# Thermophysical assessment on the feasibility of basal melting in the south polar region of Mars

Lujendra Ojha<sup>1</sup>, Jacob Buffo<sup>2</sup>, and Baptiste Raphaël Journaux<sup>3</sup>

<sup>1</sup>Rutgers University <sup>2</sup>Dartmouth College <sup>3</sup>University of Washington

February 27, 2023

#### Abstract

Bright basal reflectors in radargram from the Mars Advanced Radar for Subsurface and Ionosphere Sounding (MARSIS) of the Martian south polar layered deposits (SPLD) have been interpreted to be evidence of subglacial lakes. However, this interpretation is difficult to reconcile with the low Martian geothermal heat flow and the frigid surface temperature at the south pole. We conduct a comprehensive thermophysical evolution modeling of the SPLD and show that subglacial lakes may only form under exceptional circumstances. Subglacial lakes may form if the SPLD contains more than 60 % dust volumetrically or extremely porous ice (>30 %), which is unlikely. A thick (>100 m) layer of dirty ice (>90% dust) at the base of the SPLD are equally unlikely, such as recent magmatic intrusions at shallow depths.

#### Hosted file

956832\_0\_art\_file\_10731250\_rqllxh.docx available at https://authorea.com/users/540576/ articles/626421-thermophysical-assessment-on-the-feasibility-of-basal-melting-in-thesouth-polar-region-of-mars

#### Hosted file

956832\_0\_supp\_10731249\_rqll2h.docx available at https://authorea.com/users/540576/articles/ 626421-thermophysical-assessment-on-the-feasibility-of-basal-melting-in-the-south-polarregion-of-mars





1	
2	Thermophysical assessment on the feasibility of basal melting in the south polar region of Mars
2	Thermophysical assessment on the reasismey of basic menting in the south polar region of Mars
4	Lujendra Ojha <sup>1</sup> , Jacob Buffo <sup>2</sup> , Baptiste Journaux <sup>3</sup>
5	
6	
7	<sup>1</sup> Department of Earth and Planetary Sciences, Rutgers University, Piscataway, NJ, USA.
8	<sup>2</sup> Thayer School of Engineering. Dartmouth College. Hanover, NH, USA.
9	<sup>3</sup> Department of Earth and Space Science, University of Washington, Seattle, WA, USA.
10	
11	Corresponding Author: ( <u>luju.ojha@rutgers.edu</u> )
12	Key Points:
13 14	• The feasibility of the potential existence of a subglacial lake on the south pole of Mars is investigated using thermal models.
15 16	• Subglacial lakes cannot be reconciled with current constraints on Martian heat flow and the bulk composition of the south pole.
17 18	• Exceptional circumstances would be necessary to form subglacial lakes in the Martian south pole.

#### 19 Abstract

20 Bright basal reflectors in radargram from the Mars Advanced Radar for Subsurface and 21 Ionosphere Sounding (MARSIS) of the Martian south polar layered deposits (SPLD) have been

Ionosphere Sounding (MARSIS) of the Martian south polar layered deposits (SPLD) have been interpreted to be evidence of subglacial lakes. However, this interpretation is difficult to reconcile

with the low Martian geothermal heat flow and the frigid surface temperature at the south pole.

24 We conduct a comprehensive thermophysical evolution modeling of the SPLD and show that

subglacial lakes may only form under exceptional circumstances. Subglacial lakes may form if the

26 SPLD contains more than 60 % dust volumetrically or extremely porous ice (>30 %), which is

27 unlikely. A thick (>100 m) layer of dirty ice (>90% dust) at the base of the SPLD may also enable 28 basal melting, resembling a sludge instead of a lake. Other scenarios enabling subglacial lakes in

the SPLD are equally unlikely, such as recent magmatic intrusions at shallow depths.

#### 30 Plain Language Summary

31 Radar data from the base of the Martian south polar cap have been interpreted by some to be evidence of a liquid water lake. However, this interpretation is hard to reconcile with the 32 observation that the surface temperature of the Martian south pole is frigid (-108° C). Recently, it 33 has been proposed that the base of the Martian south polar cap contains extremely salty ice that 34 can melt at -74° C; thus, despite the freezing conditions at the Martian south pole, liquid water may 35 36 form at the base. In this paper, we create various numerical models to understand how heat is 37 transmitted through the southern polar cap of Mars and if a salty liquid water lake can form as a result. We show that exceptional requirements are necessary for heat to be transferred efficiently 38

to form liquid water at the base of the south polar cap. The conditions required for liquid water to

form at the base of the Martian southern polar cap are at stark odds with what we currently know about the composition of the polar cap. In summary, numerical model results favor alternative, dry

42 interpretations of the radar data.

# 43 1 Introduction

The south polar region of Mars is primarily composed of H<sub>2</sub>O ice in combination with minor 44 amounts of CO<sub>2</sub> ice and dust. The SPLD's gravity-derived bulk density estimate of 1100 - 1300 kg 45 m<sup>3</sup> suggests the presence of dust content of at least 9% by volume (up to 20 %) if the CO<sub>2</sub> 46 concentration is assumed to be negligible [Broquet et al., 2021; Li et al., 2012; Wieczorek, 2008; 47 48 Zuber et al., 2007]. Observations by the Shallow Radar instrument on Mars Reconnaissance 49 Orbiter also reveal small volumes of buried  $CO_2$  ice (< 1%) within the south polar layered deposits [Bierson et al., 2016; Phillips et al., 2011]. These impurities provide a glimpse into the million-year 50 51 variation in the orbital parameters of Mars leading to the differential deposition of dust and  $CO_2$  in the south polar cap. Furthermore, the presence of impurities like dust, CO<sub>2</sub>, and other possible 52 clathrates can affect the thermal structure of the south polar ice cap, possibly raising the basal 53 54 temperature to enable basal melting [Greve and Mahajan, 2005].

55

Recently, bright basal reflectors in the radargram of Ultimi Scopuli have been interpreted to be evidence of a subglacial lake system [*Lauro et al.*, 2022; *Lauro et al.*, 2021; *Orosei et al.*, 2018],

although this interpretation has been challenged by multiple authors [*Lalich et al.*, 2022; *Ojha et al.*, 2013],

*al.*, 2021; *Smith et al.*, 2021]. While subglacial lakes are common in various areas of Earth

*Livingstone et al.*, 2021, the main factor that hinders the formation of subglacial lakes on Mars is

the frigid temperature at the south pole (<165 K) [*Sori and Bramson*, 2019] and expected low

62 geothermal heat flow on present-day Mars [*Broquet et al.*, 2021; *Ojha et al.*, 2021; *Oj* 

63 2019; *Parro et al.*, 2017; *Plesa et al.*, 2018]. Therefore, augmentation of the basal temperature by 64 exceptionally high heat flow from a recent magmatic intrusion would be necessary to melt pure 65 water ice at the Martian south pole [*Sori and Bramson*, 2019]. Alternatively, if one were to 66 suppose that the base of the SPLD is composed of saline ice, then the upper limit on the 67 temperature required for melting can be relaxed to a much lower value.

68

69 Although several recent papers have provided alternative interpretations of the bright radar 70 reflections, including CO<sub>2</sub> - H<sub>2</sub>O layer boundaries [Lalich et al., 2022], saline ice [Bierson et al., 71 2021], or smectites [Smith et al., 2021], the debate has persisted. Lauro et al. (2022) asserted that 72 the subglacial lakes at the Martian south pole are composed of perchlorate brines with a eutectic temperature of  $\sim 200$  K. Further, Lauro et al. (2022) claim that basal melting can be achieved in 73 the SPLD without a recent magmatic intrusion and with reasonable assumptions about the 74 75 presence of porous ice and impurities like dust and  $CO_2$  within the SPLD. Here, we assume that the interpretation of the subglacial lake is correct. With liquid water (pure or briny) at the base of 76 the SPLD as an a priori condition, we conduct a comprehensive thermophysical evolution 77 78 modeling exercise to ascertain the conditions that can enable basal melting at the south pole of 79 present-day Mars.

## 80 2 Data and Methods

2.1 1-dimensional thermal models to assess the feasibility of basal melting due to the bulk
 impurities within the SPLD

We set up a 1-D thermophysical evolution model of the SPLD in COMSOL that simulates coupled heat transfer and phase change. COMSOL utilizes the apparent heat capacity formulation to provide an implicit method of capturing the phase change interface by solving for both phases and the heat transfer equation with effective material properties via:

87 
$$\rho C_{eq} \frac{\partial T}{\partial t} + \nabla \cdot (-k_{eq} \nabla T) = 0;$$
88 (eqn 1)  
89 where  $\rho$  (kg m<sup>3</sup>) is the SPLD density,  $C_{eq}$  (J kg<sup>-1</sup> K<sup>-1</sup>) is the effective heat capacity,  $k_{eq}$  (W m<sup>-1</sup> K<sup>-1</sup>) is  
90 the effective thermal conductivity,  $T$  (K) is temperature, and  $\nabla$  is the del operator. The latent heat  
91 of phase change (L) is taken into account by modifying the effective heat capacity (i.e., the apparent

92 heat capacity formulation) as:

93  
94 
$$c_p = \frac{1}{\rho} \left( \theta_1 \rho_1 c_{p,1} + \theta_2 \rho_2 c_{p2} \right) + L_{1 \to 2} \frac{\partial \alpha_m}{\partial T};$$

(egn 2)

where  $c_p$  is the bulk apparent heat capacity,  $c_{p,1}$  is the specific heat capacity of phase 1, and  $c_{p,2}$  is the specific heat capacity of phase 2. Here,  $L_{1\rightarrow 2}$  is the latent heat of melting which we vary with the melting temperature of the ice (see Section 2.2). The parameter  $\alpha_m$  is the mass fraction given by:

99 
$$\alpha_m = \frac{1}{2} \frac{\theta_2 \rho_2 - \theta_1 \rho_1}{\theta_1 \rho_1 + \theta_2 \rho_2}.$$
100 (eqn 3)

101 In equation 2 and 3,  $\theta_1$  and  $\theta_2$  are the mass fractions of phase 1 and phase 2 (i.e.,  $\theta_1 + \theta_2 = 1$ ). 102 As phase change occurs, the value of various thermophysical parameters are computed as a 103 volumetric average of H<sub>2</sub>O-ice ( $\theta_1$ ) and liquid water ( $\theta_2$ ) given by:

$$\rho = \theta_1 \rho_1 + \theta_2 \rho_{2;}$$
  
and  
$$k = \theta_1 k_1 + \theta_2 k_{2.}$$

105

104

106 The thermal evolution models are run for a minimum of 2 million years (Text S1). We set the 107 surface temperature at the top of the ice sheet to 165 K [Sori and Bramson, 2019]. At the base of the SPLD, a net-inward flowing heat flux boundary condition is imposed, which we vary between 5 108 109 - 60 mW m<sup>2</sup>. The accuracy of the apparent heat capacity formulation in capturing phase change is verified by comparing the output of the COMSOL simulation to the analytical solution of the 110 111 Stefan problem (Text S2; Fig. S1).

2.2 Thermophysical Parameters of water-ice and mixtures: 112

We use SeaFreeze [Journaux et al., 2020] to estimate the temperature and pressure dependent 113 thermophysical parameters of pure water-ice such as  $c_p$  and  $\rho$  (Fig. S2). The latent heat of melting 114 of ice is also temperature dependent which we estimate using SeaFreeze (Fig. S3). The thermal 115 conductivity of pure water-ice  $(k_{ice})$  is temperature dependent and can be estimated using the 116 following equation [Petrenko and Whitworth, 1999]: 117 118

120

 $k_{ice}(z) = \frac{651}{T_{(z)}}$ 119

(eqn 5)

(eqn 4)

Once the temperature-dependent thermophysical parameters of pure ice are computed, we use 121 mixing models to estimate the thermophysical parameters of water-ice and dust mixtures. We 122 123 estimate the bulk thermal conductivity of the dusty ice (Kbulk) using the heterogenous twocomponent system mixing model [Hamilton and Crosser, 1962]: 124

125

126 
$$K_{bulk}(z) = K_{ice}(z) \left[ \frac{K_{dust} + [n-1]K_{ice(z)} - [n-1]V_{dust}[K_{ice(z)} - K_{dust}]}{K_{dust} + [n-1]K_{ice(z)} + V_{dust}[K_{ice(z)} - K_{dust}]} \right].$$
127 (eqn 6)

The parameter n in equation 6 is a dimensionless, empirical shape factor set to 3. This is a valid 128 approach as long as the thermal conductivity of the discontinuous phase (i.e., dust) is less than a 129 factor of 100 different than the continuous phase (i.e., H<sub>2</sub>O ice) [Hamilton and Crosser, 1962]. V<sub>dust</sub> 130 represents the volume of dust present in the SPLD ranging between 0 and 1. Other 131 132 thermophysical parameters, such as  $c_p$  and  $\rho$  of the SPLD, are computed as volumetric averages of dust and ice (Fig. S2; Table S1). The thermal conductivity of Martian dust is set to 0.039 W m<sup>-1</sup> 133 K<sup>-1</sup>, as measured by InSight [Grott et al., 2021]. We also consider a slightly lower value of 0.015 W 134  $m^{-1}$  K<sup>-1</sup> based on laboratory measurements of the thermal conductivity of Martian dust simulants 135 [Yu et al., 2022]. The mixing models are coupled with the thermal evolution model because as the 136 137 thermal model evolves with time, the temperature at any particular depth (z) can change, affecting the thermophysical parameters of the water-ice and the bulk SPLD. 138

#### 2.3 Thermophysical Parameters of porous water-ice: 139

Porosity within the SPLD can also notably affect the thermal structure. We utilize a 140 parametereized model that relates the porosity and temperature to the thermal conductivity of 141 snow, firn, and porous ice  $k_{\text{snow-firm}}(\rho, T)$  given by [*Calonne et al.*, 2019]: 142

$$k_{\text{snow-firm}}(\rho, T) = (1 - \theta) \frac{k_{ice}(T)k_{air}(T)}{k_{ice}^{\text{ref}}k_{air}^{\text{ref}}} k_{\text{snow}}^{\text{ref}}(\rho) + \theta \frac{k_{ice}(T)}{k_{ice}^{\text{ref}}} k_{\text{firm}}^{\text{ref}}(\rho)$$
(eqn 7)

143

where  $k_{ice}(T)$  is the temperature dependent thermal conductivity of water-ice given by equation 5, 144 kair(T) is the temperature dependent thermal conductivity of air. Based on [Calonne et al., 2019], 145 we set the  $k_{ice}^{\text{ref}}$  and  $k_{air}^{\text{ref}}$  to 2.107 W m<sup>-1</sup> K<sup>-1</sup> and 0.024 W m<sup>-1</sup> K<sup>-1</sup> respectively. Because the thermal 146 conductivity of air does not notably change with temperature, we set  $k_{air}(T)$  to be equal to 147  $k_{air}^{\text{ref}}$ . Other parameters in equation 7 are parameterized as the follows: 148

$$\begin{split} \theta &= \frac{1}{\left(1 + \exp\left(-2a(\rho - \rho_{\text{transition}})\right)\right)};\\ k_{\text{firm}}^{\text{ref}}\left(\rho\right) &= 2.107 + 0.003618(\rho - \rho_{ice});\\ k_{\text{snow}}^{\text{ref}}\left(\rho\right) &= k_{\text{cal.}} = 0.024 - 1.23 \times 10^{-4}\rho + 2.5 \times 10^{-6}\rho^{2}; \end{split}$$

149

where  $\rho_{ice}$  is set to 917 kg m<sup>3</sup> and  $\rho_{\text{transition}}$  is set to 450 kg m<sup>3</sup>. The above set of equations are 150

solved as a function of temperature and density of ice  $(\rho)$ . Porosity  $(\phi)$  is computed as follows: 151

$$\emptyset = \left(1 - \frac{\rho_{ice}}{\rho_{SPLD}}\right)$$

(egn 9)

(eqn 8)

152

153 2.4 2-dimensional magmatic instrusion model

Previously, Sori and Bramson (2019) found that augmentation of the geothermal heat flux by at 154 least 72 mW m<sup>2</sup> is required for basal melting under the most favorable compositional 155 considerations. However, the size of the putative lake was not considered in that work. Here, we 156 explore the possible geometry, thermal budget, and depth of the magmatic intrusion required to 157 form a subglacial lake approximately 20 - 30 km in width [Lauro et al., 2021]. Another significant 158 difference between our magmatic intrusion model and that of Sori and Bramson. (2019) is that we 159 prescribe the magmatic intrusion with an initial temperature value instead of a heat flux. 160 Furthermore, since relatively cold country rocks would surround any recent intrusion on Mars, we 161 allow the temperature of the magmatic intrusion to cool over the timescale of the simulation (Text. 162 S3; Fig. S4). Incorporating a time-dependent cooling intrusion also allows us to assess the longevity 163 of any brines produced by an intrusion. 164

165

We consider a 50-km thick Martian crust superposed by a 1.5 km thick ice sheet resembling the 166

167 SPLD. Initially, we model the steady-state temperature profile of the crust by assuming a surface

heat flow of 30 mW m<sup>2</sup> (Text S3; Fig. S5). A magmatic intrusion of variable geometry and depth is 168

then introduced within the crust to assess its overall impact on the feasibility of basal melting. As

- 170 melting occurs within the SPLD, convection can play a dominant role; thus, we couple the heat
- transfer module within COMSOL to the laminar flow package Multiphysics to account for possible

172 convection within the melt layer. As the density between the two phases is different, we solve the173 "Weakly Compressible" form of the fluid flow equations and use a moving mesh boundary

174 condition to account for the overall volume change of the ice/water layer. The viscosity of the

- 175 liquid water is set to 0.018 Pa. s, whereas the viscosity of the H<sub>2</sub>O-ice is set to  $10^{\circ}$  Pa. s. A more
- 176 detailed explanation of the model setup is provided in Text S3. Finally, we set the initial
- temperature of the intrusion to 1500 K, a relatively high value on par with the temperature of some
- of the hottest magmas preserved on Earth [*Arndt et al.*, 2008].
- 179

# 180 3 Results and discussion

181 3.1. Can bulk impurities within the SPLD enable basal melting?

- We first consider a scenario in which we assess if the bulk impurities and porosity present within the SPLD can enable basal melting. We vary T<sub>s</sub> between 165 – 170 K and solve for the k<sub>bulk</sub> of SPLD that would allow basal melting as a function of q<sub>s</sub>. The k<sub>bulk</sub> necessary for basal melting increases with q<sub>s</sub>, and variation in T<sub>s</sub> do not notably impact the results (Fig. 1 a). It is unlikely that the regional heat flow in the south polar region exceeds 30 mW m<sup>2</sup> [*Broquet et al.*, 2021; *Ojha et al.*, 2021; *Ojha et al.*, 2019; *Parro et al.*, 2017; *Plesa et al.*, 2018], thus, k<sub>bulk</sub>  $\leq$  1.5 W m<sup>4</sup> K<sup>4</sup> is necessary for the SPLD to undergo basal melting (Fig. 1 a).
- 190

We solve for the combination of H<sub>2</sub>O-ice, porosity, and dust content that can yield  $k_{bulk} \leq 1.5$  W m<sup>-1</sup> 191 192 K<sup>1</sup>. We initially assume the bulk of the SPLD to be pore-free and solve for the volumetric dust content required for basal melting using the mixing relation for  $H_2O$ -ice and dust (eqn 6). Dust 193 content above 60 %, regardless of the two k<sub>dust</sub> value considere here, would be required for basal 194 melting to occur with a surface heat flow of 30 mW m<sup>2</sup> (Fig. 1 b). The bulk impurities necessary 195 for basal melting from our thermal simulation can be compared to the bulk composition of the 196 SPLD inferred from gravity [Broquet and Wieczorek, 2019; Li et al., 2012; Wieczorek, 2008; 197 198 Zuber et al., 2007] and radar data [Lauro et al., 2022]. While there are uncertainties in the 199 possible volume of dust present within the SPLD, it is unlikely to exceed 20 %, significantly lower than the bulk volume required for basal melting in the SPLD (Fig. 1 c). The presence of  $CO_2$  ice 200 within the SPLD can also aid in the feasibility of basal melting due to its relatively low thermal 201 conductivity of  $0.02 \text{ W m}^{-1} \text{ K}^{-1}$ . However, there is no evidence in the radar data to suggest hidden 202 voluminous (> 1%) deposits of CO<sub>2</sub> within the SPLD [Bierson et al., 2016; Phillips et al., 2011]. 203 204

We now consider if pore spaces within the SPLD can enable basal melting. Using the 205 parameterized mixing model of Calonne et al. (2019), we solve for the k<sub>bulk</sub> as a function of the 206 temperature and porosity of ice (Fig. 1 d). Regardless of the temperature, porosity in excess of 0.3 207 would be necessary for the  $k_{\text{bulk}}$  to be lower than 1.5 W m<sup>-1</sup> K<sup>-1</sup>. The required porosity for basal 208 melting can be compared with the bulk density constraint of the SPLD derived from gravity data. 209 Considering H<sub>2</sub>O-ice ( $\rho_{ice} = 930 \ kg \ m^{-3}$ ), dust ( $\rho_{dust} = 3000 \ kg \ m^{-3}$ ), and air ( $\rho_{air} =$ 210 0.02 kg  $m^{-3}$ ), to be the main components of the SPLD, bulk porosity  $\leq 0.2$  can be present within 211 the SPLD and still satisfy the gravity-derived bulk density constraint (Fig. 1 c). If a slightly lower 212  $\rho_{dust}$  of 2500 kg m<sup>3</sup> is considered, then the bulk porosity of the SPLD cannot exceed 0.12. In 213 either case, the required porosity of 0.3 for basal melting cannot be reconciled with the gravity-214





217 Figure 1. (a) Bulk thermal conductivity of the SPLD necessary for basal melting ( $T_m = 199$  K) as a function of  $q_{\cdot}$ .  $k_{bulk}$  lower than denoted by the black line would lead to basal melting. The error 218 bars show the negligible effect of T<sub>s</sub> (between 165-170 K) on the  $k_{bulk}$  required for basal 219 melting. The color shows the basal surface temperature at 1.5 km depth. (b) Similar to (a), but 220 showing the volume of dust required for basal melting as a function of the surface heat flow. The 221 effect of dust's thermal conductivity variation is negligible (notice overlap of solid and dotted line 222 values of 0.015 and 0.039 W m<sup>-1</sup> K<sup>-1</sup>). (c) Ternary diagram showing the possible mixture of pore 223 224 space, ice, and dust that can satisfy the bulk density estimate of the SPLD. (d) Thermal conductivity of ice as a function of pore-space and temperature using the mixing model of Calonne 225 et al. (2019). The black line marks the minimum porosity needed for  $k_{\text{bulk}} \leq 1.5 \text{ W m}^{-1} \text{ K}^{-1}$ . 226 227

- 3.2. Can a thin layer of impurity at the base of the SPLD enable basal melting?
- 229

We now consider if a layer of dust-rich ice (hereafter referred to as 'dirty ice') at the base of the 230 SPLD can aid in basal melting (Fig. 2 a). In this scenario, we vary the thickness (h) and the 231 volumetric dust content ( $\theta$ ) of the dirty ice layer. The SPLD above the dirty ice is assumed to have 232 20% dust volumetrically. Figure 2 (b) shows an example of the thermal evolution of a 1.5 km thick 233 ice sheet, with the bottom 300 m composed of dirty ice ( $\theta = 95$  %), and a surface heat flow of 30 234 mW m<sup>2</sup>. In this extreme scenario, the temperature within the region occupied by dirty ice 235 increases over time, and basal melting is observed at the base (region under the white contour). 236 237 We perform a parameter sweep to constrain the minimum h and  $\theta$  required under the SPLD for basal melting to occur (Fig. 2 c, d). Basal melting is only observed if the dirty ice layer contains 238

239 more than 90 % dust volumetrically, which is substantially more than the dust/ice ratio found even

in the nucleus of comets [Küppers et al., 2005]. Furthermore, the dirty ice layer's thickness must

be in excess of 100 meters for basal melting to occur. If basal melting does occur, it is much more

242 likely to occur at the base, where dirty ice is in direct contact with the crust (Fig. 2 b). A reasonable 243 variation in  $k_{dust}$  and  $T_s$  has a negligible effect on the feasibility of basal melting in this scenario (Fig.

243 variation in 2 244 2 c, d).

- 245
- 246



247 Figure 2. (a) Schematic of the case where we consider if dirty ice of varying thickness and dust content at the base of the SPLD can enable basal melting. (b) An example of a temperature 248 249 distribution of the SPLD as a function of depth and time for a surface heat flow of 30 mW m<sup>2</sup> assuming a 1.2 km thick ice sheet, superposed on a 300 m thick dirty-ice with 95% dust content. 250 251 The black line shows the 300 m mark from the base, and the white contours show the region 252 under which basal melting occurs. (c) Result from a parameter sweep that shows the minimum thickness and dust content of dirty ice layer required for basal melting. The color plot shows the 253 thickness of the resulting melt. (d) Same as (c) but assuming a  $k_{\text{dust}}$  of 0.015 W m<sup>-1</sup> K<sup>-1</sup>. (e) Similar to 254 255 (c) and (d) but showing the basal melt thickness when a 2-meter thick low conductivity layer at the top of the SPLD is considered. (f) Similar to (e), but assuming a 5-meter thick low conductivity 256 257 layer at the top of the SPLD.

258

The thermal inertia of the Martian south polar region is relatively low, suggestive of the presence of a low thermal conductivity unit such as firn or  $CO_2$  in the upper few meters [*Putzig et al.*, 2005]. The thermal conductivity of even the lowest density porous ice lies around 0.06 W·m<sup>-1</sup>·K<sup>-1</sup> [*Calonne et al.*, 2019], whereas  $CO_2$  ice can have a much lower density of 0.02

 $W \cdot m^{-1} \cdot K^{-1}$  [Sori and Bramson, 2019]. We now assume the presence of CO<sub>2</sub> ice on top of the 263 SPLD and vary its thickness to assess the effect of a low conductive layer on the feasibility of basal 264 melting. We find that a low thermal conductivity unit like  $CO_2$  ice can significantly aid basal 265 melting when a dirty ice layer is present at the base of the SPLD (Fig. 2); however, even when a 5-266 267 meter low conductive layer is assumed to be present, basal melting only occurs with dirty ice containing more than 50% dust volumetrically. An unrealistic aspect of this model is that the 268 thermal conductivity of the dust at the base of the SPLD would be equivalent to the bulk-thermal 269 conductivity of the dust measured at the surface. The kdust values reported by InSight and laboratory 270 271 measurements of the Mars simulant are for the bulk, porous dust at the surface [Grott et al., 2021; Yu et al., 2022]. The overburden pressure and pore-filling ice would reduce the pore space within 272 the dust at the base of the SPLD; thus, the results presented in Figure 2 are for a highly optimistic 273 scenario. Furthermore, if a thick layer of dirty ice does exist at the base of the SPLD, then the 274 275 resultant melt may resemble a sludge or hydrated smectite clays [Smith et al., 2021] instead of a 276 subglacial lake.

277

278 3.4. What would it require to create a 20 - 30 km wide lake under the SPLD?

279

280 A recent magmatic intrusion underneath the SPLD can augment the near-surface temperature and 281 enable basal melting. An intrusion of variable width is intruded into the crust at various depth; the 282 heat from the intrusion raises the near surface temperature of the SPLD while at the same time the temperature of the intrusion declines due to the conduction of heat to the surrounding cold 283 284 country rocks (Supplementary Movie 1; Supplementary Movie 2). The cooling rate depends on the overall geometry of the intrusion and the temperature of the surrounding rock (Fig. 3 b). In 285 286 general, larger intruded bodies at depth cool much slower due to a higher degree of insulation than smaller bodies at shallow depths (Fig. 3 b). The heat from the magmatic intrusion also affects the 287 near-surface temperature of the crust and the base of the SPLD, possibly leading to basal melting. 288 Figure 3 c – f shows the resultant width of the basal melt that can form from an intrusion of a given 289 290 width and depth over time.

291

In these models, we set the temperature of the crust by assuming a surface heat flow of 30 mW m<sup>2</sup> (Text. S3; Fig. S5). While there are many parameters at play, the key takeaway from this exercise is that intrusions need to be relatively large and close to the surface of Mars to induce any degree of

basal melting. For example, a narrow dyke-like intrusion (W = 1 - 3 km), regardless of its proximity to the surface, would not be able to provide enough heat for a 20 – 30 km wide subglacial lake as

proposed [*Lauro et al.*, 2021]. In general, an intrusion must have a similar width to the width of the

298 subglacial lake.



**Figure 3. (a)** An example of a mesh showing the magmatic intrusion (in blue) of width W at depth D within the Martian crust of thickness 50 km. We vary W and D in this experiment to ascertain their values that would enable basal melting of a 20-30 km wide lake under the SPLD. (b) The cooling rate of the magmatic intrusion of various W and D within the Martian crust as a function of time. (c) – (f) Results from parameter sweep show the width of basal melt as a function of time for magmatic intrusions of various W and D. Only large magmatic intrusions (c and d) at shallow depths can form 20-30 km wide subglacial lakes under the SPLD.

307

308 Currently, there is no way to assess if such magmatic intrusions may be present at depth under the SPLD. To date, no major Bouguer gravity anomalies or crustal magnetic field anomalies, possibly 309 310 related to an intrusion in the south polar region have been identified. Figure 3 c - f also highlights that any subglacial lakes formed by magmatic intrusion are metastable over geological timescales. 311 As the temperature of the intruded body cools down, so does the near-surface temperature, 312 refreezing any subglacial lakes. Even if their presence is confirmed, the transient nature of 313 intrusion sustained subglacial lakes, as illustrated here, has obscure implications for their 314 astrobiological significance. 315

316

317 3.5. Feasibility of basal melting with lower salinity of the SPLD

318

One of the outstanding issues with the possibility of briny subglacial lakes in the SPLD is the amount of salt required to sustain such lakes. The notion that the purported lake exists underneath the SPLD due to a high concentration of perchlorates with a eutectic temperature of 199 K is entirely speculative; no evidence currently supports that notion. Further, given the reported dimension of the main putative lake  $(20 \times 30 \text{ km})$  [*Lauro et al.*, 2022] and the minimum molal

324 concentration of salt required for supercooled brine [Toner et al., 2014], it has previously been

shown that salt masses over 10<sup>12</sup> kg would be required to form such a voluminous brine solution 325 326 under the SPLD [Oiha et al., 2021]. The perchlorate salts on Mars are derived globally from the 327 atmospheric oxidation of HCl [Catling et al., 2010; Wilson et al., 2016]; thus, the process 328 responsible for such significant sequestration of perchlorate at the SPLD remains unexplained. Given the difficulty reconciling such a large volume of perchlorate salts within the SPLD, we also 329 330 explore the thermophysical conditions required for forming subglacial lakes with higher melting 331 temperatures. Figure S6 shows that if the melting temperature were slightly higher than the 332 assumed 199 K value, basal melting would be implausible with a q of 30 mW m<sup>2</sup> for most saline ice. For example, the basal temperature of the SPLD would not exceed the eutectic temperature of 333 334  $Ca(ClO_3)_2$  and  $Mg(ClO_3)_2$  [Hanley et al., 2012] even assuming SPLD with 50 % dust. Even if the surface temperature of the SPLD were miraculously augmented to 180 K, basal melting of ice with 335 336 reasonable salinity would be unlikely (Fig. S6).

## 337 4 Conclusions

We consider several scenarios for potentially forming a subglacial lake in the SPLD. First, we show 338 that basal melting is unlikely to occur with available heat flow constraints and reasonable 339 assumptions about the presence of pore space and impurities like dust and CO<sub>2</sub>. Second, a dirty 340 ice layer at the base of the SPLD can potentially enable basal melting; however, the dirty ice layer 341 must be extremely dust rich and thick. Melt in this scenario would occur at the base resembling 342 wet soil overlain by dry dirty ice instead of a lake. Third, a recent magmatic intrusion can augment 343 the near-surface temperature and enable basal melting; however, a relatively large intrusion at 344 shallow depths would be required. Magmatic intrusion-enabled subglacial lakes would also be 345 relatively short-lived with limited astrobiological potential. Finally, basal melting of other chlorate 346 and perchlorate species is even more unlikely, even with a 50 % dust content in the SPLD. In 347 summary, exceptional circumstances would be necessary to form subglacial lakes in the SPLD. 348

#### 349 Acknowledgments

350 LO is supported by a startup grant from Rutgers University.

#### 351 Open Research

No new data is generated as part of this work. Post publication, the COMSOL model outputs will 352 deposited Zenodo and will be accessible at the following 353 be on location https://doi.org/10.5281/zenodo.7658862 (CCC) (currently restricted). For the peer-review process 354 COMSOL output is uploaded as Supporting Information. 355

- 356
- 357
- 358
- 359
- 360

361

362

#### References

- 363 Arndt, N., M. C. Lesher, and S. J. Barnes (2008), Komatiite, Cambridge university press.
- Bierson, C. J., R. J. Phillips, I. B. Smith, S. E. Wood, N. E. Putzig, D. Nunes, and S. Byrne (2016), Stratigraphy and
- evolution of the buried CO2 deposit in the Martian south polar cap, in *Geophysical Research Letters*, edited, pp.
- 366 4172-4179, doi:10.1002/2016GL068457.
- 367 Bierson, C. J., S. Tulaczyk, S. W. Courville, and N. E. Putzig (2021), Strong MARSIS Radar Reflections From the
- Base of Martian South Polar Cap May Be Due to Conductive Ice or Minerals, *Geophysical Research Letters*, 48(13),
   e2021GL093880, doi:<u>https://doi.org/10.1029/2021GL093880</u>.
- 370 Broquet, A., and M. A. Wieczorek (2019), The Gravitational Signature of Martian Volcanoes, in Journal of
- 371 Geophysical Research: Planets, edited, doi:10.1029/2019JE005959.
- Broquet, A., M. A. Wieczorek, and W. Fa (2021), The Composition of the South Polar Cap of Mars Derived From
- 373 Orbital Data, Journal of Geophysical Research: Planets, 126(8), e2020JE006730,
- doi:<u>https://doi.org/10.1029/2020JE006730</u>.
- 375 Calonne, N., L. Milliancourt, A. Burr, A. Philip, C. L. Martin, F. Flin, and C. Geindreau (2019), Thermal
- Conductivity of Snow, Firn, and Porous Ice From 3-D Image-Based Computations, *Geophysical Research Letters*,
   46(22), 13079-13089, doi:<u>https://doi.org/10.1029/2019GL085228</u>.
- 378 Catling, D. C., M. W. Claire, K. J. Zahnle, R. C. Quinn, B. C. Clark, M. H. Hecht, and S. Kounaves (2010),
- Atmospheric origins of perchlorate on Mars and in the Atacama, in *Journal of Geophysical Research: Planets*,
   edited, Wiley Online Library.
- Greve, R., and R. A. Mahajan (2005), Influence of ice rheology and dust content on the dynamics of the north-polar cap of Mars, *Icarus*, *174*(2), 475-485, doi:https://doi.org/10.1016/j.icarus.2004.07.031.
- 383 Grott, M., et al. (2021), Thermal Conductivity of the Martian Soil at the InSight Landing Site From HP3 Active
- Heating Experiments, Journal of Geophysical Research: Planets, 126(7), e2021JE006861,
- 385 doi:<u>https://doi.org/10.1029/2021JE006861</u>.
- Hamilton, R. L., and O. Crosser (1962), Thermal conductivity of heterogeneous two-component systems, *Industrial*
- 387 & Engineering chemistry fundamentals, 1(3), 187-191.
- Hanley, J., V. F. Chevrier, D. J. Berget, and R. D. Adams (2012), Chlorate salts and solutions on Mars, in
- 389 *Geophysical Research Letters*, edited, doi:10.1029/2012GL051239.
- Journaux, B., J. M. Brown, A. Pakhomova, I. E. Collings, S. Petitgirard, P. Espinoza, T. Boffa Ballaran, S. D.
- 391 Vance, J. Ott, and F. Cova (2020), Holistic approach for studying planetary hydrospheres: Gibbs representation of
- ices thermodynamics, elasticity, and the water phase diagram to 2,300 MPa, *Journal of Geophysical Research: Planets*, *125*(1), e2019JE006176.
- 394 Küppers, M., I. Bertini, S. Fornasier, P. J. Gutierrez, S. F. Hviid, L. Jorda, H. U. Keller, J. Knollenberg, D. Koschny,
- and R. Kramm (2005), A large dust/ice ratio in the nucleus of comet 9P/Tempel 1, *Nature*, 437(7061), 987-990.
- Lalich, D. E., A. G. Hayes, and V. Poggiali (2022), Explaining Bright Radar Reflections Below The South Pole of
   Mars Without Liquid Water, *Nature Astronomy*, 6(10), 1142-1146.
- 398 Lauro, S. E., E. Pettinelli, G. Caprarelli, J. Baniamerian, E. Mattei, B. Cosciotti, D. E. Stillman, K. M. Primm, F.
- Soldovieri, and R. Orosei (2022), Using MARSIS signal attenuation to assess the presence of South Polar Layered Deposit subglacial brines, *Nature communications*, *13*(1), 1-10.
- Lauro, S. E., et al. (2021), Multiple subglacial water bodies below the south pole of Mars unveiled by new MARSIS data, *Nature Astronomy*, 5(1), 63-70, doi:10.1038/s41550-020-1200-6.
- Li, J., J. C. Andrews-Hanna, Y. Sun, R. J. Phillips, J. J. Plaut, and M. T. Zuber (2012), Density variations within the
- 404 south polar layered deposits of Mars, in Journal of Geophysical Research E: Planets, edited,
- 405 doi:10.1029/2011JE003937.
- 406 Livingstone, S. J., Y. Li, A. Rutishauser, R. J. Sanderson, K. Winter, J. A. Mikucki, H. Björnsson, J. S. Bowling, W.
- Chu, and C. F. Dow (2022), Subglacial lakes and their changing role in a warming climate, *Nature Reviews Earth & Environment*, 3(2), 106-124.
- 409 Ojha, L., S. Karimi, J. Buffo, S. Nerozzi, J. W. Holt, S. Smrekar, and V. Chevrier (2021), Martian Mantle Heat Flow
- 410 Estimate From the Lack of Lithospheric Flexure in the South Pole of Mars: Implications for Planetary Evolution and
- 411 Basal Melting, Geophysical Research Letters, 48(2), e2020GL091409, doi: https://doi.org/10.1029/2020GL091409.
- 412 Ojha, L., S. Karimi, K. W. Lewis, S. E. Smrekar, and M. Siegler (2019), Depletion of Heat Producing Elements in
- the Martian Mantle, in *Geophysical Research Letters*, edited, John Wiley & Sons, Ltd, doi:10.1029/2019GL085234.

- 414 Orosei, R., et al. (2018), Radar evidence of subglacial liquid water on Mars, *Science*, *361*(6401), 490-493,
- 415 doi:doi:10.1126/science.aar7268.
- 416 Parro, L. M., A. Jiménez-Díaz, F. Mansilla, and J. Ruiz (2017), Present-day heat flow model of Mars, in *Scientific*
- 417 *Reports*, edited, doi:10.1038/srep45629.
- 418 Petrenko, V. F., and R. W. Whitworth (1999), *Physics of ice*, OUP Oxford.
- 419 Phillips, R. J., et al. (2011), Massive CO<sub>2</sub> Ice Deposits Sequestered in the South Polar Layered
- 420 Deposits of Mars, *Science*, *332*(6031), 838-841, doi:doi:10.1126/science.1203091.
- 421 Plesa, A. C., S. Padovan, N. Tosi, D. Breuer, M. Grott, M. A. Wieczorek, T. Spohn, S. E. Smrekar, and W. B.
- 422 Banerdt (2018), The Thermal State and Interior Structure of Mars, in *Geophysical Research Letters*, edited,
- 423 doi:10.1029/2018GL080728.
- 424 Putzig, N. E., M. T. Mellon, K. A. Kretke, and R. E. Arvidson (2005), Global thermal inertia and surface properties
- of Mars from the MGS mapping mission, in *Icarus*, edited, pp. 325-341, doi:10.1016/j.icarus.2004.08.017.
- 426 Smith, I. B., D. E. Lalich, C. Rezza, B. H. N. Horgan, J. L. Whitten, S. Nerozzi, and J. W. Holt (2021), A solid
- interpretation of bright radar reflectors under the Mars south polar ice, *Geophysical Research Letters*, 48(15),
  e2021GL093618.
- 429 Sori, M. M., and A. M. Bramson (2019), Water on Mars, With a Grain of Salt: Local Heat Anomalies Are Required
- for Basal Melting of Ice at the South Pole Today, *Geophysical Research Letters*, 46(3), 1222-1231,
- 431 doi:<u>https://doi.org/10.1029/2018GL080985</u>.
- 432 Toner, J. D., D. C. Catling, and B. Light (2014), The formation of supercooled brines, viscous liquids, and low-
- 433 temperature perchlorate glasses in aqueous solutions relevant to Mars, in *Icarus*, edited,
- 434 doi:10.1016/j.icarus.2014.01.018.
- 435 Wieczorek, M. A. (2008), Constraints on the composition of the martian south polar cap from gravity and
- 436 topography, in *Icarus*, edited, pp. 506-517, doi:10.1016/j.icarus.2007.10.026.
- 437 Wilson, E. H., S. K. Atreya, R. I. Kaiser, and P. R. Mahaffy (2016), Perchlorate formation on Mars through surface
- 438 radiolysis-initiated atmospheric chemistry: A potential mechanism, in Journal of Geophysical Research: Planets,
- 439 edited, pp. 1472-1487, Wiley Online Library.
- 440 Yu, W., X. Zeng, X. Li, G. Wei, and J. Fang (2022), New Martian Dust Simulant JMDS-1 and Applications to
- Laboratory Thermal Conductivity Measurements, *Earth and Space Science*, 9(1), e2020EA001347,
- 442 doi:https://doi.org/10.1029/2020EA001347.
- 443 Zuber, M. T., R. J. Phillips, J. C. Andrews-Hanna, S. W. Asmar, A. S. Konopliv, F. G. Lemoine, J. J. Plaut, D. E.
- Smith, and S. E. Smrekar (2007), Density of Mars' South Polar Layered Deposits, in *Science*, edited, pp. 1718-1719,
- 445 doi:10.1126/science.1146995.
- 446

Figure 1.







Figure 2.



Figure 3.

