# Geological carbon cycle constraints on the terrestrial hydrological response to higher atmospheric $CO_2$

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#### Abstract

How runoff will change as atmospheric  $CO_2$  rises depends upon several difficult to project factors, including  $CO_2$  fertilization, lengthened growing seasons, and vegetation greening. However, geologic records of the hydrological response to past carbon cycle perturbations indicate large increases in runoff with higher  $CO_2$ . We demonstrate that the fact that the Earth has remained habitable since life emerged sets a lower-bound on the sensitivity of runoff to  $CO_2$  changes. The recovery of the Earth system from perturbations is attributed to silicate weathering, which transfers  $CO_2$  to the oceans as alkalinity via runoff. Though many factors mediate weathering rates, runoff determines the total flux of silicate-derived cations and hence the removal flux of excess  $CO_2$ . Using a carbon cycle model that parameterizes weathering as a function of rock reactivity, runoff, temperature, and soil  $CO_2$ , we show that recovery from a perturbation is only possible if the lower-bound for the sensitivity of runoff to atmospheric  $CO_2$  is 0%/K. Using proxy data for the Paleocene-Eocene Thermal Maximum, we find that to match the marine  $d^{13}C$  record requires a runoff sensitivity greater than 0%/K and similar to estimates of the modern runoff sensitivity derived from an ensemble of Earth system models. These results suggest that the processes that enhance global runoff are likely to prevail over processes that tend to dampen runoff. In turn, that the Earth has always recovered from perturbations suggests that, though the runoff response is spatially complex, global discharge has never declined in response to warming, despite quite varied paleogeographies.

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7		ATMOSPHERIC CO <sub>2</sub>
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16	Key P	oints:
17	1.	The silicate weathering negative feedback depends upon runoff to drive recovery
18		from carbon cycle perturbations.
19	2.	Decreases in runoff with warming would lead to runaway greenhouse states.
20	3.	The sensitivity of runoff to warming has a lower-bound of 0%/K as dictated by
21		carbon cycle constraints.
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#### 23 Abstract

24 How runoff will change as atmospheric CO<sub>2</sub> rises depends upon several difficult to project 25 factors, including CO<sub>2</sub> fertilization, lengthened growing seasons, and vegetation greening. 26 However, geologic records of the hydrological response to past carbon cycle perturbations 27 indicate large increases in runoff with higher CO<sub>2</sub>. We demonstrate that the fact that the Earth 28 has remained habitable since life emerged sets a lower-bound on the sensitivity of runoff to 29 CO<sub>2</sub> changes. The recovery of the Earth system from perturbations is attributed to silicate 30 weathering, which transfers CO<sub>2</sub> to the oceans as alkalinity via runoff. Though many factors 31 mediate weathering rates, runoff determines the total flux of silicate-derived cations and hence 32 the removal flux of excess CO<sub>2</sub>. Using a carbon cycle model that parameterizes weathering as 33 a function of rock reactivity, runoff, temperature, and soil CO<sub>2</sub>, we show that recovery from a perturbation is only possible if the lower-bound for the sensitivity of runoff to atmospheric CO<sub>2</sub> 34 35 is 0%/K. Using proxy data for the Paleocene-Eocene Thermal Maximum, we find that to match 36 the marine  $\delta^{13}$ C record requires a runoff sensitivity greater than 0%/K and similar to estimates 37 of the modern runoff sensitivity derived from an ensemble of Earth system models. These 38 results suggest that the processes that enhance global runoff are likely to prevail over processes 39 that tend to dampen runoff. In turn, that the Earth has always recovered from perturbations 40 suggests that, though the runoff response is spatially complex, global discharge has never 41 declined in response to warming, despite quite varied paleogeographies.

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#### 43 Plain Language Summary

44 Runoff—the quantity of water delivered by rivers to the ocean—is likely to change as the 45 climate warms. However, because generation of runoff involves many small-scale processes, 46 it is difficult to forecast how global patterns of runoff will change as atmospheric CO<sub>2</sub> rises. 47 We take advantage of the fact that runoff is the primary mechanism that transfers excess CO<sub>2</sub> 48 from the atmosphere to the ocean for eventual burial as carbonate rock. It is this process that 49 permits the Earth to recover from major perturbations. We find that runoff must increase in a 50 warmer world; if it did not, then atmospheric CO<sub>2</sub> would continually increase in the 51 atmosphere, leading to a runaway greenhouse. This hard constraint places a lower-bound on 52 how runoff must change in response to rising atmospheric CO<sub>2</sub>.

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#### 1. Introduction

55 Runoff-the transfer of water from the continents to the oceans and to closed basins-56 is a critical component of the Earth system. Runoff (q) determines the amount of freshwater 57 available for societal use, transports nutrients from upland landscapes to lowland and marine 58 ecosystems and provides the water necessary to weather rocks that sequester atmospheric CO<sub>2</sub>. 59 Despite its importance for the functioning of the Earth system and for societies, how q will 60 change as atmospheric CO<sub>2</sub> rises remains highly uncertain. Part of this uncertainty arises 61 because the hydrological cycle is comprised of a number of different and interacting 62 components-including precipitation (P), bare-soil and vegetation-controlled evaporation (ET), and q—which may respond differently to rising atmospheric CO<sub>2</sub> and temperature (Milly 63 64 & Dunne, 2016). For some components, there are well grounded theoretical expectations for 65 how they should change. Precipitation globally is thought to increase by approximately 2%/K 66 given constraints on how much energy is available for evaporation (Held & Soden, 2006; 67 Pendergrass & Hartmann, 2014). Potential evapotranspiration (PET) over land should also 68 increase, as rising temperatures drive increases in the saturation vapor deficit. This increase in 69 *PET* is robustly predicted to be larger than the increase in *P* (Scheff & Frierson, 2014).

70 Collectively, these expectations-combined with observations that support them-71 suggest that much of the land surface will become more arid with warming (Cook et al., 2014; 72 Ficklin & Novick, 2017; Mankin et al., 2019; Milly & Dunne, 2020; Novick et al., 2016; 73 Overpeck & Udall, 2020). However, there remain two critical, linked components of the 74 hydrological cycle for which theoretical expectations and observational constraints are of low-75 confidence and poor—actual ET and q (Douville et al., 2021). ET determines how much P is 76 returned to the atmosphere and how much is partitioned to q; as a consequence, ET and P 77 determine surface water availability. Vegetation water use (*i.e.*, transpiration) is thought to be 78 the predominant component of  $ET (\sim 60\%)$  (Good et al., 2015; Schlesinger & Jasechko, 2014; 79 Wei et al., 2017); vegetation additionally impacts another 10-15% of ET due to intercepted P80 that evaporates directly from the leaf or woody surface (Wang-Erlandsson et al., 2014). Thus, 81 in addition to the physical response to CO<sub>2</sub> increase (P and PET), the global vegetation 82 response to rising CO<sub>2</sub> will further determine the trajectory of surface water availability.

However, projecting how global vegetation will change and its impact on q remains difficult. Though leaf-level transpiration is expected to decrease as CO<sub>2</sub> rises, total leaf area is expected to increase; the competition of these two processes globally will partially determine the total change in q. Satellite observations indicate that the world is overwhelmingly greening and that this greening is leading to greater *ET* (Forzieri et al., 2020; Winkler et al., 2021; Zhu

88 et al., 2016). In semi-arid regions, observations indicate this greening and associated increase 89 in ET has led to declining q (Ukkola et al., 2016), and many land-surface models—which are 90 designed to encapsulate a number of specific plant physiological process-find that increases 91 in leaf area have been substantial enough to reduce q globally (Forzieri et al., 2020; Piao et al., 92 2007). In contrast, coupled land-atmosphere models find that, while the Earth's surface greens 93 substantially, the resulting increase in ET fluxes is less than the increase in P (Lemordant et 94 al., 2018; Swann et al., 2016). Though this prediction is spatially heterogeneous (Mankin et al., 95 2019) and of low-confidence, the mean global average change in q with temperature is 96 predicted to be  $\sim 2.9 \pm 1$  %/K across the CMIP5 ensemble using the RCP8.5 scenario (Zhang 97 et al., 2014). Much of this increase is driven by widespread model agreement of increasing q98 in the northern high latitudes (Scheff et al., 2017). However, there remain serious questions 99 about how leaf-level processes and slower vegetation responses (i.e., species compositional 100 changes) are parameterized within these coupled models. As a consequence, there is no firm 101 theoretical expectation nor well-constrained observations to indicate whether global ET will 102 increase or decrease relative to P and, therefore, how globally averaged q will change as 103 atmospheric CO<sub>2</sub> rises.

104 This disagreement and uncertainty regarding the direction and magnitude of the global 105 q change as CO<sub>2</sub> rises stands in sharp contrast to both geologic observations of the hydrological 106 response to increasing CO<sub>2</sub> and theory regarding how the Earth system has recovered from 107 large carbon cycle perturbations in the geologic past. These recoveries are mediated by the 108 weathering of silicate rocks, which converts atmospheric CO<sub>2</sub> to alkalinity and permanently 109 buries carbon in marine sediments (R. A. Berner & Kothavala, 2001; Penman et al., 2020). 110 Critically, the weathering of rocks and transport of alkalinity to the ocean is thought to be 111 mediated primarily by q (Kump et al., 2000; Li et al., 2022; Maher & Chamberlain, 2014; Otto-112 Bliesner, 1995; Park et al., 2020), both to supply water to the weathering zone and to transport 113 the products of weathering to the ocean. The fact that life has persisted on Earth for more than 114 3 Ga indicates that this feedback must be particularly robust, capable of regulating Earth's climate largely independent of the precise position of the continents, extant flora and fauna, 115 116 and magnitude and nature of the various climatic perturbations that have befallen Earth over its history (Sagan & Mullen, 1972; Walker et al., 1981). Similarly, a growing body of literature 117 118 suggests that these past perturbations caused dramatic reorganizations of the terrestrial 119 hydrological cycle, recorded as thick packages of coarse-grained fluvial sediments during peak 120 warming (Chen et al., 2018; Foreman et al., 2012) and enhanced fluxes of nutrients and silica

to the ocean as reflected in widespread marine anoxia and chert deposits during many of thesecarbon cycle perturbations (Penman, 2016; Them et al., 2017).

123 Here, we exploit this fundamental negative feedback in the Earth system and its 124 relationship with q to provide a lower bound on how globally q must respond to changes in 125 atmospheric  $CO_2$ . To constrain how q must respond to changes in atmospheric  $CO_2$ , we employ 126 a version of a geological carbon cycle model that has been modified to represent the explicit 127 role of q in modulating weathered solute concentrations and fluxes. We demonstrate that the 128 operation of this weathering feedback requires that q does not decrease with rising atmospheric 129 CO<sub>2</sub> and, most likely, is positively sensitive to changes in CO<sub>2</sub>, such that the flux of weathered 130 material always increases if CO<sub>2</sub> rapidly increases in the atmosphere. We further test this model 131 against geological data from the Paleocene-Eocene Thermal Maximum (PETM), and 132 demonstrate that, to match geological proxy data from the PETM requires that globally 133 averaged q increase with atmospheric CO<sub>2</sub>. We conclude by discussing how these constraints on the sensitivity of the hydrological cycle carries profound implications for our understanding 134 135 of biogeochemical cycles in Earth's past and for Earth's future as CO<sub>2</sub> rises.

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#### 2. Background

#### 138 *2.1 Silicate Weathering and Runoff*

139 Silicate weathering is thought to be the primary process that removes CO<sub>2</sub> from the 140 coupled atmosphere-ocean system on geological timescales, neutralizing the input of CO<sub>2</sub> from 141 volcanoes (R. A. Berner & Kothavala, 2001). Silicate weathering is also thought to be sensitive 142 to climate, such that increases in the input flux of CO<sub>2</sub> and the resulting rise in atmospheric CO<sub>2</sub> causes increased silicate weathering fluxes, which removes this excess CO<sub>2</sub> from the 143 144 Earth's surface (Robert A. Berner & Caldeira, 1997; Zeebe & Caldeira, 2008). Though other 145 Earth surface processes are important for the geological carbon cycle, including the weathering 146 and burial of organic carbon and oxidation/reduction of sulfur (Hilton & West, 2020; Torres et 147 al., 2014), the evidence for these processes being sensitive to climate—and therefore capable of providing a negative feedback on atmospheric CO<sub>2</sub>—is less extensive. Further, the reaction 148 149 of silicate rocks with CO<sub>2</sub> dissolved in water (i.e., carbonic acid) is likely to have been 150 applicable for the entirety of Earth history. Critically, both data and models suggest that Earth system recovery once a perturbation ceases typically occurs within 1 Myr or less (Bowen, 151 152 2013; Colbourn et al., 2015; Lenton & Britton, 2006; Uchikawa & Zeebe, 2008).

Equation 1 describes the long-term relationship between total carbon in the oceanatmosphere system (M; mol C) and the input and output fluxes of CO<sub>2</sub> (Caves et al., 2016; Kump & Arthur, 1999):

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$$157 \qquad \frac{dM}{dt} = I - F_{silw} \tag{Eq. 1}$$

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159 where  $F_{silw}$  is the silicate weathering flux [mol C/yr] and *I* is the sum of the primary remaining 160 input and output fluxes of CO<sub>2</sub> [mol C/yr], including CO<sub>2</sub> from volcanism and solid Earth 161 degassing and organic carbon weathering and burial. On timescales longer than ocean 162 overturning, changes in the atmospheric partial pressure of CO<sub>2</sub> (*p*CO<sub>2</sub>) are proportional to 163 changes in *M*. The global silicate weathering flux is defined as (White & Blum, 1995):

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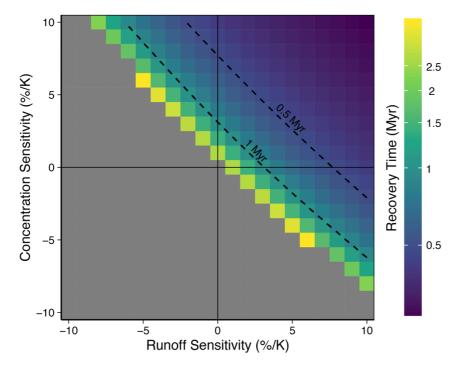
165 
$$F_{silw} = Q \times [C]_{sil}$$
(Eq. 2)

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167 where *Q* is global discharge [L/yr] and [C]<sub>sil</sub> is the global average concentration of bicarbonate 168 (HCO<sub>3</sub><sup>-</sup>) derived from the weathering of silicate rocks [mol C/L]. *Q* is related to *q* via the global 169 land area [m<sup>2</sup>] ( $Q = q \times area$ ), which, for the purposes of our simulations, we assume to be 170 constant. Consequently, for silicate weathering to act as a negative feedback on atmospheric 171 CO<sub>2</sub>, either *q*, [C]<sub>sil</sub>, or both must be sensitive to climate.

172 To demonstrate this, we use CH2O-CHOO (the Carbon-H<sub>2</sub>O Coupled HydrOlOgical 173 model), a modified version of CLiBeSO-W, which is a geological carbon cycle model that 174 calculates the mass balance of carbon and alkalinity in the ocean-atmosphere system (Caves 175 Rugenstein et al., 2019), and uses equation 2 to calculate  $F_{silw}$ . We first treat q and [C]<sub>sil</sub> as 176 being independently sensitivity to CO<sub>2</sub>-induced global warming (*i.e.*,  $\Delta q$  and  $\Delta$ [C]<sub>sil</sub> have units 177 of %/K) and assume a climate sensitivity of 4K/CO<sub>2</sub> doubling (Knutti et al., 2017). In these simulations, we inject 1000 Pg C into the atmosphere over 100 years and calculate the amount 178 179 of time required for atmospheric CO<sub>2</sub> to return to its initial value—termed the recovery time 180 [yr].

As shown in Figure 1, a negative sensitivity in either q or  $[C]_{sil}$  with temperature requires a correspondingly larger sensitivity in the other parameter. Recovery from perturbations is fastest when both parameters have a positive sensitivity, but small, positive sensitivities still result in recovery times greater than 1 Ma. Thus, to place constraints on the sensitivity of q to CO<sub>2</sub> requires knowledge of how  $[C]_{sil}$  will respond to atmospheric CO<sub>2</sub>. 186 Recent advances in biogeochemistry, however, have resulted in improved constraints on how 187 [C]<sub>sil</sub> is likely to change with climate (Godsey et al., 2009, 2019; Li et al., 2022; Maher & 188 Chamberlain, 2014; West, 2012; Winnick & Maher, 2018). Following previous work, we 189 consider here three parameters that are likely to have the greatest impact on [C]<sub>sil</sub> and which 190 are also sensitive to climate: temperature (*T*), runoff (*q*), and soil *p*CO<sub>2</sub>.



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*Figure 1*: The time required for atmospheric CO<sub>2</sub> to recover to pre-perturbation values (shading in  $10^6$  yr) as a function of the *q*-sensitivity (%/K) and [C]<sub>sil</sub>-sensitivity (%/K). Gray shading indicates that atmospheric CO<sub>2</sub> does not return to pre-perturbation values within 4 Ma postperturbation. Solid lines denote the 0% *q*- and [C]<sub>sil</sub>-sensitivities; dashed lines denote the 0.5 and 1 Myr recovery time contours.

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198 To encapsulate these three terms, we use a recently developed reactive-transport model 199 of solute generation from Maher and Chamberlain (2014) to predict [C]<sub>sil</sub> (also termed the 200 MAC model (Baum et al., 2022; Graham & Pierrehumbert, 2020). Critically, this model also 201 permits us to explicitly treat the reactivity of the land surface—set by erosion and by the type of exposed lithologies on Earth's surface (Caves Rugenstein et al., 2019; Ibarra et al., 2016; 202 203 West, 2012)—and which determines the rate at which silicate weathering occurs. These 204 equations are summarized in the Supporting Information, but are briefly described here. 205 Temperature affects [C]<sub>sil</sub> through its impact on net reaction rates, as parameterized by an Arrhenius equation. Runoff affects [C]<sub>sil</sub> via dilution, though only in weathering systems that 206 207 are kinetically limited; whether [C]<sub>sil</sub> is kinetically-limited depends upon the reactivity of the

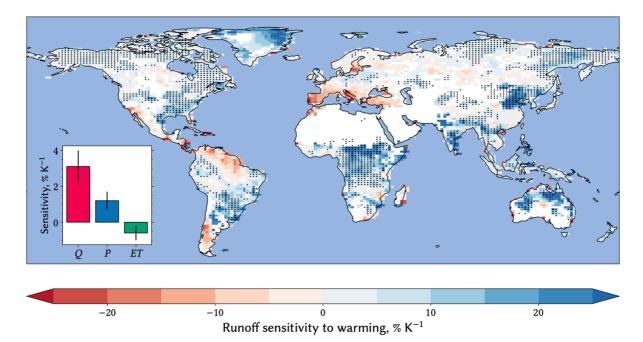
208 weathering material, which is parameterized as a function of the age of the weathering zone 209 (Ts [yr]). Low weathering zone ages, for instance, are indicative of fast supply of fresh 210 minerals, resulting in higher reaction rates, and consequently reduced kinetic limitation. Many 211 catchments today have concentrations that are invariant to changes in runoff (*i.e.*, chemostatic; 212 Godsey et al., 2009), suggesting that kinetic-limitation may not be widespread. Lastly, soil 213  $pCO_2$  modifies the maximum equilibrium  $[C]_{sil}$  ( $[C]_{sil,eq}$ ), and we assume an open weathering 214 zone CO<sub>2</sub> system (Winnick & Maher, 2018), which is a reasonable assumption given that plants 215 and microbes continually supply  $CO_2$  to the weathering zone via root and soil respiration. We 216 assume that soil pCO<sub>2</sub> scales—via a Michaelis-Menton relationship—to changes in 217 atmospheric pCO<sub>2</sub> that increase aboveground gross primary productivity (Volk, 1987). 218 Assuming a linear scaling between soil CO<sub>2</sub> and atmospheric CO<sub>2</sub> does not substantially alter 219 the conclusions presented here. This set of equations permits us to constrain [C]<sub>sil</sub> as a function 220 of several climatic parameters (*i.e.*, temperature (T), CO<sub>2</sub>, and q) and the weathering zone 221 reactivity and, therefore, treat *q*-sensitivity as the main unknown parameter.

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#### 223 2.2 Projected changes in runoff

224 To compare our CH2O-CHOO model results with state-of-the-art Earth system model 225 projections of global q change, we interrogate those models that participated in the C4MIP 226 experiments (Table S2) (Arora et al., 2013, 2020). These model versions account for a more 227 elaborate representation of land processes, such as the response of evapotranspiration to 228 warming and rising CO<sub>2</sub>. We calculate the ensemble mean runoff sensitivity to warming (%/K) 229 in the 1pctCO2 experiments (1% annual increase in atmospheric CO<sub>2</sub>) across CMIP5 and CMIP6 models. For each model, we estimate the relative *q*-sensitivity by linearly regressing 230 231 the change in *q* against the change in temperature. We find that, while the spatial distribution 232 of q change is patchy and there is poor model agreement across much of the land surface, the 233 ensemble average global q-sensitivity is  $3.1 \pm 0.9\%$ , driven by an increase in P and a slight 234 decrease in ET (Figure 2).

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237 *Figure 2*: The *q*-sensitivity (%/K) averaged across the C4MIP models (combined CMIP5 and

CMIP6). Stippling indicates where at least 75% of the models (6 of 8) agree on the sign of the runoff sensitivity. Inset shows the globally-averaged q-, P-, and ET-sensitivity (%/K)

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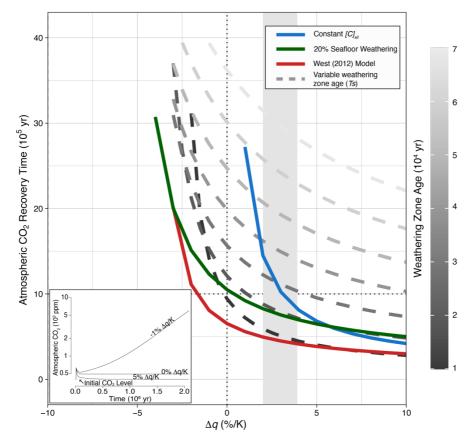
#### 3. Carbon Cycle Constraints on Runoff Changes with Warming

242 To estimate how differences in the sensitivity of runoff with warming (*i.e.*, the q-243 sensitivity) affect the long-term carbon cycle, we again use CH2O-CHOO and inject 1000 Pg 244 of carbon into the ocean-atmosphere over the span of 100 years and permit long-term carbon 245 cycle processes to restore atmospheric CO<sub>2</sub> to pre-perturbation conditions. Here, we treat the 246 *q*-sensitivity as an independent variable; however, unlike in the example above,  $[C]_{sil}$  is now 247 constrained via its relationship to atmospheric CO<sub>2</sub> through T, q, and soil pCO<sub>2</sub> (Maher & 248 Chamberlain, 2014; Volk, 1987; Winnick & Maher, 2018). We test cases where (1) we vary 249 Ts between 10,000 and 100,000 years (the average weathering zone age today is 20,000; Larsen 250 et al., 2014); (2) [C]<sub>sil</sub> is constant, following indications that many catchments on short 251 timescales are chemostatic (Godsey et al., 2009; Ibarra et al., 2016); (3) seafloor weathering-252 which is dependent only upon bottom-water temperatures and not upon q (Brady & Gíslason, 1997; Coogan & Gillis, 2018)—represents 20% of the pre-perturbation weathering flux (Hilton 253 254 & West, 2020), and; (4) silicate weathering rates are predicted using a different model, from 255 West (2012), which also contains terms for q, T, and land surface reactivity. 256 Though the precise recovery time [yr] for CO<sub>2</sub> at a given *q*-sensitivity varies depending

257 upon the value of the weathering parameters, in all cases, the timescale for atmospheric CO<sub>2</sub>

258 to return to pre-perturbation values decreases as the q-sensitivity increases. This is perhaps 259 most easily shown by the case of constant concentrations (blue line, Figure 3). In this case, 260  $[C]_{sil}$  remains constant as CO<sub>2</sub> changes and is unaffected by changes in q, T, or pCO<sub>2</sub>; 261 consequently, recovery from a perturbation is impossible once the q-sensitivity falls below 262 0%/K, given the relationship in Eq. 1. Indeed, even where  $[C]_{sil}$  is constant, recovery timescales 263 surpass 1 Ma for q-sensitivities below ~ 2.5%/K. In all other cases, both q and [C]<sub>sil</sub> are 264 permitted to vary as atmospheric CO<sub>2</sub> changes; yet, even in these other experiments, recovery times surpass 1 Ma at or near a q-sensitivity of 0%/K. This result arises because, though both 265 266 q and  $[C]_{sil}$  are linearly related to  $F_{silw}$  (Eq. 2), changes in  $[C]_{sil}$  are typically smaller than 267 changes in q as atmospheric CO<sub>2</sub> changes (Figure S1), resulting in a greater influence of q on 268 the ultimate recovery from a C-cycle perturbation. While this supposition is based upon our model of weathering, decreases in  $[C]_{sil}$  do not follow perfect dilution trajectories as q 269 270 increases, as seen in datasets that reflect short-term (Godsey et al., 2009) and long-term 271 processes (Godsey et al., 2019; Li et al., 2022), further supporting the idea that variations in 272 *q*—and not in  $[C]_{sil}$ —are the dominant control on  $F_{silw}$ .

273 The critical influence of q can be seen in how different parameterizations affect the 274 recovery time. As weathering zone age increases, the recovery time increases for a given q-275 sensitivity because  $[C]_{sil}$  becomes insensitive to changes in q, due to the relative decrease in 276 weathering zone reactivity. Similarly, if 20% of the initial, global  $F_{silw}$  occurs due to seafloor 277 basalt weathering (green line), recovery times are slightly shorter for a given q-sensitivity, 278 because a portion of [C]<sub>sil</sub>—that generated by seafloor weathering—is insensitive to changes 279 in q and only sensitive to changes in T. Below a q-sensitivity of 0%/K, seafloor weathering 280 begins to substantially reduce recovery times because T rises far more in response to the carbon 281 cycle perturbation due to the reduction in weathering on land, causing seafloor weathering to 282 become a greater proportion of total  $F_{silw}$ . The West (2012) formulation (red line) posits a faster 283 recovery time for most q-sensitivities, largely because  $F_{silw}$  is formulated not as a linear 284 combination of q and  $[C]_{sil}$  but rather as linearly dependent upon erosion rate, which is presumed to be independent of climate. This linear sensitivity to erosion rate yields a strong 285 dependence upon land surface reactivity that helps to decouple  $F_{silw}$  from changes in q. 286 Nevertheless, as in the model of Maher and Chamberlain, if erosion rate decreases (equivalent 287 288 to an increase in the weathering zone age), the West (2012) model predicts longer recovery 289 times for equivalent q-sensitivities (Figure S2).





*Figure 3*: Recovery time [10<sup>5</sup> yr] for atmospheric CO<sub>2</sub> as a function of the *q*-sensitivity (%/K) 291 292 given a 1000 Pg C-cycle perturbation over 100 years. Dashed, gray lines use the standard 293 parameters in Table S1 and variable Ts; blue line assumes globally chemostatic catchments 294 (*i.e.*, [C]<sub>sil</sub> is invariant with changes in CO<sub>2</sub>); green line assumes that seafloor weathering 295 accounts for 20% of the total alkalinity flux to the ocean (with all other parameters derived 296 from Table S1); red line uses the silicate weathering model of West (2012) using values from 297 Table S1. Vertical gray bar is the range of q-sensitivity from the C4MIP models considered 298 herein. Inset shows atmospheric CO<sub>2</sub> (log-scale) of three representative experiments from the 299 case of invariant  $[C]_{sil}$  (blue line). Given a -1% *q*-sensitivity, atmospheric CO<sub>2</sub> increases in 300 perpetuity. With 0% q-sensitivity, atmospheric CO<sub>2</sub> stabilizes but does not recover to pre-301 perturbation conditions. A q-sensitivity of 5% results in recovery to pre-perturbation CO<sub>2</sub> 302 within  $\sim$  700 kyr.

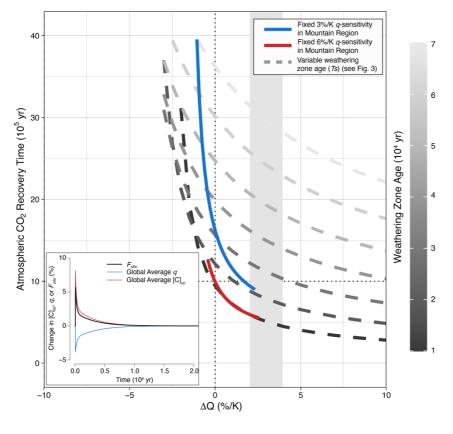
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The above analysis suggests that, for the ocean-atmosphere system to recover to preperturbation  $CO_2$  within 1 Ma, *q*-sensitivity must be greater than -2 %/K for any conceivable combination of weathering parameters; depending upon the precise model and parameterization, this value may be higher. For example, our preferred model of weathering308 using Maher and Chamberlain (2014)—requires a *q*-sensitivity of > 0 %/K for all reasonable 309 estimates of the global weathering zone age.

310 However, the above modeling results assume that the Earth's surface can be 311 characterized by single, mean values; such an assumption is likely incorrect. For example, 312 while the C4MIP ensemble global-mean q-sensitivity is  $3.1 \pm 0.9$  %/K (similar to the CMIP5 313 ensemble mean; Zhang et al., 2014), the spatial variability across large catchments is enormous, 314 spanning -20 to +40 %/K (Dai & Trenberth, 2002; Scheff et al., 2017; Swann et al., 2016) 315 (Figure 2). High, regional q-sensitivity may coincide with particularly reactive rocks (*i.e.*, low 316 weathering zone ages and consequently high [C]<sub>sil</sub>-sensitivity) yielding a different CO<sub>2</sub> 317 recovery time due to spatial co-variability in weathering zone reactivity and runoff change. 318 Indeed, consistently high tropical runoff is the basis for theories that posit that exposure of 319 reactive rocks (such as ophiolites or exposure of Large Igneous Province basalts) near the 320 equator is the primary mode of long-term atmospheric CO<sub>2</sub> control (Donnadieu et al., 2004; 321 Macdonald et al., 2019; Park et al., 2020; Swanson-Hysell & Macdonald, 2017).

322 Consequently, our estimate of the minimum q-sensitivity may be biased high because 323 we do not consider regions with coincident high q-sensitivity and high weathering zone 324 reactivity. To test this possibility, we modify CH2O-CHOO such that two distinct regions 325 contribute toward the global silicate weathering flux: Mountain Region consists of a highly 326 reactive weathering zone (Ts = 1000 yr) and contributes 50% of the initial, pre-perturbation 327 global  $F_{silw}$  and Lowland Region comprises the rest of the world and is characterized by low 328 weathering zone reactivity (Ts = 39,000 yr). Though this example is simplified, the Mountain 329 Region may be considered equivalent to the role of mountains today, which contribute half of 330 the global weathering flux (silicate and carbonate weathering combined) (Larsen et al., 2014). 331 We treat the Mountain Region as having a fixed, positive q-sensitivity and the Lowland Region 332 as having variable q-sensitivity. Globally, the q-sensitivity of mountains more likely tracks 333 changes in P than changes in ET because most precipitation in mountain ranges is partitioned 334 to q. In general, mountain ranges—particularly coastal ones—witness changes in P similar to 335 the zonal mean, such that subtropical ranges see declines in P (and likely q), whereas mid-336 latitude ranges may see changes in P equal to or greater than 7 %/K (Shi & Durran, 2014). We 337 repeat the experiment above, treating the q-sensitivity of the Low-Land Region as an independent parameter and injecting 1000 Pg C into the ocean-atmosphere system over 100 338 339 years. We test two cases: (1) where the Mountain Region has a fixed q-sensitivity of 3%/K and 340 (2) where the Mountain Region has a fixed q-sensitivity of 6%/K.

341 Even the coincidence of high weathering zone reactivity with high runoff sensitivity still requires that average global runoff sensitivity be > 0%/K in order for atmospheric CO<sub>2</sub> to 342 343 recover to pre-perturbation levels within ~ 1 Ma (Figure 4). Though a greater Mountain Region 344 q-sensitivity (6 vs. 3 %/K) results in faster recovery times, the minimum q-sensitivity is still 0 345 %/K to recover from a perturbation within 1 Myr. The different relationship between  $\Delta q$  and recovery time and compressed  $\Delta q$  variability in these simulations is due to how  $\Delta q$  is calculated 346 347 for this experiment; in contrast to the first experiments above, where  $\Delta q$  is prescribed, here 50% of the Earth experiences a fixed 3 or 6 %/K q-sensitivity. Though we vary the q-sensitivity 348 349 for the Lowland region (*i.e.*, with low weathering zone reactivity and high Ts), variations in 350 global  $\Delta q$  are damped, given the fixed q-sensitivity in the Mountain Region (*i.e.* with high weathering zone reactivity and low Ts). Further, any decreases in global q when the q-351 352 sensitivity of the Lowland Region is < 0%/K are small (see inset in Figure 2) and global average 353  $[C]_{sil}$  increases due to the effect of rising temperatures that offset the small decrease in q.



354 355

Figure 4: Recovery time  $[10^5 \text{ y}]$  for atmospheric CO<sub>2</sub> as a function of the average *q*-sensitivity (%/K) given a 1000 Pg C-cycle perturbation over 100 years and two distinct weathering regions (outlined in text). Dashed lines use the standard parameters in Table S1 with variable weathering zone age (*Ts*) (see Figure 3); solid, red line assumes a fixed Mountain Region *q*-

360 sensitivity of 6 %/K; solid, blue line assumes a fixed Mountain Region *q*-sensitivity of 3%/K. 361 In both cases, the q-sensitivity of region 2 is permitted to vary between -10 and 10 %/K. Gray shading is the C4MIP estimate of runoff sensitivity. We calculate  $\Delta q$  as the percentage change 362 in global q at the peak of the C-cycle perturbation relative to starting global q against the change 363 in global T between the peak of the perturbation and the initial condition. Inset shows how  $F_{silw}$ 364 365 (black line), global average q (blue line), and global average  $[C]_{sil}$  (red line) change during a 366 perturbation when the Lowland Region has a -10 %/K q-sensitivity and the Mountain Region 367 has a low Ts (ie, highly reactive) and a fixed q-sensitivity of 6 %/K.

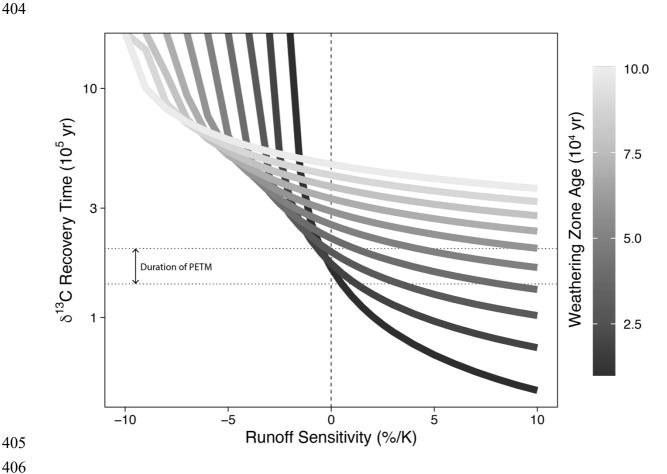
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#### 4. Geologic Constraints on q-sensitivity

We now use a well-studied past large carbon cycle perturbation—the Paleocene-Eocene 370 371 Thermal Maximum (PETM), ~ 55Ma—to constrain the q response to a large carbon cycle perturbation. Though the exact cause of the PETM remains debated, a substantial quantity of 372 373 carbon (5000-20000 Pg C) was released to the ocean-atmosphere system within 3000-10000 374 years, lowering global marine dissolved inorganic carbon (DIC)  $\delta^{13}$ C by 3–4‰ (Cui et al., 2011; Gutjahr et al., 2017; Panchuk et al., 2008; Penman et al., 2016). This negative carbon 375 376 isotope excursion is thought to have lasted ~  $170 \pm 30$  ka (Zeebe & Lourens, 2019), at which point the global marine DIC pool had recovered to pre-perturbation  $\delta^{13}$ C values. Critically, the 377 378 recovery from the PETM was likely mediated by silicate weathering-as evidenced by an excess burial of Si across the PETM (Penman, 2016)-and involved large-scale increases in 379 380 runoff in places (Chen et al., 2018; Foreman et al., 2012). The PETM thus provides an 381 opportunity to test our CH2O-CHOO model and determine the most likely q-sensitivity 382 necessary to match the duration of the  $\delta^{13}$ C negative excursion.

We force CH2O-CHOO with net carbon emissions and net emissions  $\delta^{13}C$  estimates 383 384 from Gutjahr et al. (2017), which also includes estimates of increased organic carbon burial 385 during the recovery phase (Bowen, 2013; Bowen & Zachos, 2010). As before, we prescribe a wide-range of q-sensitivity estimates that determine how runoff responds to changes in 386 387 atmospheric CO<sub>2</sub>. Because we also do not *a priori* know the average weathering zone age (*Ts*) at the PETM, we vary this parameter as well, though evidence from a range of carbon cycle 388 389 models and isotopic data indicates that bulk Earth surface weatherability was lower in the early 390 Eocene, likely reflecting higher Ts (Caves et al., 2016; Froelich & Misra, 2014; Krissansen-391 Totton & Catling, 2017; van der Ploeg et al., 2018; Vervoort et al., 2021).

392 Given these constraints, we find that, to match the duration of the PETM negative  $\delta^{13}C$ 393 excursion requires q-sensitivity to be equal to or greater than 0%/K if Ts is similar to today 394 (Figure 5). If, instead, Ts was higher in the late Paleocene, then q-sensitivity must be higher to 395 match the duration of the  $\delta^{13}$ C negative excursion. Though our model lacks a number of critical 396 carbon cycle processes that are thought to modulate the impact of rapid CO<sub>2</sub> emissions (such 397 as seafloor carbonate sediments that buffer against acidic bottom waters), these effects are 398 likely to matter primarily for the initial stages of the perturbation, rather than mediating the 399 long-term removal of CO<sub>2</sub> by silicate weathering (Penman et al., 2020). We note that, while 400 the PETM provides a useful analogue with which to understand how q responds to CO<sub>2</sub>, we do 401 not expect the *q*-sensitivity at the PETM to exactly match that of today's Earth, due to 402 paleogeographic and floral changes that have occurred during the Cenozoic that would have 403 changed precisely how much precipitation is converted to q as CO<sub>2</sub> rises.



406

Figure 5: Estimated  $\delta^{13}$ C recovery times (10<sup>5</sup> yr) as a function of the *q*-sensitivity and the 407 408 weathering zone age (grayscale). Horizontal dotted lines indicate the estimated  $\delta^{13}$ C recovery time based upon astrochronological tuning of the PETM  $\delta^{13}$ C record (Zeebe & Lourens, 2019). 409 410

411

#### 5. Implications

412 The geological record—and our understanding of how the Earth recovers from carbon 413 cycle perturbations—provides long-term constraints on how q must respond to increases in 414 atmospheric CO<sub>2</sub>. In a broad sense, our results bolster previous work pointing towards the 415 importance of runoff in modulating the long-term carbon cycle and climate (Barron et al., 1989; 416 François & Walker, 1992; Otto-Bliesner, 1995). Critically, our long-term perspective on the 417 runoff response to  $CO_2$  incorporates a number of processes that are thought to affect q but are difficult to constrain in the modern due to temporally limited instrumental records or 418 419 experiments. For example, remotely sensed products of global discharge are affected by 420 internal variability, which precludes their use currently in assessing how total runoff will 421 respond to rising atmospheric CO<sub>2</sub> (Chandanpurkar et al., 2017, 2021). Experiments that seek 422 to understand how plant responses might affect runoff—such as the Free Air CO<sub>2</sub> Enrichment 423 (FACE) experiments—are relatively short-term and the long-term response of ecosystems may 424 differ from the short-term response due to changes in nutrient demands and supply and non-425 linear plant responses to increasing CO<sub>2</sub>. In this sense, the known response of the long-term 426 carbon cycle—and how this response is linked to q—provides a constraint on how global q427 must respond to rising CO<sub>2</sub> that avoids the inherent noisiness of the relatively short 428 instrumental record.

429 Our finding that q-sensitivity must be greater than or equal to 0%/K (when using our 430 preferred model of weathering) holds implications for the processes that control changes in 431 runoff with warming. Though *PET* should increase faster than *P*, stomatal conductance changes 432 driven by CO<sub>2</sub> fertilization and increasing vapor pressure deficits may result in a greater 433 fraction of P being partitioned to q (Novick et al., 2016; Roderick et al., 2015; Swann et al., 434 2016). This effect, however, is complicated by a number of factors that may instead drive 435 greater plant water use, including a longer growing season (Mankin et al., 2019) and 436 widespread greening that increases total leaf area (Zhu et al., 2016). How these factors will 437 interact across the planet to produce a change in global q remains difficult to predict; across 438 much of the Earth's surface, for example, models frequently do not even agree on the sign of 439 the q change (Figure 2). For example, in much of the mid-latitudes, projected ensemble mean 440 runoff changes are negligible, with little model agreement on the sign of the change (Elbaum 441 et al., 2022; Scheff et al., 2017). Our results suggest that the processes on a global scale that 442 tend towards greater q with increasing atmospheric  $CO_2$  and warming—such as reductions in 443 stomatal conductance-will be predominant in a warmer world, though there may be 444 substantial spatial variability. Further, we note that our results likely hold on the millennial or 445 longer timescale, and, as such and given the potential complexity of the biospheric response to 446 warming, there may be long periods when global runoff may not increase in response to 447 warming.

448 Our findings also hold implications for understanding how paleogeographic changes 449 may have modified the strength of the silicate weathering negative feedback in the geologic 450 past. General circulation model simulations highlight that the runoff response to warming is 451 spatially complex; such results indicate the precise global q-sensitivity may vary with the 452 position of the continents (Otto-Bliesner, 1995). For example, models frequently predict that 453 subtropical areas should become drier (Burls & Fedorov, 2017; Mankin et al., 2019), 454 suggesting that a world with predominantly sub-tropical continents would perhaps see a 455 minimal increase or even decrease in global discharge. Our results suggest that either there are 456 additional mechanisms that keep global discharge from decreasing or that such a continental 457 configuration has never occurred on Earth during a major carbon cycle perturbation. If such an 458 Earth experienced a major perturbation, our results indicate that atmospheric CO<sub>2</sub> may have 459 never returned to pre-perturbation values.

460 Though we have attempted to test the robustness of our result by using different 461 parameters and even a different model of weathering (West, 2012), there are a number of 462 hypothesized feedbacks with warming and rising CO<sub>2</sub> that may impact weathering fluxes, but 463 are not-to our knowledge-yet incorporated into any model of weathering. For example, 464 warming is thought to increase the number of extreme precipitation events (Fischer & Knutti, 2016) which is likely to increase q relative to ET as P is rapidly converted to q. Though q 465 466 increases with these extreme events, much of the water may bypass the weathering zone, 467 limiting their impact on weathering. In turn, extreme events may also increase erosional 468 efficiency (Deal et al., 2017, 2018), increasing the availability of fresh minerals and thereby 469 enhancing silicate weathering by reducing average Ts. Though this process is not represented 470 in our model, we suspect that the impact of extreme events on silicate weathering is likely to 471 be minimal, as lower Ts is offset by a greater quantity of q that bypasses the weathering zone 472 via overland flow or exceptionally fast transit times. Additionally, other carbon fluxes may 473 change with a warming climate, such as erosion and burial of terrestrial organic carbon (Hilton, 474 2017) or increases in marine organic burial due to widespread anoxia (Lau et al., 2016). 475 However, these mechanisms also rely upon runoff to transport terrestrial organic matter to 476 continental shelves for burial and to delivery sufficient nutrients to drive marine anoxia (Them 477 et al., 2017), suggesting that increases in runoff are also necessary for these hypothesized 478 negative feedbacks on CO<sub>2</sub>.

479 Our work highlights the tight coupling between the hydrological cycle and the long-480 term carbon cycle and demonstrates that this coupling places a lower-bound on the globally 481 averaged runoff response to warming and rising CO<sub>2</sub>. Whereas abundant research has 482 examined the links between the hydrological cycle and the short-term carbon cycle (Forzieri et 483 al., 2020; Lemordant et al., 2018; Novick et al., 2016), our work demonstrates that the operation 484 of the long-term carbon cycle is also critically linked to the hydrological cycle. It is this linkage 485 that has acted to remove excess CO<sub>2</sub> from the ocean-atmosphere system during perturbations, 486 thus maintaining Earth's habitability. Though our knowledge of weathering continues to 487 improve, ongoing model and observational work indicates the centrality of runoff in linking 488 the weathering of silicate rocks to the ultimate burial of carbon as calcium carbonate in the 489 marine environment. Thus, the importance of runoff for weathering indicates that, on long 490 timescales, runoff must increase as CO<sub>2</sub> rises.

491

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501

#### 502 Software Availability Statement:

503 Model source code used for this research is available in Rugenstein and Winkler (2022) and

504 can be accessed at: https://doi.org/10.5281/zenodo.7219627.

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## **@AGU**PUBLICATIONS

#### AGU Advances

#### Supporting Information for

### Geologic carbon cycle constraints on the terrestrial hydrological response to higher atmospheric CO<sub>2</sub>

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#### Introduction

This supplementary file contains a detailed description of the model equations used in CH2O-CHOO and their derivation. It also contains several figures relevant to understanding the arguments in the main text as well as tables that contain model parameters, both for CH2O-CHOO and for the Earth system models that were investigated for their runoff sensitivity.

#### **Text S1. Model Framework**

We modify a carbon cycle model to incorporate runoff (q) into the equations that predict carbon mass balance. CH2O-CHOO is adapted from and is a simplified version of CLiBeSO-W (Caves Rugenstein et al., 2019), which is itself based upon aspects of the COPSE and GEOCARB models (Berner, 2006; Lenton et al., 2018; Shields and Mills, 2017). The mass balance forward model solves for the time-varying reservoirs of carbon (C) and alkalinity (A) as a function of the major fluxes of C and A to and from the oceanatmosphere, according to equations 1 and 2:

$$\frac{dc}{dt} = F_{carbw} + F_{orgw} + F_{volc} - F_{carbb} - F_{orgb}$$
(1),  
and

$$\frac{dA}{dt} = 2F_{carbw} + 2F_{silw} - 2F_{carbb}$$
(2),

where  $F_{carbw}$ ,  $F_{orgw}$ , and  $F_{silw}$  are the carbonate, organic, and silicate weathering fluxes,  $F_{volc}$  is the volcanic flux, and  $F_{carbb}$  and  $F_{orgb}$  are the carbonate and organic burial fluxes [mol C/yr].

We treat  $F_{silw}$  as a function of q and the concentration of silicate-weathering derived bicarbonate ([C]<sub>sil</sub>):

$$F_{silw} = F_{silw,0} \times R_q \times R_{[c]} \tag{3},$$

where  $F_{silw,0}$  is the initial, pre-perturbation weathering flux,  $R_q$  is the ratio of q at time t to initial  $q(q_0)$ , and  $R_{[c]}$  is the ratio of  $[C]_{sil}$  at time t to the initial  $[C]_{sil}$  ( $[C]_{sil,0}$ ). We calculate q as an exponential function of global temperature (T [K]), which is itself logarithmically related to atmospheric CO<sub>2</sub> (Knutti et al., 2017; Myhre et al., 1998):

$$q = q_0 \left(1 + \lambda_q\right)^{\Delta T} \tag{4},$$

where  $\lambda_q$  describes the sensitivity of q to T [%/K] (*i.e.*, the q-sensitivity), and  $\Delta T$  is calculated via its relationship to climate sensitivity:

$$\Delta T = S_{eq} \log_2(R_{CO2}) \tag{5}$$

where  $S_{eq}$  is the equilibrium climate sensitivity and  $R_{CO2}$  is the ratio of atmospheric CO<sub>2</sub> at time *t* (CO<sub>2,t</sub>) and the pre-perturbation CO<sub>2</sub> (CO<sub>2,0</sub>).

The value of  $[C]_{sil}$  is calculated using modified equations from Maher and Chamberlain (2014) (*i.e.*, the MAC model, following derivations in Graham and Pierrehumbert (2020)). These equations permit us to explicitly incorporate the effect of *T*, *q*, and weathering zone  $pCO_2$ —all sensitive to atmospheric CO<sub>2</sub> and climate—on  $[C]_{sil}$ .

$$[C]_{sil} = [C]_{sil,eq} \left( \frac{\frac{Dw}{q}}{1 + \frac{Dw}{q}} \right)$$
(6),

where  $[C]_{sil,eq}$  is the maximum, equilibrium concentration of silicate-derived bicarbonate and Dw is the Damköhler weathering coefficient, which is a term that encapsulates the reactivity of the weathering zone and time required to reach equilibrium. Following Maher and Chamberlain (2014), we define Dw as:

$$Dw = \frac{L_{\phi}r_{max}\frac{1}{1+Ts \times r_{eff}}}{[C]_{sil,eq}}$$
(7),

where  $L_{\phi}$  is the reactive length scale (held constant for our simulations),  $r_{max}$  is the theoretical maximum reaction rate (held constant for our simulations),  $T_s$  is the age of the weathering zone and is a key variable describing the reactivity of the weathering zone, and  $r_{eff}$  is the effective reaction rate. This effective reaction rate is defined as:

$$r_{eff} = ke^{\left[\left(\frac{Ea}{R_g}\right)\left(\frac{1}{T} - \frac{1}{T_0}\right)\right]}$$
(8),

where  $R_g$  is the universal gas constant [J/K/mol], Ea is the activation energy [J/mol], and the exponential term is an Arrhenius function to describe the effect of T on reaction rates (Brady, 1991; Kump et al., 2000). The coefficient k [y<sup>-1</sup>] encapsulates the effects of mineral surface area, molar mass, and the reference reaction rate (all assumed constant) that modulate the effect of T on  $r_{eff}$ .

Lastly,  $[C]_{sil,eq}$  is modified by the availability of reactant, which here is assumed to be dominantly CO<sub>2</sub>. We calculate this effect as a function of weathering zone  $pCO_2$  assuming open-system CO<sub>2</sub> dynamics, following Winnick and Maher (2018):

$$[C]_{sil,eq} = [C]_{sil,eq,0} (R_{CO2,wz})^{0.316}$$
(9),

where  $[C]_{sil,eq,0}$  is the pre-perturbation, initial value of  $[C]_{sil,eq}$ .  $R_{CO2,wz}$  is the ratio of weathering zone  $pCO_2$  at time t ( $WZ_{CO2}$ ) to the initial weathering zone  $pCO_2$  preperturbation ( $WZ_{CO2,0}$ ). We calculate  $R_{CO2,wz}$  using a formulation proposed by Volk (1987) that links weathering zone  $pCO_2$  with the primary source of that  $CO_2$ , which is aboveground terrestrial gross primary productivity (GPP). Here,  $WZ_{CO2}$  is calculated using an equation that links GPP,  $CO_2$  fertilization on GPP, and weathering zone  $CO_2$ :

$$WZ_{CO2} = \left[ R_{GPP} \left( 1 - \frac{CO_{2,0}}{WZ_{CO2,0}} \right) + \frac{CO_2}{WZ_{CO2,0}} \right] WZ_{CO2,0} + \left( CO_2 - CO_{2,0} \right)$$
(10).

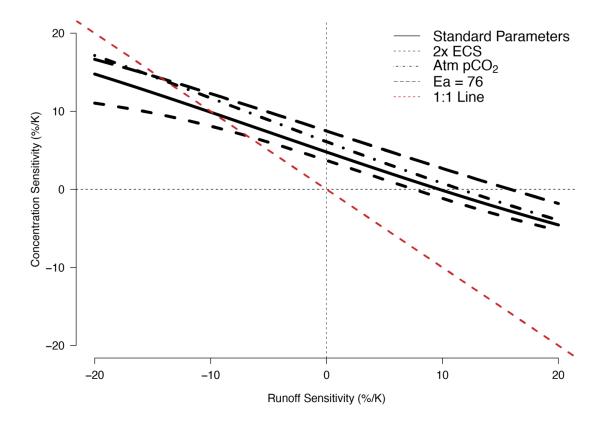
Here,  $R_{GPP}$  is the ratio of GPP at time t to the pre-perturbation GPP ( $GPP_o$ ) and the last term on the right-hand side of the equation ensures that  $WZ_{CO2}$  is always greater than atmospheric CO<sub>2</sub>. GPP is calculated using a Michaelis-Menton formulation, following Volk (1987):

$$GPP = GPP_{max} \left[ \frac{CO_2 - CO_{2,min}}{CO_{2,half} + (CO_2 - CO_{2,min})} \right]$$
(11),

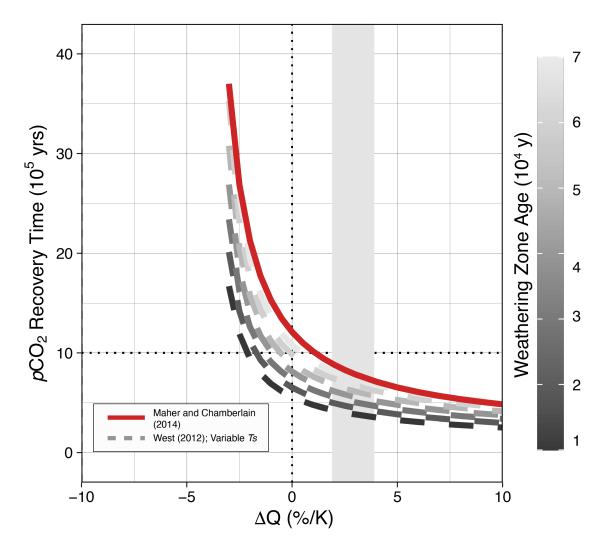
where  $GPP_{max}$  is the maximum possible global terrestrial GPP,  $CO_{2,min}$  is the  $CO_2$  at which photosynthesis is balanced exactly by photorespiration, and  $CO_{2,half}$  is the  $CO_2$  at which GPP is equivalent to 50%  $GPP_{max}$ :

$$CO_{2,half} = \left(\frac{GPP_{max}}{GPP_0} - 1\right) \left(CO_{2,0} - CO_{2,min}\right)$$
(12).

We choose a  $CO_{2,min}$  of 100 ppm based upon evidence for widespread  $CO_2$  starvation at the Last Glacial Maximum (LGM) (Prentice and Harrison, 2009; Scheff et al., 2017), which had an atmospheric  $CO_2$  of ~180 ppm. We also assume that  $GPP_{max}$  is equal to twice  $GPP_0$ , though our results are insensitive to this parameter. Lastly, we assume that  $WZ_{CO2,0}$  is 10x larger than  $pCO_{2,0}$  given evidence that soil  $CO_2$  is typically elevated above atmospheric levels by approximately an order of magnitude (Brook et al., 1983). **Figure S1.** The percentage change in [C]<sub>sil</sub> as atmospheric CO<sub>2</sub> doubles as a function of the *q*-sensitivity, using the Maher and Chamberlain (2014) formulation of weathering. Generally, [C]<sub>sil</sub> increases even if *q* increases, unless *q* increases are so large (8-16 %/K) that increasing *q* causes dilution that overwhelms the effect from rising temperatures and increasing soil *p*CO<sub>2</sub>. However, changes in [C]<sub>sil</sub> are less than 1:1 with changes in *q*, as indicated by the lower slope of the black lines relative to the 1:1 line (red). Solid line uses the standard parameters (Table S1); dashed line assumes an 8° K/CO<sub>2</sub> doubling equilibrium climate sensitivity; dot-dashed line assumes that soil *p*CO<sub>2</sub> scales 1:1 with atmospheric *p*CO<sub>2</sub> (rather than using the Volk (1987) formulation that relates soil *p*CO<sub>2</sub> to atmospheric *p*CO<sub>2</sub>); long dashed line uses a higher activation energy (*Ea* = 76 kJ/mol).



**Figure S2.** Effect of variations of weathering zone age (Ts) on the recovery time using the West (2012) formulation of weathering zone fluxes. Red line uses the standard values (Table S1) and the Maher and Chamberlain (2014) formulation of weathering fluxes. Gray vertical bar indicates the range of runoff sensitivity estimated using the C4MIP model ensemble (see main text).



**Table S1**. Parameters used in the CH2O-CHOO model, organized by model subcomponent. Sources are listed if different from the original model formulation in Maher and Chamberlain (2014) or West (2012).

Parameters	Optimized Values	Units	Source		
	Clima	te parameters			
Earth System Sensitivity	4	K/CO <sub>2</sub> doubling	Knutti et al. (2017)		
Initial Earth Surface T	15	K			
<b>q</b> <sub>0</sub>	0.3	m/yr	Manabe et al. (2004)		
pCO <sub>2</sub>	400	ppm			
рН	8.15	_			
	Carbon cycle mo	odel parameters			
F <sub>silw,0</sub>	6	10 <sup>12</sup> mol C/yr	Moon et al. (2014)		
Fcarbw,0	12	10 <sup>12</sup> mol C/yr	Gaillardet et al. (1999)		
Forgw,0	6	10 <sup>12</sup> mol C/yr	Berner (2006)		
F <sub>carbb,0</sub>	18	10 <sup>12</sup> mol C/yr	Milliman and Droxler (1996		
Forgb,0	6	10 <sup>12</sup> mol C/yr			
Fvolc,0	6	1012 mol C/yr	Wallmann (2001)		
Maher ar	nd Chamberlain (	(2014) model parame	eters		
r <sub>eff</sub>	8.7	10 <sup>-6</sup> mol/m <sup>2</sup> /yr			
т	270	g/mol			
A	0.1	m²/g			
r <sub>max</sub>	1085	µmol/L/yr			
$L_{\phi}$	0.1	m			
Ts	2	10 <sup>4</sup> yr	Larsen et al. (2014)		
Ceq	374	µmol/L			
Ea	38	kJ/mol			
	West (2012	) model parameters			
κ	2.6	10-4			
Kw	7.6	10 <sup>-5</sup>			
Z	8.9	-			
Xm	0.09	-			
σ	0.89	-			

Models	MPI-ESM1- 2-LR	UKESM1-0- LL	CNRM- ESM2-1	IPSL- CM6A-LR	ACCESS- ESM1-5	NorESM1- ME	CESM1- BGC	IPSL- CM5A-LR
Generation	CMIP6	CMIP6	CMIP6	CMIP6	CMIP6	CMIP5	CMIP5	CMIP5
No. of PFTs	13	13	16	15	13	16	16	5
Land Model	JSBACH3.2	JULES-ES- 1.0	ISBA-CTRIP	ORCHIDEE, branch 2.0	CABLE2.4 with CASA- CNP	CLM4	CLM4	ORCHIDEE
Land resolution	1.8° ×1.8°	1.875° × 1.25°	1.4° × 1.4°	2.5° × 1.3°	1.875° × 1.25°	$2.5^\circ \times 1.9^\circ$	$0.9^{\circ}\times1.2^{\circ}$	$2^{\circ} \times 0.5^{\circ} - 2^{\circ}$
Dynamic vegetation	Yes	Yes	No	No	No	No	No	No
Nitrogen cycle	Yes	Yes	No	No	Yes	Yes	Yes	No
Fire	Yes	No	Yes	No	No	Yes	Yes	Yes

Table S2. Overview of CMIP5/6 models included in this study (Arora et al., 2020, 2013)