Wintertime nitrate formation pathways in the North China Plain: Importance of N2O5 heterogeneous hydrolysis

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Abstract

Nitrate aerosols, formed via nitric acid (HNO) to balance inorganic cations in the particle phase, have constituted a major fraction of fine particulate matters (PM) during wintertime haze events in the North China Plain(NCP), with a progressively increasing contribution to PMmass. HNOis produced through homogeneous and heterogeneous pathways in the atmosphere, but the contribution of the two pathways to nitrate remains elusive. Simulations of a wintertime haze event in the NCP using a source-oriented WRF-Chem model reveal that the homogeneous and heterogeneous pathways contribute 48.4% and 51.6% of near-surface nitrate mass on average, respectively. The heterogeneous pathway dominates the nighttime HNOproduction in the planetary boundary layer, with an average contribution of 83%. Although NOis photolytically liable during daytime, the heterogeneous NOhydrolysis still contributes 10% of HNO. Our study highlights the significantly important role of NOheterogeneous hydrolysis in the nitrate formation during wintertime haze days.

1 Wintertime nitrate formation pathways in the North China Plain: 2 Importance of N₂O₅ heterogeneous hydrolysis 3 456789 Lang Liu^{1,4}, Naifang Bei², Bo Hu³, Jiarui Wu^{1,4}, Suixin Liu^{1,4}, Xia Li^{1,4}, Ruonan Wang^{1,4}, Zirui Liu³, Jiaoyang Yu¹, Min Zuo^{1,4}, Zhenxing Shen², Junji Cao^{1,4}, Xuexi Tie¹, and Guohui Li^{1,4*} ¹Key Lab of Aerosol Chemistry and Physics, SKLLQG, Institute of Earth Environment, Chinese Academy of Sciences, Xi'an, 710061, China ²School of Human Settlements and Civil Engineering, Xi'an Jiaotong University, Xi'an, 710049, China 10 ³State Key Laboratory of Atmospheric Boundary Layer Physics and Atmospheric Chemistry, Institute of 11 Atmospheric Physics, Chinese Academy of Sciences, Beijing, 100029, China 12 ⁴CAS Center for Excellence in Quaternary Science and Global Change, Xi'an, 710061, China 13 14 *Correspondence to: Guohui Li (<u>ligh@ieecas.cn</u>) 15 16 Abstract: Nitrate aerosols, formed via nitric acid (HNO₃) to balance inorganic cations in the 17 particle phase, have constituted a major fraction of fine particulate matters (PM_{25}) during

wintertime haze events in the North China Plain (NCP), with a progressively increasing 18 19 contribution to PM_{25} mass. HNO₃ is produced through homogeneous and heterogeneous 20 pathways in the atmosphere, but the contribution of the two pathways to nitrate remains 21 elusive. Simulations of a wintertime haze event in the NCP using a source-oriented 22 WRF-Chem model reveal that the homogeneous and heterogeneous pathways contribute 48.4% 23 and 51.6% of near-surface nitrate mass on average, respectively. The heterogeneous pathway 24 dominates the nighttime HNO₃ production in the planetary boundary layer, with an average 25 contribution of 83%. Although N₂O₅ is photolytically liable during daytime, the 26 heterogeneous N₂O₅ hydrolysis still contributes 10% of HNO₃. Our study highlights the 27 significantly important role of N₂O₅ heterogeneous hydrolysis in the nitrate formation during 28 wintertime haze days.

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30 **Plain Language Summary:** Nitrate aerosols have become a major fraction of fine particulate 31 matters during wintertime haze events in the North China Plain (NCP) with implementation 32 of strict emission mitigation measures since 2013. We quantify the contribution of nitrate 33 formation pathways during a wintertime haze event in the NCP using a source-oriented 34 WRF-Chem model. The homogeneous and heterogeneous pathways dominate the nitrate formation during daytime and nighttime respectively. The N₂O₅ heterogeneous hydrolysis 35 36 also plays an appreciable role in the nitrate formation even during daytime under weak 37 sunlight and high humidity conditions.

39 1 Introduction

40 In recent ten years, persistent and pervasive haze with high levels of fine particulate 41 matters (PM_{2.5}) frequently engulfs the North China Plain (NCP) during wintertime, caused by 42 a synergy of massive anthropogenic emissions and atmospheric processes (An et al., 2019). 43 High loadings of PM_{2.5} in the atmosphere not only adversely affect human health, ecosystems, 44 and visibility, also perturb regional weather and climate (Lelieveld et al., 2015; IPCC, 2013). 45 Therefore, since 2010, the Chinese government has made great efforts to decrease emissions 46 of air pollutants, and the air quality has been improved considerably (Zheng et al., 2018; 47 Zhang et al., 2019). However, field observations have demonstrated that the nitrate 48 contribution to PM_{2.5} mass has progressively increased and nitrate aerosols have constituted a 49 major fraction of PM_{2.5} in recent several years (Tao et al., 2017; Sun et al., 2015; Zhang et al., 50 2013).

51 Nitrate aerosols are formed via nitric acid (HNO₃) to balance inorganic cations in the 52 particle phase, which is mainly produced through the homogenous pathway of OH+NO₂ and 53 the heterogeneous pathway of the N₂O₅ heterogeneous hydrolysis uptake on surfaces of deliquescence aerosols (Chang et al., 2016; Brown et al., 2016; Lowe et al., 2015; Chang et 54 55 al., 2011; Bertram & Thornton, 2009). The N₂O₅ hydrolysis is the most important pathway 56 during nighttime since N₂O₅ is photolytically liable, or even plays a considerable role in 57 nitrate formation during heavy haze events with weak sunlight and high relative humidity 58 (RH) (Brown et al., 2016).

Xue et al. (2014) have reported the average nitrate production rate of 1.36 µg m⁻³ h⁻¹
based on field observations during haze events in Hong Kong. According to N₂O₅
measurements in Beijing, Wang et al. (2017) have estimated that the potential nitrate
contribution of the N₂O₅ hydrolysis is about 52%. Using the stable isotope analysis method,
Wang et al. (2019) have found that the homogeneous and heterogeneous pathways contribute

64 66% and 34% of the nitrate mass in Beijing, respectively. Model simulations have further 65 highlighted the importance of the N_2O_5 hydrolysis to nitrate formation (Riemer et al., 2003; 66 Kim et al., 2014; Lowe et al., 2015; Liu et al., 2019). However, those model evaluations are 67 generally based on the brute force method (BFM), i.e. turning off the homogeneous or 68 heterogeneous pathway in simulations. The BFM can be used to evaluate the importance of 69 the certain pathway, but has flaws in quantifying its contribution, considering interactions of 70 complicated physical and chemical processes in the atmosphere (Zhang & Ying, 2011).

71 The source-oriented method coupled with air quality models (AQMs) can track the 72 formation of particulate and gas pollutants produced from various sources or pathways 73 through one mathematical simulation of emissions, chemical reactions, gas-to-particle 74 conversion, transport and deposition (Ying & Kleeman, 2006). The source-oriented AQMs 75 have been widely used to quantify contributions of emission sources or formation pathways 76 to PM_{2.5} or ozone (O₃) (Ying & Krishnan, 2010; Zhang & Ying, 2011; Hu et al., 2012; Hu et 77 al., 2015; Hu et al., 2017; Wang et al., 2018; Ying et al., 2018; Qiao et al., 2018; Liu et al., 78 2019). However, the quantitative study of nitrate formation pathways based on the 79 source-oriented AQMs has rarely been reported.

In the present study, a source-oriented WRF-Chem model has been developed and
applied to quantify contributions of homogeneous and heterogeneous pathways to nitrate
formation during wintertime of 2016 in the NCP.

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84 2 Model and method

85 2.1 WRF-Chem model and configuration

The source-oriented AQM used in the study is based on a specific version of the WRF-Chem model (Grell et al., 2005) with modification by Li et al. (2010; 2011a; 2011b; 2012). The specific WRF-Chem model includes a flexible gas phase chemical module with 89 consideration of different chemical mechanisms and the CMAQ aerosol module (AERO5) 90 developed by US EPA (Foley et al., 2010). The organic aerosols (OA) are simulated using the 91 volatility basis-set (VBS) modeling method, with the secondary OA (SOA) contributions 92 from glyoxal and methylglyoxal. ISORROPIA (Version 1.7) is used to predict the inorganic 93 aerosols, calculating the composition and phase state of an ammonium-sulfate-nitrate-water 94 inorganic aerosol in thermodynamic equilibrium with gas phase precursors (Nenes et al., 95 1998). Detailed description about the specific WRF-Chem model can be found in Supporting 96 Information (SI).

97 The source-oriented method introduces additional chemical species to represent 98 formations from various pathways, providing a direct and quantitative determination of the 99 contribution of each pathway (Ying & Krishnan, 2010). A detailed description of the method 100 can be found in previous studies (Ying & Kleeman, 2006; Ying & Krishnan, 2010; Zhang & 101 Ying, 2011). In the atmosphere, the reactions of OH with NO₂, NO₃ with hydrocarbons 102 (HCs), and the N₂O₅ heterogeneous hydrolysis dominate the HNO₃ formation. In the study, 103 two reactive tagged gas-phase species (HNO3 A and HNO3 B) are introduced to track the 104 homogeneous and heterogeneous HNO3 formation, respectively (Table S2), and the 105 corresponding aerosol-phase species (ANO3_A and ANO3_B) are also introduced to trace 106 the nitrate aerosol formation from each pathway. The detailed reactions of HNO3_A and 107 HNO3_B can be found in Table S1. We use ISORROPIA to distribute the NH₃/ammonium, 108 HNO₃/nitrate, and water between the gas and aerosol phases as a function of total sulfate, 109 total ammonia, total nitrate, relative humidity and temperature (Nenes et al., 1998). The 110 apportionment for nitrate aerosols follows the mass conversion of N(+VI) from each 111 formation pathway, when the total nitrate is distributed between the gas and aerosol phases by 112 ISORROPIA after one time step integration, as shown in Figure 1. The reaction rate constant 113 of N₂O₅ heterogeneous hydrolysis on surfaces of deliquescent aerosols is calculated using the 114 reaction probability $(\gamma_{N_2O_5})$. The parameterization of $\gamma_{N_2O_5}$ used in this study follows 115 Riemer et al. (2009), considering effects of organic coating on the N₂O₅ hydrolysis uptake 116 (Liu et al., 2019).

The source-oriented WRF-Chem model is used to simulate a heavy haze episode from 16 to 31 December 2016 in the NCP. Figure S1 shows the model simulation domain, and the detailed model configuration can be found in Table S2. The observation data sets used to evaluate the model performance include 3-hourly measurements of meteorological parameters in Beijing, and hourly on-line measurements of air pollutants released by Ministry of Ecology and Environment of China, and 12-hourly filter measurements of sulfate, nitrate, and ammonium in Beijing, Tianjin, Shijiazhuang and Cangzhou (Figure S1).

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125 3 Results and Discussions

Generally, the source-oriented WRF-Chem model performs reasonably well in simulating meteorological fields, air pollutants, and inorganic aerosols compared to measurements in the NCP (Figures S2-S5). Detailed model validation can be found in SI. The good model performance provides a reliable base for further quantifying the nitrate formation pathways.

131 **3.1** Contribution of the homogeneous and heterogeneous pathways to nitrate

Figure 2 presents the simulated temporal variation of average near-surface nitrate concentrations from the homogeneous and heterogeneous pathways in the NCP from 16 to 31 December 2016. The homogeneous and heterogeneous pathways contribute 48.4% (15.1 μ g m⁻³) and 51.6% (16.3 μ g m⁻³) of the nitrate mass on average in the NCP during the episode, respectively. During the stage of the haze formation and development from 16 to 19 December, the homogeneous pathway is the dominant nitrate source, with an average contribution of 63.0%. Meanwhile, the nitrate contribution of the heterogeneous pathway 139 progressively increases from around 10% to 50%. After 20 December, the heterogeneous 140 pathway plays a more important role in nitrate formation than the homogeneous pathway, 141 accounting for 57.0% of nitrate mass on average. Figure 3 presents the spatial distributions of 142 average near-surface nitrate contributions of the homogeneous and heterogeneous pathways 143 during the episode. Contributions of the homogeneous pathway to the nitrate formation are more than 12 μ g m⁻³ in the NCP, exceeding 27 μ g m⁻³ in the south of Hebei. Similar to the 144 145 spatial distribution of the homogeneous pathway, the nitrate contribution of the heterogeneous pathway exceeds 15 μ g m⁻³ in the NCP and exceeds 27 μ g m⁻³ in highly 146 147 industrialized cities, such as Shijiazhuang, Xingtai, and Zhengzhou.

148 Model results have shown that the heterogeneous pathway plays a significantly 149 important role in nitrate formation during the haze episode, which is consistent with field 150 observations in the NCP. Wang et al. (2017) have calculated the daily average nitrate 151 formation potential from the heterogeneous N₂O₅ hydrolysis based on the field measurements 152 in Beijing, reporting that the heterogeneous pathway accounts for 52% of the total nitrate 153 formation. Moreover, Wang et al. (2019) have evaluated the different pathways to nitrate 154 contribution in Beijing using stable oxygen isotopes analysis methods. They have found that 155 around 34% on average and up to 54% of nitrate mass are produced by the N₂O₅ hydrolysis 156 in winter. However, Liu et al. (2019) have evaluated nitrate contributions of the N₂O₅ hydrolysis in Beijing-Tianjin-Hebei (BTH) using the BFM based on the WRF-Chem model, 157 158 showing that the N₂O₅ hydrolysis produces about 30% of nitrate mass, which is much lower 159 than that in the present study.

160 In order to settle the large inconsistency, we have further used the BFM to assess nitrate 161 contributions of the heterogeneous pathway in the present study, by differentiating 162 simulations with and without the heterogeneous pathway (referred to as BASE and HETO 163 case, respectively) (Figure S6). The result demonstrates that about 30.8% of near-surface 164 nitrate aerosols are contributed by the heterogeneous pathway in the NCP during the episode, 165 indicating that the BFM might not be suitable for evaluation of the importance of different 166 pathways to nitrate formation. When both the homogeneous and heterogeneous pathways are 167 considered in the BASE case, the HNO₃ formed from each pathway competes for inorganic 168 cations in the particle phase, and the nitrate apportionment between the two pathways 169 depends on the total nitrate (N(+VI)) produced by each pathway. However, when the 170 heterogeneous pathway is not considered in the HETO case, only the HNO₃ formed from the 171 homogeneous pathway balances inorganic cations in the particle phase, without competition 172 with that from the heterogeneous pathway. Therefore, more HNO₃ from the homogeneous 173 pathway exists in the particle phase in the HETO case than the BASE case. When comparing 174 the BASE and HETO case, the nitrate contribution of the heterogeneous pathway is 175 underestimated, and the underestimation relies on the inorganic cations in the particle phase.

176 **3.2** HNO₃ production rate of the homogeneous and heterogeneous pathways

177 Generally, the heterogeneous N₂O₅ hydrolysis mainly occurs during nighttime 178 considering that N₂O₅ is formed from the reaction of NO₂+NO₃ and NO₃ and N₂O₅ are both 179 photolytically liable, causing very low levels of daytime N₂O₅. The N₂O₅ heterogeneous 180 hydrolysis constitutes the most important nitrate source during nighttime, but its role during 181 daytime remains elusive, particularly during haze days with weak solar radiation and high 182 RH. Furthermore, the reaction of OH+NO₂ dominates daytime HNO₃ production but becomes 183 an insignificant HNO₃ source during nighttime due to the absence of OH. The NO₃ reaction 184 with HCs has large potentials to be an important HNO₃ source because of increased NO₃ and 185 HCs concentrations during nighttime. However, as a bulk method, ISORROPIA distributes 186 the gas-phase HNO₃ and particulate-phase nitrate according to the total nitrate, which 187 includes contributions of accumulation and local production. Therefore, it is not possible to 188 determine the nitrate contribution of each pathway during daytime or nighttime using 189 concentrations of the tagged HNO₃ and nitrate aerosols (Figure 2).

190 We have further included two additional tagged species to represent the *in-situ* HNO₃ 191 production of the homogeneous and heterogeneous pathways in the source-oriented 192 WRF-Chem model, to provide temporal variations of the HNO₃ contribution of each pathway. Figure 4 presents the diurnal profile of the average HNO₃ production rate (P_{HNO_3}) of the 193 194 homogeneous and heterogeneous pathways in the planetary boundary layer (PBL) over the NCP during the episode. The average P_{HNO_3} of the homogeneous pathway is 85.6 ppt h⁻¹ in 195 196 the PBL during the episode, with a HNO₃ contribution of 49.8%. On December 21 and 26, the P_{HNO_3} of the homogeneous pathway decreases significantly due to reduction in the 197 198 downward solar radiation (Figure S7). Although N₂O₅ is highly photolabile, the daytime HNO_3 production rate of the N_2O_5 hydrolysis is nonzero and exceeds 25 ppt h^{-1} on December 199 21. The average P_{HNO_3} of the heterogeneous pathway is 86.4 ppt h⁻¹ in the PBL, with a 200 201 HNO₃ contribution of 50.2% in the NCP during the episode, which is a little bit higher than that of the homogeneous pathway. It is worth noting that the calculated P_{HNO_3} ratio of the 202 203 homogeneous to heterogeneous pathways in the PBL are well consistent with that of the 204 near-surface nitrate contribution between the two corresponding pathways.

Figure 5 shows the diurnal cycle of the average P_{HNO_3} of homogeneous and 205 206 heterogeneous pathways in the PBL over the NCP during the episode. During nighttime, from 207 18:00 to 07:00 Beijing Time (BJT) of the next day, the heterogeneous pathway dominates the 208 HNO₃ production, with an average HNO₃ contribution of 83.0%. Because of lack of OH due 209 to absence of O₃ and peroxide photolysis, the nighttime HNO₃ contribution of the 210 homogeneous pathway is mainly attributed to the NO₃ reaction with HCs, attaining 17.0%. 211 During daytime, from 08:00 to 17:00 BJT, the homogeneous pathway accounts for 89.9% of 212 the HNO₃ production on average. The P_{HNO_3} increases with enhancement of solar radiation 213 in the morning and peaks around noontime. The heterogeneous pathway plays an appreciable

role in the daytime HNO₃ production, with a contribution of 10.1%. The P_{HNO_3} of the heterogeneous pathway increases rapidly after sunset, peaks at approximately 20:00 BJT, and then drops slowly, which is caused mainly by decrease of O₃ due to continuous NO_x emissions during nighttime.

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219 4 Conclusion

In the study, a source-oriented WRF-Chem model is developed and used to simulate a heavy haze episode from 16 to 31 December 2016 in the NCP, to verify contributions of the homogeneous and heterogeneous pathways to nitrate aerosols. The nitrate homogeneous pathway includes reactions of OH with NO₂ and NO₃ with HCs, and the heterogeneous pathway is referred to as the N_2O_5 hydrolysis uptake on surfaces of deliquescence aerosols. The model generally performs well in simulating meteorological parameters, air pollutants and inorganic aerosols against observations in the NCP.

227 On average, during the episode in the NCP, the homogeneous and heterogeneous 228 pathways contribute 48.4% and 51.6% of the near-surface nitrate mass, respectively. During 229 nighttime, the N₂O₅ hydrolysis dominates the HNO₃ formation in the PBL, with a 230 contribution of 83.6%, while the homogeneous pathway accounts for 89.9% of daytime 231 HNO₃ formation. The homogeneous and heterogeneous pathways play almost same role in 232 the HNO₃ formation in the PBL. It is worth noting that the N₂O₅ hydrolysis also plays an 233 appreciably role in the daytime HNO₃ formation, with an average contribution of 10.1%, 234 particularly under conditions with weak sunlight. Our research highlights the significant 235 importance of the heterogeneous pathway to nitrate formation.

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website http://www.meteomanz.com. The hourly observations of PM2.5, O3, SO2, NO2, and 239 CO concentrations are released by Ministry of Ecology and Environment of China and can be 240 241 downloaded from the website http://106.37.208.233:20035. This work is financially 242 supported by the National Key R&D Plan (Quantitative Relationship and Regulation 243 Principle between Regional Oxidation Capacity of Atmospheric and Air Quality 244 (2017YFC0210000)) and National Research Program for Key Issues in Air Pollution Control 245 (DQGG0105). The authors would like to acknowledge helpful discussions with Professor 246 Jianlin Hu.

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249 References

- An, Z. S., Huang, R. J., Zhang, R. Y., Tie, X. X., Li, G. H., Cao, J. J., et al. (2019). Severe haze in northern China: A synergy of anthropogenic emissions and atmospheric processes. *Proceedings of the National Academy of Sciences of the United States of America*, 116, 8657-8666. https://doi.org/10.1073/pnas.1900125116.
- Bei, N., Wu, J., Elser, M., Feng, T., Cao, J., El-Haddad, I., et al. (2017). Impacts of meteorological uncertainties on the haze formation in Beijing–Tianjin–Hebei (BTH) during wintertime: a case study. *Atmospheric Chemistry and Physics*, 17, 14579–14591. https://doi.org/10.5194/acp-17-14579-2017.
- Bertram, T. H. & Thornton, J. A. (2009). Toward a general parameterization of N₂O₅ reactivity on aqueous particles: the competing effects of particle liquid water, nitrate and chloride. *Atmospheric Chemistry and Physics*, 9, 8351–8363. https://doi.org/10.5194/acp-9-8351-2009.
- Brown, S. S., Dubé, W. P., Tham, Y. J., Zha, Q., Xue, L., Poon, S., et al. (2016). Nighttime chemistry at a high altitude site above Hong Kong. *Journal of Geophysical Research–Atmospheres*, 121, 2457–2475. https://doi.org/10.1002/2015JD024566.
- 263 Chang, W. L., Bhave, P. V., Brown, S. S., Riemer, N., Stutz, J., & Dabdub, D. (2011). Heterogeneous
 264 Atmospheric Chemistry, Ambient Measurements, and Model Calculations of N₂O₅: A Review.
 265 Aerosol Science and Technology, 45, 665–695. https://doi.org/10.1080/02786826.2010.551672.
- Chang, W. L., Brown, S. S., Stutz, J., Middlebrook, A. M., Bahreini, R., Wagner, N. L., et al. (2016).
 Evaluating N₂O₅ heterogeneous hydrolysis parameterizations for CalNex 2010. *Journal of Geophysical Research–Atmospheres*, 121, 5051–5070. https://doi.org/10.1002/2015JD024737.
- Foley, K. M., Roselle, S. J., Appel, K. W., Bhave, P. V., Pleim, J.E., Otte, T. L., et al. (2010). Incremental testing of the Community Multiscale Air Quality(CMAQ) modeling system version 4.7. *Geoscientific Model Development*, 3,205–226. https://doi.org/10.5194/gmd-3-205-2010.
- 272 Grell, G. A., Peckham, S. E., Schmitz, R., McKeen, S. A., Frost, G., Skamarock, W. C., et al. (2005). Fully
 273 coupled "online" chemistry within the WRF model. *Atmospheric Environment*, 39, 6957–6975.
 274 https://doi.org/10.1016/j.atmosenv.2005.04.027.
- Guo, S., Hu, M., Zamora, M. L., Peng, J., Shang, D., Zheng, J., et al. (2014). Elucidating severe urban haze formation in China. *Proceedings of the National Academy of Sciences of the United States of America*, 111, 17373–17378. https://doi.org/10.1073/pnas.1419604111.
- Hu, J., Howard, C. J., Mitloehner, F., Green, P. G., & Kleeman, M. J. (2012). Mobile source and livestock feed contributions to regional ozone formation in Central California. *Environmental Science* & *Technology*, 46, 2781–2789. https://doi.org/10.1021/es203369p.
- Hu, J., Wang, P., Ying, Q., Zhang, H., Chen, J., Ge, X., et al. (2017). Modeling biogenic and anthropogenic secondary organic aerosol in China. *Atmospheric Chemistry and Physics*, 17, 77–92. https://doi.org/10.5194/acp-17-77-2017.
- Hu, J., Wu, L., Zheng, B., Zhang, Q., He, K., Chang, Q., et al. (2015). Source contributions and regional transport of primary particulate matter in China. *Environmental Pollution*, 207, 31–42. https://doi.org/10.1016/j.envpol.2015.08.037.
- IPCC (Intergovernmental Panel on Climate Change). (2013). Climate Change 2013: The Physical Science
 Basis. Contribution of Working Group I to the Fifth Assessment Report of the Intergovernmental
 Panel on Climate Change. Cambridge University Press, New York, 571-657.
- Kim, Y. J., Spak, S. N., Carmichael, G. R., Riemer, N., & Stanier, C. O. (2014). Modeled aerosol nitrate
 formation pathways during wintertime in the Great Lakes region of North America. *Journal of Geophysical Research–Atmospheres*, 119, 12420–12445. https://doi.org/10.1002/2014JD022320.
- Lelieveld, J., Evans, J. S., Fnais, M., Giannadaki, D., & Pozzer, A. (2015). The contribution of outdoor air pollution sources to premature mortality on a global scale. *Nature*, 525, 367-371. https://doi.org/10.1038/nature15371.

- Li, G., Bei, N., Tie, X., & Molina, L. T. (2011a). Aerosol effects on the photochemistry in Mexico City during MCMA-2006/MILAGRO campaign. *Atmospheric Chemistry and Physics*, 11, 5169–5182. https://doi.org/10.5194/acp-11-5169-2011.
- Li, G., Lei, W., Bei, N., & Molina, L. T. et al. (2012). Contribution of garbage burning to chloride and PM_{2.5} in Mexico City. *Atmospheric Chemistry and Physics*, 12, 8751–8761. https://doi.org/10.5194/acp-12-8751-2012.
- Li, G., Lei, W., Zavala, M., Volkamer, R., Dusanter, S., Stevens, P., et al. (2010). Impacts of HONO sources on the photochemistry in Mexico City during the MCMA-2006/MILAGO Campaign. *Atmospheric Chemistry and Physics*, 10, 6551–6567. https://doi.org/10.5194/acp-10-6551-2010.
- Li, G., Zavala, M., Lei, W., Tsimpidi, A. P., Karydis, V. A., Pandis, S. N., et al. (2011b). Simulations of organic aerosol concentrations in Mexico City using the WRF-Chem model during the MCMA-2006/MILAGRO campaign. *Atmospheric Chemistry and Physics*, 11, 3789–3809. https://doi.org/10.5194/acp-11-3789-2011.
- Liu, L., Wu, J., Liu, S., Li, X., Zhou, J., Feng, T., et al. (2019). Effects of organic coating on the nitrate formation by suppressing the N₂O₅ heterogeneous hydrolysis: a case study during wintertime in Beijing–Tianjin–Hebei (BTH). *Atmospheric Chemistry and Physics*, 19, 8189–8207.
 https://doi.org/10.5194/acp-19-8189-2019.
- Lowe, D., Archer-Nicholls, S., Morgan, W., Allan, J., Utembe, S., Ouyang, B., et al. (2015). WRF-Chem
 model predictions of the regional impacts of N₂O₅ heterogeneous processes on night-time chemistry
 over north-western Europe. *Atmospheric Chemistry and Physics*, 15, 1385–1409.
 https://doi.org/10.5194/acp-15-1385-2015.
- Nenes, A., Pandis, S. N., & Pilinis, C. (1998). ISORROPIA: A new thermodynamic equilibrium model for multiphase multicomponent inorganic aerosols. *Aquat Geochem*, 4, 123–152. https://doi.org/10.1023/a:1009604003981.
- Qiao, X., Ying, Q., Li, X., Zhang, H., Hu, J., Tang, Y., et al. (2018). Source apportionment of PM_{2.5} for 25
 Chinese provincial capitals and municipalities using a source-oriented Community Multiscale Air
 Quality model. *Science of the Total Environment*, 612, 462–471.
 http://dx.doi.org/10.1016/j.scitotenv.2017.08.272.
- Riemer, N., Vogel, H., Vogel, B., Schell, B., Ackermann, I., Kessler, C., et al. (2003). Impact of the heterogeneous hydrolysis of N₂O₅ on chemistry and nitrate aerosol formation in the lower troposphere under photosmog conditions. *Journal of Geophysical Research–Atmospheres*, D4, 4144. https://doi.org/10.1029/2002JD002436.
- Riemer, N., Vogel, H., Vogel, B., Anttila, T., Kiendler-Scharr, A., & Mentel, T. F. (2009). Relative
 importance of organic coatings for the heterogeneous hydrolysis of N₂O₅ during summer in Europe.
 Journal of Geophysical Research–Atmospheres, D17307. https://doi.org/10.1029/2008JD011369.
- Sun, Y. L., Wang, Z. F., Du, W., Zhang, Q., Wang, Q. Q., Fu, P. Q., et al. (2015). Long-term real-time measurements of aerosol particle composition in Beijing, China: seasonal variations, meteorological effects, and source analysis. *Atmospheric Chemistry and Physics*, 15, 10149–10165. https://doi.org/10.5194/acp-15-10149-2015.
- Tao, J., Zhang, L., Cao, J., & Zhang, R. (2017). A review of current knowledge concerning PM_{2.5} chemical composition, aerosol optical properties and their relationships across China, *Atmospheric Chemistry and Physics*, 17, 9485–9518. https://doi.org/10.5194/acp-17-9485-2017.
- Wang, H., Lu, K., Chen, X., Zhu, Q., Chen, Q., Guo, S., et al. (2017). High N₂O₅ concentrations observed in urban Beijing: Implications of a large nitrate formation Pathway. *Environmental Science &Technology Letters*, 4, 416–420. https://doi.org/10.1021/acs.estlett.7b00341.
- Wang, P., Ying, Q., Zhang, H., Hu, J., Lin, Y., & Mao, H. (2018). Source apportionment of secondary organic aerosol in China using a regional source-oriented chemical transport model and two emission inventories. *Environmental Pollution*, 237, 756–766. https://doi.org/10.1016/j.envpol.2017.10.122

- Wang, Y. L., Song, W., Yang, W., Sun, X. C., Tong, Y. D., Wang, X. M., et al. (2019). Influences of atmospheric pollution on the contributions of major oxidation pathways to PM2.5 nitrate formation in Beijing. Journal of Geophysical Research–Atmospheres, 124, 4174–4185. https://doi.org/10.1029/2019JD030284.
- Xue, J., Yuan, Z., Lau, A. K. H., & Yu, J. Z. (2014). Insights into factors affecting nitrate in PM_{2.5} in a polluted high NOx environment through hourly observations and size distribution measurements. *Journal of Geophysical Research–Atmospheres*, 119, 4888–4902. https://doi.org/10.1002/2013JD021108.
- Ying, Q., Feng, M., Song, D., Wu, L., Hu, J., Zhang, H., et al. (2018). Improve regional distribution and source apportionment of PM_{2.5} trace elements in China using inventory-observation constrained emission factors. *Science of the Total Environment*, 624, 355–365. https://doi.org/10.1016/j.scitotenv.2017.12.138.
- 356 Ying, Q., & Kleeman, M. J. (2006). Source contributions to the regional distribution of secondary particulate matter in California. *Atmospheric Environment*, 40, 736–752. https://doi.org/10.1016/j.atmosenv.2005.10.007.
- Ying, Q., & Krishnan, A. (2010). Source contributions of volatile organic compounds to ozone formation
 in southeast Texas. *Journal of Geophysical Research–Atmospheres*, 115, D17306.
 https://doi.org/10.1029/2010JD013931.
- 362 Zhang, H., & Ying, Q. (2011). Contributions of local and regional sources of NO_x to ozone concentrations
 363 in Southeast Texas. *Atmospheric Environment*, 45, 2877–2887.
 364 https://doi.org/10.1016/j.atmosenv.2011.02.047.
- 365 Zhang Q., Zheng Y., Tong, D., Shao, M., Wang, S., Zhang, Y., et al. (2019). Drivers of improved PM_{2.5} air quality in China from 2013 to 2017. *Proceedings of the National Academy of Sciences of the United States of America*, 116(49), 24463–24469. https://doi.org/10.1073/pnas.1907956116.
- Zhang, R., Jing, J., Tao, J., Hsu, S., Wang, G., Cao, J., et al. (2013). Chemical characterization and source apportionment of PM_{2.5} in Beijing: seasonal perspective. *Atmospheric Chemistry and Physics*, 13, 7053–7074. https://doi.org/10.5194/acp-13-7053-2013.
- Zheng, B., Tong, D., Li, M., Liu, F., Hong, C., Geng, G., et al. (2018). Trends in China's anthropogenic emissions since 2010 as the consequence of clean air actions. *Atmospheric Chemistry and Physics*, 14095-14111. https://doi.org/10.5194/acp-18-14095-2018.
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379	Figure Captions
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381 382 383 384 385	Figure 1 Conceptual scheme of source apportionment for the nitrate formation pathway. A: homogenous pathway; B: heterogeneous pathway; Subscript g: gas phase; Subscript a: aerosol phase; Superscript T: total; $S(+VI)_a$: sulfate; $N(+V)_a$: nitrate aerosol; $N(+V)_g$: nitric acid; $N(-III)_a$: ammonium aerosol; $N(-III)_g$: ammonia; t: current time; δt : integration time step.
386 387	Figure 2 Simulated diurnal profiles of the average near-surface nitrate concentration from the homogeneous and heterogeneous pathways in the NCP from 16 to 31 December 2016.
388 389 390	Figure 3 Spatial distributions of average near-surface nitrate concentrations from the (a) homogeneous and (b) heterogeneous pathway in the NCP from 16 to 31 December 2016.
391 392 393	Figure 4 Simulated temporal variations of the average <i>in-situ</i> HNO ₃ production rate of the homogeneous and heterogeneous pathways in the PBL over the NCP from 16 to 31 December 2016.
394 395	Figure 5 Diurnal cycle of the average <i>in-situ</i> HNO ₃ production rate of the (a) homogeneous and (b) heterogeneous pathway in the PBL over NCP from 16 to 31 December 2016.
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Figure 1 Conceptual scheme of source apportionment for the nitrate formation pathway. A:

homogenous pathway; B: heterogeneous pathway; Subscript g: gas phase; Subscript a: aerosol phase; Superscript T: total; $S(+VI)_a$: sulfate aerosol; $N(+V)_a$: nitrate aerosol; $N(+V)_g$: nitric acid; $N(-III)_a$: ammonium aerosol; $N(-III)_s$: ammonia; t: current time; δt : integration time step.



Figure 2 Simulated diurnal profiles of the average near-surface nitrate concentration from thehomogeneous and heterogeneous pathways in the NCP from 16 to 31 December 2016.



422 Homogeneous pathway (ug m³)
423 Figure 3 Spatial distributions of average near-surface nitrate concentrations from the (a) homogeneous and (b) heterogeneous pathway in the NCP from 16 to 31 December 2016.
426 Homogeneous and (b) heterogeneous pathway in the NCP from 16 to 31 December 2016.



433 Figure 4 Simulated temporal variations of the average *in-situ* HNO₃ production rate of the

- 434 homogeneous and heterogeneous pathways in the PBL over the NCP from 16 to 31
- 435 December 2016.



Figure 5 Diurnal cycle of the average *in-situ* HNO₃ production rate of the (a) homogeneous
and (b) heterogeneous pathway in the PBL over NCP from 16 to 31 December 2016.