Microbial Models for Simulating Soil Carbon Dynamics: A Review

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Abstract

Soils store the largest amount of carbon (C) in the biosphere, and the C pool in soil is critical to the global C balance. Numerous microbial models have been developed over the last few decades to represent microbial processes that regulate the responses of soil organic carbon (SOC) to climate change. However, the representation of microbial processes varies, and how microbial processes are incorporated into SOC models has not been well explored. Here, we reviewed 70 microbial models to characterize the microbial processes incorporated into SOC models and analyzed variations in mechanistic complexity. We revealed that (1) four processes (microbial decomposition, mineral interaction, microbial mortality, and transition between active and dormant microbial states) are commonly incorporated in microbial models, (2) ~47% of models simulate only one (i.e., microbial decomposition) and 33% simulated two microbial processes: microbial decomposition and mineral interaction, (3) more than 80% microbial models use nonlinear mathematical equations, such as forward Michaelis-Menten kinetics, to represent SOC decomposition, (4) the concept of persistence of SOC due to its intrinsic properties has been replaced by organo-mineral interaction (~39% of microbial models) that protects SOC from decomposition, and (5) various temperature and moisture modifiers and pH effects have been used to explain the environmental effect on microbial processes. Finally, we propose a roadmap for SOC model improvement. In the future, to realistically incorporate microbial processes into Earth System Models, it is imperative to identify experimental evidence on rate limitation processes and firmly ground model structure on the field and laboratory data.

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1 2	Microbial Models for Simulating Soil Carbon Dynamics: A Review					
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8	Key Points:					
9	1. Recent model development indicates a growing interest in incorporating microbial					
10	processes					
11	2. Diversity in mathematical equations indicates uncertainty in translating current					
12	knowledge of microbial processes into models					
13	3. Field observational data are critical to validate the mechanistic representation of					
14	microbial processes					
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22 Abstract

23 Soils store the largest amount of carbon (C) in the biosphere, and the C pool in soil is critical to 24 the global C balance. Numerous microbial models have been developed over the last few 25 decades to represent microbial processes that regulate the responses of soil organic carbon (SOC) to climate change. However, the representation of microbial processes varies, and how microbial 26 27 processes are incorporated into SOC models has not been well explored. Here, we reviewed 70 microbial models to characterize the microbial processes incorporated into SOC models and 28 analyzed variations in mechanistic complexity. We revealed that (1) four processes (microbial 29 30 decomposition, mineral interaction, microbial mortality, and transition between active and dormant microbial states) are commonly incorporated in microbial models, (2) ~47% of models 31 simulate only one (i.e., microbial decomposition) and 33% simulated two microbial processes: 32 33 microbial decomposition and mineral interaction, (3) more than 80% microbial models use nonlinear mathematical equations, such as forward Michaelis-Menten kinetics, to represent SOC 34 decomposition, (4) the concept of persistence of SOC due to its intrinsic properties has been 35 replaced by organo-mineral interaction (~39% of microbial models) that protects SOC from 36 decomposition, and (5) various temperature and moisture modifiers and pH effects have been 37 used to explain the environmental effect on microbial processes. Finally, we propose a roadmap 38 for SOC model improvement. In the future, to realistically incorporate microbial processes into 39 Earth System Models, it is imperative to identify experimental evidence on rate limitation 40 41 processes and firmly ground model structure on the field and laboratory data.

42

44 1. Introduction

45 Soils store the largest amount of organic carbon (C) in terrestrial ecosystems (Lehmann & 46 Kleber, 2015). Thus, even a small change in soil C turnover could have significant consequences 47 for atmospheric CO₂ concentrations and the stability of the global climate system (Luo et al., 2016; Schmidt et al., 2011). It is estimated that the global soil organic carbon (SOC) pool size at 48 49 a depth of 1 m is 1417–1469 Pg C (Hiederer & Köchy, 2011), which is nearly three times the 50 amount of C stored in plant biomass (Schlesinger & Bernhardt, 2013) and twice the amount of C in the atmosphere (Schmidt et al., 2011). Therefore, it is crucial to understand and simulate the 51 52 critical processes underlying the dynamics of SOC to accurately forecast its responses to future changes in climate and land management (Amelung et al., 2020). However, the current process-53 based models have very high uncertainty in estimating the response of global SOC to climate 54 55 change (Fan et al., 2021; Todd-Brown et al., 2013; Wieder et al., 2013). These uncertainties result partly from inadequate representations of ecosystem processes that control the exchanges 56 of water, energy, and C between land ecosystems and the atmosphere (Hao et al., 2015; Wieder 57 58 et al., 2013) and partly from the uncertainties in estimating the SOC model parameters (Abramoff et al., 2022; Luo & Schuur, 2020; Luo et al., 2016). 59 Developing models that accurately simulate belowground processes is challenging for soil, 60 environmental, and earth sciences (Hinckley et al., 2014; Todd-Brown et al., 2013; Wieder et al., 61 2015a). Currently, SOC dynamics in Earth System Models (ESMs) are mostly represented by 62 conventional SOC models that do not explicitly simulate microbial activity or soil microbial 63 communities. Instead, these models strongly emphasize the relationship between SOC chemical 64 recalcitrance and soil C storage (Wieder et al., 2014; Zeng et al., 2006), assuming that respired 65

66 CO₂ is proportional to the soil C pool size (Davidson et al., 2014; Wieder et al., 2015a). These

conventional SOC models implicitly represent microbial activities under the assumptions that (1)
microbes respond so quickly to changes in substrate availability that their abundance never limits
the decomposition rate (Schimel, 2001), (2) microbial and other ecosystem properties as
expressed by parameters in models are invariant across wide environmental and edaphic
conditions and through time (Luo & Schuur, 2020), and (3) microbial communities have
functional equivalence allowing them to optimally process the available SOC (Bradford &
Fierer, 2012; Wieder et al., 2015a).

Theoretically, it is known that microbial processes fundamentally regulate the decomposition and 74 75 stabilization of SOC (Davidson et al., 2014). Therefore, in the past few decades, researchers have incorporated various microbial processes to improve the simulation of future C-cycle-climate 76 feedback (Wieder et al., 2013) and as a result, numerous microbial models have been developed 77 78 over the last few decades to simulate microbial regulation on the response of SOC to climate change. Here we define a microbial model as a soil biogeochemical model that simulates at least 79 one discrete microbial biomass pool that controls the decomposition rate of SOC. Studies to date 80 indicate large variations among microbial models in the capacity to simulate and predict SOC 81 82 dynamics, possibly due to their variations in model structure and representations of various processes in models. Since the 1970s, many microbial models have been developed, for example, 83 the Schimel model (Schimel & Weintraub, 2003), the enzyme-driven model (Allison et al., 84 2010), ReSOM (Tang & Riley, 2015), and MIND (Fan et al., 2021). Several studies have 85 reported contrasting findings when SOC dynamics were compared between conventional and 86 microbial models. For example, one study compared a conventional SOC model (similar to the 87 CENTURY model) with microbial models (EC1 and EC2) to simulate soil respiration from a 88 laboratory-based pulsed drying-rewetting experiment, revealed that microbial controls on SOC 89

90	decomposition improved the model's ability to capture the observed pulsed soil respiration
91	(Lawrence et al., 2009). However, other studies reported similar or amplified uncertainty in SOC
92	responses to climate change when incorporating microbial control on SOC decomposition, which
93	might be due to complex mechanisms in microbial processes and the challenges of
94	parametrization (Shi et al., 2018; Sulman et al., 2018). For example, by selecting suitable
95	environmental response functions and an improved parameterization method, conventional SOC
96	models could also capture the pulse dynamics of soil heterotrophic respiration similarly well
97	with microbial models (Zhou et al., 2021). In addition, the uncertainty of the MIMICS microbial
98	model in projecting long-term SOC was >10 times greater than that in the conventional Century-
99	type model, possibly because the complex model structure and a large number of parameters
100	increased uncertainty due to feedback in the model dynamics (Shi et al., 2018).
101	Although several studies have reviewed SOC models (Chertov et al., 2007; Frissel & Van Veen,
102	<u>1981; McGill, 1996; Molina & Smith, 1997; Paustian, 1994; Paustian et al., 1997; Smith et al.,</u>
103	1998; Wieder et al., 2015a; Xu et al., 2016), comprehensive synthesis and analysis of microbial
104	processes incorporated into SOC models are lacking. Microbial models vary a lot in terms of
105	representations of microbial processes and their incorporations into SOC models. To provide an
106	overview of the status of microbial models, in this study, we reviewed 70 microbial models
107	developed over the last few decades (Table 1). We first examine the history of microbial model
108	development and the trend of the microbial processes incorporated. Then, we provide a
109	comprehensive overview on each of the microbial processes that substantially overlap across
110	microbial models and their mechanistic representations into SOC models. For each process, we
111	include the mathematical equations adopted in the models and the environmental factors that
112	influence them. We finish the review with the challenges associated with microbial models and

- some recommendations that would be beneficial for better model development in estimating
- 114 SOC dynamics.

Table 1: Microbial models for SOC decomposition and the four processes incorporated along with the decomposition formulation. ZO: Zero-order; FO: First-order; FMM: Forward Michaelis Menten; RMM: Reverse Michaelis-Menten (The names of the models are determined based on two criteria: (1) if the model has a name in the original publication, that name will be used to represent the model; (2) if the model has no name, the last name of the first author will be used to name the model, for example, Fatichi model.)

Microbial Model	Decomposition mechanism	Microbial decomposition	Transition between active and dormant microbial state	Mineral interaction	Microbial mortality	References
Averill model	RMM	Yes	No	No	No	<u>Averill (2014)</u>
Barot model	FO	Yes	No	No	No	Fontaine and Barot (2005)
Blagodastsky model	FO	Yes	No	No	No	Blagodatsky et al. (2010)
CLM-Microbe	FMM	Yes	No	No	No	Wieder et al. (2013)
CMAX framework	RMM	Yes	No	No	No	<u>Xu et al. (2014)</u>
COMISSION	RMM	Yes	No	Yes	No	Ahrens et al. (2015)
CORPSE	RMM	Yes	No	Yes	Yes	<u>Sulman et al. (2014)</u>
C-STABILITY	Multiplicative	Yes	No	No	No	Sainte-Marie et al. (2021)
DAMM-MCNiP	ECA	Yes	No	No	No	Abramoff et al. (2017)
DecoBio v1.0	FMM	Yes	No	No	No	Xenakis and Williams (2014)
DEMENT	FMM	Yes	No	No	No	Allison (2012)
DNDC	ZO	No	No	No	No	<u>Li et al. (1994)</u>
DORMANCY	FMM	Yes	Yes	No	No	<u>He et al. (2015)</u>
DORMANCY 2.0	FMM	Yes	Yes	No	No	<u>Liu et al. (2019)</u>
EC1	RMM	Yes	No	No	No	Lawrence et al. (2009)
EC2	RMM	Yes	No	No	No	Lawrence et al. (2009)
EcoSMMARTS	RMM	Yes	Yes	No	Yes	Brangarí et al. (2020)
Ecosys	FMM	Yes	No	No	No	<u>Grant et al. (1993)</u>
EEZY	RMM	Yes	No	No	No	Moorhead et al. (2012)
Enzyme driven model	FMM	Yes	No	No	No	Allison et al. (2010)
Fatichi	FMM	Yes	No	No	No	<u>Fatichi et al. (2019)</u>
FOND	ZO	No	No	Yes	Yes	<u>Fan et al. (2021)</u>
GDM	FMM	Yes	No	No	No	Moorhead and Sinsabaugh (2006)
GENDEC	ZO	No	No	No	Yes	Moorhead and Sinsabaugh (2000)
German	FMM	Yes	No	No	No	<u>German et al. (2012)</u>
Hagerty	FMM	Yes	No	No	No	<u>Hagerty et al. (2018)</u>
He model	FMM	Yes	No	No	No	<u>He et al. (2014)</u>
JSM	RMM	Yes	No	Yes	Yes	<u>Yu et al. (2020)</u>
Kaiser	FMM	Yes	No	No	Yes	Kaiser et al. (2014)
LIDEL	ZO	No	No	No	No	<u>Campbell et al. (2016)</u>
Manzoni	RMM	Yes	No	No	No	<u>Manzoni et al. (2021)</u>
MEMS v1.0	ZO	No	No	Yes	No	Robertson et al. (2019)
MEND	FMM	Yes	No	Yes	No	<u>Wang et al. (2013)</u>
MEND_dor	FMM	Yes	Yes	Yes	No	<u>Wang et al. (2015)</u>
MESDM	FMM	Yes	Yes	No	No	Zhang et al. (2022)

MiCNiT	Multiplicative	Yes	No	No	No	<u>Blagodatsky et al. (2011)</u>
MIC-TEM_Hao	FMM	Yes	No	No	No	<u>Hao et al. (2015)</u>
MIC-TEM_Zha	FMM	Yes	No	No	No	Zha and Zhuang (2018)
MIC-TEM-dormancy	FMM	Yes	Yes	No	No	Zha and Zhuang (2020)
MiFe	Logistic	Yes	No	Yes	No	<u>Liao et al. (2022)</u>
Millennial model	FMM	Yes	No	Yes	No	Abramoff et al. (2018)
Millennial V2.0	RMM	Yes	No	Yes	No	Abramoff et al. (2022)
MIMICS	FMM	Yes	No	Yes	No	Wieder et al. (2014)
MIMICS_D	FMM	Yes	No	Yes	No	<u>Zhang et al. (2020)</u>
MIMICS-CN v1.0	FMM	Yes	No	Yes	No	Kyker-Snowman et al. (2020)
MIMICS-DB	FMM	Yes	No	Yes	No	<u>Zhang et al. (2020)</u>
MIMICS-DBT	FMM	Yes	No	Yes	No	Zhang et al. (2020)
MIND	FMM	Yes	No	Yes	Yes	<u>Fan et al. (2021)</u>
MOMOS	ZO	No	No	No	Yes	<u>Pansu et al. (2010)</u>
MySCaN	ZO	No	No	No	No	<u>Orwin et al. (2011)</u>
NCSOIL	RMM	Yes	No	No	No	<u>Hadas et al. (1998)</u>
ORCHIMIC v1.0	ECA	Yes	Yes	Yes	Yes	<u>Huang et al. (2018)</u>
ORCHIMIC v2.0	FMM	Yes	Yes	Yes	Yes	<u>Huang et al. (2021)</u>
Parnas	FMM	Yes	No	No	No	<u>Parnas (1975)</u>
Phoenix	Density-	Yes	No	Yes	No	<u>McGill et al. (1981)</u>
	dependent					
Resat	FMM	Yes	No	No	No	<u>Resat et al. (2012)</u>
RESOM	ECA	Yes	No	Yes	No	Tang and Riley (2015)
ReSom vNN	ECA	Yes	No	Yes	No	<u>Abramoff et al. (2019)</u>
ReSom vTD	ECA	Yes	No	Yes	No	<u>Abramoff et al. (2019)</u>
ReSom vTI	ECA	Yes	No	Yes	No	<u>Abramoff et al. (2019)</u>
ReSom vTN	ECA	Yes	No	Yes	No	<u>Abramoff et al. (2019)</u>
RothC	ZO	No	No	No	No	Coleman and Jenkinson (1996)
SCAMPS	FMM	Yes	No	No	No	<u>Sistla et al. (2014)</u>
Schimel model	RMM	Yes	No	No	No	Schimel and Weintraub (2003)
SOCRATES	ZO	No	No	No	No	<u>Grace et al. (2006)</u>
SOMic v1.0	RMM	Yes	No	Yes	No	Woolf and Lehmann (2019)
SOMKO	Density-	Yes	Yes	No	No	<u>Gignoux et al. (2001)</u>
OVMDHONN	dependent	V	N	N	NL	
SYMPHUNY TDIDLEV MICDODE	FU	Y es	INO Var	INO	INO No	Perveen et al. (2014)
IKIPLEX_MICKOBE	FIMIM	Y es	Yes	Yes	INO	<u>vvang et al. (2017)</u>
VERBERNE	ZO	No	No	Yes	No	<u>Verberne et al. (1990)</u>



slow during the late 20th century. Only 8



Figure 1: Historic development of microbial models since 1975 (a and b); Percentage of microbial models with consideration of major microbial processes. The percentage was calculated as the number of models considering each process divided by the total number of published models in each time period (c)

microbial models were developed during the last 25 years of the 20th century (Fig 1b), and the
treatment of microbial biomass was often indistinguishable from the active pool of conventional
SOC models, such as VERBERNE, GENDEC, DNDC, and RothC microbial models. Microbial
models started to receive more attention, mainly after Schimel and Weintraub (Schimel &
Weintraub, 2003) proposed the Reverse Michaelis-Menten kinetics derived from the Langmuir
sorption isotherm theory and explicitly represented the extracellular enzyme (ENZ) pool in their

139	model. Subsequently, several studies explored additional ecological interactions between
140	microorganisms and SOC. For example, it was previously thought that the long-term persistence
141	of SOC was because of the recalcitrant chemical property of SOC, such as humic substances that
142	were considered large, complex macromolecules and the most stable component of SOC
143	(Lützow et al., 2006). However, recent studies suggested that the recalcitrant components
144	represent only a small fraction of total OM, and the molecular property alone does not control
145	the persistence of SOC (Kleber & Johnson, 2010; Sutton & Sposito, 2005). Rather, mineral
146	surfaces predominantly influence the decomposition of SOC by altering SOC concentration and
147	its mobilities (Greenland, 1965). Mineral particles in soil adsorb SOC onto its surfaces by
148	forming various chemical bonds that prevent SOC accessibility from microbes (McGill et al.,
149	1981; McLaren & Peterson, 1965), resulting in explicit consideration of the mineral interaction
150	process in the models.
151	Likewise, relatively recent development in microbiology and genomics uncovered that under
152	natural environmental conditions, soil microbes exist in three physiological states: dead, alive,
153	and dormant microbes (Gignoux et al., 2001; Mason et al., 1986; Wang et al., 2014a). Thus, a
154	significant increase in the trends of the number of microbial models (Fig. 1a and 1b) and
155	consideration of microbial processes controlling SOC decomposition were observed (Fig 1c). For
156	example, 22 and 35 microbial models were developed during the periods of 2007 to 2014 and
157	2015 to 2022, respectively (Fig 1a). Microbial processes such as microbial mortality and

dynamic active-dormant microbial states are relatively less studied than the microbial processes

related to decomposition and mineral association because of our recent knowledge development

160 of microbial physiological states and the limitation in their measurements *in situ*.

162 **3. Model Representation of Microbial Processes**

163 Early on, empirical fitting of a first-order model to 164 SOC decomposition required multiple pools so that fractions of SOC decayed with different 165 turnover rates (Woolf & Lehmann, 2019). Such 166 167 multi-pool models, derived from empirical results, reflect a conceptual paradigm that different types 168 169 of SOC have different representative 170 turnover rates. Although many microbial processes are suggested to be essential for 171 controlling SOC cycling in the literature 172 (Lehmann et al., 2020; Schmidt et al., 2011), 173 there is less agreement about the best 174 mathematical formulations to represent these 175 processes (Table 2, (Sulman et al., 2018; Wieder 176 177 et al., 2015a).



Figure 2: Number of microbial models (a) with major processes incorporated into microbial models; (b) simulating the number of microbial processes

Our review of the 70 microbial models revealed that four microbial processes are widely incorporated into SOC models: microbial decomposition, mineral interaction, microbial mortality, and transition between active and dormant microbial states (Fig 2a; Table 1). Among the 70 microbial models, almost 50% simulate two processes (i.e., microbial decomposition), and one-third of the microbial models simulate two processes in a single model. The combinations of the two processes are mostly between microbial decomposition and mineral interaction (Fig 2b;

- Table 1). In this section, we focus on these four microbial processes that have been incorporated
- into most of the microbial models.

186 3.1 Microbial Decomposition of SOC

187 Microbial decomposition is a critical process in the soil C cycle because it is the primary 188 189 pathway through which CO₂ fixed by plants 190 is returned to the atmosphere (Zhang et al., 2022). Therefore, microbial models have 191 192 taken diverse approaches to represent the decomposition process (Fig. 1c, Table 2). 193 There is a consensus among microbial 194 models that microbes produce ENZ to degrade 195 complex SOC into dissolved organic carbon 196



SOC decomposition reaction mechanism

Figure 3: Number of microbial models (a) adopted various SOC decomposition reaction mechanisms; (b) representing linear and nonlinear decomposition kinetics

(DOC) through catalysis, take up DOC, convert the assimilated C into microbial biomass for 197 growth, and release CO₂ through respiration (Sinsabaugh et al., 2008; Zhang et al., 2022). Two 198 pathways are used for the representation of the decomposition of SOC: enzymatic- and microbial 199 biomass-mediated decomposition (Fig 2a). The major difference between these two pathways is 200 that enzymatic-mediated decomposition models simulate an explicit ENZ pool, assuming ENZ 201 production is controlled by both substrate concentration and microbial community structure 202 (Sistla et al., 2014) and directly couple SOC decomposition to the ENZ activity instead of 203 microbial biomass (Table 2). For example, microbial models such as the Averill model, DAMM-204 MCNiP, DEMENT, DORMANCY, EC1, EC2, EcoSMMARTS, EEZY, Enzyme driven model, 205 Fatichi, German, Hagerty, HE model, Kaiser, MEND dor, MESDM, MIC-TEM Hao, MIC-206

207	TEM_Zha, MIC-TEM-dormancy, MEND, ORCHIMIC (v1.0, v2.0), Resat, RESOM (v1, v2, v3,
208	v3, v4, v5), SCAMPS, Schimel model, and TRIPLEX_MICROBE (Table 1) consider a separate
209	ENZ pool in addition to microbial biomass pool (Table 2). In contrast, microbial biomass-
210	mediated decomposition models (e.g., Barot model, Blagodastsky model, C-Stability, CLM-
211	Microbe, CMAX framework, COMISSION, CORPSE, DecoBio v1.0, DORMANCY 2.0,
212	Ecosys, GDM, JSM, Manzoni, MiCNiT, MiFe, Millennial model, Millennial v2.0, MIMICS,
213	MIMICS (v1, v2, v3, v4), MIND, NCSOIL, Parnas, SOMic v1.0, SOMKO, and SYMPHONY
214	(Table 1) do not explicitly simulate the pool of ENZ. Instead, they implicitly assume the
215	enzymatic catalysis of SOC to drive the rate of SOC decomposition. We consider these pathways
216	to be separate processes in our analyses to preserve the uniqueness of their model structures and
217	their process representations (Table 2).
218	Further, we classified the microbial decomposition of SOC into six types based on the equations
219	used: 1) zero-order, 2) first-order, 3) forward Michaelis-Menten (FMM), 4) Reverse Michaelis-
220	Menten (RMM), 5) Equilibrium Chemistry Approximation (ECA), or 6) Other (Fig 3a). Types 1
221	and 2 are linear-type whereas types 3-6 are nonlinear models. These formulations differ
222	functionally with different fundamental assumptions on whether the decomposition of SOC is
223	limited by substrate availability, microbial biomass (or ENZ), or both, and on how these
224	components are linked with decomposition. For example, formulations 1 (i.e., zero-order) and 2
225	(i.e., first-order) are represented by simple mathematical equations (consisting of only one
226	parameter). In contrast, the nonlinear microbial model family (i.e., formulations 3-6) is
227	represented by various complex mathematical equations with a large number of parameters. It is
228	worth to note that more than 80% of the microbial models used nonlinear kinetics (formulations
229	3-6) to represent SOC decomposition (Fig 3b; Table 1).

230 3.1.1 Zero-Order Microbial Model (ZO)

The zero-order microbial model is not a function of microbial biomass although the model
simulates a discrete microbial biomass pool. Such a model may be a function of substrate as:

233

237

$$D_c = kS \tag{1}$$

where D_c is the rate of decomposition of C, S is the substrate of SOC, and k is the coefficient of the decomposition rate.

Of the 70 models we reviewed, 10 models simulated microbial biomass as one (e.g., FOND,

GENDEC, LIDEL, MEMS v1.0, MOMOS, RothC, VERBERNE) or more components (e.g.,

238 bacteria and fungi pools in MySCaN, protected and unprotected microbial C in SOCRATES,

labile and resistant microbial C in the DNDC model) of the SOC pool (Fig 3a). However, the

240 microbial component was only taken as a substrate of decomposition rather than as a decomposer

that could modify the rates of decomposition in these models. Thus, the SOC decomposition

reaction rate becomes zero-order with respect to microbial biomass (Eq 1). This formulation is

similar to the conventional SOC models in which each substrate with specific quality has its own

244 microbial community associated with it, and the microbial community is presumably in

equilibrium with the substrate most of the time. Therefore, decomposition is only limited by

- substrate (S) (McGill & Myers, 1987; Wutzler & Reichstein, 2008). Furthermore, the
- 247 contribution of soil microbes to SOC decomposition is implicitly included in the apparent

248 decomposition rate coefficients of different C pools (McGill, 1996; Paustian, 1994) when a

- 249 model is parameterized. Due to this treatment of soil microbes, models are independent of
- temporal and spatial variations in the soil microbial community and may lack the flexibility to

simulate the effects of land-use or climate change that impact soil microbial biomass andactivities on SOC decomposition (Fang et al., 2005).

253 **3.1.2** First-Order Microbial Model (FO)

In the first-order microbial models, the decomposition of SOC depends linearly on the size of the 254 microbial biomass pool (Eq. 1 in Table 2). Only 3 (Barot, Blagodastsky, and SYMPHONY 255 256 models) out of the 70 microbial models used first-order kinetics for microbial biomass (Fig 3a), as the assumption of this type of model, that is, the substrate is the only limiting factor for SOC 257 decomposition was questioned in the Barot model and later was adopted in Blagodastsky and 258 SYMPHONY models (Blagodatsky et al., 2010; Fontaine & Barot, 2005; Perveen et al., 2014). 259 These models considered that the decomposition of recalcitrant SOC was limited by the 260 extracellular enzymes instead of the quantity of substrate and assumed that the quantity of ENZ 261 is proportional to the size of the microbial biomass pool. The SOC decomposition increases 262 linearly with the size of the microbial pool (Eq. 1 in Table 2), resulting in first-order kinetics 263 264 with respect to the microbial biomass.

265 3.1.3 Forward Michaelis-Menten (FMM)

266 The first-order (linear) models have been challenged on the grounds that SOC breakdown

depends on the amount of SOC as well as on microbial components (Fang et al., 2005; Schimel

- 268 <u>& Weintraub, 2003</u>), thus resulting in nonlinear decomposition rates. This tight coupling
- between the substrate and biological processes is necessary, in particular, while modeling short-
- term C and nitrogen (N) dynamics (<u>Blagodatsky et al., 1998</u>), while it might also be relevant in
- 271 medium- (Whitmore, 1996) and long-term analyses (Smith et al., 1998). Based on the
- assumption that the decomposition rate of SOC is limited by the substrate or the microbial pool

(or the enzyme pool), various mathematical equations were used to describe the decompositionof SOC (Table 2).

275 The FMM kinetics assumes that substrate availability is the rate-limiting factor in

decomposition, i.e., the decomposition rate saturates as the substrate available for decomposition

277 rises (Wieder et al., 2015a). In FMM kinetics, the SOC decomposition rate varies linearly with

the microbial biomass (or enzyme pool) and nonlinearly with the substrate. The representation of

279 SOC decomposition in microbial models was dominated by FMM kinetics: 31 out of 70

280 microbial models (~44%) used the FMM kinetics for SOC decomposition (Eqs. 2-6 in Table 2).

281 3.1.4 Reverse Michaelis-Menten (RMM)

282 In contrast to FMM kinetics, in RMM kinetics, the decomposition reaction rate changes linearly

with the amount of substrate and saturates with the enzyme pool (or microbial biomass) (Schimel

284 <u>& Weintraub, 2003</u>). The assumptions underlying the RMM kinetics are (1) the size of SOC pool

is sufficiently large enough such that the amount of ENZ (or microbial biomass), rather than the

substrate, is the rate-limiting factor for SOC decomposition, (2) the maximum binding capacity

of enzymes is proportional to the concentration of the substrate, and (3) the resulting

decomposition rate is proportional to the amount of bound enzyme (Moorhead & Weintraub,

289 2018; Schimel & Weintraub, 2003). The Averill model, CMAX framework, COMISSION,

290 CORPSE, EC1, EC2, EcoSMMARTS, EEZY, JSM, Manzoni, NCSOIL, Schimel model, and

291 SOMic v1.0 models have all adopted RMM kinetics for the SOC decomposition (Eqs. 7-12 in

Tables 2). However, some of these models, such as the CMAX framework, COMISSION,

293 CORPSE, JSM, Manzoni, NCSOIL, and SOMic v1.0, do not separately simulate ENZ pools but

instead assume that ENZ production linearly depends on microbial biomass because it is

challenging to measure ENZ production and these models focus on simulating C pools that aremeasurable.

297 **3.1.5** Equilibrium Chemistry Approximation (ECA)

ECA is a relatively new decomposition mechanism proposed by Tang and Riley (2013). It was derived from the first-order approximation of the full equilibrium chemistry formulation of a consumer-substrate network that can account for multiple consumers (i.e., microbes and minerals) and multiple substrates as:

$$D_c = \frac{V_{max} * S * E}{k_m + S + E}$$
(2)

303 where V_{max} is the maximum decomposition rate, k_m is the half-saturation constant, and E is the 304 extracellular enzyme

The assumptions underlying the ECA kinetics are (1) there is no binding between substrates or between consumers, and (2) once an enzyme-substrate complex is formed, it will not bind with another substrate or consumer to form new complexes (<u>Tang & Riley, 2013</u>). Eq. 2 demonstrates a reaction that has only one enzyme interacting with one substrate.

309 It should be noted that the decomposition rate becomes FMM when the substrate changes

significantly while the enzyme concentration is much lower than the substrate, such that $k_m + E$

is almost constant. On the other hand, when the substrate concentration is much higher than the

enzyme concentration, such that the microbial process barely changes the total substrate

313 concentration in the temporal window of interest, $k_m + S$ is almost constant, and eq. 2 is reduced

to RMM (Tang & Riley, 2013). ReSOM and its subsequent versions, DAMM-MCNiP and

ORCHIMIC v1.0, used the ECA mechanism (Eqs. 7, 13-14 in Table 2).

316

318 3.1.6 Other Mechanisms

Although the mechanisms mentioned above were the most commonly used in microbial models, 319 other mechanisms have been used in some instances. For example, Phoenix and SOMKO models 320 321 use microbial density-dependent SOC decomposition and assume that the increase in the ratio of microbial C to structural C slows down the activity of microbes because of increased competition 322 among microbes for nutrients and space (Eqs. 15-16 in Table 2). Subsequently, to avoid such a 323 heavy nonlinear model parameterization and also assuming the low concentrations of SOC, some 324 models (e.g., C-Stability and MiCNiT) use a multiplicative expression that still couples microbes 325 and SOC, but the decomposition of SOC varies linearly with microbial biomass and substrate 326 (Eqs. 17, Table 2) (Manzoni & Porporato, 2007). Furthermore, a recent study (Liao et al., 2022) 327 was conducted to understand the most probable mechanisms behind the observed nonlinear 328 329 patterns of lignin decomposition. The study reported that neither the conventional nor FMM nonlinear models could simulate the observed nonlinear patterns of lignin decomposition (Huang 330 et al., 2019). Instead, the data-driven approach revealed that time-dependent growth and 331 mortality functions expressed by logistic equations in the microbial-iron (MiFe) model better 332 represented observed CO₂ release from lignin decomposition than models assuming either first-333 order or FMM (Eq. 18 in Table 2). 334

335 **3.2 Mineral Interaction**

SOC interaction with mineral surfaces is a critical process for the stabilization of SOC because plant- and microbially-derived SOC can be protected from decomposition through the formation of complex organo-mineral interactions (Abramoff et al., 2019). Sorption and desorption are the two processes that regulate the amount of DOC available to microbes for decomposition (Huang et al., 2018). The majority of SOC models simulate the mineral interaction implicitly by modifying the SOC decomposition rate with an empirical factor based on the clay fraction

342 (Abramoff et al., 2019; Coleman & Jenkinson, 1996; Sulman et al., 2014; Wieder et al., 2013).

- However, it is becoming increasingly clear that the persistence and decomposition of SOC are
- interconnected with the physical environment, organic-mineral interactions, and both local biotic
- and abiotic factors (<u>Newcomb et al., 2017</u>). In addition, a significant proportion of stable SOC is
- derived from simple C rather than chemically resistant compounds (<u>Cotrufo et al., 2013</u>),
- 347 suggesting molecular structure alone does not control the long-term stability of SOC (Schmidt et
- 348 <u>al., 2011</u>). Sorption is a rapid process that occurs within seconds to minutes and thus occurs more
- rapidly than microbial decomposition (Kothawala et al., 2008; Qualls & Haines, 1992).
- 350 Therefore, the long residence time or the stabilization of SOC are commonly attributed to an
- 351 interaction between DOC, microbially derived C, or intact plant compounds with mineral
- 352 surfaces, which provide reactive sites for physical and chemical stabilization, thus preventing
- degradation of SOC by microbes (Grant et al., 2022; Kleber et al., 2015; Kleber et al., 2007;
- 354 <u>Lehmann & Kleber, 2015; Schmidt et al., 2011</u>). Factors influencing the formation and stability
- 355 of protected C include the chemistry of OM, texture, and structure of soils, physicochemical
- properties and abundance of soil minerals, pH, the ionic strength of the soil water, temperature,
- and moisture (Abramoff et al., 2018; Feng et al., 2016).
- In our analysis, 27 out of 70 microbial models explicitly represented the mineral interaction with
- 359 SOC (Fig 2a; Table 1). However, not all of these 27 models explicitly simulate the sorption and
- desorption processes simultaneously. For example, CORPSE, FOND, MEMS, MIND, and
- 361 VERBERNE do not explicitly represent desorption; instead, they simulate the net sorption of C
- to mineral surfaces, meaning that when the mineral-associated organic matter (MAOM) pool
- 363 reaches saturation, the net transfer of C from DOC to MAOM can be negative, i.e., C is
- transferred from MAOM to DOC (Eqs. 20, 21, 23, 30 in Table 2).

365	Environmental and biotic controls on sorption varied greatly among models. Factors considered
366	that could influence the sorption process include DOC (or SOC), MAOM, maximum sorption
367	capacity (Q_{max} , depending on clay and silt content), soil temperature and moisture, and microbial
368	necromass (i.e., mass from microbial death and subsequent lysis and fragmentation of microbes).
369	In most microbial models, the maximum sorption of SOC depended on the amount of DOC
370	available, the availability of sorption sites, and the sorption capacity (Eqs. 19-33 in Table 2).
371	Thus, the rate of SOC sorption increases when the DOC content is higher, and the sorption sites
372	are unoccupied. In addition, models such as JSM, Millennial v1.0, and ORCHIMIC (v1.0 and
373	v2.0) introduced temperature rate modifiers for the sorption process (Eq. 22, 26, 31 in Table 2).
374	Only JSM and Millennial (v1.0 and v2.0) models simulated the effect of soil moisture on the
375	sorption process (Eqs. 22, 25-26 in Table 2). In most microbial models, particulate organic
376	carbon (POC) and DOC were competing for the mineral surfaces. However, in some microbial
377	models, such as FOND, Millennial (v1.0 and v2.0), MIMICS (v1-4), and MIND, microbial
378	necromass was also competing for the mineral surfaces (Eqs. 21, 25-30 in Table 2).
379	Environmental and biotic controls on desorption also varied among models but in less complex
380	ways than their controls on sorption. Microbial models such as COMISSION, JSM, MEND,
381	MEND_dor, Millennial (v1.0 and v2.0), MIMICS(v1-v5), MIND, ORCHIMIC (v1.0 and v2.0)
382	and SOMic v1.0 explicitly represented the desorption process (Eq. 19, 22, 24-29, 31, 33 in Table
383	2). The desorption mostly depended on the amount of C sorbed to the mineral surfaces and Q_{max} .
384	However, some models modulated the desorption process by adding temperature (JSM,
385	Millennial (v1.0), and ORCHIMIC (v1.0 and v2.0) or moisture functions (JSM, Millennial v1.0,
386	and v2.0).

388 3.3 Microbial Mortality

389 Although the microbial models reviewed in this study consider the carbon pool of microbial 390 biomass separately, most microbial models did not explicitly represent the microbial necromass 391 pool with a different decomposition rate from plant residue. For example, in the Millennial model, a fraction of microbial necromass and plant residues (such as root exudates and leaf 392 393 leachate) enter into the same C pool, i.e., low molecular weight carbon, which follows the same 394 decomposition pattern (Abramoff et al., 2018). The microbial necromass pool mainly consists of microbially derived SOC, such as dead microbes and extracellular compounds released from the 395 396 dead microbes, that have a faster decomposition rate than the plant residues (Huang et al., 2018). Several studies reported that soil microbes have different structural and chemical compositions 397 from plant litter, which could result in their different decomposition rates (Kögel-Knabner, 2002; 398 Liang et al., 2017). For example, the global mean C:N ratio of microbial biomass (~7) (Xu et al., 399 2013) is much lower than that of plant litter (~53) (Yuan & Chen, 2009), which may cause 400 decoupling of C and N if microbes prioritize SOC with high N content to meet their demands. 401 Consequently, the microbial assimilation of high N-containing SOC for the growth of microbial 402 biomass may lead to different decomposition rates between microbial necromass and plant 403 residues because of varying chemical structures and characteristics of microbially- and plant-404 derived SOC (Kögel-Knabner, 2002; Liang et al., 2017). For example, in the GENDEC model, 405 the decomposition rate of microbial necromass is higher than that of the plant residues because 406 the N-content of microbial necromass is relatively higher than that of the plant residues 407 (Moorhead & Reynolds, 1991). 408

Although the C pool size of active microbial biomass in the soil is minimal (<2% of total SOC,

410 (<u>Dalal, 1998</u>), microbial necromass may accumulate over a long period of time, and it can

411	contribute to a significant proportion of SOC if (1) the turnover rate of microbial biomass is
412	higher than the input rate of plant litter (Liang et al., 2011; Simpson et al., 2007; Zhu et al.,
413	2020), (2) the chemical composition of microbial necromass is not labile, (3) mineral matrix of
414	the soil protects microbially derived SOC (Dwivedi et al., 2017; Mikutta et al., 2006; Miltner et
415	al., 2012; Torn et al., 1997). According to the previous studies, the contribution of microbial
416	necromass to SOC can range from 24-80% of SOC (Khan et al., 2016; Liang et al., 2019; Liang
417	& Balser, 2011; Miltner et al., 2012). Therefore, the role of microbial necromass in the formation
418	of SOC cannot be ignored when considering microbial biomass as a decomposer in microbial
419	models (Fan et al., 2021; Kögel-Knabner, 2002).
420	It is widely known that different microbial groups differ in their chemical composition. For
421	instance, the cell walls of fungi are composed of a high proportion of recalcitrant polymers (e.g.,
422	protein and melanin), whereas bacterial cell walls are made up of carbohydrates (Kögel-Knabner,
423	2002). However, despite the differences in the cell wall composition of microbial groups, the
424	decomposition rates of necromass of different microbial groups in the soil have been found to be
425	similar (<u>Throckmorton et al., 2012</u>). In our review, we found 10 out of 70 microbial models,
426	including CORPSE, EcoSMMARTS, FOND, GENDEC, JSM, Kaiser, MIND, MOMOS, and
427	ORCHIMIC (v1.0 & v2.0) explicitly represented a separate microbial necromass pool under the
428	assumption that decomposition of microbial necromass is 1) similar among different microbial
429	groups, but 2) different from that of plant residues. However, the mechanistic representation of
430	microbial necromass in a microbial model still poses some challenges that is discussed in section
431	5.2.

434 **3.4** Active and Dormant Microbial Dynamics

435 In a given environment, at any given time, microorganisms can be in any of the physiological 436 states: active, dormant, or dead (Mason et al., 1986). Therefore, distinguishing these states in the 437 microbial models may be important to modeling SOC accurately. The active fraction of microbial communities play a significant role in ecologically important processes like SOC 438 439 decomposition and nutrient cycling (Blagodatsky et al., 2000). However, when environmental conditions are unfavorable for growth, for example, when there is not enough substrate, microbes 440 may reduce metabolic activities from low to zero to prevent biomass loss and may enter into 441 dormant states (Lennon & Jones, 2011; Stolpovsky et al., 2011). The dormant microbes do not 442 play the same roles as those active microbes, and dormancy is considered an evolutionary 443 strategy that preserves genotypes until conditions improve to allow replication (Price & Sowers, 444 2004). The maintenance cost of C in dormant microbes can be two to three orders of magnitude 445 lower than that of metabolically active microorganisms (Anderson & Domsch, 1985a; Anderson 446 & Domsch, 1985b). 447 It is important to represent active versus dormant microbes in microbial models to accurately 448 simulate SOC dynamics, given the variations in substrate and environmental conditions over 449 time and space. With seasonal variations in substrate availability, temperature, and moisture, 450 many soils have slow SOC turnover rates. Even when some resources are abundant at a time, the 451 spatial and temporal complexity of soils may lead to disproportionate distributions of other 452 potentially limited resources, which can dramatically increase the dormancy rates. High 453 dormancy rates may be a defining characteristic of soil systems when spatial and temporal 454 complexity is paired with various resource distributions across species within a community. 455 Therefore, an understanding of dormancy could improve the prediction on how active microbes 456

457 contribute to ecosystem processes like decomposition and nutrient cycling (Blagodatsky et al.,

458 <u>2000; Wang et al., 2014a</u>).

Despite the potential importance, it is challenging to study microbial dormancy because there is 459 460 no single method available to measure individual microbial physiological states: active, dormant, or dead simultaneously; instead, a combination of various techniques has been used to quantify 461 microbial states (Wang et al., 2014a). In microbial models, generally, there are two methods used 462 to depict physiological states (Wang et al., 2014a): one is to separate total live microbial biomass 463 into two pools: active and dormant (Table 1); and another is to directly regard the active fraction 464 (i.e., a ratio of active to total live microbial biomass) as a state variable (Blagodatsky et al., 465 1998). However, despite the limited ability to distinguish between active, dormant, and dead 466 microbial biomass, a wealth of studies suggest that in a given microbial community, the majority 467 468 of microbes may be dormant under natural circumstances (Blagodatsky et al., 2000; Yarwood et al., 2013). For example, in a Typic Argiudoll soil from Argentinean Pampa, only 3.8–9.7 % of 469 total microbial biomass is in the active state (Alvarez et al., 1998); however, 0.02–19.1 % and 470 9.2-24.2 % of total microbial biomass are in active states in the subkurgan paleosoils of different 471 ages and modern background soils, respectively (Khomutova et al., 2004). Other studies reported 472 that under natural soil conditions, the fraction of active microbial biomass is usually below 50%473 of total live microbial biomass (Lennon & Jones, 2011; Stenstrom et al., 2001; Van de Werf & 474 Verstraete, 1987). Thus, not including dormancy from the microbially-driven ecosystem 475 processes could result in incorrect estimates of total live microbial biomass, leading to 476 inaccuracies in model parameterization and forecasts of SOC (Wang et al., 2014a). 477 In our review, only 10 out of 70 models explicitly simulated microbial transformation between 478 479 active and dormant states (Brangarí et al., 2020; Gignoux et al., 2001; He et al., 2015; Huang et

480	al., 2018; Huang et al., 2021; Liu et al., 2019; Wang et al., 2015; Wang et al., 2017; Zha &
481	Zhuang, 2020; Zhang et al., 2022), (Table 1). SOMKO was one of the first microbial models that
482	distinguish active and dormant microbial biomass (Gignoux et al., 2001). In SOMKO, the
483	direction of net flux from the active to the dormant state depends on the maintenance
484	requirement relative to substrate availability. If the substrate availability is less than the
485	maintenance requirement, there is a positive net flux from the active to the dormant pool and
486	vice versa. Later, MEND_dor introduced the rates of dormancy and reactivation of microbial
487	biomass (Wang et al., 2014a) into the MEND model (Wang et al., 2013). Following Wang et al.
488	(2013), a few more microbial models were developed by adopting the MEND_dor dormancy
489	framework to simulate SOC decomposition. For example, ORCHIMIC (v1.0 and v2.0) and
490	TRIPLEX_Microbe microbial models have incorporated the MEND_dor dormancy framework
491	along with the following assumptions: (1) the dormancy $(B_{a\to d})$ and reactivation rates $(B_{d\to a})$ are
492	proportional to the active and dormant biomass pool sizes, respectively; (2) when substrate
493	concentration is very high, $B_{a\to d} \to 0$ and $B_{d\to a} \ge 0$; (3) when substrate concentration is very
494	low, $B_{a\to d} \ge 0$ and $B_{d\to a} \to 0$; (4) both transformation processes are governed by the maximum
495	specific maintenance rate for active microbes since the maintenance energy cost is the critical
496	factor determining the dormancy strategy (Lennon & Jones, 2011; Wang et al., 2014a). Unlike
497	the above-mentioned microbial models that consider the substrate dependence of dormancy, the
498	microbial dormancy in the microbial models DORMANCY 2.0, EcoSMMARTS, and MESDM
499	is also affected by soil moisture content (Table 2). Such microbial models were developed to
500	simulate the soil respiration in soil moisture-limited conditions to capture the drying-rewetting
501	effect (i.e., Birch effect) under the assumptions that the soil water content determines the overall

- 502 microbial performance and changes in soil water content can alter the physiological state of a
- 503 portion of the microbes (<u>Brangarí et al., 2020</u>; <u>Zhang et al., 2022</u>).

504 Table 2: Mathematical formulations of the four processes: Microbial Decomposition (Eqs. 1-18), Mineral Interaction (Eqs. 19-33), Microbial Mortality (Eqs. 34-505 37), and Transition between Active and Dormant States (Eqs. 38-42)

Equations	Ecological description	Models
1. $D_c = k^*M$	A function of microbial biomass (M)	Barot model; BLAGODATSKY; SYMPHONY
2. $D_c = V_{max} * M * \frac{s}{K_m + s}$	The function of microbial biomass (M) and substrate(S)	Parnas; MIND; GDM; German; CLM-Microbe; MIMICS: MIMICS-CN v1.0: MIMICS-D. MIMICS-
		DB, MIMICS-DBT; Ecosys; DecoBio v1.0
3. $D_c = V_{max} * E * \frac{s}{K_m + s}$	Function of Extracellular enzyme (E) and substrate(S)	Enzyme-driven model; Fatichi; Hagerty; He model; Kaiser; MEND_dor; MESDM; MEND; MIC-TEM; Resat; SCAMPS; TRIPLEX_Microbe; DEMENT; ORCHIMIC v2.0
4. $D_c = V_{max} * \frac{v_{temp-15}}{10} * E * \frac{s}{K + s} * (120 - CN_{soil})$		DORMANCY; MIC-TEM-dormancy
5. $D_c = V_{max} * \frac{s}{K_c + s} * \frac{M}{K_m + M} * f(T, W)$	Double Michaelis-Menten kinetics	Millennial model
6. $D_c = k^* M^* \frac{c}{(\kappa_c + c)} \frac{\sigma_2}{(\kappa_{o_2} + \sigma_2)}$	Function of DOC and O_2 (dissolved oxygen concentration in water)	DORMANCY v2.0; DAMM; MIC-TEM_Hao
7. $D_c = V_{max} * S* \frac{E}{K_m * f(T) + S + E} * f(pH, W, T, clay)$	Function of S, E, clay content, soil pH, temperature (T), and moisture (W)	ORCHIMIC v1.0
8. $D_c = k^* S^* \frac{M}{K_m + M}$	Reverse Michaelis-Menten. Millennial V2 includes moisture modifier function, f(W)	CMAX framework; NCSOIL; COMISSION; Millennial V2
9. $D_c = V_{max} * S * \frac{E}{K_m + E}$	A function of S and E	JSM; Schimel model; EEZY; Averill model; Manzoni; SOMic v1.0
10. $D_{c} = V_{max} * S * \frac{E}{K_{m}+E} * f(T, W)$	A function of S and E, T, and W	EC1, EC2
11. $D_{c} = V^{*}S^{*}\frac{E}{K_{m}+E} f(W)$		EcoSMMARTS
12. $D_c = k^* S^* \frac{M/S}{K_m + \frac{M}{S}} f(W)$	Function of S and ratio of M and S	CORPSE
13. $D_c = V_{max} * \frac{S*E}{K_m + S + E}$	ECA	DAMM-MCNiP
14. $D_c = V_{max} * S * \frac{E}{K_{es}(1 + \frac{S}{K_{es}} + \frac{E}{K_{es}} + \frac{Min}{K_{me}})}$	A function of S, E, and mineral particle (Min)	ReSOM; ReSOM vNN, ReSOM vTN , ReSOM vTD, ReSOM vTI
15. $D_c = k^* \frac{1}{1 + K_1(\frac{M}{S})K_2} * M * f(T, W)$	Microbial density-dependent SOC decomposition	Phoenix
16. $D_c = (1 - e^{-k\frac{M}{s}})S$	Exponentially related to microbial biomass (M)	SOMKO
17. $D_c = k^*M^*S$		C-STABILITY; MiCNiT
18. $D_c = \frac{\theta_1}{1 + e^{-\theta_2(t - \theta_3)}}$	θ_1 , θ_2 , θ_3 are maximum rate, growth rate and lag phase	MiFe
19. Sorption = K_{ads}^* DOC *(q_{max} - C_{MAOM}) Desorption = $K_{des}^* C_{MAOM}$	The availability of sorption sites limits the sorption rate Langmuir isotherm	COMISSION
20. Creat comption = SOC*f(clay) - $\frac{cp}{cp}$	The rate of protected C formation is proportional to the	CORPSE
$= net_{solption} = (\pi f) f$	amount of unprotected C pool. τ is the residence time of protected C.	
21. $C_{MAOM} = (1 - f_{BNF}) k_B M + k C_{DN} - R K_{NF} C_{MAOM}$	k is the decomposition rate of microbes derived DOC (C_{DN})	FOND

		R is the ratio of decomposition rate of C_{MAOM} to fast pool of microbial necromass	
22.	Sorption = $K_{ads}^* DOC^* e^{\frac{-Ea}{R} \left(\frac{1}{T} - \frac{1}{T_{ref}}\right)^* \frac{W}{dz}} Q_{max}$	dz is soil depth Langmuir isotherm	JSM
	Desorption = $K_{des} * e^{\frac{-Ea}{R} \left(\frac{1}{T} - \frac{1}{T_{ref}}\right) * \frac{W}{dz}} C_{MAOM}$		
23.	Net Sorption = DOC* $(\frac{\frac{(K \cdot Q_{max} \cdot DOC)}{1 + (K \cdot DOC)} - \mathcal{C}_{MAOM}}{Q_{max}})$	Sorption and desorption are not simulated separately, Q_{max} maximum sorption capacity, Langmuir isotherm	MEMS v1.0
24.	Sorption = $K_{ads}^* (1 - \frac{Q}{Q_{max}}) * DOC$	Q is adsorbed phase of DOC K_{adc} and K_{dsc} are sorption and desorption rate	MEND; MEND_dor; TRIPLEX Microbe
	Desorption = $K_{des}^*(\frac{q}{q_{max}})$		_
25.	Sorption = DOC* $\left(\frac{k_{lm}*Qmax*DOC}{1+(K_{lm}*DOC)}-C_{MAOM}\right)$ f(T, W) + k _m *M*f(T,	K _{lm} is binding affinity O _{max} is maximum sorption capacity	Millennial
	W) + $k_b * f(T, W)(1-p_a) * C_{aggregate}$	BD is bulk density L is LMWC	
	Desorption = $V_{ma} * \frac{C_{MAOM}}{K + C_{MAOM}} (1 - \frac{C_{aggregate}}{A_{max}}) f(T, W)$	A_{max} is the maximum capacity of C in soil aggregates k_m is sorption rate of microbial biomass	
	$K_{\rm int} = 10^{(-0.186 pH - 0.216)}$	$_{\rm L}^{\rm m}$ is rate of breakdown	
26	$Q_{\text{max}} = BDIO(c_1 c_2) + c_2$	K ₁ is the hinding affinity	Millennial V2
26.	Sorption = $K_{lm} * DOC * (1 - \frac{1}{Q_{max}}) f(W) + p_b K_{bd} * M^2$ + (1 - n) k * C * F(W)	%claysilt is the clay and silt content in percent and a	
	(1 - Pa) (b + Caggregate + 1 (V))	coefficient (p_c) K _w is desorption coefficient	
	Desorption = $K_{ld} \frac{c_{MAOM}}{c_{MAOM}} + (1-p_a) k_{ma} c_{MAOM} f(W)$	Depth is site-level sampling depth in m	
	$K_{\rm lm} = e^{-p1pH - p2} K_{\rm ld}$	k_{ma} is the aggregate formation rate from MAOM	
	$Q_{max} = depth*BD %claysilt*p_c$	p _a is the proportion of aggregate C allocated to POM Langmuir isotherm	
27.	Sorption = f_1 *Input + f_2 *M Desorption = $1.5*10^{-5}$ * k_d * $e^{-1.5*f_{clay}}$	k_d is the coefficient of desorption rate	MIMICS MIMICS-CN v1.0
28.	Sorption = f_1 *Input + f_2 *M	K_{dp} iss the coefficient for tuning the relationship between	MIMICS-D (Kyker-Snowman et al., 2020)
20	Desorption = $1.5*10^{-5*}k_d*e^{-1.5*j}clay*e^{k_dp*t_MAOM}$ Sorption = f_* Input + f_*M	the desorption and C_p pool.	MIMICS-DB
29.	Desorption = $1.5*10^{-5*}k_d*e^{-1.5*f_{clay}}*e^{k_{dp}*C_{MAOM}}*e^{k_{bs}*BS}$	desorption	MIMICS-DD
30.	$C_{net_sorption} = (1 - f_{BNF}) * k_B * M - R * \frac{V_{max} * M + C_{MAOM}}{K_M + C_{MAOM}}$	f_{BNF} is proportion of fast pool in microbial biomass, k_B is average mortality rate. M is microbial biomass	MIND (Fan et al., 2021)
		R is the ratio of decomposition rate of C_{MAOM} to fast pool	
		of microbial necromass	
31.	Sorption = K _{ads} *DOC * $e^{\frac{-Ea_{ads}}{R} \left(\frac{1}{T} - \frac{1}{T_{ref}}\right)} * \left(1 - \frac{C_{MAOM}}{Q_{max}}\right)$	Function of temperature Arrhenius equation	ORCHIMIC v1.0; ORCHIMIC v2.0
	Desorption = $K_{des} * e^{\frac{-Ea_{des}}{R} \left(\frac{1}{T} - \frac{1}{T_{ref}}\right)} * \left(\frac{C_{MAOM}}{Q_{max}}\right)$		

32.	$Q_{\max} = \frac{x}{m} = k^* C^{1/n}$	X is grams of OM adsorbed, m is weight of soil, k is sorption constant, C is DOC	Phoenix
33.	Sorption = $f_{sorb} * k*DOC$ Desorption = $f_{desorb}*C_{MAOM}$	f_{sorb} is the sorption coefficient, k is the rate constant for the combined processes of microbial uptake and sorption	SOMic v1.0
34.	D* B _a	Death rate of microbes is fraction (D) of their active biomass (B _a)	ORCHIMIC v1.0; ORCHIMIC v2.0
35.	$C_N = D^*M$	Function of mortality rate, and microbial biomass	CORPSE; MIND; FOND; EcoSMMARTS
36.	$(k_1+k_3)^*M+k_2^*\gamma$	k1, k2, k3, and γ are the minimum microbial death rate, growth-related microbial death rate, and mortality associated with the change in soil water potential, and microbial growth, respectively.	GENDEC
37.	$m = \frac{1}{M_{CS}} * m_f$	M_{CS} is the maximum cell size of a microbe, m_f is a factor relating mortality rate to the inverse of maximum microbial biomass	Kaiser
38.	$R_{a-d} = k_{tran} * 1/(1 + (\frac{Sesample}{Sehalf})^{b}) * B_{a}$ $R_{d-a} = k_{tran} * 1/(1 + a * (\frac{Sehalf}{Sesample})^{b}) * B_{d}$	k_{tran} is the maximum transition rate constant, Se_{sample} is effective moisture saturation of sample, Se_{half} is the saturation at which R equals $0.5*K_{tran}$. B_a and B_d are active and dormant microbes, respectively.	DORMANCY 2.0
39.	$\begin{split} R_{a\text{-}d} &= k_i * (1 - \chi_a \xi_c) * B_a \\ R_{d\text{-}a} &= k_d * \Gamma_m * \xi_c * B_d \end{split}$	k_i and k_d are the maximum specific cell activation and deactivation rates. χ_a is the coefficient of water stress, ξ_c is the saturation coefficient of DOC, Γ_m is the coefficient of drought-legacy on microbes.	EcoSMMARTS
40.	$\begin{cases} S_{MBC} = B_d \cdot \frac{dW}{dt} & when \ \frac{dW}{dt} > 0 \\ S_{MBC = (B_a + B_{MBC_{WP}}) \cdot \frac{dw}{dt}} & when \ \frac{dW}{dt} < 0 \end{cases}$	S_{MBC} is microbial biomass transformation rate due to water content, $C_{MBC_{WP}}$ is potential active microbial biomass, respectively.	MESDM
41.	$R_{a-d} = \begin{bmatrix} 1 - \frac{DOC}{(K_D + DOC)} \end{bmatrix}^* m_R * B_a$ $R_{d-a} = \frac{DOC}{(K_D + DOC)} \end{bmatrix}^* m_R * B_d$	$m_{R} is$ the specific maintenance rate of B_{a}	ORCHIMIC v1.0; ORCHIMIC v2.0; MEND_dor; TRIPLEX_Microbe
42.	$R_{a-d} = (1 - \phi)^* m_R^* \frac{t^{emp-15}}{10} B_a^* B_a$ $R_{d-a} = \phi^* m_R^* \frac{t^{emp-15}}{10} B_d$	ϕ is the directly accessible substrate for microbial assimilation	DORMANCY; MIC-TEM-dormancy

506 4. Environmental Control on Microbial Processes

Many environmental factors affect microbial processes, including soil temperature, moisture, pH,
redox potential, and oxygen availability. This review mainly focuses on soil temperature,
moisture, and pH because they are commonly incorporated into microbial models (Table 3).
Among the 70 models we reviewed, 38 included temperature, 12 included soil moisture, and 7
included pH.

Temperature. In the microbial models, the temperature dependency of microbial 512 processes was simulated using four mathematical functions: (1) Q_{10} functions, (2) Arrhenius 513 functions, (3) Generalized Poisson function, and (4) Arctangent function. Of these functions, the 514 Arrhenius function is most widely used among most microbial models, followed by the Q_{10} 515 function. The Arrhenius function represents an increase in SOC decomposition with temperature 516 and dependence on substrate quality through the activation energy (Zhang et al., 2014). Only the 517 518 SOMic v1.0 model was found to use a Generalized Poisson function, determined by fitting data 519 from an incubation experiment conducted in the laboratory in which cellulose was labeled and decomposed at three different temperatures (Burke et al., 2003; Parton et al., 1987; Sorensen, 520 521 1981). Similarly, only the Millennial model was found to use the arctangent function, which 522 predicts a decline in temperature sensitivity with increasing temperature (Abramoff et al., 2018). 523 These two temperature response functions were taken from CENTURY and DAYCENT, respectively (Burke et al., 2003). 524

Moisture. Modeling the response of microbial communities to pulse moisture dynamics is challenging because moisture controls complex physical and biological interactions in soil and has significant direct and indirect impacts on the decomposition rates (Lawrence et al., 2009). Soil moisture is a critical factor controlling SOC decomposition because, at high water content,

529	O ₂ becomes a limiting factor, whereas, at low water content, diffusion is constrained by thin and
530	discontinuous water films (Abramoff et al., 2017; Abs & Ferrière, 2020; Sihi et al., 2018).
531	Various mathematical functions were used to represent the responses of microbial processes to
532	soil moisture; for example, the moisture response variables vary widely by including the function
533	of soil water potential, water holding capacity, or soil water content (Table 3).
534	pH. A small number of microbial models considered the effect of soil pH on microbial
535	processes (7 out of 70 models). Soil pH has a significant impact on mineral surfaces and SOC
536	availability to microbes. At high pH, the sorption capacity of mineral surfaces is reduced
537	drastically, which means that less SOC will be sorbed on the mineral surfaces, and most of the
538	SOC will be available to microbes for decomposition (<u>Abramoff et al., 2022</u>).
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552 Table 3: Environmental Control on Microbial Processes

Model	Temperature	Moisture	рН	Note
MEMS v1.0	Q ₁₀ =2, T _{ref} = 13.5 C			
GENDEC	2< Q ₁₀ >3 T _{ref} = 25 C	SM= α - μlog(-Ψ)	N/A	α and μ are intercept and slope of soil moisture effect on decay rate, Ψ is soil water potential
MOMOS	Q ₁₀ =2.2 T _{ref} =28 C	f(θ) = a + b θ/WHC	N/A	Moisture correction factor (f(θ)) WHC is water holding capacity, a and b range between 0 and 1
MIND; ORCHIMIC v1.0; ORCHIMIC v2.0; MEND; TRIPLEX-Microbe (f(T) & pH only); Microbial-Enzyme model (f(T) only)	$f(T) = e^{\frac{-Ea}{R}(\frac{1}{T} - \frac{1}{Tref})}$ T _{ref} = 285.15 K	f(θ)= max[0.25, min(1, -1.1*θ ² + 2.4*θ -0.29)]	$f(pH)= e^{\frac{-(pH-pH_{opt})^2}{pH_{sen}^2}}$ $pH_{opt} = 6$ $pH_{sen} = 1.66$	R is ideal gas constant, O is soil moisture (%), pH _{opt} is optimal pH for substrate decomposition pH _{sen} is the sensitivity parameter of substrate decomposition
SOCRATES	Q ₁₀ Temp factor, TF= 0.177 $e^{(0.069{ m T})}$	Moisture factor, MF= 0.0598*MAP ^{0.279}	N/A	T is mean annual air temperature (C)
MySCaN	$e^{\left(3.36*\frac{(T-40)}{(T+31.79)}\right)}$	N/A	N/A	Temperature response: Arrhenius function
ReSOM	$K_{EQ}(T) = K(T_0) e^{\left[\frac{-\Delta G_{EQ}}{R} \left(\frac{1}{T} - \frac{1}{T_0}\right)\right]}$ $K_{NEQ}(T) = K_{NEQ}(T_0) \frac{T}{T_0} e^{\left[\frac{-\Delta G_{EQ}}{R} \left(\frac{1}{T} - \frac{1}{T_0}\right)\right]}$ $f_{act} = \frac{1}{1 + e^{\left(-\frac{n\Delta G_{EQ}}{RT}\right)}}$	N/A	N/A	K_{EQ} is temperature-dependent equilibrium reactions; K_{NEQ} is temperature dependent non- equilibrium reactions; f_{act} temperature dependent fraction of active enzyme ReSOM vTN: KEQ=0 ReSOM vTD: In eq K _{EQ} (T), ΔG_{EQ} = -20 kJ/mol ReSOM vTI: In eq K _{EQ} (T), ΔG_{EQ} = 20 kJ/mol
DAMM-MCNiP; DEMENT; Ecosys; CORPSE	$e^{\frac{-Ea}{RT}}$	N/A	N/A	Arrhenius equation
Millennial	$\frac{t_2 + \frac{t_3}{\pi} \operatorname{atan}[\pi(T - t_1)]}{t_2 + \frac{t_3}{\pi} \operatorname{atan}[\pi t_4 (T_{ref} - t_1)]}$	$\frac{1}{1+w_1e^{(-w_2RWC)}}$	10 ^(-0.186pH-0.216)	t_1 and t_2 are x-axis and y-axis locations of the inflection point (°C), respectively t_3 is the distance from maximum to minimum point and t_4 is the slop of the line at the inflection point w_1 and w_2 are empirical parameters RWC relative water content T_{ref} is the reference temperature (30 °C)
Millennial v2.0	$a_{x}e^{-\frac{Ea}{R(T+273.15)}}$	$(rac{ heta}{arphi})^{0.5}$	K_{lm} = $e^{-p_1p_H-p_2} K_{ld}$	K_{lm} and K_{ld} are binding affinity and desorption coefficient, p1 and p2 are sorption coefficient, θ is volumetric water content, ϕ is matric potential, Arrhenius equation
CMAX framework	$f(T) = \begin{cases} 0, & T \le Ts_{min} \\ Q_{10s}^{T - Ts_{ref}} & T > Ts_{min} \end{cases}$	$f(M) = \frac{\log(\frac{M_{min}}{M})}{\log(\frac{M_{min}}{M_{max}})}$	N/A	Ts_{min} and Ts_{ref} are minimum and reference temperatures for the substrate decomposition M_{min} and M_{max} are the minimum and maximum

		when, $M_{min} \le M$ $\le M_{max}$		moisture for substrate decomposition
CLM-Microbe; Enzyme-driven model; German; He model; MIC-TEM-Hao; MIC-TEM-Zha; Hagerty; MESDM; DAMM; SCAMPS	$V_{max} = V_{max0} * e^{\left(-\frac{E_a}{R^*(temp+273)}\right)}$ $K_m = K_{mslope} * temp + K_{m0}$	N/A	N/A	Arrhenius equation
MIMICS-CN v1.0; MIMICS-D; MIMICS- DB; MIMICS-DBT	$V_{max} = e^{(V_{slope}*T+V_{int})*}a_{v}*V_{mod}$ K = $e^{(K_{slope}*T+V_{int})*}a_{k}*K_{mod}$	N/A	N/A	Arrhenius equation
JSM	$f(T, W) = e^{\frac{-Ea}{R}(\frac{1}{T} - \frac{1}{Tref})} * \frac{W}{soil_{depth}}$	N/A	N/A	Arrhenius equation
DecoBio v1.0; DORMANCY; MIC-TEM- DORMANCY	$Q_{10}^{\frac{T-15}{10}}$	N/A	N/A	
SOMic v1.0	$f(T)= f_t \left(\frac{T_{max} - T}{T_{max} - T_{opt}}\right)^2 e^{\frac{0.2}{2.63} \left(1 - \left(\frac{T_{max} - T}{T_{max} - T_{opt}}\right)^{2.63}\right)}$ $f_t = 4.99$ $T_{max} = 45 \ ^{\circ}C$ $T_{opt} = 35 \ ^{\circ}C$	If amd>0.444max_md, θ =1.0 If amd \leq 0.444max_md θ =0.2 +0.8 $\frac{\max_{max_{md}} - amd}{0.556\max_{md}}$	N/A	Max_md is maximum possible soil moisture deficit Generalized Poisson function for temperature modifier
EC1, EC2	N/A	<u>WHC</u> 60%	N/A	Water Holding Capacity (WHC)

553 5. Challenges and Recommendations

554	Despite the diverse representations of microbial processes and appropriate simulations of the
555	microbial responses to perturbations by microbial models, conventional SOC models remain the
556	backbone of SOC modeling in most applications, including ESMs, partly because microbial
557	models have not yet demonstrated their reliability to provide robust predictions over varying
558	environmental conditions and long-time scales (Wieder et al., 2013; Woolf & Lehmann, 2019).
559	Moreover, the microbial models were intended to represent the SOC dynamics better than the
560	conventional SOC model (Wang et al., 2014b) with the belief that microbial models may be
561	appropriate to describe the C cycling under variable environmental conditions (Schimel, 2001;
562	Schimel & Weintraub, 2003). However, it poses several challenges, including (1) the lack of
563	experimental evidence for the rate-limitation processes, (2) the lack of observational data to
564	constrain model parameters, and (3) model complexity due to a large number of parameters.
565	5.1 Experimental Evidence for Rate-Limitation Processes in Microbial Models
566	It is known that SOC is decomposed mainly as a result of ENZ produced by microbes, and it has
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566 567 568 569	It is known that SOC is decomposed mainly as a result of ENZ produced by microbes, and it has been demonstrated that microbes can degrade almost all SOC, irrespective of the chemical composition of SOC, if it is physically accessible to microbes (Kleber, 2010; Lützow et al., 2006; Woolf & Lehmann, 2019). Microbial models are mainly based on the assumption that the
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566 567 568 569 570 571 572 573	It is known that SOC is decomposed mainly as a result of ENZ produced by microbes, and it has been demonstrated that microbes can degrade almost all SOC, irrespective of the chemical composition of SOC, if it is physically accessible to microbes (Kleber, 2010; Lützow et al., 2006; Woolf & Lehmann, 2019). Microbial models are mainly based on the assumption that the SOC decomposition rate is limited by either microbial biomass or ENZ or both (Allison et al., 2010). However, a few studies report that in soil, microbial activities do not limit the rate of SOC decomposition; instead, abiotic processes are rate-limiting (Kemmitt et al., 2008). A common way for abiotic processes to control SOC decomposition is through physical protection that
566 567 568 569 570 571 572 573 573	It is known that SOC is decomposed mainly as a result of ENZ produced by microbes, and it has been demonstrated that microbes can degrade almost all SOC, irrespective of the chemical composition of SOC, if it is physically accessible to microbes (Kleber, 2010; Lützow et al., 2006; Woolf & Lehmann, 2019). Microbial models are mainly based on the assumption that the SOC decomposition rate is limited by either microbial biomass or ENZ or both (Allison et al., 2010). However, a few studies report that in soil, microbial activities do not limit the rate of SOC decomposition; instead, abiotic processes are rate-limiting (Kemmitt et al., 2008). A common way for abiotic processes to control SOC decomposition is through physical protection that limits microbial access to substrates (Dungait et al., 2012; Kemmitt et al., 2008; Schimel &

of microbes and their enzyme production never limits microbial processes, and microbial
communities will always rapidly adapt to the available substrate and subsidence of
environmental stress (<u>Schimel, 2001</u>). Thus, it is imperative to conduct experimental studies to
examine the assumption on the rate limitation processes by microbial biomass or enzyme
activity.

581 5.2 The Lack of Observational Data to Estimate Model Parameters

A lack of observational data is one of the most significant constraints to the validation of 582 mechanistic descriptions of microbial processes and the parameterization of microbial models. 583 Model development and data collection are generally separate activities, and their integration is 584 critical for the advancement of science (De Kauwe et al., 2014; Luo et al., 2012; Peng et al., 585 2011; Xu et al., 2016). In addition, the performance of a model is usually assessed by comparing 586 simulations against a set of empirical observations derived from independent experiments 587 (Moorhead & Sinsabaugh, 2006). When processes are made explicit rather than implicit, it is 588 589 essential to test the validity of those assumptions against the reality provided by data (Schimel, 2001). However, the incorporation of microbial processes increases the complexity of models 590 and enlarges the number of model parameters, which can be challenging to empirically measure. 591 For example, FMM, RMM, and ECA kinetics use two kinetic parameters, the maximum specific 592 reaction rate (V_{max}) and half saturation constant (K_m) . There are very few estimates of the K_m for 593 enzyme pools in explicitly enzyme-represented microbial models (Lawrence et al., 2009; 594 Moorhead & Sinsabaugh, 2006) or of the V_{max} or K_m for substrates (Wang et al., 2013). In 595 addition, observational data of pool size is critical to constrain rate processes (Xu et al., 2006). 596 However, the inability to measure active, dormant, and dead microbial biomass in situ 597 simultaneously (see section 3.4) and difficulty in differentiating microbial necromass C from 598

599	nonmicrobial C (Liang et al., 2019) present challenges in validating these processes. Similarly,
600	the measurement and evaluation of the stability of various SOC-mineral interactions in different
601	soils are challenging due to difficulty with the fractionation of SOC bound to different minerals
602	in situ (Lützow et al., 2006). Therefore, most of the parameter values used by microbial models
603	are primarily laboratory-based (Sulman et al., 2014; Wieder et al., 2015a; Wieder et al., 2013;
604	Wieder et al., 2015b) or assumed by the researchers (Wang et al., 2013), which may not reflect
605	the actual conditions in the field. As model parameterization is one of the three elements towards
606	realistic model predictions (Luo & Schuur, 2020), the research community needs to collect
607	observational data for estimating model parameters. When extensive SOC data were used to
608	estimate spatially varying parameters, the conventional Community Land Model (CLM4.5) can
609	well simulate spatial and vertical patterns of SOC storage over the US continent (Tao & Luo,
610	2022: Tao et al., 2020).

611 **5.3 Potential Improvements of Microbial Models**

While the incorporation of microbial processes increases model complexity, several strategies 612 can reduce the mismatch between model complexity and observational data. First, we need 613 targeted, precise data collection strategies because more data does not necessarily produce a 614 615 better-constrained model. Coordinated efforts between modelers and empiricists can return data maximally useful to constrain a model (Keenan et al., 2013; Richardson et al., 2010). Second, 616 inaccurate parameterization is emerging as one of the major causes of mismatches between 617 models and data (Luo & Schuur, 2020). Therefore, model improvements should include 618 optimization algorithms that calibrate model parameters with data, such as data assimilation 619 techniques (Luo et al., 2016; Wang & Chen, 2013). Third, an alternative approach is model 620 reduction techniques that can simplify complex models without the loss of key model processes 621

or the ability to integrate real data. Some commonly used model complexity reduction techniques include conversation analysis, nondimensionalisation, model decomposition (Snowden et al., 2017), and Manifold Boundary Approximation Method (MBAM) (Transtrum & Qiu, 2014). For example, a recent study applied the MBAM technique to a highly complex microbial model to demonstrate the systematic reduction of model complexity to match the information content of different datasets and thereby could explain fundamental controlling mechanisms in each dataset (Marschmann et al., 2019).

629 6. Summary

During the past three decades, SOC models have increasingly considered microbial controls on C 630 cycling to identify mechanisms that govern C fluxes. Although the first model was developed in 631 the 1970s, our review of 70 microbial models revealed that the majority of microbial models 632 were developed since 2000, likely in sync with the acceleration of molecular methods to study 633 soil microbes. Most of the microbial models incorporated one or more of four microbial 634 processes: microbial decomposition, mineral interactions, microbial mortality, or transition 635 between active and dormant microbial states. Among the four processes, microbial mortality and 636 dormancy were the least studied. 637

Our analysis synthesized a diverse suite of mathematical formulations used to represent
microbial processes across the 70 models. For example, 18 types of mathematical expressions
are used to describe the decomposition of SOC across the 70 microbial models. The majority of
models used nonlinear equations, especially forward Michaelis-Menten kinetics. Similarly, 15
types of mathematical expressions were devised to explain mineral interactions. Most microbial
models simulated sorption and desorption processes separately, while a few simulated net
sorption. The numerical expressions for microbial mortality were similar across models, with

645 mortality proportional to microbial biomass. The 10 models that simulated dormancy used one of 646 5 equations to describe the dynamics of active and dormant microbial states as a function of the 647 environmental stresses of soil temperature and moisture.

648 We identified the three major environmental factors incorporated in the microbial processes: soil temperature, soil moisture, and the effect of soil pH on the sorption capacity of minerals and/or 649 650 SOC availability to microbes. We propose that future model structures could benefit from the following considerations: (1) focusing on measurable soil pools so that model performance can 651 be evaluated against observational data, for example, (2) utilizing model-data integration 652 653 approaches to help identify the most probable mechanisms underlying system behavior, and (3) employing a rigorous statistical method, such as data assimilation, that helps improve the model 654 655 performance by optimizing the model parameters.

In conclusion, advancements in microbial genomics and computational sciences have improved 656 our understanding of the microbial processes governing SOC dynamics, and microbial models 657 658 bloomed in the past few decades to incorporate such advanced understandings. However, the diversity in mathematical equations used to incorporate microbial processes and the lack of 659 observational data to validate these processes limit the translation of the current knowledge on 660 SOC processes into models. Thus, to realistically incorporate microbial processes into ESMs, a 661 significant challenge for future research is to design experiments that could quantify the key 662 processes involved in the formation of various SOC pools in different soils. 663

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