

Characterizing the N₂O isotopomer behavior of two N-disturbed soils using natural abundance and isotopic labelling techniques

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Abstract

Nitrous oxide (N₂O), a potent greenhouse gas that contributes significantly to climate change, is emitted mostly from soils by a suite of microbial metabolic pathways that are nontrivial to identify, and subsequently, to manage. Using either natural abundance or enriched stable isotope methods has aided in identifying microbial sources of N₂O, but each approach has limitations. Here, we conducted a novel pairing of natural abundance and enriched assays on two dissimilar soils, hypothesizing this pairing would better constrain microbial sources of N₂O. We incubated paired natural abundance and enriched soils from a corn agroecosystem and a subalpine forest in the laboratory at 10-95% soil saturation for 28 hr. The natural abundance method measured intramolecular site preference (SP) from emitted N₂O, whereas the enriched method measured emitted ¹⁵N₂O from soils amended with ¹⁵N-labelled substrate. The isotopic composition of emitted N₂O was measured using a laser-based N₂O isotopic analyzer, yielding three key findings. First, isotopic signatures from natural abundance and enriched N₂O generally agreed in interpretation. Second, our novel pairing of isotopic methodologies refined understanding of microbial N-transformations in drier agricultural soil. In the 50% saturation agricultural soil, nitrification might have been deemed an important process based on SP alone, but enrichment helped reveal that its contribution to N₂O emissions was minor. Finally, we quantified, to our knowledge for the first time, persistent (>50%) β-position-specific enrichment in emitted ¹⁵N₂O, which is far in excess of SP-level fractionation expectations. This counter-intuitive enrichment pattern raises the possibility of previously unrecognized N-transformations in these soils.

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1 **Characterizing the N₂O isotopomer behavior of two N-disturbed soils using natural**
2 **abundance and isotopic labelling techniques**

3

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12 **Key Points:**

- 13 • Pairing natural abundance and enriched stable isotope methods improves confidence in
14 disentangling microbial sources of N₂O
- 15
- 16 • N₂O isotopic signatures from natural abundance or ¹⁵N-enriched soils generally agreed in interpretation of
17 microbial source process
- 18
- 19 • Consistent position-specific enrichment patterns from emitted enriched N₂O may reveal
20 previously unrecognized soil N-transformations

21

22

23

24 **Abstract**

25 Nitrous oxide (N₂O), a potent greenhouse gas that contributes significantly to climate change, is emitted
26 mostly from soils by a suite of microbial metabolic pathways that are nontrivial to identify, and
27 subsequently, to manage. Using either natural abundance or enriched stable isotope methods has aided in
28 identifying microbial sources of N₂O, but each approach has limitations. Here, we conducted a novel
29 pairing of natural abundance and enriched assays on two dissimilar soils, hypothesizing this pairing would
30 better constrain microbial sources of N₂O. We incubated paired natural abundance and enriched soils from a
31 corn agroecosystem and a subalpine forest in the laboratory at 10-95% soil saturation for 28 hr. The natural
32 abundance method measured intramolecular site preference (SP) from emitted N₂O, whereas the enriched
33 method measured emitted ¹⁵N₂O from soils amended with ¹⁵N-labelled substrate. The isotopic composition
34 of emitted N₂O was measured using a laser-based N₂O isotopic analyzer, yielding three key findings. First,
35 isotopic signatures from natural abundance and enriched N₂O generally agreed in interpretation. Second,
36 our novel pairing of isotopic methodologies refined understanding of microbial N-transformations in drier
37 agricultural soil. In the 50% saturation agricultural soil, nitrification might have been deemed an important
38 process based on SP alone, but enrichment helped reveal that its contribution to N₂O emissions was minor.
39 Finally, we quantified, to our knowledge for the first time, persistent (>50%) β-position-specific enrichment
40 in emitted ¹⁵N₂O, which is far in excess of SP-level fractionation expectations. This counter-intuitive
41 enrichment pattern raises the possibility of previously unrecognized N-transformations in these soils.

42 **Plain Language Summary**

43 Microbes in soils respire the greenhouse gas nitrous oxide (N_2O), which contributes to global
44 warming. Respiration is a chemical reaction comparable to breathing. When environmental
45 conditions like air temperature or soil moisture change, microbes respire N_2O differently, but
46 often in ways we can neither easily anticipate nor identify. Fortunately, microbes all respire some
47 naturally “heavy” forms of N_2O molecules, called heavy isotopes. Microbes emit different
48 amounts of heavy isotopes when they respire different ways. To combat global warming, we
49 aimed to better understand microbial N_2O emissions by bringing together two methods to
50 measure heavy N_2O isotopes: natural abundance and labelling. Natural abundance measures the
51 heavy isotopes microbes naturally emit, and labelling is when scientists feed microbes
52 isotopically heavy food to trace how they respire the heavy isotopes. We put soils in sealed jars
53 for 28 hours, and then measured the emitted heavy N_2O from each method with a specialized
54 beam of laser-light. We better identified how microbes emitted N_2O when we used natural
55 abundance *and* labelling at the same time. And so, two methods are better than one. Moving
56 forward, we can use both methods to more precisely identify how microbes emit N_2O and better
57 control global warming.

58 1. Introduction

59 Nitrous oxide (N₂O) is a far more potent greenhouse gas (GHG) than other biogenic
60 GHGs (Ravishinkara et al. 2009, Ciais et al. 2013). On a per molecule basis, N₂O has a warming
61 potential 298x greater than carbon dioxide (CO₂) and 34x greater than methane (CH₄) (Alvarez
62 et al. 2012, Rector et al. 2018). This is problematic because the atmospheric N₂O concentration
63 has risen an unprecedented 20% since the Industrial Revolution began (Ciais et al. 2013).
64 Microbial metabolism of synthetic and manure-based nitrogen (N) fertilizers in agricultural soils
65 are largely responsible for this sharp rise in atmospheric N₂O (Davidson 2009, Park et al. 2012,
66 Smith 2017). In fact, approximately 70% of N₂O is emitted from soils, but primarily from soils
67 with an abundant inorganic N supply, hereafter classified as “N-disturbed” soils. In N-disturbed
68 soils, N supply exceeds soil carbon (C) availability, which typically manifests in heightened
69 microbial N metabolism and excess N₂O emissions (Davidson 2009).

70 Multiple microbial metabolic pathways can generate N₂O, and so it can be difficult to
71 identify the process(es) responsible for emissions (Snider 2011, Zhang et al. 2016, Ibraim et al.
72 2018, Wong et al. 2020). Microbes can emit N₂O via nitrification, dissimilatory nitrate reduction
73 to ammonium (DNRA), denitrification (bacterial and fungal), nitrifier-denitrification, co-
74 denitrification, and anaerobic ammonia oxidation (anammox; Butterbach-Bahl et al. 2013).
75 However, identifying the microbial source process(es) responsible for emissions is particularly
76 challenging because each process is sensitive to a variety of spatially and temporally variable
77 factors such as soil physical and chemical properties, climate, moisture, availability of N
78 substrate, and microbial community composition and activity (Jenny 1980, Wrage et al. 2004,
79 Park et al. 2011, Toyoda et al. 2015, van Groenigen et al. 2015, Congreves et al. 2019, Denk et
80 al. 2019). To further complicate matters, processes can co-occur, sometimes even being
81 performed by the same soil microbe (Wen et al. 2016, Sanford et al. 2012). Taken together, this

82 consort of variables makes it difficult to comprehensively identify the microbial processes
83 responsible for N₂O emissions over space and time. To manage rising N₂O levels, we must be
84 able to better identify which microbial pathways are responsible for emissions, especially in N-
85 disturbed soils.

86 Stable isotopes have proven central to understanding the biochemical source processes of
87 N₂O (Baggs 2008, Baggs 2011, Snider 2011, Snider et al. 2015, Hu et al. 2015, Yu et al. 2020).
88 Both natural abundance and enriched methods have been used to quantify these sources (Perez et
89 al. 2006, Ostrom and Ostrom 2012, Scriber et al. 2012, Ostrom and Ostrom 2017, Yamamoto et
90 al. 2017). Isotopic enrichment has proved a reliable and robust method for partitioning among
91 the better-studied N₂O-generating processes, nitrification and denitrification (Wrage et al. 2004,
92 Mathieu et al. 2006, Wagner-Riddle et al. 2008, Russow et al. 2009). By amending a given soil
93 with isotopically labelled ¹⁵NH₄⁺ or ¹⁵NO₃⁻, researchers can reveal when nitrification or
94 denitrification dominates by tracing the emitted enriched N₂O back to the enriched substrate.
95 However, ¹⁵N additions are somewhat limiting in that they only partition between the two broad
96 classes of processes.

97 In contrast, a number of studies have now characterized the natural abundance
98 intramolecular distribution of ¹⁵N in the N₂O molecule to delineate among multiple N₂O-
99 producing processes (Sutka et al. 2006, Chen et al. 2016). Like many biochemical
100 transformations, stable isotope fractionation occurs during N₂O production (Menyailo and
101 Hungate 2006, Vieten et al. 2007, Lewicka-Szczebak et al. 2015, Snider et al. 2015). This causes
102 differential accumulation of heavy N in either the central, α-position N, or the terminal, β-
103 position N, in the linear N₂O molecule (^βN=^αN=O, Yoshida and Toyoda 2000). The
104 intramolecular distribution, or difference in δ¹⁵N between δ¹⁵N^α and δ¹⁵N^β isotopomers, is termed

105 site preference (SP, Yoshida and Toyoda 2000). Since the early 2000s, a number of pure culture,
106 lab, and field studies have shown that many N₂O-generating processes reliably yield consistent
107 SP values (Toyoda et al. 2005, Sutka et al. 2006, Well et al. 2006, Perez et al. 2006, Baggs 2008,
108 Park et al., 2011, Snider 2011, Maeda et al., 2015). To date, researchers have used SP to
109 disentangle nitrification via ammonium oxidizing Archaea (AOA) or ammonium oxidizing
110 bacteria (AOB), fungal vs. bacterial denitrification, and nitrifier denitrification (Wrage et al.
111 2004, Sutka et al. 2006, Wu et al. 2016, Wrage-Mönnig et al. 2018, Rohe et al. 2020). However,
112 there are disagreements in the literature about the robustness of this method for multiple reasons.
113 First, interlaboratory calibration for N₂O isotopic standards remains an ongoing issue (Mohn et
114 al. 2014, Ostrom and Ostrom 2017, Ostrom et al. 2018, Harris et al. 2020). Second, SP values
115 have been reported to overlap among very different source processes (e.g., nitrification and
116 fungal denitrification, Decock and Six 2013, Xia et al. 2013, Hu et al. 2015, Wenk et al. 2016,
117 Yamamoto et al. 2017, Yu et al. 2020). And third, N₂O → N₂ reduction during denitrification
118 enriches δ¹⁵N^α and can thus confound SP values (Koster et al. 2013, Mohn et al. 2014, Ostrom
119 and Ostrom 2017, Lewicka-Szczebak et al. 2020, Stuchiner et al. 2020). To overcome the
120 individual limitations of enriched and natural abundance studies, we hypothesize that pairing
121 both methods on the same soils could enable better partitioning among N₂O-generating source
122 processes.

123 Here, we used natural abundance and enriched methods on the same soils to characterize
124 the agreement between the two methods. To our knowledge, this is the first study to deploy the
125 two methods jointly for comparison. We performed natural abundance and enriched laboratory
126 incubations on soils from two N-disturbed sites: a corn agroecosystem with varying levels of N-
127 fertilization and irrigation, and a subalpine forest impacted by ~30 years of atmospheric N

128 pollution. We also manipulated soil moisture in each incubation to promote diverse N₂O-
 129 generating microbial metabolic pathways. By incubating different N-disturbed soils under both
 130 natural abundance and isotopically enriched conditions, we aimed to reveal gaps in our
 131 understanding about these isotopic approaches, and better understand the microbial N-
 132 transformations occurring in different soils. If effective, these methods in conjunction could be
 133 another strategy for constraining the global N₂O budget.

134 2. Materials and Methods

135 2.1 Field sampling and soil characterization

136 2.1.1 Site descriptions

137 We collected soils from two contrasting, N-disturbed ecosystems in Colorado: a corn
 138 field and a subalpine forest. Soil properties and treatments for each site are summarized Table 1.

139 **Table 1.** Properties and treatment descriptions (as applicable) for all sites. Percent soil organic C
 140 (SOC) and soil organic N (SON), and microbial biomass C and N were measured in June 2018.
 141 Growing season irrigation or N application was averaged across all the same agricultural
 142 treatments plots, and the subalpine values are for the entire watershed surrounding that subalpine
 143 environment. The n-value for soil treatment corresponds to the number of plots that samples
 144 were collected from. The n-values for all other measurements correspond to the number of
 145 technical replicates from each bulked soil sample. Error bars represent \pm one SE from the mean.

Site	Soil treatment	Abbreviated soil name	Irrigation (mm/growing season)	Total N application rate (kg/ha/yr)	% SOC	% SON	Microbial Biomass C ($\mu\text{g C/g dry soil}$)	Microbial Biomass N ($\mu\text{g N/g dry soil}$)
Agriculture	High N High Water (n = 4)	HNHW	497	266	1.21 (\pm 0.162) (n = 4)	0.077 (\pm 0.003) (n = 4)	9.00 (\pm 0.600) (n = 4)	0.918 (\pm 0.340) (n = 4)
Agriculture	High N Low Water (n = 3)	HNLW	441	270	1.36 (\pm 0.036) (n = 4)	0.098 (\pm 0.017) (n = 4)	5.92 (\pm 1.08) (n = 4)	0.301 (\pm 0.081) (n = 4)
Agriculture	Low N High Water (n = 4)	LNHW	497	172	1.28 (\pm 0.060) (n = 4)	0.093 (\pm 0.002) (n = 4)	6.59 (\pm 1.75) (n = 4)	0.554 (\pm 0.390) (n = 4)
Agriculture	Low N Low Water (n = 3)	LNLW	441	154	1.47 (\pm 0.090) (n = 4)	0.102 (\pm 0.004) (n = 4)	11.5 (\pm 2.51) (n = 4)	0.934 (\pm 0.320) (n = 4)
Subalpine	None (n = 4)	Subalpine	N/A	3-3.5	7.82 (\pm 1.20) (n = 4)	0.355 (\pm 0.086) (n = 4)	18.2 (\pm 3.16) (n = 3)	1.37 (\pm 0.490) (n = 3)

146

147 *Agricultural site*

148 We collected soil from the Limited Irrigation Research Farm (LIRF) located north of
149 Greeley, Colorado (described in detail in Zhang and Yemoto 2018). In 2016, experimental
150 treatments were applied across plots that measured approximately 6 x 20 m. These treatments
151 manipulated soil irrigation and fertilizer N to assess the impact on corn crop yield (detailed in
152 Table 1). The experimental design used 14 fully randomized treatment blocks with high or low
153 irrigation rates, and high or low additions of urea and NO_3^- fertilizer. Thus, our treatments
154 included: high N high water (HNHW), high N low water (HNLW), low N high water (LNHW),
155 and low N low water (LNLW).

156 The high and low irrigation treatments provided crops with 100 and 65% of
157 evapotranspiration (ET) met during the late vegetative and maturation growth stages. The high
158 and low N additions were 250 and 130 kg/ha and were applied as a combination of liquid urea
159 and NO_3^- in the irrigation water, which resulted in some discrepancies in the amount of fertilizer
160 applied to plots, depending on HW or LW appointment. Liquid urea was applied in ~22 kg/ha
161 drip fertigation aliquots throughout the vegetative growth stages (N. Flynn, *pers comm*).

162
163 *Subalpine site*

164 The Loch Vale Watershed (LVWS) is located in Rocky Mountain National Park (RMNP)
165 on the eastern edge of the Front Range in Colorado, USA, between 3100 and 4000m elevation
166 (described in detail in Heath and Baron 2014). LVWS is subjected to atmospheric N deposition
167 due to easterly winds carrying inorganic N from agricultural, vehicle, and industrial sources
168 along the Colorado Front Range into the park. The N falls primarily as wet deposition (Baron et
169 al. 2000). The LVWS receives ~3-3.5 kg/ha/y of wet N deposition, which has previously been
170 found to alter ecosystem processes (Baron et al. 2000; Booth et al. 2016; Oleksy et al. 2020). The

171 watershed receives ~105 cm of precipitation per year, with ~50 cm falling in the summer months
172 (Heath and Baron 2014).

173 We collected soils from the subalpine forest at ~3200 m, sampling from four randomly
174 selected GPS coordinates where the conditions appeared undisturbed by human foot traffic. No
175 sampling was conducted on the long-term N fertilization plots (Boot et al. 2016).

176 **2.1.2 Soil collection and analyses**

177 All soils were collected either in the second week of June 2018 or the third week of July
178 2018. Soils were collected using a 5cm-diameter soil auger to a depth of ~20 cm. Six cores were
179 collected randomly throughout each sampling plot and bulked into gallon Ziploc bags. Bags were
180 placed on ice in the field to minimize microbial activity, and then refrigerated at 4 C upon return
181 to the lab. Within 24 hr after sampling, soils were sieved to 2mm and homogenized by treatment.
182 Ziploc bags containing the processed soil were frozen at -18 C. All incubations and analyses
183 were performed within three months after soils were collected.

184 Prior to freezing soils, we performed KCl extractions and calculated soil gravimetric
185 water content. To quantify soil NH_4^+ and NO_3^- levels, we mixed 10 g soil subsamples with 50 mL
186 2M KCl, and they were shaken at 250 rpm for 1 hr, settled overnight, and then gravity filtered.
187 Extracts were frozen and thereafter analyzed colorimetrically using an Alpkem FIA wet
188 chemistry system (O.I. Analytical, College Station TX). We determined gravimetric water
189 content by drying subsamples to a constant weight in a 105 C oven.

190 After soils had been frozen, we measured soil pH, soil organic C and N (SOC and SON),
191 and soil microbial biomass C and N (MBC and MBN). We created slurries of 1:10 soil to DI
192 water and then measured soil pH with a benchtop meter (Thermo Scientific Orion Star™ A211
193 Benchtop pH Meter, Waltham, MA, USA). Frozen soil subsamples were dried in a 60 C oven

194 and then ground for SOC and SON analysis with a LECO Tru-Spec CN analyzer (Leco Corp.,
195 St. Joseph, MI, USA). Microbial biomass was extracted from 20 g frozen soil subsamples in 100
196 mL 0.5M K₂SO₄, and then solubilized in 1% chloroform. MBC and MBN were measured using a
197 Shimadzu Total Organic Carbon analyzer that also measures ON (Shimadzu Scientific
198 Instruments, Wood Dale, IL, USA). MBC and MBN were calculated as the differences between
199 1% chloroform slurry C or N and the extractable OC or ON concentration. Extraction efficiency
200 was corrected at $k_{EC} = 0.45$ and $k_{EN} = 0.46$ (Zeglin et al. 2013). SOC, SON, MBC, and MBN
201 were only measured for soils collected in June 2018.

202 Frozen soils were overnight shipped on ice to Ward Laboratories Inc. (Kearney, NE,
203 USA), where soil phospholipid fatty acids (PLFAs) were extracted, saponified, and methylated
204 to form fatty acid methyl esters (FAMES) as in Allison et al. 2007. FAMES were then identified
205 and quantified using a capillary gas chromatograph. Ward Laboratories summed individual
206 FAMES into the following functional groups: Gram + bacteria, Gram – bacteria, arbuscular
207 mycorrhizal fungi, saprophytes, protozoa, and undifferentiated using methods as in Allison et al.
208 2007, and afterwards we estimated total fungi:bacteria ratios for each soil.

209 **2.1.3 Determination of soil saturation**

210 To manipulate soil moisture, we first determined the soil maximum water holding
211 capacity (WHC). By sieving our soils, we broke down all soil pore structures, so % water-filled
212 pore space (WFPS), which is used to calculate field capacity, was of no use here (Farquharson
213 and Baldock 2008). Instead, we thawed subsamples of the frozen, field-moist soil and amended
214 them with DI water until fully saturated, or at maximum WHC. Then, we dried samples to a
215 constant weight in a 105 C oven and calculated saturated water content by dividing the g water in
216 the subsample by the subsample dry soil mass.

217 To determine the desired soil saturation for a given sample, we multiplied the desired
218 percent soil saturation by the saturated water content. Then, to determine the g of water to add to
219 a given sample, we multiplied the difference between the subsample GWC and desired soil
220 saturation by the g soil incubated.

221 **2.1.4 Comparison of soil properties at two time points**

222 For our experiments, we incubated soils under isotopically enriched conditions and at
223 natural abundance. Due to the limited volume of soil we were permitted to collect in June 2018,
224 we collected more soil in July 2018. As such, the isotopically enriched incubations were
225 performed on June soils, and only on the subalpine soil and the HNHW soil from the agricultural
226 site. Dissimilarly, the natural abundance incubations were performed on all the soil types (see
227 Table 1) we collected in July.

228 To account for differences in soil properties from June to July that could impact
229 microbial activities, we compared all measured soil properties for the soils used in enriched and
230 natural abundance incubations (subalpine and HNHW) with t-tests. Since the other agricultural
231 soils were not used in June, we did not compare how their properties changed from June to July.

232 **2.2 Soil incubations**

233

234 **2.2.1 Soil amendments and incubation setup**

235 Soils were separated into two treatment groups: those amended with isotopically enriched
236 substrate and those held at natural abundance. For all incubations, the frozen soil equivalent of
237 50g of dry soil (this varied by site and soil treatment) was weighed into 0.5 L Ball jars and
238 refrigerated overnight to thaw. Prior to amendment, all soils were removed from the refrigerator
239 and warmed slowly to room temperature to reduce disruption of microbial cellular membranes
240 (Boot et al. 2016).

241 Soils that were isotopically enriched were amended with 98 AP excess $^{15}\text{NH}_4\text{Cl}$ or
242 $\text{Na}^{15}\text{NO}_3$ at an application rate of less than 3% soil NH_4^+ or NO_3^- to prevent fertilization. The
243 stable isotope tracer was dissolved in DI water and pipetted over the soils at a constant
244 application rate, and then additional DI water was distributed by pipette over the soils to bring
245 soils to the desired saturations (10-95% soil saturation). After all liquid was added to a given
246 soil, it was mixed thoroughly to ensure sufficient distribution of stable isotope tracer and
247 homogeneous saturation.

248 Soils that were held at natural abundance were brought to desired saturations either via
249 additions of DI water or through air drying on ice to reduce microbial activity. For the soils that
250 needed to be air dried, they were weighed periodically (~every 10 min) to track moisture loss
251 until the desired dryness was reached.

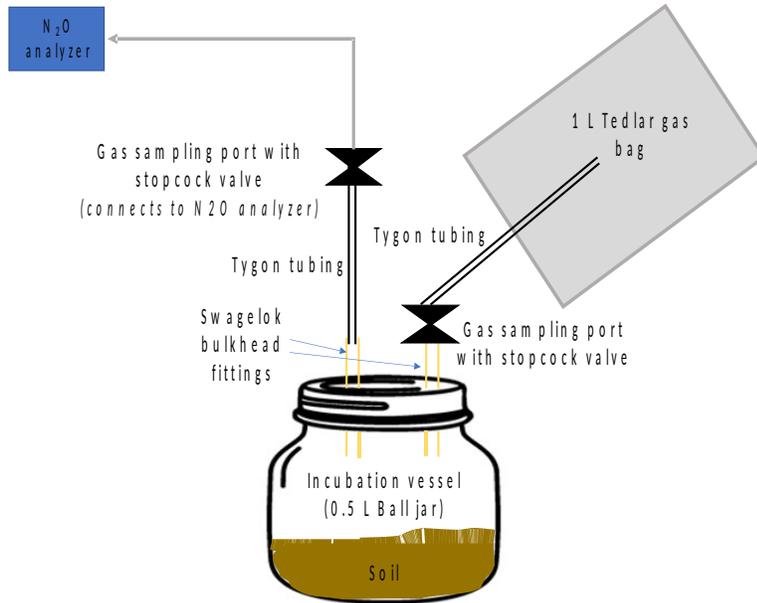
252 After soils were brought to their desired saturation and amended with tracer (if
253 applicable), jars were sealed for incubation (Diagram 1). It is methodologically challenging to
254 balance the need for a small headspace (thus maximizing final N_2O concentration) with the need
255 to remove large volumes of air for isotopic analysis from a hard-sided incubation jar (more
256 details in Section 3). We compromised with a design that included a 0.5L jar and a 1L gas bag
257 connected to the jar headspace. The jar lid was drilled to contain two ports with Swagelok
258 bulkhead fittings. One fitting vented to $\frac{1}{4}$ in Tygon tubing with a two-way stopcock for gas
259 sampling. The other fitting vented to a luer slip that could be fitted with a two-way stopcock
260 attached to a 1 L Tedlar gas bag.

261

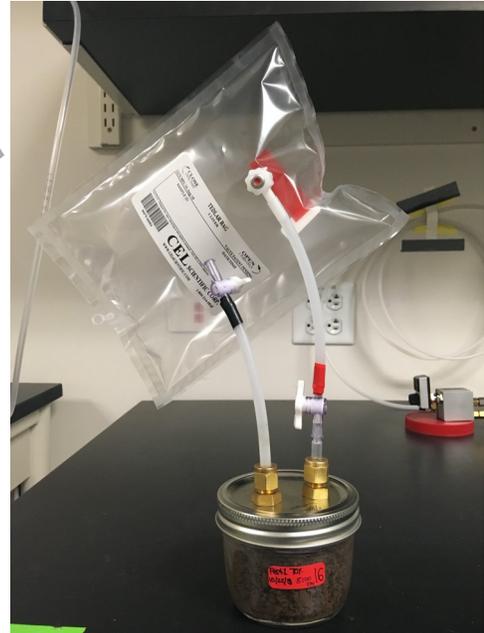
262

263 **Diagram 1.** Incubation jar-gas bag apparatus. Panel **a** is a conceptual model of the apparatus,
 264 and panel **b** shows a photograph of the apparatus.

a)



b)



265

266

267 At the outset of the incubation, we flushed all jars and filled associated gas bags from a
 268 cylinder of medical-grade compressed air to provide a uniform starting gas background (Airgas
 269 Industries, etc.). Soils were incubated on in an interior lab countertop, at approximately 24 C for
 270 28 hr. At the end of the incubation period, we mixed the jar and gas bag air with a 60 mL syringe
 271 to homogenize the headspaces. At the time of sampling, we connected the gas analyzer
 272 instrument system directly to the headspace-gas bag apparatus. Removal of jar headspace air thus
 273 emptied the gas bag and jar headspaces together, keeping the jar air pressure at atmospheric
 274 levels.

275

276 2.2.2 Measurements of N₂O concentration and isotopic compositions

277 After 28 hours, incubation vessels were attached to our laser-based Los Gatos Research

278 (LGR) N₂O isotopic analyzer. Gas was sampled from each incubation vessel for 12-15 minutes

279 (or until the N₂O concentration stabilized). Samples were attached to the analyzer upstream of a
 280 Nafion-Carbosorb-Silica gel scrubbing system to remove CO₂, H₂O vapor, and VOCs from the
 281 sampling stream. This measure is taken to minimize the optical peak-broadening effects inherent
 282 to laser-based analyzers, as these effects can decrease the accuracy of reported N₂O
 283 concentrations. For further details, see Stuchiner et al. (2020).

284 We previously described the instrumental determination of N₂O concentration and
 285 isotopic composition (Stuchiner et al. 2020). Briefly, our analyzer measures concentrations of
 286 N₂O (¹⁴N¹⁴N¹⁶O) and all its isotopomers (¹⁴N¹⁵N¹⁶O, ¹⁵N¹⁴N¹⁶O, and ¹⁴N¹⁴N¹⁸O) using cavity
 287 enhanced laser absorption spectroscopy (Los Gatos Research N₂O Isotopic Analyzer model 914–
 288 0027; ABB-Los Gatos Research, Mountain View, CA, USA) in a continuous flow-through
 289 system without pre-concentration of the incoming gas.

290 All raw concentration data for each sample was exported to Excel (version 16.46) where
 291 it was trimmed to include only the N₂O and isotopomer concentrations after readings had
 292 stabilized. N₂O concentration and each isotopomer was calibrated using the model described in
 293 Stuchiner et al. (2020), and then converted to δ notation. We calculated the δ values using
 294 standard notation and the ¹⁵N/¹⁴N ratio of atmospheric N₂ (0.0036765) or the ¹⁸O/¹⁶O in VSMOW
 295 (0.0020052). All δ values are in ‰. The δ¹⁸O values are reported in the Supplemental
 296 Information (Figure S1).

297

298 Equations 1-3 can be used to determine the δ values for the N₂O isotopomers:
 299

300 (1) $\delta N^{\alpha} = \dots$

301
 302 (2) $\delta N^{\beta} = \dots$

303
 304 (3) $\delta O = \dots$
 305

306 And to determine SP we used Equation 4:
307

$$308 \quad (4) SP = \delta_{\square}^{15} N^{\alpha} - \delta^{15} N^{\beta}$$

309

310 At low N₂O concentrations the SP data was consistently unreliable or out of the
311 biologically plausible range, so we did not use any SP data from those incubations. Despite
312 generally good precision and accuracy at N₂O concentrations above 2 ppm, we have observed
313 that the instrument periodically reported SP values that were far outside of the plausible range
314 (Stuchiner et al. 2020). According to Hu et al. (2015), this plausible range goes from -30 to 50%,
315 but we decided to extend the range from -40 to 65% because there is uncertainty surrounding the
316 “true” biologically plausible SP range owing to ambiguity in isotopomer calibration and
317 analytical precision (Ostrom and Ostrom 2017, Stuchiner et al. 2020). Thus, we excluded 10 of
318 the 59 samples included in our analysis, where they reported biologically implausible SP values.
319 The complete SP dataset is presented in Table S1. The majority of these implausible values were
320 associated with relatively low N₂O concentrations (Figure S3).

321

322 **2.2.3 Leak test of incubation apparatus**

323 A separate test was performed to assess the gastight seals of the incubation apparatus used
324 in this experiment (Diagram 1). Twelve incubation apparatus were filled with zero-grade air
325 (80:20 N₂:O₂ blend, Airgas Industries) as described in Section 2.2.1 and injected with 1 mL of 99
326 atom percent (AP) ¹⁵N¹⁵N¹⁶O using a 3 mL syringe, raising the N₂O concentration to ~500ppb and
327 the $\delta^{15}\text{N}^{\text{bulk}}$ to ~6300‰. Six apparatus were sampled for N₂O and its isotopomer concentrations 1
328 hr after preparation (T0) and the remaining six apparatus were sampled 48 hr after preparation
329 (T48). All samples were taken on our LGR N₂O isotopic analyzer. Change in total N₂O
330 concentration was not significant, while change in ¹⁵N enriched N₂O was < 2.3% for all
331 isotopomers.

332 2.3 Data analysis

333 All raw data was collected and collated in Excel, and then all statistical analyses were
 334 performed in RStudio (version 4.0.2 (2020-06-22) -- "Taking Off Again" © 2020 The R
 335 Foundation for Statistical Computing). Differences among inherent soil properties and treatment
 336 effects were examined using ANOVAs and t-tests. Residuals were examined for departure from
 337 normality, and all N₂O production data for the enriched incubations were log-transformed to
 338 meet assumptions of normality in residuals. All predicted N₂O production data for the natural
 339 abundance incubations were also log-transformed to meet assumptions of normality in residuals.

340 The logistic regressions used to predict N₂O production rates were fitted in RStudio with
 341 the package `dr4pl`.

342 3. Results

343 3.1 Soil properties

344 Soils from the agricultural and subalpine sites differed sharply in biogeochemical
 345 properties. The June soils are summarized in Table 2. There were notable differences in most soil
 346 properties between the HNHW and subalpine June soils ($p < 0.05$ in all cases), excluding
 347 fungi:bacteria ratios, which were not significantly different.

348 **Table 2.** Biogeochemical properties of HNHW and subalpine soils collected in June 2018. The
 349 n-value for each measurement corresponds to the number of technical replicates within each
 350 bulked soil treatment. Error bars represent \pm one SE from the mean.

351

Soil treatment	NO ₃ ⁻ (µg N/g dry soil)	NH ₄ ⁺ (µg N/g dry soil)	Fungi:Bacteria (% fungi/% bacteria)	Soil pH
HNHW	80.3 (± 6.51) (n = 10)	3.03 (± 1.09) (n = 10)	0.193 (± 0.035) (n = 4)	7.90 (± 0.06) (n = 8)
Subalpine	1.06 (± 0.420) (n = 9)	5.35 (± 1.29) (n = 9)	0.192 (± 0.030) (n = 3)	5.14 (± 0.08) (n = 9)

352

353 The July soils are summarized in Table 3. The soil NO₃⁻ concentrations were notably
 354 higher in all agricultural soils compared to the subalpine soil ($p < 0.001$). Interestingly, pairwise
 355 comparisons illustrate that only the HNHW soil had a significantly higher NO₃⁻ concentration

356 compared to the other agricultural soils ($p < 0.05$ in all cases). The other treatments of the
 357 agriculture soil did not substantially influence NO_3^- concentration, regardless of HN or LN
 358 appointment. This is likely due to high concentrations of dissolved NO_3^- in the irrigation water.

359 **Table 3.** Biogeochemical properties of agricultural and subalpine soils collected in July 2018.
 360 The n-value for each measurement corresponds to the number of technical replicates within each
 361 bulked soil treatment. Error bars represent \pm one SE from the mean.

Soil treatment	NO_3^- ($\mu\text{g N/g dry soil}$)	NH_4^+ ($\mu\text{g N/g dry soil}$)	Fungi:Bacteria (% fungi/% bacteria)	Soil pH
HNHW	77.9 (± 4.74) (n = 12)	1.80 (± 0.23) (n = 12)	0.142 (± 0.008) (n = 4)	8.15 (± 0.03) (n = 8)
HNLW	19.5 (± 6.19) (n = 9)	2.82 (± 0.04) (n = 9)	0.177 (± 0.014) (n = 3)	8.12 (± 0.04) (n = 6)
LNHW	13.3 (± 1.31) (n = 12)	1.80 (± 0.14) (n = 12)	0.188 (± 0.023) (n = 4)	8.16 (± 0.05) (n = 8)
LNLW	8.51 (± 0.74) (n = 9)	2.07 (± 0.07) (n = 9)	0.183 (± 0.013) (n = 3)	8.32 (± 0.04) (n = 6)
Subalpine	0.500 (± 0.13) (n = 18)	5.11 (± 0.74) (n = 18)	0.267 (± 0.020) (n = 8)	4.72 (± 0.04) (n = 18)

362
 363 Pairwise comparisons also demonstrate that all agricultural soils had significantly less
 364 NH_4^+ compared to the subalpine soil ($p < 0.001$ in all cases), but none of the agricultural soils
 365 differed in NH_4^+ concentration.

366 There were some differences in fungi:bacteria ratios for soils sampled in July. While
 367 there were no differences in fungi:bacteria within the agricultural soils, the HNHW soil had a
 368 significantly lower fungi:bacteria ratio from the subalpine soil ($p = 0.0016$), and the HNHW,
 369 LNHW, and LNLW soils had borderline significantly lower fungi:bacteria ratios compared to the
 370 subalpine soil ($p = 0.045$, $p = 0.056$, $p = 0.071$, respectively).

371 Soil pH was slightly basic for all agricultural soils and slightly acidic for the subalpine
 372 soil ($p < 0.001$).

373 Finally, we compared the HNHW and subalpine soil properties from June to July. We
 374 determined no significant differences in soil NO_3^- , NH_4^+ , or fungi:bacteria within each soil from
 375 June to July ($p > 0.05$ in all cases). However, both soils differed in pH from June to July, with
 376 the HNHW soil becoming more basic ($p = 0.005$) and the subalpine soil becoming more acidic (p
 377 < 0.001).

378

379 **3.2 Soils held at natural abundance**380 **3.2.1 N₂O production rate in natural abundance soils**381 N₂O production rates showed a clear threshold response to variation in soil moisture382 (Figure 1). At low soil moistures (10-50% soil saturation), N₂O production rate was relatively

383 low and constant. However, at approximately 60% soil saturation, we observed a marked

384 increase in N₂O production rate (Figure 1). We fit the following four-parameter logistic model to

385 the data to characterize these response curves (Equation 5):

386

387
$$(5) y = d + \frac{a-d}{1 + \left(\frac{x}{c}\right)^b}$$

388

389

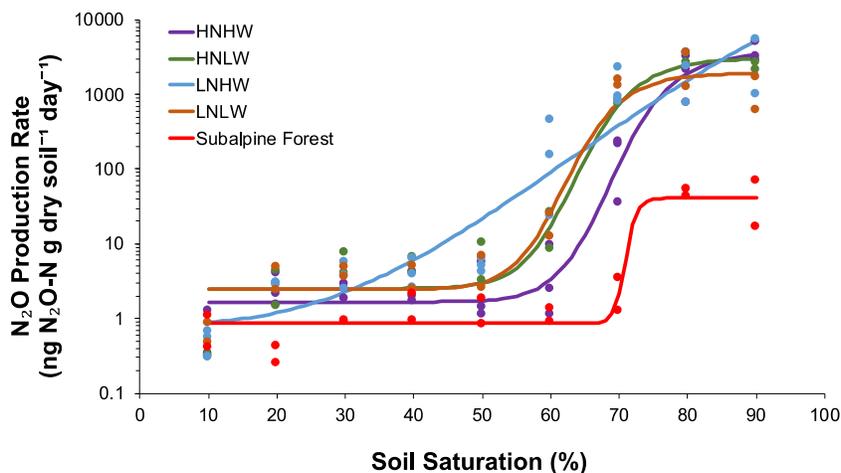
390 Where a is the minimum value, b is the slope of the line, c is the inflection point on the line

391 (halfway point between a and d), and d is the maximum value. Each of these values correspond

392 to a biologically relevant N₂O production parameter (Table 3).

393

394



395

396 **Figure 1.** N₂O production rates for all soils that were held from 10 to 90% soil saturation. Data

397 points correspond to observed production rate values and fitted lines are modelled predictions of

398 N₂O production rate for all soils. All fits were modelled using logistic regressions. For the

399 observed data, $n = 27$ for HNHW, $n = 26$ for LNHW, and $n = 18$ in all other cases. *Note the log-*
 400 *scaled y-axis.*

401

402 The results of these regressions are in Table 2. All logistic fits were strong (R^2 range from

403 0.88 to 0.94). These fits allowed us to compare the mean water content where soil N_2O

404 production rate flipped from low production rates to high production rates (Table 4). For most

405 agricultural soils, the transition point was at approximately 60% soil saturation, and for the

406 subalpine soil the transition point was at approximately 70% soil saturation (Figure 1).

407

408 **Table 4.** Summary of fit data comparing N_2O production rate data to estimated N_2O production
 409 rate data using logistic regressions. The R^2 values correspond to simple linear regressions
 410 comparing real production rate data to estimated production rate data, and the p-values
 411 correspond to each R^2 value. The values a-d correspond to the parameters estimated from each
 412 logistic regression. Parameter values occur from Equation 5. *Note the parameters and R^2 values*
 413 *result from log-transformed N_2O production rate data.*

414

Site	Treatment	Sample size (n)	R^2	Minimum N_2O production rate (a)	Slope (b)	Soil saturation transition point (c)	Maximum N_2O production rate (d)
<i>Agricultural</i>							
	HNHW	27	0.94	0.511	16.3	69.2	8.22
	HNLW	18	0.92	0.924	15.3	64.0	8.03
	LNHW	26	0.90	-0.170	2.65	83.5	15.8
	LNLW	18	0.94	0.906	15.6	62.3	7.58
<i>Subalpine forest</i>							
	Subalpine	18	0.88	-0.155	89.8	70.9	3.71

415

416 It is worth noting that the soil saturation transition point is notably higher for the LNHW

417 soil. The poorer model fit for this soil is likely due to one very high observed N_2O production

418 rate value that pulls the modeled values up, which is which the LNHW curve does not flatten at

419 high soil moistures, like the other curves do.

420 3.2.2. Intramolecular site preference (SP) at natural abundance

421 Across all soils, intramolecular SP generally decreased as percent soil saturation

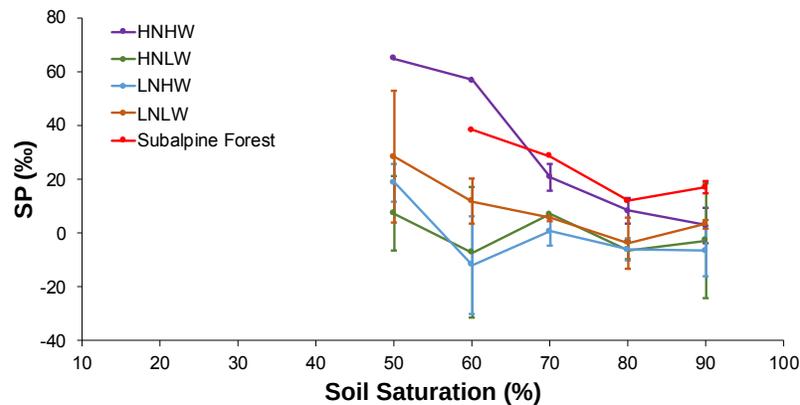
422 increased (Figure 2). Linear regression across all data points was significant ($p = 0.01$) with an

423 R^2 of 0.25. However, at 90% saturation, SP increased for HNLW, LNLW, and the subalpine soil

424 (Figure 2). Across all soil saturations, HNHW tended to have more enriched SP values than other

425 soils. While there is only data for the subalpine soil starting at 60% soil saturation, this soil also
 426 tended to have more enriched SP values. Across all saturations, HNLW, HNLW, and LNLW
 427 soils reported SP values between the HNHW and subalpine SP endpoints (Figure 2).

428



429

430 **Figure 2.** Intramolecular SP for all soils from 50-90% soil saturation. SP values were only used
 431 from this soil saturation range because values from lower soil saturations were deemed
 432 unreliable or not biologically realistic. Error bars represent \pm one SE from the mean. *Note not all*
 433 *data points have error bars, as multiple soils have $n = 1$ at certain soil saturations.* All sample
 434 size information is summarized in Table S2.

435

436 While all soils followed the same general patterns with increasing saturation, there was a
 437 substantial degree of variability among SP values at each saturation level (Figure 2). At 50% soil
 438 saturation, SP ranged from 65‰ for HNHW to 7‰ for HNLW. Interestingly, as saturation
 439 increased, the degree of difference among SP values decreased, however soils deviated from this
 440 trend at 90% saturation (Figure 2). At this soil saturation the range of values widened. The
 441 subalpine soil had an SP of 17‰, whereas the LNHW soil had an SP of -7‰.

442

443 Together, the similar magnitudes and directions of SP values across soils indicate that
 444 microbes were likely performing similar N-transformations at each saturation level. However,
 445 variation among values within each saturation level could indicate soil-specific differences in
 446 microbial behavior. These distinctions may be able to help better elucidate finer-scale differences
 447 in microbial metabolism across soils.

447

448 **3.3 Isotopically amended soils**

449 **3.3.1 N₂O production rate in isotopically amended soils**

450 There was no difference in N₂O production rate within each soil saturation level. This

451 indicates that there was no treatment effect from the ¹⁵NO₃⁻ vs. ¹⁵NH₄⁺ amendments (Figure 3).

452 N₂O production rate was highest in the 90% saturation agricultural soils (1650 and 1928

453 ng N₂O-N/g dry soil/day from soils amended with ¹⁵NH₄⁺ and ¹⁵NO₃⁻, respectively), but markedly

454 decreased in the 50% saturation soils (3.87 and 3.78 ng N₂O-N/g dry soil/day from soils

455 amended with ¹⁵NH₄⁺ and ¹⁵NO₃⁻, respectively; Figure 3).

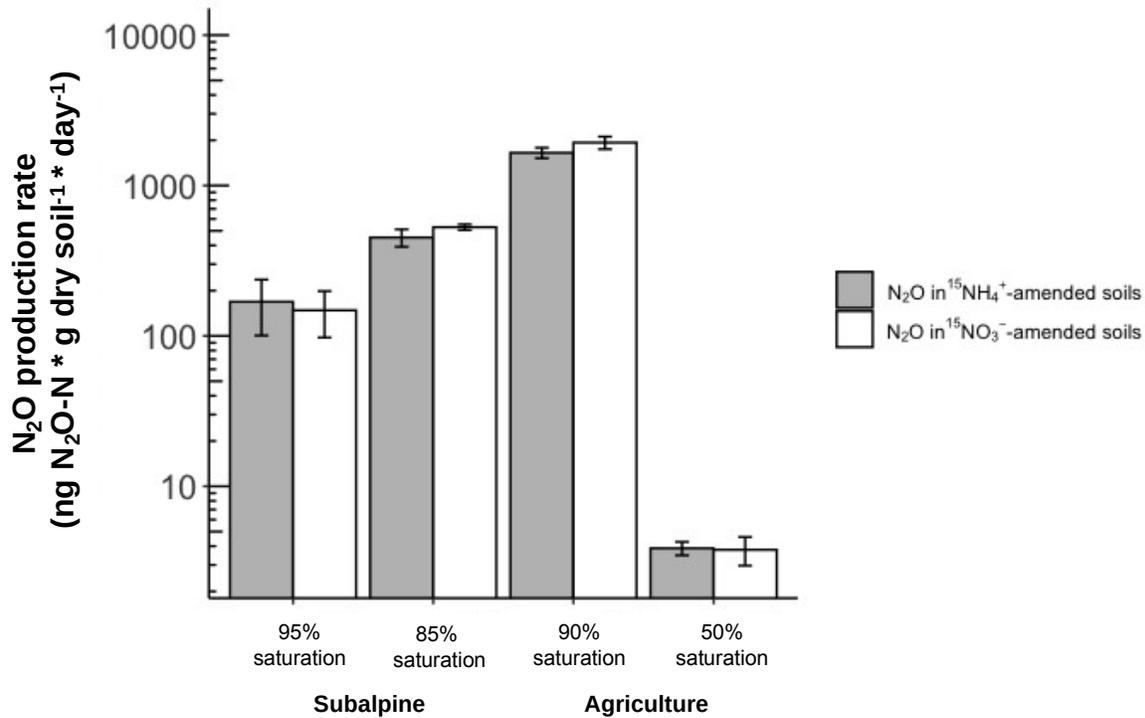
456 The subalpine soil had intermediate N₂O production rates at both saturations (Figure 3).

457 N₂O production rate was higher in the 85% saturated soils (451 and 528 ng N₂O-N/g dry soil/day

458 from soils amended with ¹⁵NH₄⁺ and ¹⁵NO₃⁻, respectively), and lower in the 95% saturated soils

459 (168 and 147 ng N₂O-N/g dry soil/day from soils amended with ¹⁵NH₄⁺ and ¹⁵NO₃⁻, respectively),

460 suggesting that N₂O → N₂ reduction could be an important process in the wettest subalpine soils.



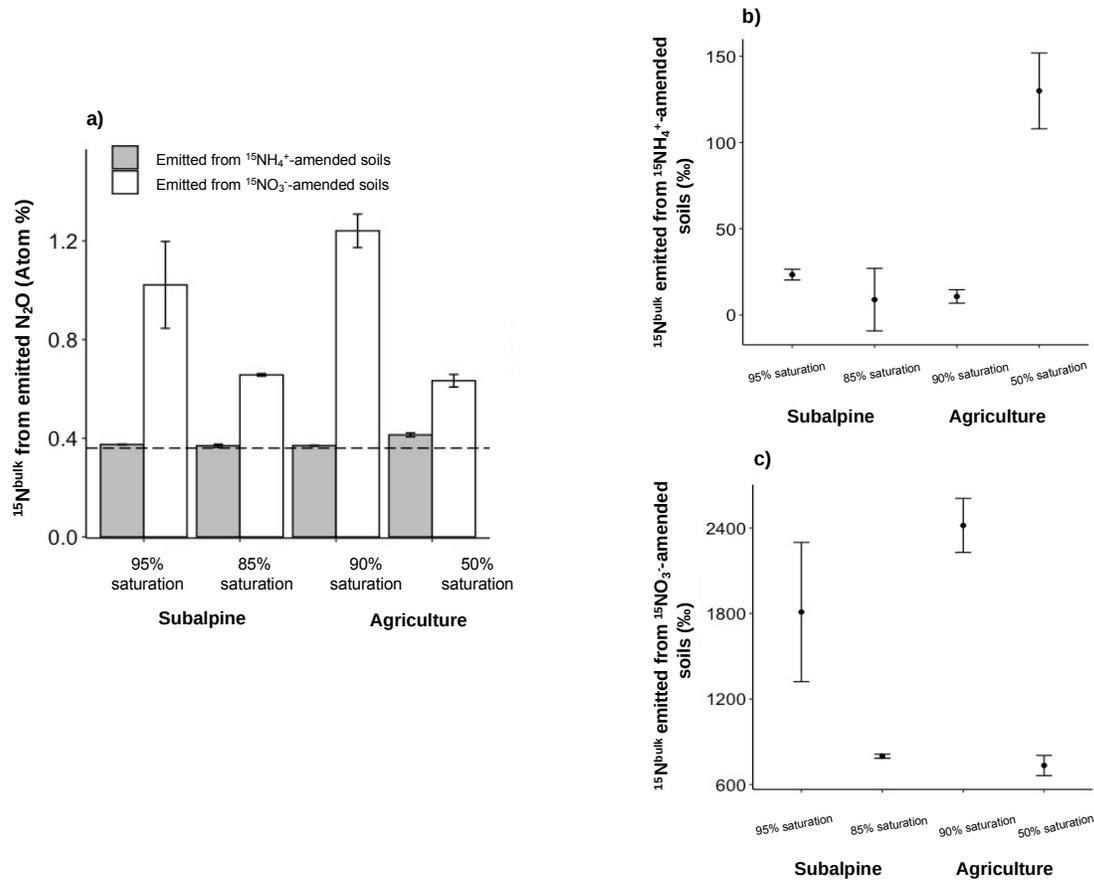
461
 462 **Figure 3.** N₂O production rates from agricultural and subalpine soils across all soil moisture and
 463 isotopic enrichment treatments. In all cases n = 4 except n = 3 for the ¹⁵NO₃⁻-amended 50%
 464 saturation agricultural soil. Error bars represent ± one SE from the mean. *Note the log scale y-*
 465 *axis.*

466
 467

468 3.3.2 Tracing ¹⁵N^{bulk} signatures to source partition among microbial processes

469 Across all soil saturations, the majority of ¹⁵N-label was emitted from soils amended with
 470 ¹⁵NO₃⁻ compared to soils amended with ¹⁵NH₄⁺ (Figure 4; p < 0.001). In the ¹⁵NO₃⁻-amended soils,
 471 the largest ¹⁵N^{bulk} signature (AP) was emitted from the 90% saturation agricultural soil, the
 472 smallest ¹⁵N^{bulk} signature was emitted from the 50% saturation agricultural soil, and intermediate
 473 between those two were the subalpine soils, with the 95% saturation soil emitting a greater ¹⁵N^{bulk}
 474 signature than the 85% saturation subalpine soil (Figure 4a).

475

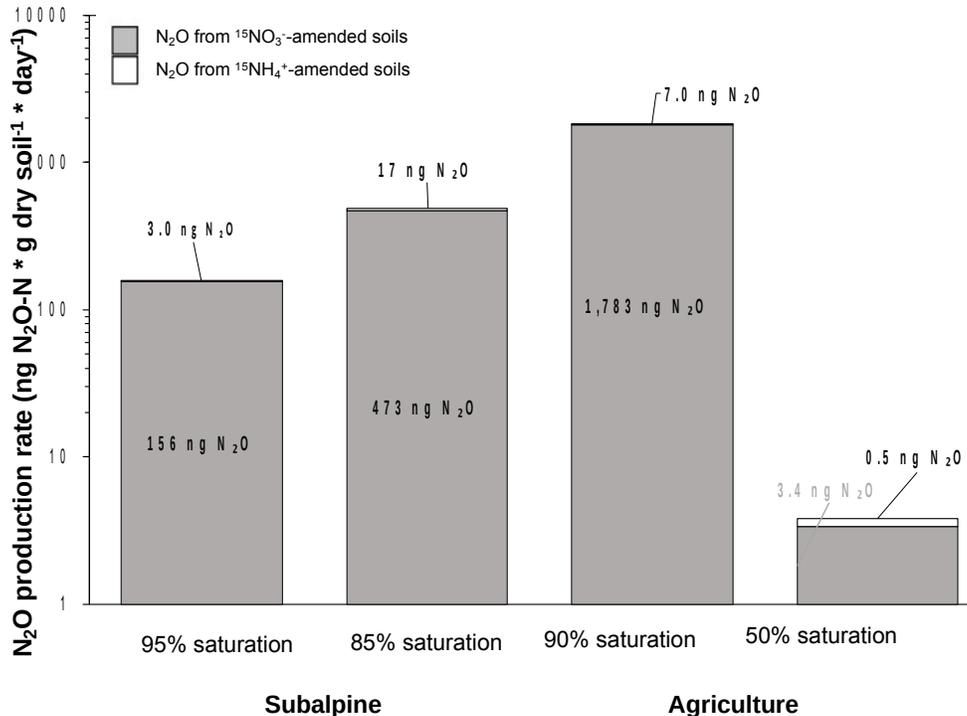


476
 477 **Figure 4.** *Panel a:* $^{15}\text{N}_2\text{O}_{\text{bulk}}$ in AP emitted from $^{15}\text{NH}_4^+$ or $^{15}\text{NO}_3^-$ amended soils. *Panel b:*
 478 $\delta^{15}\text{N}_2\text{O}_{\text{bulk}}$ emitted from $^{15}\text{NH}_4^+$ amended soils. *Panel c:* $\delta^{15}\text{N}_2\text{O}_{\text{bulk}}$ emitted from $^{15}\text{NO}_3^-$ amended
 479 soils. In all cases $n = 4$ except $n = 3$ for the $^{15}\text{NO}_3^-$ -amended 50% saturation agricultural soil.
 480 Error bars represent \pm one SE from the mean. The dashed horizontal line in Panel a indicates the
 481 AP threshold for isotopic enrichment above natural abundance (0.36 %).
 482

483 In the $^{15}\text{NH}_4^+$ -amended soils, the $^{15}\text{N}_{\text{bulk}}$ signatures for the two subalpine soils and the
 484 90% saturation agricultural soil were mostly homogenous and concordant with natural
 485 abundance isotopic enrichment (3 to 45‰; Figure 4c). In contrast, the $^{15}\text{N}_{\text{bulk}}$ signature for the
 486 50% saturation soil was significantly more enriched than the other $^{15}\text{NH}_4^+$ -amended soils ($p <$
 487 0.001), and reported a mean AP and δ value substantially above ambient (0.41% and 130‰,
 488 respectively, Figure 4a, 4c). Because soils were amended with highly enriched substrate (99 AP
 489 excess $^{15}\text{NH}_4^+$), this indicates that NH_4^+ was not an important substrate for microbial processes in

490 three of the four soils at these moisture levels, but it was at least tractably used for microbial
 491 metabolism in the 50% saturation agricultural soil, albeit not appreciably (Figure 4c).

492



493

494 **Figure 5.** N₂O production rates partitioned by NH₄⁺-consuming vs. NO₃⁻-consuming pathways.
 495 In most cases, the majority of emitted N₂O is generated from microbial NO₃⁻ use, although a
 496 detectable fraction of emitted N₂O is generated via microbial NH₄⁺ use in the ¹⁵NH₄⁺-amended
 497 50% saturation agricultural soil. In all cases, n = 4, except n = 3 for the ¹⁵NO₃⁻-amended 50%
 498 saturation agricultural soil. Error bars represent ± one SE from the mean.

499

500

501 We also calculated the total N₂O production rate for each soil partitioned by N₂O emitted
 502 from ¹⁵NO₃⁻ or ¹⁵NH₄⁺-amended soil. This analysis further illustrates that most emitted N₂O came
 503 from NO₃⁻ transformations, but a minute fraction of emitted N₂O came from NH₄⁺

504 transformations (Figure 5). Together, Figures 4 and 5 show that 1) the majority of emitted N₂O

505 was generated via NO₃⁻ transformations, 2) there is some variability in ¹⁵N^{bulk} signatures and N₂O

506 production rate among the ¹⁵NO₃⁻-amended soils, and 3) a small, but non-negligible fraction of

507 N₂O was emitted from NH₄⁺ transformations. While this is not surprising, given that the soils are

507 all held at different saturations, this variability suggests that microbial activity likely differs
 508 among the soils.

509

510 **3.3.3 Isotopic enrichment in the $\delta^{15}\text{N}^\alpha$ vs. $\delta^{15}\text{N}^\beta$ position of the emitted N_2O**

511 The location of isotopic enrichment (α -position or β -position) in the emitted N_2O varied
 512 among soils and by type of isotopic amendment. We chose to present this data by quantifying the
 513 percent of emitted N_2O enriched in the β -position ($^{15}\text{N} - \text{N} - \text{O}$) from each soil by calculating the
 514 atom percent enrichment (APE) in each isotopomer, using Equations 6 and 7, where AP^{15}N^x is
 515 the $\text{AP}^{15}\text{N}^\alpha$ or $^{15}\text{N}^\beta$, and AP_{std} corresponds to the AP of $^{15}\text{N}/(^{15}\text{N}+^{14}\text{N})$ in N_2 -Air (i.e., 0.003663):

516

$$517 \quad (6) \text{ APE} \in \beta \text{ position} = \text{AP}^{15} \text{N}^\beta - \text{AP}_{\text{std}}$$

$$518 \quad \text{APE} \in \alpha \text{ position} = \text{AP}^{15} \text{N}^\alpha - \text{AP}_{\text{std}}$$

519

520

$$521 \quad (7) \text{ Fraction of enriched } ^{15}\text{N} \in \beta \text{ position} = \frac{\text{APE } \beta \text{ position}}{\text{APE } \beta \text{ position} + \text{APE } \alpha \text{ position}}$$

522

523 On the converse, the fraction of N_2O enriched in the α -position ($\text{N} - ^{15}\text{N} - \text{O}$) from each
 524 soil would be 1 - fraction β -position enrichment.

525 The majority of the emitted enriched N_2O derived from the $^{15}\text{NO}_3^-$ -amended soils, so we
 526 chose to focus on those soils for our analysis of α and β -position enrichment (Figure 4).

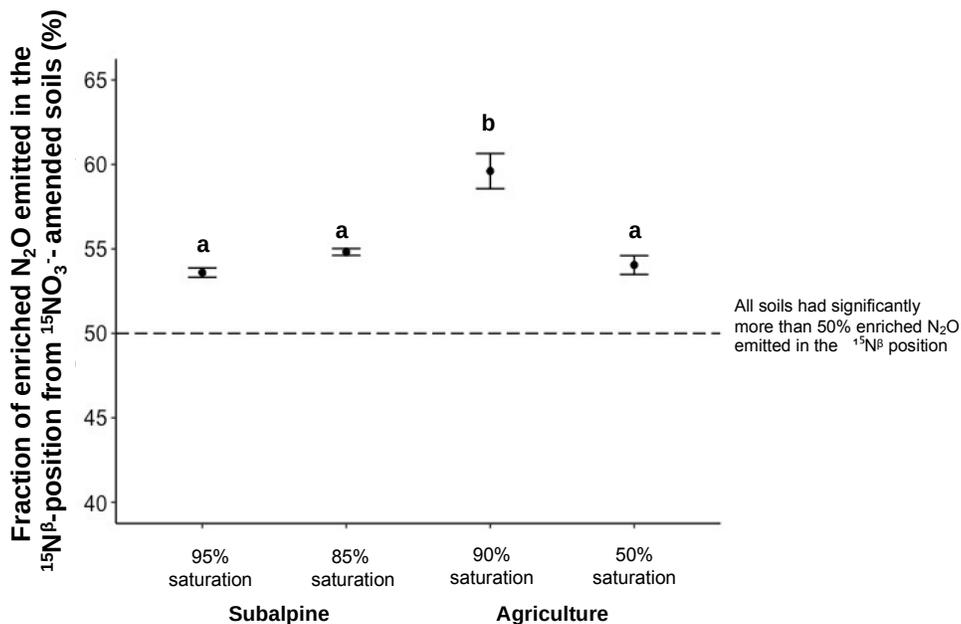
527 Information about the α and β -position enrichment from the $^{15}\text{NH}_4^+$ -amended soils is available in
 528 the Supplemental Information (Figure S2).

529 In the $^{15}\text{NO}_3^-$ -amended soils, the 90% saturation agricultural soil emitted significantly
 530 more N_2O enriched in the β -position than the other three soils ($p < 0.001$; Figure 6) as

531 characterized by highly negative SP values (Table S3). Conversely, neither the two subalpine
 532 soils nor the 50% saturation agricultural soil emitted significantly different amounts of N_2O

533 enriched in the β -position. The 90% saturation agricultural soil emitted 60% of β -position-

534 enriched N_2O , whereas the 50% saturation agricultural soil, the 95% saturation subalpine soil,
 535 and the 85% saturation subalpine soil emitted 54, 54, and 55% β -position-enriched N_2O ,
 536 respectively (Figure 6). Meaning, the 90% saturation agricultural soil was the least enriched in
 537 the α -position, whereas the other three soils had comparable amounts of α -position enrichment.
 538 However, one-sample t-tests compared each soil's fraction of emitted $^{15}N^{\beta}$ to 50%; all soils
 539 emitted significantly more than 50% β -position-enriched N_2O ($p < 0.05$ in all cases).



540
 541 **Figure 6.** Percent of emitted N_2O enriched in the β -position ($^{15}N-N-O$) from each $^{15}NO_3^-$ -
 542 amended soil. In all cases, $n = 4$, except $n = 3$ for the $^{15}NO_3^-$ -amended 50% saturation agricultural
 543 soil. Bars represent a 95% CI around the mean. For all soils, significantly more than half of the
 544 enriched N_2O emitted was enriched in the β -position (all percentages significantly above 50%
 545 dotted line). Different letters illustrate significantly different means. *Note: the percent β -position*
 546 *enriched N_2O was only considered for the $^{15}NO_3^-$ -amended soils because those soils emitted*
 547 *almost all the isotopically enriched N_2O .*

548

549

550 4. Discussion

551 By incubating N-disturbed soils under both natural abundance and enriched conditions,

552 we sought to better understand how well these isotopic methods align, and which microbial

553 source processes were driving N_2O emissions. To our knowledge, this is the first study to pair

554 enriched and natural abundance N stable isotope methods directly to examine N_2O production

555 pathways. Methodologically, we acknowledge that this study has some limitations, but it has
556 allowed us to make some interesting insights into microbial N₂O-generating activity. In
557 particular, there is generally good agreement between natural abundance and enriched
558 approaches, with some interesting departures. This work also led to new, unique observations of
559 position-specific enriched isotopomers, which both raise questions about biogeochemical
560 pathways and suggest a future tool for tracing pathways of N₂O emission.

561

562

563 **4.1 Modelling N₂O production rate**

564 The general response of our natural abundance incubated soils to variation in soil
565 moisture is consistent with previous studies, such that N₂O production rates showed a strong
566 threshold behavior with much higher rates of emissions at higher soil moisture (Figure 1; Parton
567 et al. 1996, Del Grosso et al. 2000, Li and Aber 2000, Parton et al. 2001, Ni et al. 2011, Taylor et
568 al. 2017, Ji et al. 2018, Song et al. 2019,). While the specific logistic regression we applied is
569 not important, this type of model fit allows us to extract useful parameters about N₂O production
570 capacity from these diverse soils (Table 4).

571 Our findings are consistent with classic interpretations of microbial N₂O metabolism,
572 such as the hole-in-the-pipe (HIP) and the “anaerobic balloon,” wherein the transition from
573 aerobic to anerobic metabolism is consistent with a shift from nitrification to denitrification, and
574 this is largely dictated by soil WFPS (Firestone and Davidson 1989, Li and Aber 2000,
575 Butterbach-Bahl et al. 2013). While we used percent soil saturation, not WFPS, our findings
576 align with Davidson (1993), who illustrates that from WFPS 60 to ~85%, most of the N-gas flux
577 is emitted as N₂O. This likely also helps to explain why the modelled N₂O production rate
578 stabilized for most of our soils at 90% saturation (Figure 1). However, it is worth noting that the
579 maximum N₂O production rates differed quite a bit among soil treatments, which is likely

580 modulated by other variables that effect the $\text{N}_2\text{O}:\text{N}_2$ emission ratio, such as soil NO_3^- or OC
581 (Firestone and Davidson 1989; Figure 1). Overall, the patterns we observed support our approach
582 of manipulating soil moisture to induce diverse N_2O production pathways and supports our
583 exploration of N_2O stable isotope responses.

584

585 **4.2 Patterns in natural abundance SP**

586 The trend we observed for SP in natural abundance incubations (Fig. 2) was clear and
587 may also suggest finer-scale differences in soil microbial behavior. The observed decline in SP
588 as soils became more saturated is consistent with literature SP values that indicate a transition
589 from nitrification to denitrification as soils get wetter (Bergstermann et al. 2011, Denk et al.
590 2017, Congreves et al. 2019, Ding et al. 2019). It is worth noting that in some cases $n = 1$ or 2
591 replicates per treatment as a consequence of instrument errors leading to biologically implausible
592 SP values (Stuchiner et al. 2020). Despite the small sample size, the pattern we observe showing
593 a transition from larger to smaller SP with increasing soil moisture is coherent, both within and
594 among soils (Figure 2).

595 That said, there is some nuance in the SP patterns that might provide insights into soil
596 microbial N_2O metabolism. We observed a SP of 65‰ for the 50% saturation HNHW soil
597 (Figure 2). Although higher SP values in drier soils are often consistent with nitrification, this
598 value falls above the reported range for nitrification of 9 to 46‰ (Löscher et al. 2012, Hu et al.
599 2015, Denk et al. 2017, Ding et al. 2019, Lewicka-Szczebak et al. 2020). Higher SP values can
600 be indicative of fungal denitrification (~20 to 45‰), but previous research suggests fungal
601 denitrification is unlikely to be dominant at this dry soil moisture level, and there were no large
602 soil aggregates to host fungal denitrification in our finely-seived soil (Hu et al. 2015, Maeda et
603 al. 2015, Denk et al. 2017). Recently, Wong et al. (2020) reported similarly high SP values (83‰

604 $\pm 25\%$) in marine sediments and posit that multiple biotic and abiotic processes proceeding
605 through multiple cycles could account for such enrichment. Additionally, it is also possible that a
606 yet-identified SP, like that for DNRA, could be responsible for this isotopic signature, since
607 DNRA has been reported in soils under drier conditions, where NO_3^- might be more bioavailable
608 (Rütting et al. 2011, Schmidt et al. 2011, Zhang et al. 2015, Friedl et al. 2018).

609 We also see generally more enriched SP values for HNHW and subalpine soils across all
610 saturations. In the subalpine soil, we were able to measure biologically plausible SP values
611 starting at 60% saturation, when denitrification tends to become more important (Groffman
612 2012, Fang et al. 2015, Cardenas et al. 2017, Schlüter et al. 2019, Thilarkarathna and Hernandez-
613 Ramirez 2021). Fungal denitrification could be responsible for more enriched SP values from the
614 subalpine soil, as the subalpine soil had greater fungi:bacteria ratio than the other soils (Table 3).
615 In contrast, the HNHW soil has the lowest fungi:bacteria ratio (Table 3). Thus, this pattern in
616 HNHW soils support Wong et al.'s (2020) assertion that more enriched SP values are due to
617 multiple processes co-occurring and proceeding through multiple cycles.

618 Lastly, we observed increases in SP in HNLW, LNLW, and the subalpine soil at 90%
619 saturation (Figure 2). These increases support previous findings that SP increases as $\text{N}_2\text{O} \rightarrow \text{N}_2$
620 reduction becomes more important, which is typical in anoxic and more reducing conditions
621 (Ostrom et al. 2010, Decock and Six 2013, Wu et al. 2016, Gaimster et al. 2018, Congreves et al.
622 2019).

623 In sum, these more subtle patterns in SP values suggest nuance in microbial N
624 biogeochemistry and/or N_2O production pathways among soils that may have been missed if the
625 assays had been of the more binary (i.e., nitrification vs. denitrification) ^{15}N isotope labeling
626 type.

627 **4.3 Isotopically enriched vs. natural abundance soils**

628 In general, our use of ^{15}N -labelled inorganic substrates to create the enriched incubations
629 corroborate our findings from natural abundance incubations. The wetter isotopically enriched
630 soils (85, 90, and 95% saturation) emitted almost all N_2O from NO_3^- transformations, rather than
631 from NH_4^+ transformations, suggesting denitrification (Figure 5). This aligns with the literature
632 and our natural abundance findings (Figure 2): (1) previous studies show wetter, anoxic soils use
633 NO_3^- as substrate for denitrification, (2) the classic range of bacterial denitrification SP goes from
634 -11 to 0‰, and (3) our natural abundance HNHW and subalpine soils also show a general
635 decrease in SP as they get wetter, indicating denitrification (Sutka et al. 2006, Butterbach-Bahl et
636 al. 2013, Rohe et al. 2017, Congreves et al. 2019).

637 However, our isotopically enriched findings depart from dogmatic expectations at lower
638 soil moisture, where the natural abundance data were likewise surprising. Specifically, while we
639 expected the drier, more oxic 50% saturation HNHW soil to generate most of its N_2O from
640 nitrification, the overwhelming majority of the N_2O came from $^{15}\text{NO}_3^-$ transformations (Figure 5,
641 Russow et al. 2009, Ball et al. 2010, Parker and Schimel 2011, Wu et al. 2017, Tan et al. 2018).
642 As noted in the previous section, the SP value was also unusually high (65‰) for this soil at 50%
643 saturation. Together, these data suggest that co-occurring processes, such as DNRA and
644 bacterial denitrification in anoxic microsites, could have given rise to the high $^{15}\text{NO}_3^-$ derived
645 N_2O in the enriched soil and the uncharacteristically enriched SP in the natural abundance soil
646 (Palacin-Lizarbe et al. 2019, Wong et al. 2020) Also, despite the one month time difference
647 between enriched and natural abundance assays, it seems likely that the soils were behaving
648 similarly, even after sieving the soil (hence breaking down natural soil pore structures) and
649 creating artificial incubation conditions. For example, soil NH_4^+ and NO_3^- levels were very
650 similar over time, and N_2O production rates were comparable (Tables 2 and 3, 3.6 vs. 2.8 ng

651 N₂O-N/g dry soil/day). Thus, although NO₃⁻ may not be the dominant source for N₂O in drier
652 field conditions, it clearly was for this incubation, and the SP data suggest an interesting
653 combination of processes leading to the N₂O emissions that we observed (Perez et al. 2006,
654 Opdyke et al. 2009, Ostrom et al. 2010, Park et al. 2011). Clearly, the combined deployment of
655 enriched and natural abundance methods created here a richer understanding of processes than
656 would be generated by either approach alone.

657

658 **4.4 Position-specific enrichment in isotopically labelled soils**

659 By isotopically labelling soils with ¹⁵NH₄⁺ or ¹⁵NO₃⁻, we were able to determine, and
660 report for the first time, position-specific ¹⁵N-enrichment in the emitted N₂O from incubations.
661 The majority of the emitted N₂O was from ¹⁵NO₃⁻-amended soils, as we discussed above, but
662 within that emitted N₂O, 54-60% of the ¹⁵N appeared in the β-position (Figure 6). This begs the
663 question: what can this strong degree of position-specific enrichment tell us? It would be ideal if
664 isotopically enriched isotopomers and SP could be informative of microbial N₂O-generating
665 pathways because these enriched signatures are far less ambiguous than natural abundance
666 signatures (Wagner-Riddle et al. 2008; Zhang et al. 2015, Snider et al. 2017). Because most of
667 the ¹⁵N-enrichment in the emitted N₂O derived from the wetter (85, 90, and 95% saturation)
668 ¹⁵NO₃⁻-amended soils, bacterial denitrification was likely an important N₂O-generating pathway
669 in these oxygen-poor soils (Figure 4; Parkin et al. 1987, Barnard et al. 2005, Baggs 2011, Krause
670 et al. 2017, Smith 2017, Gaimster et al. 2018). At natural abundance, bacterial denitrification
671 tends to correspond to low to negative SP, which aligns with the greater β-position enrichment
672 we observed ($SP = \delta^{15}N^{\alpha} - \delta^{15}N^{\beta}$, so if $\delta^{15}N^{\beta}$ is bigger than $\delta^{15}N^{\alpha}$ SP will be smaller; Toyoda et al.
673 2005, Toyoda et al. 2015). However, our enriched samples showed strongly negative SP values

674 (e.g., in the negative hundreds, see Table S3), whereas at natural abundance SP does not
675 typically get much lower than -30‰ (Wu et al. 2019; Hu et al. 2015).

676 Currently, we do not have an explanation for these strongly negative SP values from the
677 enriched incubations, but three possibilities emerge. First, isotopically labelled enrichment may
678 not scale proportionately with natural abundance enrichment. For example, Andriukonis and
679 Gorokhova (2017) show that phytoplankton growth rate appreciably decreases as ¹⁵N enrichment
680 in the environment increases, likely due to the kinetic isotope effect (KIE). Although enrichment
681 did not appear to impact N₂O production rate, as N₂O production rates between the enriched and
682 natural abundance incubations are comparable, perhaps it impacted the isotopic composition of
683 the emitted N₂O (Figure 1, Figure 3). Second, an inverse isotope effect, like Yang et al. (2014)
684 observed, wherein the β-position N-atom binds more strongly to the active site on a
685 denitrification enzyme, could have resulted in greater β-position enrichment in the emitted N₂O.
686 Finally, this could be an indication of microbial N-transformations we have not yet discovered or
687 do not yet fully understand biochemically. Future research should evaluate whether greater
688 enrichment in the β-position occurs reliably under particular conditions. If strong patterns in
689 position-specific enrichment are broadly observed, then this measure, like SP, could become
690 another tool for understanding which microbial processes give rise to N₂O.

691

692 **4.5 Concluding remarks**

693 We sought to better understand how incubating the same soils under natural abundance
694 and isotopically enriched conditions can better inform our ability to study microbial N₂O-
695 generating processes. Our study yielded both confirmatory and novel insights. Our natural
696 abundance incubations demonstrated that intramolecular SP decreases as soils transition from
697 dry to wet, which generally aligns with nitrification to denitrification SP values from the

698 literature. Our isotopically enriched incubations yielded isotopic signatures that mostly agreed
699 with the natural abundance isotopic signatures, which lends further credence to natural
700 abundance SP methods. While we acknowledge that it can be logistically challenging for labs to
701 perform natural abundance and isotopically enriched experiments in conjunction, our work
702 demonstrates that it can result in more robust trust in N₂O isotopic datasets for determining
703 microbial N₂O-generating pathways. As we improve our abilities to measure stable isotopes, it
704 may become more useful to compare-contrast natural abundance and enriched signatures because
705 natural abundance methods have the advantage of being less invasive, whereas enriched methods
706 have the advantage of being, usually, less ambiguous.

707 Scientific research has made ample strides over the past ~30 years to improve our usage
708 of stable isotopes for constraining the N₂O budget, but there remains need to continue to bolster
709 confidence in these methods. Our study illustrates that pairing natural abundance and isotopically
710 enriched soil incubations can reveal gaps in our understanding of microbial N-transformations
711 and different isotopic methodologies. We encourage future studies to consider pairing these
712 methods, and we also encourage researchers to assess position-specific N₂O enrichment, as this
713 may be yet another emergent property of the N₂O molecule that can inform us about microbial
714 N-metabolism.

715 **Data Availability Statement**

716 We have uploaded all finalized data to www.mountainscholar.org Accession number
717 forthcoming.

718 *Prepublication note: These data are embargoed, pending publication of this manuscript.*

719 **Acknowledgments, Samples, and Data**

720 The authors declare no conflicts of interest in this work. All data supporting this manuscript are
721 available via www.Mountainscholar.org. Due to limited remaining availability of frozen soil, the
722 authors will instead be happy to assist interested researchers to obtain fresh samples from our
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 727

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