum pressure) 228 at which the mantle is melting decreases with time (i.e. secular cooling). However, at 229 any given time, melt is produced from regions of the mantle with highly variable $T_{\rm p}$, which 230 encompass the $T_{\rm p}$ of the mantle sources estimated in this study. The Gusev basalts are 231 the only primitive basalts whose location is known with certainty. Additionally, NWA 232 7034 and the depleted shergottites have recently been suggested to have originated from 233 Karratha and 09-000015 craters, respectively (Lagain et al., 2021, 2022). Under all three 234 locations and at the appropriate—and highly contrasting— crystallization ages, the T_p 235 236 of the mantle sources would have been nearly identical and in the range 1525-1562 °C (Fig. 3c,d and S3). This confirms that despite the overall decrease in mantle temper-237 ature with time, a limited basaltic sample suite can record near-constant mantle tem-238 perature. The thermochemical evolution model predicts that the mantle temperature 239 should first increase due to the decay of radioactive elements and peak at the Noachian/Hesperian 240 transition before slowly decreasing (e.g., Plesa et al., 2022). This thermal maximum is 241 not recorded by the 3.7 billion years old Gusev basalts but seems consistent with our re-242 interpretation of the $T_{\rm p}$ of Hesperian volcanic provinces (1520–1660 °C; Baratoux et al., 243

2011), assuming that the average mantle is relatively MgO-rich (Mg# of 79; Yoshizaki
& McDonough, 2020).

Perhaps the main discrepancy between the thermochemical evolution model and 246 the MAGMARS constraints is the shallow depth of melting that we estimate for the source 247 of shergottites, which is predicted to be well within the lithospheric mantle (Plesa et al., 248 2022). Filiberto (2017) noted that if a larger amount of olivine fractionation had taken 249 place, the primary melts of shergottites could have been in equilibrium with the convect-250 ing mantle at 3–5 GPa. While this pressure of melting is more consistent with the thick 251 lithosphere of the late Amazonian (Fig. S3), such melt compositions would require a high 252 $T_{\rm p}$ of 1710 ± 73 °C, which exceeds significantly the maximum $T_{\rm p}$ achievable by thermal 253 evolution models at that time (Fig. 3d). Therefore, we consider it more likely that the 254 $T_{\rm p}$ of the sources was low (1470–1520 °C) and that the pressure of melting derived from 255 MAGMARS simulations (1.6 \pm 0.5 GPa) does not represent the average pressure of melt-256 ing but simply the final pressure of equilibration with the mantle. If shergottites formed 257 in the Tharsis region (e.g., Lagain et al., 2021), deeply-sourced primary melts could have 258 re-equilibrated with a warm lithospheric mantle, locally heated by magmas, at the base 259 of the crust (110–130 km; Wieczorek et al., 2022). 260

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5.2 Origin of the mantle sources and their variable concentrations of incompatible elements

The mantle source of Clast VI (NWA 7533) could be nearly identical to the prim-263 itive mantle (Fig. 2), as previously suggested based on rare-earth element (REE) mod-264 eling (Humayun et al., 2013). All other mantle sources are depleted in CaO and Al_2O_3 265 relative to the various primitive mantle compositions proposed in literature (DW85, LF97, 266 YM20 and K22). One possibility is that these mantle sources represent melting residues 267 from which 10-20 wt.% melt had been removed prior to producing the melts that even-268 tually formed the primitive basalts used in this study (Fig. 2a). However, the concen-269 trations of alkalis and other incompatible elements (e.g., TiO₂, P₂O₅) are, in most cases, 270 too high at a given Al_2O_3 concentration, regardless of the style (batch vs. fractional) and 271 pressure of melting (Fig. 2b-d). Only the Adirondack basalts are consistent in detail with 272 the simple re-melting of a mantle residue, following ± 10 wt.% prior melting of a prim-273 itive mantle (see also Collinet et al., 2021). Other processes must be invoked to explain 274 the chemical variability of the remaining mantle sources. 275

The Columbia Hills basalts are often assumed to be related to the Adirondack basalts, 276 as both groups were analyzed by Spirit at Gusev crater. Compared to the Adirondack 277 basalts, they are rich in alkali elements as well as other incompatible elements $(TiO_2,$ 278 P_2O_5) and poor in CaO and Al_2O_3 (Fig. 1). McSween, Ruff, et al. (2006) suggested that 279 the Columbia Hills basalts could have derived from melts similar to the Adirondack basalts 280 by fractional crystallization. The higher incompatible element concentrations (e.g., K, 281 P, Ti) of the Columbia Hills basalts have also been suggested to result from the contam-282 ination of Adirondack-like primitive melts by a crustal component (Schmidt & McCoy, 283 2010). However, crustal assimilation and fractional crystallization (AFC) of basaltic melts 284 should lower markedly the MgO concentrations (and Mg#; Ostwald et al., 2022). As the 285 Mg# of the Columbia Hills and Adirondack basalts are similar, most workers now re-286 gard them as two sets of near-primary melts (Schmidt & McCov, 2010; Filiberto & Das-287 gupta, 2011; Collinet et al., 2015). Schmidt and McCoy (2010) proposed that the high 288 K_2O content of the Columbia Hills basalts could be accounted for by melting a fertile 289 mantle source with a higher K_2O content compared to the Dreibus and Wänke (1985) 290 composition. According to their model, the Adirondack basalts would be slightly younger 291 and produced by re-melting the same region of the mantle. However, the similarly low 292 CaO and Al_2O_3 concentrations of their sources (Fig. 2a) suggest that both the Adiron-293 dack and Columbia Hills basalts were derived from depleted mantles, affected by 10–20 294 wt.% prior melting at ~ 3.0 GPa. Metasomatism has been invoked to reconcile the high 295

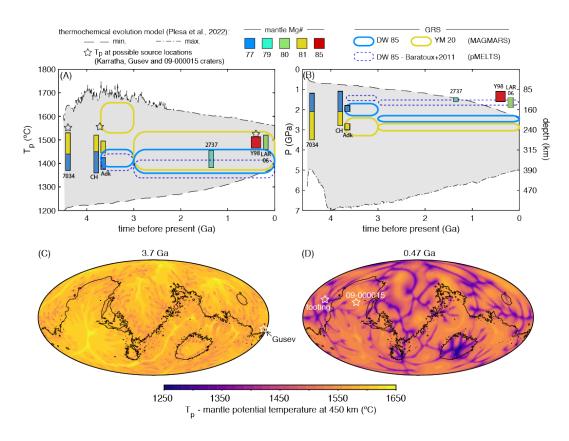


Figure 3. Temporal evolution of $T_{\rm p}$ (A) and the average pressure and depth of melting or conditions of mantle-melt re-equilibration (B). The rectangles represent the sources of the basaltic compositions listed in Table 1. The rounded fields are the sources of the GRS volcanic provinces of Baratoux et al. (2011), re-calculated with MAGMARS. The black lines represent the evolution of the potential temperatures and pressures of the part of the mantle that is affected by partial melting in the thick-crust geodynamical model of Plesa et al. (2022). The minimum pressure of melting (dashed line in B) can be interpreted as the thinnest thermal lithosphere observed anywhere on the planet. Panels C and D represent regional variations in $T_{\rm p}$ for this geodynamical model at the time of Gusev basalt (C) and depleted shergottites (D) crystallization. At their possible source locations (white stars, see text for references), the $T_{\rm p}$ are nearly identical: 1562 vs. 1525 °C (see also Fig. S3).

water and incompatible element concentrations of nakhlites-chassignites with their Sr-296 Nd isotopic compositions indicative of ancient depleted sources (Goodrich et al., 2013; 297 McCubbin et al., 2013; Day et al., 2018) and could also help explain the high K_2O con-298 centrations analyzed in numerous rocks from Gale crater (e.g., Schmidt et al., 2014). Sim-299 ilarly, we posit that the relative enrichment of incompatible elements in the Columbia 300 Hills basalts (alkali elements as well as elements like P and Ti that are less mobile in flu-301 ids) could be explained by the secondary addition of low-degree melts to a Adirondack-302 like mantle source. The highest possible K_2O concentrations that we calculate for the 303 Columbia Hills mantle sources are in the range 0.13–0.15 wt.%. This is much smaller than 304 the percent level K_2O concentrations of highly metasomatized and phlogopite-bearing 305 terrestrial peridotites (e.g., Condamine & Médard, 2014) but similar to other intraplate 306 peridotites containing no hydrous phases (e.g., Smith et al., 1993). The source of the Columbia 307 Hills basalts was likely affected by low degrees of cryptic metasomatism and was thus 308 not significantly hydrated. 309

The isotopic systematics of Martian meteorites suggest the existence of a magma 310 ocean that crystallized early in Mars' history (e.g., Elkins-Tanton et al., 2005; Debaille 311 et al., 2008; Kruijer et al., 2017; Bouvier et al., 2018). Some of the resulting heterogene-312 ity was never erased by convection and ancient mantle sources were affected by partial 313 melting and formed the shergottites as recently as 170 million years ago (Moser et al., 314 2013; Wu et al., 2021; McFarlane & Spray, 2022). The major and incompatible element 315 concentrations of the sources of shergottites must in part reflect the processes of magma 316 ocean crystallization. For example, the superchondritic CaO/Al₂O₃ ratio of shergottites 317 has been suggested to result from the fractionation of majorite in the deep mantle (Borg 318 & Draper, 2003). Here, we find that the sources of shergottites had mildly superchon-319 dritic CaO/Al_2O_3 ratios that could have appeared at low pressure, following 15–20 wt.% 320 melting of the primitive mantle (Fig. 2a). A 20 wt.% depletion from a primitive man-321 tle is also sufficient to decrease the incompatible element concentrations to levels iden-322 tical to those of the source of depleted shergottites (Fig. 2b-d). In this case, however, 323 the melting residue only reaches a Mg# of 77 (when starting from a DW85 mantle) to 324 81 (YM20), following 20 wt.% of melting. The much higher Mg# of the source of Y 980659 325 (85–86) remains easier to explain if it formed as a magma ocean cumulate (e.g., Borg 326 & Draper, 2003; Elkins-Tanton et al., 2005). The enriched shergottites have higher con-327 centrations of incompatible elements. Their composition in radiogenic isotopes indicates 328 that the enriched signature is most likely derived from evolved residual melts that were 329 trapped in mantle cumulates during the crystallization of an early MMO, rather than 330 from crustal assimilation (e.g., Borg & Draper, 2003; Symes et al., 2008; Debaille et al., 331 2008; Brandon et al., 2012; Ferdous et al., 2017; Armytage et al., 2018). This could also 332 explain the slightly higher concentrations of minor incompatible elements that we cal-333 culate for the source of enriched shergottites (Fig. 2). 334

6 Conclusions

The mantle temperature of the sources that gave rise to known primitive basalts 336 appears to have remained relatively stable through time ($T_{\rm p}$ of 1400–1500 °C). This could 337 be due to a sampling bias. The higher mantle $T_{\rm p}$ (~1600 °C) of the Hesperian volcanic 338 provinces (Baratoux et al., 2011), recalculated with MAGMARS and assuming a man-339 tle with Mg# of 79 or higher (Yoshizaki & McDonough, 2020; Khan et al., 2022), hint 340 at a significant secular cooling $(>100 \ ^{\circ}\text{C})$ as expected from thermochemical evolution mod-341 els (Plesa et al., 2022). The shergottite melts were likely produced at pressures greater 342 than 3 GPa but re-equilibrated with the lithospheric mantle at 1-2 GPa, for example 343 at the base of the thick Tharsis crust. 344

With the exception of the source of NWA 7034 and paired rocks, the mantle sources of known Martian basalts were poorer in Al₂O₃ and CaO compared to primitive mantle compositions (e.g., Dreibus & Wänke, 1985; Yoshizaki & McDonough, 2020). The com-

positions of the sources of Gusev crater basalts that we calculate do not explicitly re-348 quire a magma ocean stage and could represent simple depleted mantle reservoirs affected 349 by 10–20 wt.% prior melting (Adirondack basalts) or depleted mantle reservoirs re-fertilized 350 by fluids and low-degree silicate melts (Columbia Hills basalts). On the other hand, the 351 major element composition of the source of depleted shergottites cannot be easily ex-352 plained by partial melting alone and suggest, along with their Sr-Nd-Hf isotope system-353 atics, that they formed as mantle cumulates during the crystallization of the MMO. The 354 sources of enriched shergottites are consistent with trapping a more evolved residual melt. 355 It is also possible that the relative enrichment of the minor incompatible elements (Na, 356 K, Ti and P) of the Columbia basalts is a vestige of magma ocean processes and does 357 not result from metasomatism. But regardless of its origin, this relative enrichment is 358 limited, with concentrations of Na₂O, P₂O₅, and TiO₂ not exceeding the range displayed 359 by primitive mantle compositions. 360

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368 Open Research

The data used for the discussion and figures is summarized in the supplementary material (Table S1-S3) and available in full at https://doi.org/10.5281/zenodo.7691390 (Collinet et al., 2023)

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Figure 1.

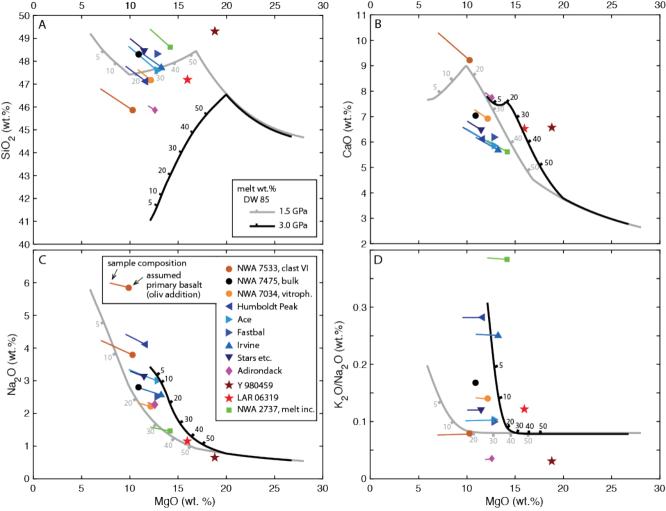


Figure 2.

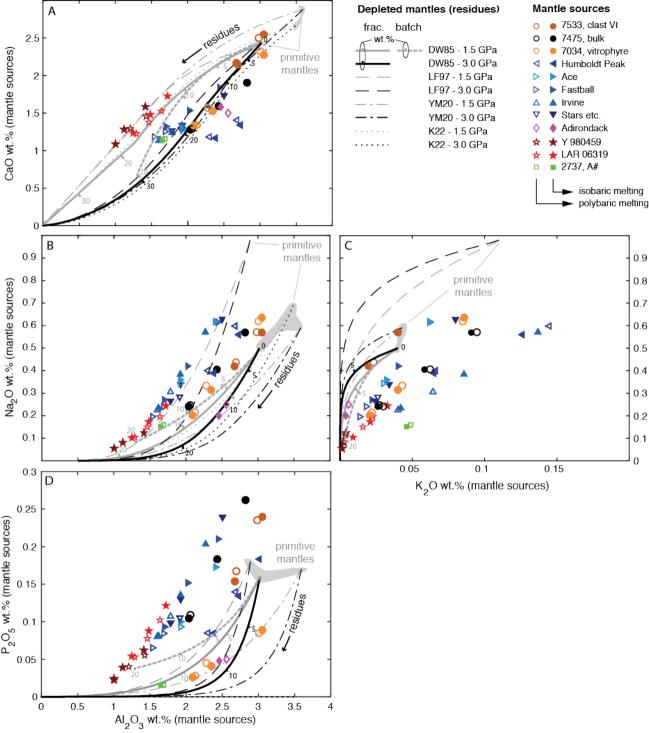
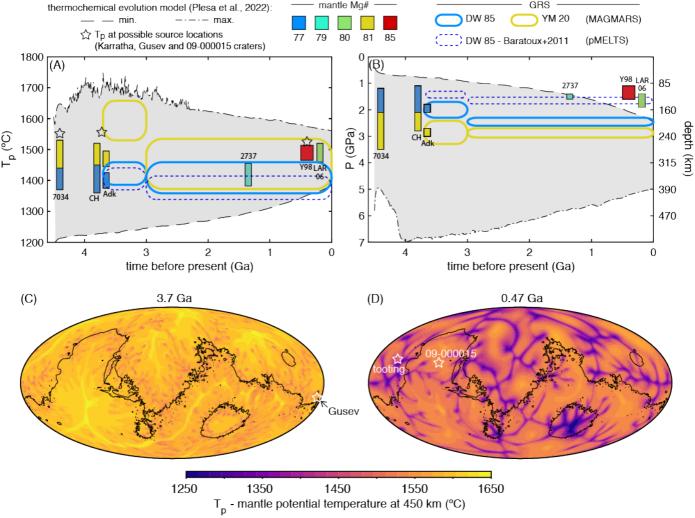


Figure 3.



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Supporting Information for

The temperature and composition of the mantle sources of Martian basalts

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- 2. Figure S3: temperature profile below the possible locations of primitive Martian basalts

Additional Supporting Information (Files uploaded separately)

- 1. Supplementary Tables S1–S3 (with submission)
- 2. MAGMARS results used in figure 3 to match the GRS volcanic provinces of Baratoux et al. (2011) (data repository)
- 3. MAGMARS scripts and files used to produce figure S1–S2 (data repository) data repository: https://doi.org/10.5281/zenodo.7691390

S1 Non-uniqueness and model uncertainties

MAGMARS uncertainties are discussed in detail in section 3.2 of Collinet et al. (2021) and propagated by combining analytical, experimental, and model uncertainties. They are provided as fixed average uncertainties for the composition of melts produced by melting any mantle composition, at any pressure and temperature: ± 1.3 wt.% SiO₂, 0.21 wt.% TiO₂, 0.38 wt.% Al₂O₃, 0.11 wt.% Cr₂O₃, 0.85 wt.% FeO, 0.06 wt.% MnO, 0.75 wt.% MgO, 0.61 wt.% CaO, 0.28 wt.% Na₂O, 0.12 wt.% K₂O, 0.12 wt.% P₂O₅, for lherzolite melting. To quantify the uncertainties on the mantle composition and melting conditions that can produce a specific primary basalt, we run a large number of simulations while varying systematically the parameter space (black contours in Figure S1). We retain only the simulations that produce a melt identical to the basalt Fastball (representative example) within the model uncertainties stated above (blue circles). Figure S1 can be compared to Figure 2 and shows that despite the substantial uncertainties associated with the method, the conclusions of the study remain unchanged.

Another large set of simulations is filtered assuming that the model uncertainties are smaller than reported in Collinet et al. (2021) (Figure S2). While it is possible that the MAGMARS model uncertainties are overly conservative, the goal here is simply to isolate model uncertainties and non-uniqueness. Even assuming a small model uncertainty, there is a large array of possible mantle sources and melting conditions that can produce nearly identical basaltic melts (the composition of Fastball in this case).

For a given basaltic melt (e.g., Fastball), the most refractory mantle sources are associated with the lowest melt fraction and lowest mantle temperatures due to the smaller release of latent heat of melting (Fig. S1a,b). To produce the same average basaltic liquid by polybaric melting as by isobaric melting, melting must start deeper and extend to a shallower region of the mantle, where the solidus temperature is low. The T_p is therefore lower compared to the isobaric case (Table S2, Fig. S1a, S2a, also see Fig. 2 in Collinet et al., 2021).

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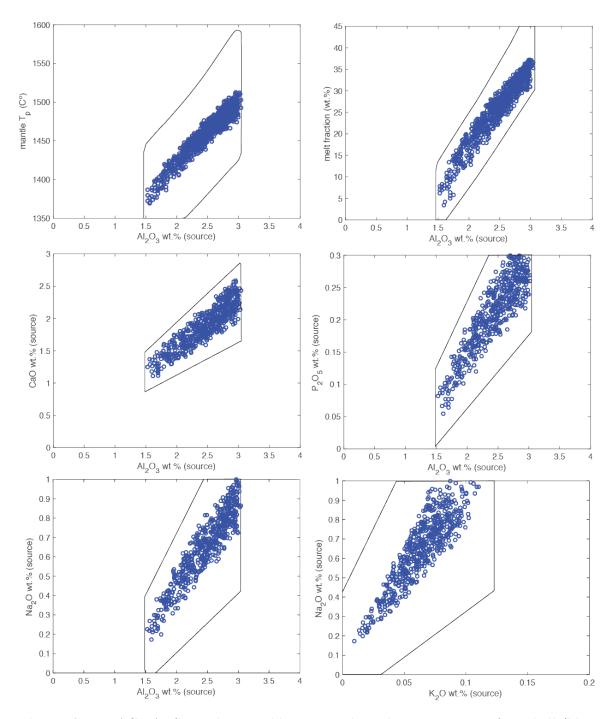


Figure S1: MAGMARS simulations able to reproduce the composition of Fastball (blue circle) assuming the average model uncertainty of Collinet et al. (2021): ± 1.3 wt.% SiO₂, 0.21 wt.% TiO₂, 0.38 wt.% Al₂O₃, 0.11 wt.% Cr₂O₃, 0.85 wt.% FeO, 0.06 wt.% MnO, 0.75 wt.% MgO, 0.61 wt.% CaO, 0.28 wt.% Na₂O, 0.12 wt.% K₂O, 0.12 wt.% P₂O₅. The black envelope represents the conditions sampled by 105000 MAGMARS simulations.

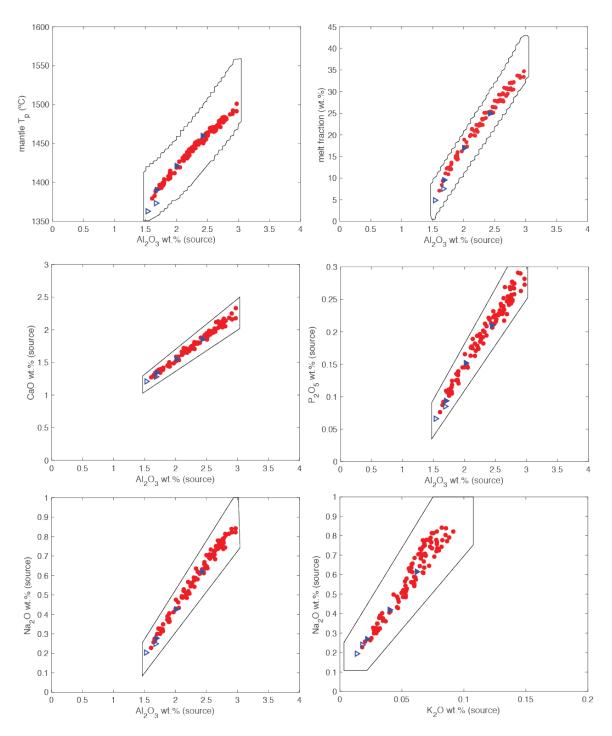


Figure S2: MAGMARS simulations able to reproduce the composition of Fastball assuming a low uncertainty of ± 0.5 wt.% SiO₂, 0.05 wt.% TiO₂, 0.15 wt.% Al₂O₃, 0.11 Cr₂O₃, 0.25 wt.% FeO, 0.06 MnO, 0.25 wt.% MgO, 0.15 wt.% CaO, 0.07 wt.% Na₂O, 0.03 wt.% K₂O, 0.05 wt.% P₂O₅ (red circles). The open and closed blue triangles represent the sample sources reported in Figure 2 and Table S2. The black envelope represents the conditions sampled by 375000 MAGMARS simulations.

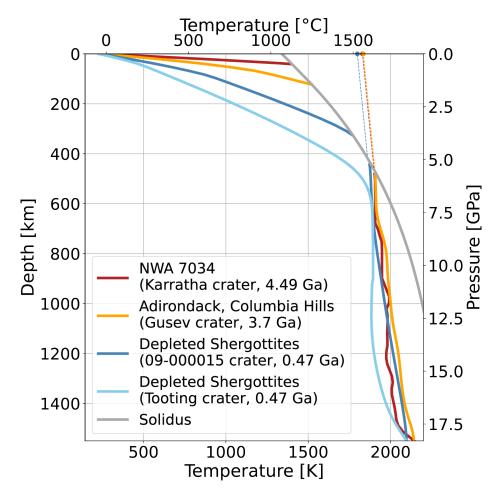


Figure S3: Temperature profiles of the mantle below Gusev crater, and the possible location of the sources of depleted shergottites (09-000015, Tooting; Lagain et al. (2021)) and NWA 7034 (Karratha; Lagain et al. (2022)) at the time of their crystallization, from the thick-crust model of Plesa et al. (2022). The mantle below Tooting crater does not reach the solidus of Ruedas and Breuer (2017) (grey line). Below the other 3 craters, the solidus is crossed at ~5 GPa and the mantle potential temperatures are nearly identical (filled circles on the upper temperature axis): 1556 °C (Karratha), 1562°C (Gusev) and 1525 °C (09-000015).