Pore-scale Bubble Population Dynamics of CO2-Foam at Reservoir Pressure

Benyamine Benali¹, Tore Lyngås Føyen², Zachary Alcorn¹, Malin Haugen¹, Jarand Gauteplass¹, Anthony R Kovscek³, and Martin Anders Fernø¹

¹University of Bergen ²Sintef ³Stanford University

November 24, 2022

Abstract

The flow of CO2 foam for mobility control in porous media is dictated by the foam texture, or bubble density, which is commonly expressed as the number of bubbles per unit of flowing gas. In most high-pressure laboratory studies of foam in porous media, the local foam texture cannot be determined due to opaque flow systems. Here, we unlock real-time foam texture dynamics at high pressure (100 bar) by utilizing a realistic pore network with an extended field of view. We identified snap-off as the dominant foam generation mechanism, with additional fining of foam texture caused by backward foam propagation. Foam coalescence during continuous CO2 injection resulted in large gas channels parallel to the general flow direction that reduced the overall foam apparent viscosity. A large fraction of the CO2 foam remained trapped ($X_{-t} > 0.97$) and stationary in pores to divert CO2 flow and increase sweep efficiency. The gas mobility was calculated from the fraction of trapped bubbles at the pore-scale, and the apparent foam viscosity was in agreement with similar injection test performed at core-scale. Hence, improved understanding of CO2 foam texture evolution (n_f) can strengthen the validation of numerical foam models for physical upscaling of flow phenomena, instrumental in the development of field scale implementation of CO2 foam for in carbon utilization and storage applications.

1 Pore-scale Bubble Population Dynamics of CO₂-Foam at Reservoir Pressure

Benyamine Benali¹, Tore L. Føyen^{1,2,*}, Zachary Paul Alcorn¹, Malin Haugen¹, Jarand Gauteplass¹, Anthony R. Kovscek³ and Martin A. Fernø¹

- ⁴ ¹ Department of Physics and Technology, University of Bergen, Norway
- ⁵ ² SINTEF Industry, Norway
- 6 ³ Department of Energy Resources Engineering, Stanford University, USA
- 7 Benyamine Benali (<u>benyamine.benali@uib.no</u>)

8 Key Points:

- Pore-scale CO_2 foam bubble flow, texture and trapping dynamics at high pressure.
- A comprehensive laboratory investigation of essential foam features and phenomena
- 11 during dynamic flow supported by the population balance foam model framework.

12 Abstract

The flow of CO_2 foam for mobility control in porous media is dictated by the foam texture, or 13 bubble density, which is commonly expressed as the number of bubbles per unit of flowing gas. 14 In most high-pressure laboratory studies of foam in porous media, the local foam texture cannot 15 be determined due to opaque flow systems. Here, we unlock real-time foam texture dynamics at 16 high pressure (100 bar) by utilizing a realistic pore network with an extended field of view. We 17 identified snap-off as the dominant foam generation mechanism, with additional fining of foam 18 texture caused by backward foam propagation. Foam coalescence during continuous CO₂ 19 injection resulted in large gas channels parallel to the general flow direction that reduced the 20 overall foam apparent viscosity. A large fraction of the CO₂ foam remained trapped ($X_t > 0.97$) 21 22 and stationary in pores to divert CO_2 flow and increase sweep efficiency. The gas mobility was calculated from the fraction of trapped bubbles at the pore-scale, and the apparent foam viscosity 23 was in agreement with similar injection test performed at core-scale. Hence, improved 24 understanding of CO_2 foam texture evolution (n_f) can strengthen the validation of numerical 25 foam models for physical upscaling of flow phenomena, instrumental in the development of field 26 scale implementation of CO₂ foam for in carbon utilization and storage applications. 27

28 **1 Introduction**

Foam is a promising method for reducing CO_2 mobility in enhanced oil recovery (EOR) 29 and CO₂ storage processes. CO₂ foam for mobility control can improve reservoir sweep 30 efficiency, mitigate gravity override, and reduce viscous fingering for increased oil recovery 31 (Enick et al. 2012) and CO₂ storage potential (Føyen et al. 2020). Foam is a thermodynamically 32 unstable two-phase system of dispersed gas bubbles separated by continuous aqueous films 33 (lamellae), stabilized by a foaming agent such as surfactant and/or nanoparticles. The geometry 34 of the foam bubbles are dictated by the porous media they are contained within. Lamellae span 35 entire pores, orthogonally to the general flow direction, reducing the mobility of gaseous phase. 36 The aqueous phase remains continuous and is therefore unaffected by foam (Kovscek & Radke 37 38 1994). Despite successful field-scale tests (Alcorn et al. 2019; Blaker et al. 1999; Chou et al. 1992), others have been unsuccessful (Stephenson et al. 1993) due injectivity issues and 39 difficulty attributing additional displacement specifically to CO₂ foam, which is related to a 40 limited understanding of foam generation, flow dynamics, and coalescence at reservoir 41 conditions. Therefore, this study aims to unlock real-time foam dynamics at high pressure (100 42 bar) by utilizing a realistic pore network with an extended field of view. 43

Population balance foam models provide a mechanistic framework where the kinetics of 44 foam generation and coalescence is mathematically expressed to track the number of foam 45 46 bubbles (Kovscek et al. 1995). The number of bubbles in the system is used to calculate the mobility of the gas, however, the calculation is not straightforward. Crucial input in the 47 calculation must be estimated, which is challenging in opaque cylindrical core plugs and sand 48 packs. Such input includes quantifying the trapped gas fraction (X_t) , which requires dual gas-49 phase tracer measurements (Tang & Kovscek 2006). In addition, the texture of flowing foam 50 bubbles (n_t) , relies on measurements of produced foam which may not represent foam in porous 51 media (Ettinger & Radke 1992; Hou et al. 2013). 52

Foam is generated in porous media by three main mechanisms, including *Roof snap-off*,
 lamella mobilization/division and lamella leave-behind (Ransohoff & Radke 1988; Rossen 2003;

Chen et al. 2005; Kovscek et al. 2007). The former two are considered prevailing mechanisms 55 56 with lamellae orthogonal to flow direction and effective flow impediment (Kam & Rossen 2003). The third mechanism results in lamellae oriented parallel to flow direction, and an ineffective 57 continuous gas foam (Friedmann et al. 1991; Kovscek & Radke 1994). Several studies have 58 suggested that a minimum threshold pressure gradients (or minimum flow velocities) need to be 59 exceeded for foam generation to be initiated (Ransohoff & Radke 1988; Rossen & Gauglitz 60 1990; Rossen et al. 1994; Gauglitz et al. 2002). The existence of minimum threshold pressure 61 gradients can have unfavorable consequences for foam generation and propagation far from 62 injection wells, where the pressure gradients are low. However, foam can be generated by Roof 63 snap off independent of pressure gradients at sharp increases in permeability (Rossen 1999; Shah 64 et al. 2019). During foam flow the pressure gradients can vary and fluctuate in space and time, 65 resulting in local pressure gradients for the creation of lamellae (Kam & Rossen 2003). Positive 66 feedback processes can also occur during foam generation because foam increases the pressure 67 gradients and subsequently causes the generation of more foam (Føyen et al. 2020). 68

Foam coalescence reduces the number of bubbles in porous media by three mechanisms: 69 coarsening by diffusion (Ostwald ripening), capillary suction drainage (Kovscek & Radke 1994) 70 and gravitational liquid drainage. Coarsening by diffusion occurs due to the transport of gas from 71 smaller bubbles (small radius, high curvature), with a higher internal pressure, to larger bubbles 72 with lower internal pressure, causing the smaller bubbles disappear (Saint-Jalmes 2006; 73 Marchalot et al. 2008). Capillary suction drainage occurs when the water saturation approaches a 74 saturation value where the lamellae are no longer stable, because the capillary pressure exceeds 75 the maximum disjoining pressure of the foam film and drains the lamellae (Jiménez & Radke 76 1989; Falls et al. 1989; Farajzadeh et al. 2015). 77

Gas mobility reduction by foam is caused by increased effective gas viscosity (μ_g^f) and decreased effective gas relative permeability (k_{rg}^f). The combined effect can be assessed by adapting Darcy's law for foam flow (Kovscek & Radke 1994).

$$u_g = \frac{k k_{rg}^{\dagger} \nabla p_g}{\mu_g^{\dagger}} \tag{1}$$

81 where u_g is the superficial gas flow velocity, k is the absolute permeability, k_{rg}^f is the effective 82 gas-foam relative permeability, μ_g^f is the effective viscosity of flowing foam and ∇p_g is the gas 83 pressure gradient.

Most quantification of foam strength relies on the measured pressure gradient. However, the pressure gradient depends on flow velocity (*u*) and absolute permeability (*k*), which lack generality, often making it unsuitable for comparison purposes. Foam apparent viscosity (μ_{app}) accounts for differences in absolute permeability and flow velocity and is commonly reported:

$$\mu_{app} = \frac{k}{u} * \nabla p_g \tag{2}$$

For foam flowing through the same pores where individual bubbles are constantly coalesced and regenerated (Ettinger & Radke 1992), the effective gas viscosity increases. The increased effective gas viscosity is caused by the viscous shear when lamellae move along pore walls (Hirasaki & Lawson 1985) and through pore throats (Falls et al. 1989). The effective viscosity of flowing foam (μ_g^f) is proportional to the bubble density (n_f) (Kovscek & Radke 1994; Fried 1961), and is typically shear-thinning with respect to interstitial velocity (v_f) , and can be described by (Kovscek et al. 1995):

$$\mu_g^f = \mu_g + \frac{\alpha * n_f}{v_f^c} \tag{3}$$

where μ_g is the gas viscosity, α is a system dependent scaling constant and *c* has been estimated to be approximately 1/3 (Hirasaki & Lawson 1985).

In addition to increasing the effective gas viscosity, trapped foam reduces the gas-foam 97 relative permeability (k_{rg}^{f}) . The flowing bubble trains will occupy the largest pores (Radke & Gillis 1990) and a (Stone 1970) model for three-phase relative permeability is applicable to 98 99 estimate the effective gas-foam relative permeability. By considering the flowing (S_{ta}) and 100 trapped (S_{tg}) gas as two pseudo saturations, we have a three-phase system, with water as the third 101 phase which is most wetting and the flowing gas-foam is the least wetting phase. The relative 102 103 permeability of the least (and most) wetting phase depends only on its own saturation and is the same as its two-phase relative permeability. Therefore, the effective gas-foam relative 104 permeability (k_{rq}^{f}) equals the no-foam gas relative permeability at the flowing gas saturation S_{fg} 105 (Kovscek & Radke 1994). 106

107 The trapped gas fraction, X_t , is commonly used to compute the trapped gas saturation (S_{tg} 108) from gas saturation (S_q):

$$X_t = \frac{S_{tg}}{S_g} \tag{4}$$

109 Continuous-gas foam occurs when lamella generation becomes insufficient to maintain 110 bubble densities, resulting in gas flow mainly through one or several interconnected channels not 111 impeded by lamellae (Falls et al. 1989). The mobility reduction will be less compared to 112 discontinuous foam because $n_f=0$. However, the trapped bubbles will still reduce the mobility 113 of the moving gas, as the gas-foam relative permeability (k_{ra}^f) is reduced.

Direct visual observation using microscopy during flow of fluids through porous media 114 can be obtained using two-dimensional flow cells termed micromodels. Micromodels allow 115 observation and characterization of individual fluid bubbles or ganglion and are particularly 116 117 useful for foam studies for a range of important foam phenomena including generation (Géraud et al. 2017; Kovscek et al. 2007), coarsening (Jones et al. 2018a; Marchalot et al. 2008), texture 118 (Rognmo et al. 2019; Rangel-German & Kovscek 2006), gas trapping (Lv et al. 2018; Jones et al. 119 120 2018b), flow (Géraud et al. 2016) and flow diversion (Khoshkalam et al. 2019; Gauteplass et al. 2015). Most microfluidic studies investigated decoupled foam phenomena in a limited field of 121 view (FoV), whereas a complete assessment of foam in porous media requires combined 122 observations of foam generation, coarsening, and texture. 123

This work presents a comprehensive laboratory investigation of essential foam features and phenomena during dynamic flow at reservoir conditions over a representative area, supported by the framework developed for population balance foam models. In addition, we investigate the foam generation and stabilization when adding nanoparticles at two differentconcentrations (150 and 1500 ppm) to the foaming solution.

129 2 Materials and Methods

130 <u>2.1 Fluid preparation</u>

Brine (3.5 wt.% NaCl) was used for all aqueous phases (Table 1) and was filtered 131 through a 0.45 µm cellulose acetate filter to remove particles before mixing with other 132 components or injected it through the micromodel. Two foaming agents were used: Surfonic 133 L24-22A (a non-ionic surfactant with linear ethoxylated alcohol, Huntsman), and Levasil CC301 134 (a surface-modified spherical silica nanoparticle, Nouryon). Three foaming solutions were 135 prepared by mixing brine with surfactant, nanoparticle or a hybrid combination of both (Table 136 1). CO₂ of 99.999% purity was used during the micromodel foam injections. The pore space was 137 cleaned between each injection cycle using 2-propanol-water azeotrope (IPA). 138

Table T Composition of aqueous solutions.					
Aqueous Solution	Concentration, Component				
Brine	35000 ppm, NaCl in distilled water				
SF5000	5000 ppm , <i>Surfonic L24-22</i>				
SF5000 + NP1500	5000 ppm , Surfonic L24-22, + 1500 ppm , Levasil CC30				
SF5000 + NP150	5000 ppm, <i>Surfonic L24-22,</i> + 150 ppm, <i>Levasil CC301</i>				

877000 ppm, 2-propanol in distilled water

139 Table 1 Composition of aqueous solutions.

140 <u>2.2 Micromodel and holder</u>

IPA

The micromodel consists of an etched silicon wafer with a realistic porous structure bonded to an 141 optically transparent borosilicate glass. The chemical characteristics of crystalline silicon (silicon 142 wafer) and borosilicate glass are similar to sandstone (mainly quartz), which are chemically inert 143 to injected fluids. Deep reactive ion etching resulted in vertical pore walls and sharp edges. 144 resembling grain shapes found in real reservoir rocks generated from Berea sandstone thin 145 sections. However, when 3D porous media in a real reservoir rock is simplified to 2D, some 146 modifications were made to the pattern to connect pores which were isolated, resulting in higher 147 permeability and porosity relative to Berea sandstone. Complete production procedures can be 148 found in (Buchgraber et al. 2012). The rectangular porous pattern (27 mm X 21.40 mm) was 149 equipped with flow ports in each corner, with inlet and outlet fluid distribution channels (200 µm 150 width) that connected ports 1 and 2, and ports 3 and 4 (Figure 1). The porous pattern (27000 151 grains) constitutes 36 (4 x 9) repetitions of pore network with 749 unique grains with shapes 152 from a thin-section image of natural sandstone; a rock type extensively used in laboratory studies 153 of oil production, foam and CO₂ storage. The porous pattern can be considered as a simplified 154 two-dimensional projection of real pore structures, with connected pores that allow flow and 155 discontinuous, irregularly shaped grains that provide flow tortuosity. 156







Micromodel properties are listed in Table 2. The porosity was quantified as the ratio of areal 163 pore space to the total area (grains + pore space) and was calculated to be 0.61 from a stitched, 164 165 high-resolution images of the whole porous pattern. The pore volume (PV) with constant etching depth of 30 μm was 11.1 μL . The grain size distribution of the unique pattern with 749 grains 166 ranged between 100 to 79 000 μm^2 and the pore throat (defined as the shortest pore space 167 distance between two adjacent grains) distribution ranged between 10 to 200 µm (Figure 1). The 168 constant etching depth resulted in square- or rectangular-shaped pore throat cross-sections. The 169 absolute permeability of the micromodel was 2.97 D. 170

Parameter	Value
Width	27 mm
Length	21.40 mm
Depth	30 µm
Porosity	0.61
Permeability	2.97 D
Pattern repetition	36
Unique grains	749
Total grains	27000
Grain size	$100-79000 \mu m^2$
Pore throat length	10-200μm

171	Table	2	Micromodel	properties.
-----	-------	---	------------	-------------

The micromodel was positioned in a depression in the bottom part of the two-part aluminum 172 micromodel holder with flow ports aligned with threaded connections sealed with O-rings. The 173 top part, with an open window for direct observation, was attached to the bottom with eight 174 175 screws using 0.20 Nm force. The system can be pressurized to 150 bar without external confinement pressure. 176

2.3 Experimental setup and procedure 177

178 The micromodel holder was positioned on a motorized scanning stage below the microscope (Axio Zoom. V16, Zeiss) equipped with a diffuse ring-illuminator. The microscopic zoom, 179 focus, illuminator intensity, imaging, and the motorized stage was controlled with the Zeiss 180 proprietary software. Microscope settings (light intensity, aperture and shutter time) were 181 optimized for image processing and kept constant for all experiments. The software performed 182 shading corrections and focus adjustments for any misalignments. Fluid injection and production 183 184 utilized a series of valves and pumps (Figure 2) and the pore pressure was kept at a minimum of 100 bar using a backpressure system. The injected and produced fluids were considered 185 incompressible at the system temperature (ambient), and pressure gradients during flow (tens of 186 millibars) were not expected to change fluid properties and flow rates. Four unsteady state CO_2 187 injections were performed (three with foaming solutions listed in Table 1, and one using brine 188 for baseline) using the following procedure: 189

- i. Pre-saturate the micromodel with foaming solution 190
- ii. Inject dense phase CO₂ at a constant volumetric flow rate (4 $\mu L/min$) into port1 with port 191 2 closed and ports 3 and 4 open and kept at 100 bars using the backpressure system. 192



193

Fig 194 . 2 Experimental setup used for the pore-scale foam studies using etched-silicon wafer micromodels. The micromodel holder was 195 positioned underneath the microscope (not shown). Two high precision plunger pumps (Quizix Q5000-2.5K and Quizix Q5000-196 10K) were used for fluid injection. The Q5000-2.5K pump was used in single cylinder injection mode, with CO₂ and cleaning 197 solvents (IPA) in each of the two cylinders, whereas the Q5000-10K was solely used for injection of surfactant. The N $_2$ 198 accumulator tank was used to regulate the backpressure: a 1 L pressure vessel was filled with N₂ and received the produced fluids 199 (water and dense liquid CO₂) during injections. The produced fluids were vented from the pressure vessel during the preparation 200 between the micromodel foam experiments.

2.4 Image processing and analysis 201

The motorized scanning stage and software enabled imaging of the entire porous pattern (27 mm 202 X 21.40 mm) with high spatial resolution (4.38 μ m/pixel) by stitching multiple overlapping 203 images (Figure 1). Image acquisition time of the porous pattern (121 separate images) was 1 min 204 13 sec and time-lapsed series captured the position of each individual foam bubble every 1 min 205 206 15 sec. The following three-step image segmentation process was performed:

- i. 207 Thresholding the images using a low (dark) threshold value by utilizing the significantly darker colored grain structure compared to the pore space. The grains were then drawn 208 black on a white empty image. 209
- Thresholding the images using a high (bright) threshold value by utilizing the white ii. 210 colored (due to the diffuse ring-illuminator) gaseous-aqueous interfaces and grain walls. 211 The bright areas were then painted black on the white empty image. 212
- 213 iii. The rest of the white empty image is the discrete bubbles. This resulted in a binary image with grains, grain walls and gaseous-aqueous interfaces as black, and discrete bubbles as 214 white. 215

The final segmented image (Figure 3) shows the discrete bubbles colored red and grains colored 216

blue for illustrative purposes. The bubbles were further described using functions available in the 217 Python library OpenCV (Bradski 2000) as continuous contours with the following features: 218

center of mass (coordinates), area, perimeter length, orientation, bounding boxes, and minimum 219

enclosing ellipse with minor and major axis. These characteristics were used in the analysis of 220

foam flow and bubble generation/coalescence. 221



222 223

Fig. 3 Illustration of the segmentation process. Left: Raw image (2.2 mm x 1.3 mm) of the micromodel with foam occupying the porous pattern. Right: Segmented image, blue areas are the grain structure, red areas are discrete foam 224 225 bubbles 226

227 **3** Results and discussion

Pore-scale foam flow phenomena were studied using temporal bubble density mapping, bubble 228 229 number (N_i) and bubble size during foam generation and coalescence. Temporal bubble numbers during foam generation and coalescence (N_{bubble}) were normalized to the baseline (N_{baseline}) for 230 three foaming solutions (Figure 4). Foam generation increased N to approximately 90 to 100 231 times of the baseline, reaching peak values after approximately 5 PV of CO₂ injected. Bubble 232 coarsening (decreased N) was the dominating coalescence mechanism. The foam generation and 233 stabilization were not affected by adding nanoparticles to the foaming solution as the temporal 234

bubble numbers for the three injections had same trend. Detailed discussion of results from foam generation (3.1), foam flow (3.2), and foam decay (3.3) is below.



Fig. 4 Development in normalized bubble number (with respected to the baseline). The bubbles are sorted into logarithmical bubble size groups: a) all bubbles; b) <u>small</u> bubbles with area $i \, 10^3 \, \mu m^2$; c) <u>intermediate</u> bubbles with area between $10^3 \, i \, 10^4 \, \mu m^2$; d) <u>large</u> bubbles with area $i \, 10^4 \, \mu m^2$. Sorting the number of bubbles based on size shows that the number of intermediate size bubbles are increasing at the expense of small bubbles after 5 PV of CO₂ injection. The baseline and the foam injections are smoothed with a running average of 51 and 11, respectively.

243 <u>3.1 Foam generation</u>

Foam generation occurred primarily by snap-off in regions with sufficient liquid saturation and 244 sharp constrictions between pore throats and bodies. Subsequent flow diversion increased CO₂ 245 flow velocity locally, resulting in additional foam generation. Bubble density maps initially show 246 low foam generation with a rapid increase when the injected CO₂ reached the outlet flow channel 247 between 0.4 and 1.3 PV CO₂ injected (Figure 5). This resulted in backward (from outlet to inlet) 248 foam propagation; previously reported and attributed to foam becoming stronger (Simjoo & 249 Zitha 2020; Almajid et al. 2019; Apaydin & Kovscek 2001). The determining mechanisms for 250 foam strengthening are not clear, and it has hypothesized to occur by foam transitioning from 251 weak to strong state, or by favorable conditions for snap-off in the transition zone. We cannot 252 rule out or support either hypotheses. Backward propagation of the foam front does, however, 253 indicate that snap-off was the prevailing generation mechanism during the initial period because 254 lamella mobilization and division is a secondary generation mechanism that requires the 255 256 presence of foam and can only result in forward-propagation. The secondary generation mechanism may have contributed to refinement of the foam texture and regeneration to increase 257 bubble densities after the initial period. The observed foam behavior is likely important for long-258 distance foam generation where pressure gradients are likely to be small (Hirasaki et al. 1997; 259 Szafranski et al. 1998). Foam generation at sharp permeability contrasts can also occur some 260 distance from the injection well at lower pressure gradients if sufficient foaming solutions are 261 262 available. Additionally, super-linear increases in apparent viscosity during core scale unsteady

- state experiments, indicate that positive feedback behavior (Føyen et al. 2020) is likely related to
- the backward propagation mechanism observed at the pore-level.



Fig. 5 Bubble density mapping during foam generation (up to 5 PV of CO₂ injection) for three foam solutions listed

in Table 1. Foam generation initiated between 0.4 and 1.3 PV of CO₂ injected at the sharp permeability contrast
 (outlet flow channel, bottom side of density maps), resulting in higher foam density (brighter color) in the outlet
 regions of porous pattern. Foam propagated backwards (outlet to inlet) resulting in higher foam density through the
 micromodel. The resolution of the spatially resolved hexagonal binning plot is 150 X 90.

272 <u>3.2 Flowing and trapped foam</u>

Flowing regions were characterized by moving bubble trains and continuous gas channels. 273 Movement of bubble trains occurred with temporary flow suspension, restart and randomly 274 relocated (Figure 6). Open gas channels extended several pore lengths and emerged from 275 inadequate upstream bubble regeneration to replace displaced bubbles downstream. Open 276 channels were irregularly filled with bubbles and reemerged in the same or new pathways. 277 Trapped bubbles adjacent to open or partially filled channels changed from low energy 278 configuration (no curvature, image a) to texture with less capillary resistance (with curvature, 279 images c and f). The observation of large regions of trapped bubbles support that lamellae 280 moving to a minimum energy configuration (at rest) will have a greater capillary resistance to 281 remobilization (Jones et al. 2018a; Hou et al. 2013). Hence, the lamella in stationary regions will 282 not move when exposed to the same pressure gradient as the lamella in the flowing region. 283



a) OPEN channel



b) PARTIALLY open channel



c) FILLED channel



d) REEMERGED BRANCHED channel

- **Fig. 6** Dynamics of open gas channels during bubble generation: red indicates path of an open channel (without bubbles) and orange indicates where the open channel fills with bubbles. Sequential images show: a) path of an open channel (t = 33.8 PV); b) channel becomes partially open (t = 34.2 PV); c) channel becomes filled with bubbles (t = 34.7 PV); d) a new path emerges that branches from the original path, now partially filled with bubbles (t = 35.1 PV). The images are from the SF5000 injection at the same field of view (2.2 mm x 2.2 mm).
- The density of the flowing bubbles (n_f) is used to calculate the effective gas viscosity in population balance foam models and is commonly assumed to be equal to the density of trapped bubbles (n_t) (Kovscek and Radke 1994). With a limited field of view (2 mm x 2 mm, **Figure 6**),
- assessment of local n_f becomes ambiguous:
- i. In open channels (images a and d), the CO₂ flow is not impeded by bubbles and $n_f = 0$, *i.e.* $n_f < n_t$
- ii. In filled channels (image c, n_f is higher than n_t in surrounding region, *i.e.* $n_f > n_t$
- iii. Partially filled channels (image b), n_f is the mean value of the filled and the open part of the channel, and depends on the relative size between the two parts and the bubble density

Hence, a foam representative elementary volume (foam-REV) must be defined that provides a n_f 300 representative for the whole system (Hill 1963). Dynamics of bubble density, gas channels, 301 flowing bubbles and trapped bubbles were therefore mapped in the entire pore space using 302 hexagonal binning plot (Figure 7). Open gas channels were defined as large bubbles exceeding a 303 size of 25% of micromodel length. Flowing bubbles were identified by temporal changes in 304 bubble position and size, individually determined for all bubbles: bubble position changed if the 305 306 relocation exceeded 20% of its minor axis (defined by an enclosing eclipse) between time-lapse images, and bubble size changed if its area increased/decreased more than 20%. Temporal 307 changes in bubble position and size and the position of open gas channels were based on 308 observations between 5 image frames, corresponding to 1.7 PV CO₂ injected. Trapped bubbles 309 (without change in position and size) were found by combining the bubble density and flowing 310 bubble maps: trapped = density / (flow +1). It is not known if the size of the whole porous 311 312 pattern (27 mm x 21.40 mm) is sufficient to define foam-REV. Still, we observed that the length of the open channels remained constant for extended periods (Figure 8). Therefore, when the 313 relationship between flowing and stationary regions is constant, we can estimate a mean 314 representative flowing bubble density. 315

316

The gas mobility in the presence of foam can be found with knowledge of trapped bubbles 317 318 (reduced gas relative permeability) and moving bubble trains (increased effective gas viscosity). The latter can be ignored when open gas channels spanned the porous pattern (Equation 3 $(n_f=0)$) 319 $\rightarrow \mu_g^f = \mu_g = 0.081 cP$). Hence, gas mobility can be determined directly from the fraction of trapped gas (X_t) , that is estimated from the open gas channel area (**Figure 8**) or flowing region 320 321 area (Figure 7). Note that the flowing regions were determined over 1.7 PV and the fraction of 322 trapped bubbles is underestimated due to the irregular relocation of bubble trains and continuous 323 gas channels (Figure 7). The gas mobility can be determined when the majority of flow occurs 324 in open gas channels because they can be used to estimate X_t by subtracting the volume (area) 325 constituting the open gas channels (Figure 8): with open gas channels occupying between 0.75 326 327 to 1.75% of the pore space, X_t ranged between 0.98 and 0.99 (assuming $S_a = 0.8$). As mentioned earlier, the effective gas-foam relative permeability equals the no-foam gas relative permeability 328

- at the flowing gas saturation, and by assuming a linear relationship between relative permeability and saturation, the gas-foam relative permeability (k_{ra}^{f}) can be calculated to range between 0.008
- and saturation, the gas-roam relative permeability (κ_{rg}) can be calculated to range between 0.008 and 0.016. Foam apparent viscosity has been calculated using equation 1 and 2 to range between
- 332 5.1 to 10.1 cP that corroborates core floods at similar conditions (Føven et al. 2020).
- 333

The effective gas viscosity from bubble trains cannot be ignored when open gas channels do not 334 span the porous pattern and must be taken into account to assess the gas mobility reduction for 335 such cases using Equation 1. Direct quantification is not obtainable from the images because the 336 parameter α that depends on the foaming system and germination sites (Pancharoen et al. 2012) 337 338 cannot be estimated from the unsteady state injections reported here. However, adequate measured differential pressure can potentially be used to estimate α , which is a parameter used in 339 population balance modeling (Kovscek et al. 1995; Eide et al. 2020). Direct quantification of α 340 by micromodel studies can, therefore, be valuable for foam agent screening, as it is less time 341 consuming than core floods, and can be used as input for population balance modeling. 342



Fig. 7 Pore-scale foam dynamics with spatial bubble density, open gas channels, flowing bubbles and trapped
bubbles using hexagonal binning plots (150 x 90) Left column: bubble density (number of bubbles in each hexagon)
with mapped open gas channels (lines). Center column: flowing bubbles. Right column: Trapped bubbles. Location
of open gas channels, flowing and trapped bubbles were calculated each 1.7 PV CO₂ injected. The micromodel was
pre-saturated with the foaming solution (SF5000) prior to CO₂ injection (top to bottom)





3.3 Foam decay 354

Foam coalescence became dominant at later times (after approximately 5 PV of CO₂ injected, 355 Figure 4) with overall decreasing bubble numbers. We rank coalescence mechanisms in our 356 porous media from the three known bubble coalescence mechanisms: *i*. coarsening by diffusion 357 (Ostwald ripening), *ii.* liquid film drainage by capillary or gravity forces, and *iii.* gravitational 358 liquid drainage. From the observed flow pattern changes due to foam regeneration (discussed in 359 360 detail below) we conclude that capillary liquid film drainage is not prevailing because it requires low liquid saturations not consistent with observations. Gravitational liquid drainage is 361 considered negligible in the horizontally oriented system with 30 μm etching depth. Therefore, 362 we consider Ostwald ripening the prevailing coalescence mechanism, where intermediate-sized 363 bubbles increased at the expense of smaller bubbles (Figure 4). Smaller bubbles disappeared due 364 to diffusive gas transport from bubbles with higher Laplace pressure (small bubble radii) to 365 366 bubbles with lower Laplace pressure (high bubble radii). Fick's law determines diffusion rate, and it is expected that Ostwald ripening will depend on time. The development in bubble number 367 and mean bubble area (A) agrees with Von Neumann's law (Saint-Jalmes 2006; Marchalot et al. 368 2008) that describes coarsening in trapped two-dimensional foam (Figure 9). The development 369 in mean bubble area and the ratio $N_o/N(t)$ scales linearly with time (PV injected), where N(t) is 370 371 the number of bubbles at a time t and N_0 is the initial number of bubbles before foam coalescence becomes dominant. N_{o} is the greatest observed bubble number and marked the transition between 372 when the rate of foam coalescence exceeded rate of foam generation. The transition can be 373 recognized when $N_o/N(t)=1$ (after approximately 5 PV CO₂ injected), and the relationship 374 $N_o/N(t)$ increases as the foam texture becomes coarser. We focus our foam coarsening analysis 375 between 5 and 30 PV injected as foam was regenerated between 35 and 45 PV injected, likely 376 caused by mobilization of foaming solution from system dead ends. 377





.9 Increasing $N_o/N(t)$ with decreasing number of bubbles (*left axis, solid line*) and increasing mean bubble area **i** A > i (*right axis, filled area*) during foam coalescence by Ostwald ripening for three foaming solutions. Both the mean bubble area and the ratio $N_o/N(t)$ scaled linearly with time (PV injected) and agrees with Von Neumann's law for two-dimensional foams. The associated regression coefficient (slope) are included on top of the curves

In trapped, two-dimensional foams, coarsening is expected to stop when foam lamellae are 383 located at the pore-throats with zero curvature, *i.e.* equal bubble pressures without pressure 384 gradients to drive further diffusion and coarsening (Jones et al. 2018a). Hence, A and $N_o/N(t)$ 385 scale with time only during a limited initial coarsening period (Jones et al. (2018a) reported 1500 386 sec), before the lamellae come to rest at the pore throats. The presence of large gas channels and 387 slow foam regeneration sustained foam coarsening from approximately 5 PV CO₂ injected. The 388 presence of large gas channels spanning several pores observed in our system (Figure 9) were 389 effectively large bubbles with lower pressure than surrounding bubbles that promoted sustained 390 Ostwald ripening. The prolonged coarsening caused the mean bubble areas for the three 391 injections to increase in an average sense as 44 μm^2 per PV injected over minimum 25 PV (4125 392 sec). The rate of increase was 0.26 $\mu m^2 s^{-1}$, compared with 1.14 $\mu m^2 s^{-1}$ for two-dimensional 393 trapped foam during a limited initial coarsening period (Jones et al. 2018a). The difference 394 between stationary and flowing systems is expected as regeneration of foam does not occur in a 395 stationary system. Additionally, different gases and pressures were used, and the solubility of the 396 gaseous phase in aqueous phase influence diffusion and subsequent coarsening. 397

398 4 Conclusions

Bubble flow, texture and trapping dynamics for dense CO_2 foam were studied at high pressure at 399 the pore scale using a realistic pore network. The aggregated effect of foam generation, 400 401 coarsening, and texture was contextualized through the population balance model framework. Roof snap-off at sharp permeability contrasts was the dominant foam generation mechanism with 402 additional refining of foam texture caused by backward foam propagation. Coarsening during 403 404 continuous CO₂ injection occurred due to Ostwald ripening. Foam generation and stabilization were not affected by adding nanoparticles to the foaming solution. Flowing regions were 405 characterized by moving bubble trains and open gas channels. Stationary regions with trapped 406 407 foam had a greater capillary resistance than moving bubble trains. The gas mobility in the presence of open gas channels and trapped bubbles was estimated by quantifying the fraction of 408 trapped bubbles, corresponding to an apparent foam viscosity between 5.1 to 10.1 cP. 409 Assessment of foam texture $(n_f \text{ and } n_f)$ needs to be performed over a significantly large area of 410 the porous pattern to define a foam representative elementary volume. 411

412 **5 Data availability**

The foam injection datasets, including segmented images, used during the current sturdy are available at http://dx.doi.org/10.17632/5d37nbzf9s.1, an open-source online data repository hosted by Mendeley Data (Benali & Føyen 2020). The raw foam injection micromodel images are available from the corresponding author on reasonable request.

417 6 Abbreviation and Nomenclature

Carbon dioxide	CO_2
field of view	FoV
PV	Pore volume
foam representative elementary volume	foam-REV
Permeability	k
Effective gas-foam relative permeability	k^f_{rg}
Gas viscosity	μ_g
Effective viscosity of flowing foam	μ^f_g
Superficial gas flow velocity	u _g
Foam apparent viscosity	μ_{app}
Gas pressure gradient	∇p_g
Gas saturation	S _g
Flowing gas saturation	S _{fg}
Trapped gas saturation	S_{tg}
Flowing bubble density	n _f
Trapped bubble density	n _t
Total number of bubbles	N _{bubble}
Baseline number of bubbles	N _{baseline}
Initial number of bubbles before foam coalescence	N _o
mean bubble area	A

trapped gas fraction	X _t
System dependent scaling constant	α

418 7 Acknowledgements

- 419 The authors wish to acknowledge the Research Council of Norway for financial support (project
- no. 249742, 268216, 294886 and 301201) for funding of PhD candidate Benyamine Benali and
- 421 Dr. Zachary Paul Alcorn, and support from industry partners Nouryon, Shell E&P, Total E&P,
- 422 and Equinor.

423 **8 References**

- Alcorn, Z. P., Fredriksen, S. B., Sharma, M., Rognmo, A. U., Føyen, T. L., Fernø, M. A., &
 Graue, A. (2019). An Integrated Carbon-Dioxide-Foam Enhanced-Oil-Recovery Pilot
 Program With Combined Carbon Capture, Utilization, and Storage in an Onshore Texas
 Heterogeneous Carbonate Field. SPE Reservoir Evaluation & Engineering, 22(04), 14491466. doi:10.2118/190204-pa
- Almajid, M.M., Nazari, N., Kovscek, A.R.: Modeling Steady-State Foam Flow: Hysteresis and
 Backward Front Movement. Energy & Fuels (2019).
 https://doi.org/10.1021/acs.energyfuels.9b01842
- 432 Apaydin, O.G., Kovscek, A.R.: Surfactant Concentration and End Effects on Foam Flow in
- 433 Porous Media. Transport in Porous Media (2001). https://doi.org/10.1023/A:1010740811277
- 434 Benali, B., Føyen, T.: [Dataset] Micromodel foam.... In. Mendeley Data, (2020)
- Blaker, T., Celius, H.K., Lie, T., Martinsen, H.A., Rasmussen, L., Vassenden, F.: Foam for Gas
 Mobility Control in the Snorre Field: The FAWAG Project. Paper presented at the SPE
 Annual Technical Conference and Exhibition, Houston, Texas, 1999/1/1/
- 438 Bradski, G.: The opency library. Dr Dobb's J. Software Tools 25, 5 (2000)
- Buchgraber, M., Al-Dossary, M., Ross, C.M., Kovscek, A.R.: Creation of a dual-porosity
 micromodel for pore-level visualization of multiphase flow. Journal of Petroleum Science
 and Engineering (2012). https://doi.org/10.1016/j.petrol.2012.03.012
- Chen, M., Yortsos, Y.C., Rossen, W.R.: Insights on foam generation in porous media from pore network studies. Colloids and Surfaces A: Physicochemical and Engineering Aspects
 (2005). https://doi.org/10.1016/j.colsurfa.2005.01.020
- Chou, S., Vasicek, S., Pisio, D., Jasek, D., Goodgame, J.: CO2 Foam Field Trial at North Ward Estes. Paper presented at the SPE Annual technical conference and exhibition,
- Eide, Ø., Fernø, M., Bryant, S., Kovscek, A., Gauteplass, J.: Population-balance modeling of
 CO2 foam for CCUS using nanoparticles. Journal of Natural Gas Science and
 Engineering (2020). https://doi.org/10.1016/j.jngse.2020.103378
- Enick, R.M., Olsen, D.K., Ammer, J.R., Schuller, W.: Mobility and Conformance Control for
 CO2 EOR via Thickeners, Foams, and Gels -- A Literature Review of 40 Years of
 Research and Pilot Tests. Paper presented at the SPE Improved Oil Recovery
 Symposium, Tulsa, Oklahoma, USA, 2012/1/1/
- Ettinger, R.A., Radke, C.J.: Influence of Texture on Steady Foam Flow in Berea Sandstone. SPE
 Reservoir Engineering (1992). https://doi.org/10.2118/19688-PA
- Falls, A.H., Musters, J.J., Ratulowski, J.: The Apparent Viscosity of Foams in Homogeneous
 Bead Packs. SPE Reservoir Engineering (1989). https://doi.org/10.2118/16048-PA

- Farajzadeh, R., Lotfollahi, M., Eftekhari, A.A., Rossen, W.R., Hirasaki, G.J.H.: Effect of
 Permeability on Implicit-Texture Foam Model Parameters and the Limiting Capillary
 Pressure. Energy & Fuels (2015). https://doi.org/10.1021/acs.energyfuels.5b00248
- 461 Fried, A.N.: Foam-Drive Process for Increasing the Recovery of Oil. In. University of North
 462 Texas Libraries, Washington D.C., (1961)
- Friedmann, F., Chen, W.H., Gauglitz, P.A.: Experimental and Simulation Study of High Temperature Foam Displacement in Porous Media. SPE Reservoir Engineering (1991).
 https://doi.org/10.2118/17357-PA
- Føyen, T., Brattekås, B., Fernø, M.A., Barrabino, A., Holt, T.: Increased CO2 storage capacity
 using CO2-foam. International Journal of Greenhouse Gas Control (2020).
 https://doi.org/10.1016/j.ijggc.2020.103016
- Gauglitz, P.A., Friedmann, F., Kam, S.I., Rossen, W.R.: Foam generation in homogeneous
 porous media. Chemical Engineering Science (2002). https://doi.org/10.1016/S0009 2509(02)00340-8
- Gauteplass, J., Chaudhary, K., Kovscek, A.R., Fernø, M.A.: Pore-level foam generation and flow
 for mobility control in fractured systems. Colloids and Surfaces A: Physicochemical and
 Engineering Aspects (2015). https://doi.org/10.1016/j.colsurfa.2014.12.043
- Géraud, B., Jones, S.A., Cantat, I., Dollet, B., Méheust, Y.: The flow of a foam in a twodimensional porous medium. Water Resources Research (2016).
 https://doi.org/10.1002/2015wr017936
- 478 Géraud, B., Méheust, Y., Cantat, I., Dollet, B.: Lamella Division in a Foam Flowing through a
 479 Two-Dimensional Porous Medium: A Model Fragmentation Process. Physical Review
 480 Letters (2017). https://doi.org/10.1103/PhysRevLett.118.098003
- Hill, R.: Elastic properties of reinforced solids: Some theoretical principles. Journal of the
 Mechanics and Physics of Solids (1963). https://doi.org/10.1016/0022-5096(63)90036-X
- Hirasaki, G.J., Lawson, J.B.: Mechanisms of Foam Flow in Porous Media: Apparent Viscosity in
 Smooth Capillaries. Society of Petroleum Engineers Journal (1985).
 https://doi.org/10.2118/12129-PA
- Hirasaki, G.J., Miller, C.A., Szafranski, R., Tanzil, D., Lawson, J.B., Meinardus, H., Jin, M.,
 Londergan, J.T., Jackson, R.E., Pope, G.A., Wade, W.H.: Field Demonstration of the
 Surfactant/Foam Process for Aquifer Remediation. Paper presented at the SPE Annual
 Technical Conference and Exhibition, San Antonio, Texas, 1997/1/1/
- Hou, J., Du, Q., Li, Z., Pan, G., Lu, X., Zhou, K.: Experiments on foam texture under high
 pressure in porous media. Flow Measurement and Instrumentation (2013). <u>https://doi.org/</u>
 10.1016/j.flowmeasinst.2013.05.002
- Jiménez, A.I., Radke, C.J.: Dynamic Stability of Foam Lamellae Flowing Through a Periodically
 Constricted Pore. In: Oil-Field Chemistry, vol. 396. ACS Symposium Series, vol. 396,
 pp. 460-479. American Chemical Society, (1989)
- Jones, S.A., Getrouw, N., Vincent-Bonnieu, S.: Foam flow in a model porous medium: I. The effect of foam coarsening. Soft Matter (2018a). https://doi.org/10.1039/C7SM01903C
- Jones, S.A., Getrouw, N., Vincent-Bonnieu, S.: Foam flow in a model porous medium: II. The
 effect of trapped gas. Soft Matter (2018b). https://doi.org/10.1039/C7SM02458D
- Kam, S.I., Rossen, W.R.: A Model for Foam Generation in Homogeneous Media. SPE-195310 PA (2003). https://doi.org/10.2118/87334-PA

- Khoshkalam, Y., Khosravi, M., Rostami, B.: Visual investigation of viscous cross-flow during
 foam injection in a matrix-fracture system. Physics of Fluids (2019).
 https://doi.org/10.1063/1.5079482
- Kovscek, A.R., Patzek, T.W., Radke, C.J.: A mechanistic population balance model for transient
 and steady-state foam flow in Boise sandstone. Chemical Engineering Science (1995).
 https://doi.org/10.1016/0009-2509(95)00199-F
- Kovscek, A.R., Radke, C.J.: Fundamentals of Foam Transport in Porous Media. In: Foams:
 Fundamentals and Applications in the Petroleum Industry, vol. 242. Advances in
 Chemistry, vol. 242, pp. 115-163. American Chemical Society, (1994)
- Kovscek, A.R., Tang, G.Q., Radke, C.J.: Verification of Roof snap off as a foam-generation
 mechanism in porous media at steady state. Colloids and Surfaces A: Physicochemical
 and Engineering Aspects (2007). https://doi.org/10.1016/j.colsurfa.2007.02.035
- Lv, M., Liu, Z., Ji, C., Jia, L., Jiang, Y.: Investigation of Pore-Scale Behaviors of Foam Flow in a
 Polydimethylsiloxane Micromodel. Industrial & Engineering Chemistry Research (2018).
 https://doi.org/10.1021/acs.iecr.8b03366
- Marchalot, J., Lambert, J., Cantat, I., Tabeling, P., Jullien, M.C.: 2D foam coarsening in a
 microfluidic system. EPL (Europhysics Letters) (2008). https://doi.org/10.1209/0295 5075/83/64006
- Moffitt, P., Pecore, D., Trees, M., Salts, G.: East Vacuum Grayburg San Andres Unit, 30 Years
 of CO2 Flooding: Accomplishments, Challenges and Opportunities. Paper presented at
 the SPE Annual Technical Conference and Exhibition,
- Pancharoen, M., Fernø, M., Kovscek, A.: Modeling foam displacement in fractures. Journal of
 Petroleum Science and Engineering (2012). https://doi.org/10.1016/j.petrol.2012.11.018
- Radke, C.J., Gillis, J.V.: A Dual Gas Tracer Technique for Determining Trapped Gas Saturation
 During Steady Foam Flow in Porous Media. Paper presented at the SPE Annual
 Technical Conference and Exhibition, New Orleans, Louisiana, 1990/1/1/
- Rangel-German, E.R., Kovscek, A.R.: A micromodel investigation of two-phase matrix-fracture
 transfer mechanisms. Water Resources Research (2006).
 https://doi.org/10.1029/2004wr003918
- Ransohoff, T.C., Radke, C.J.: Mechanisms of Foam Generation in Glass-Bead Packs. SPE
 Reservoir Engineering (1988). https://doi.org/10.2118/15441-PA
- Rognmo, A.U., Fredriksen, S.B., Alcorn, Z.P., Sharma, M., Føyen, T., Eide, Ø., Graue, A.,
 Fernø, M.: Pore-to-Core EOR Upscaling for CO2 Foam for CCUS. SPE-195310-PA
 (2019). https://doi.org/10.2118/190869-PA
- Rossen, W.R.: Foam Generation at Layer Boundaries in Porous Media. SPE-195310-PA (1999).
 https://doi.org/10.2118/59395-PA
- Rossen, W.R.: A critical review of Roof snap-off as a mechanism of steady-state foam
 generation in homogeneous porous media. Colloids and Surfaces A: Physicochemical and
 Engineering Aspects (2003). https://doi.org/10.1016/S0927-7757(03)00309-1
- Rossen, W.R., Gauglitz, P.A.: Percolation theory of creation and mobilization of foams in porous
 media. AIChE Journal (1990). https://doi.org/10.1002/aic.690360807
- Rossen, W.R., Shi, J., Zeilinger, S.C.: Percolation modeling of foam generation in Porous media.
 AIChE Journal (1994). https://doi.org/10.1002/aic.690400618
- Saint-Jalmes, A.: Physical chemistry in foam drainage and coarsening. Soft Matter (2006).
 https://doi.org/10.1039/B606780H

- Shah, S.Y., Wolf, K.-H., Pilus, R.M., Rossen, W.R.: Foam Generation by Capillary Snap-Off in
 Flow Across a Sharp Permeability Transition. SPE-195310-PA (2019).
 https://doi.org/10.2118/190210-PA
- Simjoo, M., Zitha, P.L.J.: Modeling and Experimental Validation of Rheological Transition
 During Foam Flow in Porous Media. Transport in Porous Media (2020).
 https://doi.org/10.1007/s11242-019-01251-9
- Stephenson, D. J., Graham, A. G., & Luhning, R. W. (1993). Mobility Control Experience in the
 Joffre Viking Miscible CO2 Flood. SPE Reservoir Engineering, 8(03), 183-188.
 doi:10.2118/23598-pa
- Stone, H.L.: Probability Model for Estimating Three-Phase Relative Permeability. SPE-2116-PA
 (1970). <u>https://doi.org/10.2118/2116-PA</u>
- Szafranski, R., Lawson, J.B., Hirasaki, G.J., Miller, C.A., Akiya, N., King, S., Jackson, R.E.,
 Meinardus, H., Londergan, J.: Surfactant/foam process for improved efficiency of aquifer
 remediation. In, Darmstadt 1998. Structure, Dynamics and Properties of Disperse
 Colloidal Systems, pp. 162-167. Steinkopff
- Tang, G.-Q., Kovscek, A.: Trapped gas fraction during steady-state foam flow. Transport in porous media (2006). https://doi.org/10.1007/s11242-005-6093-4